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Enhancement of Thermoelectric Figure-of-Merit by Resonant States of Aluminum Doping in Lead Selenide

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By adding aluminium (Al) into lead selenide (PbSe), we successfully prepared n-type PbSe thermoelectric materials with a figure-of-merit (*ZT*) of 1.3 at 850 K. Such high *ZT* is achieved by a combination of high Seebeck coefficient caused by very possibly the resonant states in the conduction band created by Al dopant and low thermal conductivity from nanosized phonon scattering centers.

Broader context

Thermoelectric devices directly convert heat to electricity, thus they are important for harvesting natural heat as well as waste heat. For efficient devices, high figure-of-merit (*ZT*) materials are desired. We report here n-type lead selenide (PbSe) thermoelectric materials with *ZT* of 1.3 at 850 K. These materials are prepared by adding aluminium (Al) in PbSe during ball milling and hot pressing. Al, as a dopant in PbSe, possibly creates resonant states in the conduction band and causes increase in the local density of states (DOS) near the Fermi level. As a result, the Seebeck coefficients of Al added PbSe samples are about $40\sim100\%$ higher than the predicted values by the simple parabolic band model and about 40% higher than the Cl-doped reference PbSe sample without resonant states. Furthermore, using ball milling and hot pressing technique, the structure of our samples contains features such as Pb depleted discs, small grains, and ~10 nm subgrains that are effective for phonon scattering, and result in a much lower lattice thermal conductivity of $0.6\sim0.7$ Wm⁻¹K⁻¹ at 850 K. Further improvement could be expected by even better control of the Fermi level through optimization of addition of other doping materials.

Thermoelectric (TE) materials and devices have drawn significant interests and attentions for the past decades due to their potential applications in converting solar energy and waste heat into electricity¹. The performance of thermoelectric materials depends on dimensionless figure-of-merit, $ZT = S^2 \sigma T / (\kappa_{lat} + \kappa_{carr})$, where *S*, σ , *T*, κ_{lat} , and κ_{carr} are Seebeck coefficient, electrical conductivity, absolute temperature, lattice thermal conductivity, and electronic thermal conductivity, respectively². Improvements on *ZT* can be achieved by either increasing the power factor ($S^2 \sigma$) or reducing the thermal conductivity, or even better, by simultaneously increasing the power factor and reducing the thermal conductivity³. Potential ways to improve the power factor are through increasing the Seebeck coefficient (*S*) via band modification⁴ or engineering⁵, energy barrier filtering⁶⁻⁸, additional carrier pocket⁹, and resonant states¹⁰, or enhancing the mobility such as modulation doping¹¹. Among these possibilities, resonant states and additional carrier pockets have led to net increased *ZT* values via the power factor enhancements⁹, ¹⁰. In contrast, many methods have been found to be effective in reducing lattice thermal conductivity, for example, alloy scattering, boundary scattering, nanocomposites or nanoinclusion routes, etc. Simultaneously achieving resonant states to enhance the Seebeck coefficient thus the power factor and nanoscale grain boundary scattering to reduce the lattice thermal conductivity have not been reported before.

As one of the most studied bulk TE materials, PbTe based materials get a significantly enhancement recently by introducing various nano inclusions or modifying the band structure, and reach a high ZT value of $1.5 \sim 1.8$, such as p-type PbTe_{1-x}Se_x:Na $(\sim 1.8)^5$, PbTe-Sr₂Te:Na $(\sim 1.7)^{12}$, PbTe:Tl $(\sim 1.5)^{4.6}$, NaPb_mSbTe_{2+m} $(\sim 1.7)^{13}$, and n-type AgPb_mSbTe_{2+m} $(\sim 1.7)^{14}$ and PbTe-PbS:PbI₂ $(\sim 1.5)^{15}$. All these materials are based on Te. However, Te is extremely rare in the earth crust, just 0.001ppm, which is even less than 0.004 ppm for Au and 0.005 ppm for Pt This will limit the wide applications of PbTe-based materials. To develop materials without Te or less Te will be very urgent and important. Although PbS and PbSe are very similar to PbTe in crystal structure, band structure, and many other properties¹⁷, they are much less studied than PbTe because of the concerns that the smaller size of S or Se and lighter atomic mass in comparison with Te potentially result in higher lattice thermal conductivity.

