

MIT Open Access Articles

Recent and future trends in synthetic greenhouse gas radiative forcing

The MIT Faculty has made this article openly available. *Please share* how this access benefits you. Your story matters.

Citation: Rigby, M., R. G. Prinn, S. O'Doherty, B. R. Miller, D. Ivy, J. Muhle, C. M. Harth, et al. "Recent and Future Trends in Synthetic Greenhouse Gas Radiative Forcing." Geophysical Research Letters 41, no. 7 (April 4, 2014): 2623–2630. © 2014 American Geophysical Union

As Published: http://dx.doi.org/10.1002/2013gl059099

Publisher: Wiley Blackwell

Persistent URL: http://hdl.handle.net/1721.1/99158

Version: Final published version: final published article, as it appeared in a journal, conference proceedings, or other formally published context

Terms of Use: Article is made available in accordance with the publisher's policy and may be subject to US copyright law. Please refer to the publisher's site for terms of use.







Geophysical Research Letters

RESEARCH LETTER

10.1002/2013GL059099

Kev Points:

- Measurements of all the major synthetic greenhouse gases have been compiled
- These measurements have been used to infer recent global emissions trends
- Based on these trends, future emissions scenarios have been investigated

Supporting Information:

- · Readme
- · Synthetic GHG scenario development
- · Synthetic GHG inverse method
- · C3F8 emissions
- · C2F6 emissions
- SO2F2 emissions
- SF6 emissions
- · NF3 emissions
- HFC365mfc emissions
- HFC32 emissions
- · HFC245fa emissions
- · HFC23 emissions
- · HFC227ea emissions
- · HFC152a emissions
- HFC143a emissions
- HFC134a emissions
 HFC125 emissions
- HCFC22 emissions
- · HCFC142b emissions
- HCFC141b emissions
- CH3CCl3 emissions
- CFC12 emissions
- CFC115 emissions
- CFC114 emissions
- CFC113 emissions
 CFC11 emissions
- CF4 emissions
- CCl4 emissions

Correspondence to:

M. Rigby, matt.rigby@bristol.ac.uk

Citation:

Rigby, M., et al. (2014), Recent and future trends in synthetic greenhouse gas radiative forcing, *Geophys. Res. Lett.*, 41, 2623–2630, doi:10.1002/2013GL059099.

Received 19 DEC 2013 Accepted 15 FEB 2014 Accepted article online 18 FEB 2014 Published online 4 APR 2014 Corrected 2 JUN 2014

This article was corrected on 2 JUN 2014. See the end of the full text for details.

Recent and future trends in synthetic greenhouse gas radiative forcing

M. Rigby^{1,2}, R. G. Prinn², S. O'Doherty¹, B. R. Miller³, D. Ivy⁴, J. Mühle⁵, C. M. Harth⁵, P. K. Salameh⁵, T. Arnold⁵, R. F. Weiss⁵, P. B. Krummel⁶, L. P. Steele⁶, P. J. Fraser⁶, D. Young¹, and P. G. Simmonds¹

¹School of Chemistry, University of Bristol, Bristol, UK, ²Center for Global Change Science, Massachusetts Institute of Technology, Cambridge, Massachusetts, USA, ³Earth System Research Laboratory, NOAA, Boulder, Colorado, USA, ⁴Department of Earth Atmospheric and Planetary Science, Massachusetts Institute of Technology, Cambridge, Massachusetts, USA, ⁵Scripps Institution of Oceanography, University of California, San Diego, La Jolla, California, USA, ⁶Centre for Australian Weather and Climate Research, CSIRO Marine and Atmospheric Research, Aspendale, Victoria, Australia

Abstract Atmospheric measurements show that emissions of hydrofluorocarbons (HFCs) and hydrochlorofluorocarbons are now the primary drivers of the positive growth in synthetic greenhouse gas (SGHG) radiative forcing. We infer recent SGHG emissions and examine the impact of future emissions scenarios, with a particular focus on proposals to reduce HFC use under the Montreal Protocol. If these proposals are implemented, overall SGHG radiative forcing could peak at around 355 mW m⁻² in 2020, before declining by approximately 26% by 2050, despite continued growth of fully fluorinated greenhouse gas emissions. Compared to "no HFC policy" projections, this amounts to a reduction in radiative forcing of between 50 and 240 mW m⁻² by 2050 or a cumulative emissions saving equivalent to 0.5 to 2.8 years of CO₂ emissions at current levels. However, more complete reporting of global HFC emissions is required, as less than half of global emissions are currently accounted for.

1. Monitoring Global Trends in Synthetic Greenhouse Gases

In 2012, the major long-lived synthetic greenhouse gases (gases with no significant natural sources and lifetimes of at least 1 year) were responsible for 350 ± 10 mW m⁻² of direct radiative forcing (RF), 19% as large as the increase in RF due to CO₂ since the preindustrial era [Hall et al., 2012] (note that the uncertainty in SGHG RF excludes radiative transfer assumptions, which are estimated to be of the order of 10% [Forster et al., 2007]). This group consists of a few tens of compounds, which we break down into the following groups (in the order of their current contribution to RF): chlorofluorocarbons (CFCs) and other strongly ozone depleting substances (ODS) (which include CCl₄ and CH₃CCl₃ here), hydrochlorofluorocarbons (HCFCs), hydrofluorocarbons (HFCs), fully fluorinated GHGs (FFGHGs) (consisting of perfluorocarbons, SF₆, and NF₃), and SO₂F₂. The CFCs and their replacements, the HCFCs and HFCs, are primarily used in refrigeration, air conditioning, and foam blowing [e.g., Montzka et al., 2011]. FFGHGs are emitted during aluminum manufacture (primarily CF_4) and are used