Recently, a theoretical calculation of PbSe by Parker et al.¹⁶ showed that due to the flat feature of energy diagram in valence band, the p-type PbSe may possibly reach $ZT\sim2$ at 1000 K if it is heavily doped. Based on this suggestion, Wang et al. doped Na into PbSe¹⁷, and got a peak ZT value of around 1.2 at 850 K, which is 15% lower than 1.4 of Na doped PbTe. Comparing to p-type, the highest ZT of n-type PbSe was limited to $\sim 0.9^{18}$, even in the recently reported Ga and In doped PbSe¹⁹.

In this paper, we present the thermoelectric properties of the unexpected n-type PbSe samples by addition of Al prepared by ball milling and hot pressing. The highest ZTs we achieved are ~1.3 due to the combined effects of high Seebeck coefficient and low thermal conductivity, which is comparable to 1.4 of n-type Iodine doped PbTe with no special nano inclusions²⁰. Intuitively, Al with less valence electron than Pb, it should be acceptor when it substitutes Pb. However, our experimental results show that Al addition to PbSe make it n-type unexpectedly. To our knowledge, this is the first report regarding the effect of Al doping on thermoelectric properties of lead chalcogenides, and also the first observation on enhancement of Seebeck coefficient by creating resonant states in n-type PbSe.

Element Pb (granules, 99.99%), Se (granules, 99.999%), and Al (powder, 99.999%) from Alfa Aesar were weighed according to the stoichiometry of PbSe:Al_x (x=0, 0.005, 0.01, 0.02, and 0.03, with higher concentration also experimented but not presented here due to worse properties), PbSe:Cl_x (x=0.002, 0.01), PbSe:I_x (x=0.005) and loaded in a stainless jar with stainless balls for mechanical alloying by a high energy ball mill SPEX 8000D (SPEX SamplePrep). The alloyed nanopowders were then loaded into a graphite die with an inner diameter of 12.7 mm, and finally consolidated by direct current (DC) induced hot pressing^{3,21,22}.

The thermal diffusivities of the as-pressed disk samples were measured by a laser flash system (NETZSCHLFA 457), and the specific heat was determined by a DSC system (NETZSCH DSC 200-F3). Thermal conductivity was then calculated as the product of the thermal diffusivity, specific heat, and volumetric density that was determined by the Archimedes' Method. Bar samples with dimensions of about $2\times2\times12$ mm were cut from the disk for measuring the Seebeck coefficient and electrical conductivity simultaneously by a four-point probe system (ULVAC ZEM-3). Thin disk samples of around 0.5 mm in thickness were prepared for Hall measurements on a Lakeshore system (Hall Effect System7712A). The structures of the as-pressed samples were studied by X-ray diffraction (Rigaku RU300) using Cr radiation (K_a: 2.8976 Å), field emission scanning electron microscopy (SEM, JEOL-6340F), and transmission electron microscopy (TEM, JEOL-2010F).

A first principles calculation was conducted using the PWscf code to investigate the effect of Al on PbSe. In our calculation, fully relativistic pseudo-potentials are used. We first construct a $2 \times 2 \times 2$ PbSe supercell with 64 atoms, then we put 1 Al atom into the system (~1.5%) either substituting Pb or Se or sitting at the interstitial place. We relax the systems separately and perform the calculation on the relaxed systems, namely Al replaces Pb (AlPb₃₁Se₃₂) or Al replaces Se (Pb₃₂Se₃₁Al) or Al goes to interstitial site (AlPb₃₂Se₃₂). The formation energy was also calculated to find out the most favourable location for Al among all the three possible cases.

Dense samples (at least 96.3% of the theoretical density) of PbSe:Al_x with different Al doping concentrations (x=0, 0.005, 0.01, 0.02, 0.03, and 0.04) were prepared, and their X-ray diffraction patterns are shown in Fig. 1, which indicate that all the samples are crystallized in rock salt structure. In the range of x=0~0.03, no visible impurity phase could be detected. Impurity phase in samples with x higher than 0.03 were indeed detected, resulting in worse thermoelectric properties. The lattice parameters of all the samples were calculated by using Rietveld refinement, which was performed using an X'Pert HighScore Plus software (PANalytical, X'Pert Pro). The lattice parameter *a* of all the samples is 6.130-6.133 Å with error bar of 0.002Å. No clear doping concentration dependent increase or decrease in lattice parameter was observed.

Figure 2a plot the room temperature Seebeck coefficient of Al doped PbSe as a function of carrier concentration. It is noted that the Seebeck coefficients of our PbSe:Al_x ($x=0\sim0.03$) samples are all negative showing the nature of n-type conduction. It was reported that Indium could work as n-type dopant in PbSe^{19, 23}. However, the reason is still unknown. We guess Al and In may get in to the interstitial site, similar with Cu in Bi₂Te_{2.7}Se_{0.3}, acting as n-type dopant ²⁴. However, we did not observe any lattice parameter increase in Al doped PbSe, meaning interstitial case is not correct. So substitution of Al for Pb or Te may be the case. A theoretical calculation clearly showed that Al replacing Pb is the reasonable result and causing both n-type conduction and resonant states.



Fig. 1 XRD spectra of PbSe:Al_x samples with different Al concentrations of x = 0, 0.005, 0.01, 0.02, 0.03, and 0.04.

The theoretical Pisarenko curves for both n- and p-type materials were calculated using a simple parabolic band model $(SPB)^{25}$, as shown in Eqs. (1) and (2). We choose acoustic phonon scattering as the dominance electron scattering mechanism (s = -1/2), and the effective mass of $m^*/m_0 = 0.28$ according to PbSe:Na¹⁷.

$$S = \pm \frac{k_B}{e} \left(\frac{(5/2+r)F_{3/2+r}(\eta)}{(3/2+r)F_{1/2+r}(\eta)} - \eta \right)$$
(1)
$$n_H = 4\pi \left(\frac{2m^* k_B T}{h^2} \right)^{3/2} \frac{F_{1/2}(\eta)}{R_H}$$
(2)

In Eqs. (1) and (2), *S*, *k*_B, *e*, and *r* are Seebeck coefficient, Boltzmann constant, electron charge, and scattering parameter, respectively. The $F_i(\eta)$ is the Fermi integral and η is the reduced Fermi energy. In Fig. 2a, Seebeck coefficient of the pure PbSe sample (no Al added with a carrier concentration of ~3.1×10¹⁸ cm⁻³), Cl-doped PbSe samples (0.2% and 1% Cl), I doped PbSe sample (0.5% I) lies right on the n-type Pisarenko curve, indicating good agreement with the expectation of SPB model. The reference samples with Cl^{18, 23} for n-type, Na17 for p-type also fit the Pisarenko curve. In other words, there is no evidence of enhancement of Seebeck coefficient by possible resonant states in normal n-type PbSe with Cl or I dopants. However, the situation for Al in PbSe is totally different: when the carrier concentration increases to 1.78~3.96×10¹⁹ cm⁻³ upon 0.5 to 3 at.% Al doping, the Seebeck coefficient almost remains constant at around -111 µVK⁻¹, about 40~100% higher than that from the SPB model and experimental values of PbSe:Cl samples with the same carrier concentration. Similar observation was also found in p-type PbTe:Tl samples⁴, which was attributed to the increase of the local DOS near Fermi level by presence of resonant states. The resonant state is an additional carrier pocket close to the major carrier band⁹, which is believed to increase to the S²n and hence the power factor.



Fig. 2 (a) Dependence of Seebeck coefficient on carrier concentration at RT. Solid lines are theoretical Pisarenko curve with $m^*/m_0 = 0.28$. Filled circles and partially filled triangles represent results of this study. (b) local DOS of $Pb_{32}Se_{32}$ and $AIPb_{31}Se_{32}$; (c) local p-DOS for each atom: AI, Pb, Se density of state of $AIPb_{31}Se_{32}$. A resonant sate was shown near the conduction band edge of PbSe caused by AI. The valence band top was chosen as the zero energy in (b) and (c).

In order to find out which lattice site (Pb, Se, interstitial) in PbSe is more favourable for Al atom and why the Seebeck coefficient remains the same with increased carrier concentration, we conducted first-principle calculations based on density function theory with $2\times2\times2$ supper cell. Firstly, the formation energy of two most possible cases, Al substituting Pb and the interstitial site, was calculated. The formation energy for the substitution case (AlPb₃₁Se₃₂) is defined as $E_{sub}=E(AlPb_{31}Se_{32})-E(Pb_{31}Se_{32})-E(Al)$, while for the interstitial case, it is $E_{int}=E(AlPb_{32}Se_{32})-E(Pb_{32}Se_{32})-E(Al)$. Finally, we get the formation energy of substitution case ($E_{sub}=-0.45$ Ry), lower than that of the interstitial case ($E_{int}=-0.18$ Ry), which means that the substitution case is more energy favourable. Based on the band structure calculation, the density of state (DOS) for both substitution and interstitial case was obtained. A small bump near the conduction band is found in the substitution case as compared with the undoping case as shown in Fig. 2b. A partial DOS for each atom including Al, Pb, and Se is shown in Fig. 2c. A similar resonant state was also observed in Al doped Pb_{1-x}Al_xTe^{26,27}. It is clearly suggested that a resonant state in the conduction band is caused by the 1.5% Al in Pb_{1-x}Al_xSe. However, we did not observe similar resonant state near the band edge in the interstitial case (i.e., AlPb₃₂Se₃₂) according to our first principle calculation. Combining the experimental and theoretical results, we come to a conclusion that Al likely replaces Pb rather than sits at the interstitial site.



Fig. 3 Temperature dependent thermoelectric properties of PbSe:Al_x (x=0, 0.005, 0.01, 0.02, and 0.03)(a) electrical resistivity, (b) Seebeck coefficient, and (c) power factor. The inset of (a) is the electrical resistivity of pure PbSe sample without Al dopant. The dash line is for the 0.2% Cl-doped PbSe.

It is reported that the modification of band edge also can be realized by nano inclusion^{28,29}, such as the AgSbTe₂ nanoinclusion in the PbTe matrix³⁰. Since such nano inclusions are widely observed in nanocomposite made by ball milling and hot pressing^{3, 21, 22}, the Al related nano inclusion may be another possible reason for the unchanged Seebeck coefficient increasing carrier concentration. Besides the possible band structure changing by resonant state or nano inclusion, there is another possibility, i.e. an increase in scattering parameter *r*. This possible reason for the enhanced Seebeck coefficient can be interpreted from the Mott expression of Seebeck coefficient which is shown as following,

$$S = \frac{\pi^2}{3} \frac{k_B}{e} k_B T \left\{ \frac{1}{n} \frac{dn(E)}{dE} + \frac{1}{\mu} \frac{d\mu(E)}{dE} \right\}_{E=E_F}$$
(3)

where n, μ , E, and E_F are carrier concentration, mobility, energy of carriers, and Fermi level, respectively. The increase in $d\mu(E)$

 $\frac{d\mu(L)}{dE}$ is equivalent to an increased scattering parameter r. The changing in r can be achieved from the resonant scattering

which involves electron jumping in and out between the normal band and the quasi-local impurity state. Unfortunately, we do not have enough evidence for the existence of such resonant scattering in our Al doped PbSe samples.

The detailed thermoelectric properties of PbSe:Al samples are presented in Fig. 3 together with the data of 0.2% Cl-doped PbSe sample for comparison. The Cl-doped PbSe is made by using the same method. It is found that Al is a much weaker n-type dopant than Cl. The carrier concentration of 0.2% Cl-doped PbSe $(3.0 \times 10^{19} \text{ cm}^{-3})$ is comparable to that of 2% Al-doped PbSe $(2.7-2.9 \times 10^{19} \text{ cm}^{-3})$. However, the Seebeck coefficient (-80 μ VK⁻¹) of the 0.2% Cl-doped PbSe is much lower than that (-106 μ VK⁻¹) of the 2% Al-doped PbSe, which demonstrates that PbSe:Al has larger effective mass than the 0.2% Cl-doped PbSe. This fact could be a direct evidence for the existence of resonant state. Owing to the enhanced Seebeck coefficient, a significantly higher power factor of 2030 μ Wm⁻¹K⁻² was obtained in the 2% Al-doped PbSe compared with 1440 μ Wm⁻¹K⁻² for the 0.2% Cl-doped PbSe.

Total thermal conductivities of the studied samples, demonstrated in Fig. 4, were calculated using $\kappa = \rho DC_p$, where ρ is volumetric density, *D* the diffusivity (Fig. 4a), and C_p the specific heat (Fig. 4a). In calculation of κ , the C_p values of x=0.01 are used for x=0.005, 0.01 samples, and that of x=0.03 are used for x=0.02, 0.03 samples. The thermal conductivity of PbSe:Al samples are ~0.9 Wm⁻¹K⁻¹ at 850 K. The lattice thermal conductivity is calculated by subtracting the electronic contribution from the total thermal conductivity ($\kappa_{lat} = \kappa - \kappa_{carr} = \kappa - L\sigma T$, where *L* is the Lorenz number). Here, the Fermi energy related Lorenz number is calculated by using the Fermi energy extracted from Seebeck coefficient within a simple parabolic band model, and the acoustic phonon scattering as the major carrier scattering mechanism³¹. As a result, the Lorenz number at room temperature is ranged from 1.61×10^{-8} to 1.87×10^{-8} V²K⁻² depending on the Al concentration. The κ_{lat} of pure PbSe sample at RT is 1.81 Wm⁻¹K⁻¹, consistent with the reported data¹⁷. The κ_{lat} of Al-doped samples is lower and decreases with Al concentration. It is very interesting to note that the Al-doped samples have κ_{lat} as low as $0.58 \sim 0.69$ Wm⁻¹K⁻¹ at 850 K. This low κ_{lat} is related to the microstructures discussed below. The highest *ZT* of around 1.3 is achieved at x=0.01 at 850 K, which is ~300% and 40% higher than that of the un-doped and Cl-doped samples, respectively.



Fig. 4. Temperature dependent thermoelectric properties of PbSe:Al_x (x=0, 0.005, 0.01, 0.02, and 0.03), x values are wrong in the figures (a) diffusivity coefficient with the specific heat C_p as the inset, (b) thermal conductivity with the lattice thermal conductivity as the inset, and (c) figure of merit.

In order to elucidate the reasons for such a low lattice thermal conductivity of PbSe:Al samples, extensive microscopy study was conducted by SEM and TEM. The following microscopic characteristics could be essential to theoretically explain the ultralow lattice thermal conductivity. Firstly, SEM microphotograph in Fig. 5a shows that the average grain size of PbSe:Al samples is less than 1 µm, which was analyzed by linear intercept particle analyzing method³², and confirmed by TEM bright field image shown in Fig. 5b. Secondly, Fig. 5a and 5b show that there are many nanosized pores (sphere-like, around 50 nm in diameter) distributing along the grain boundaries (mostly) and inside the grains (some), with the latter pores indicated by arrows in Fig. 5a and Fig. 5b. These small crystalline grains and the nanosized pores could mainly result in the reduced lattice thermal conductivity due to strong scattering to the phonons. Thirdly, it is interesting to observe that there are a lot of Pb-depleted disks lying on the {100} planes in PbSe:Al grains with diameter around 2~5 nm and thickness less than 0.5 nm, as shown in Fig. 5c. With a volumetric density of about 9×10¹⁷ cm⁻³, the Pb-depleted disks in Fig. 5c may cause significant lattice strain fields (3-20 nm) on both sides of the disks along their normal directions (see the arrow directions). This kind of Pb-depleted disks has been reported in our previous work³³ in PbTe system. Due to the lack of Pb atomic layers in the crystal, both the disks and the related lattice strain fields could also exert additional scattering effect on phonon transporting and lead to the further reduced lattice thermal conductivity. Fourthly, the atomic scale microphotographs in Fig. 5d and 5e clearly show the contrast differences related to atomic columns and even a tilt contrast column, as separately pointed out by the arrows in Fig. 5e. Although Fig. 5e is only a modified HRTEM lattice image rather than a structural image, it is suggested that these weaker white spots were caused by the Al doping. Accordingly, the atomically scaled inhomogeneity might affect the phonon scattering in each grain. Finally, along the [111] direction, Fig. 5f shows that there are many distinctive lattice areas, named subgrains (enclosed by white circles), in each grain. It is noted that these nanoscaled subgrains are separated by obscure boundary regions. It is suggested that these subgrain boundaries may come from ball milling and hot pressing process.



Fig. 5 (a) SEM microphotograph of freshly fractured surface, (b) bright field TEM image of typical PbSe:Al microstructure, (c) HRTEM image of the Pb-depleted disks lying on the {100} crystal planes, (d) atomic scaled lattice image showing the edge of the crystal foil in (c), (e) enlarged and modified HRTEM image from (d), and (f) nanoscaled subgrains and the subgrain boundaries.

Conclusions

In summary, with the enhancement of Seebeck coefficient due to the increase of local DOS near Fermi level by Al created resonant states, and the reduction of the lattice thermal conductivity by numerous distinctive microstructures, such as Pb depleted discs, small grains, and ~10 nm subgrains, we report an enhanced ZT~1.3 at 850 K, which is ~300% and ~40% higher than the undoped and Cl-doped PbSe samples, respectively. By accurately controlling the alignment between Fermi level and resonant states through other more appropriate dopants, further improvement on power factor and ZT could be expected.

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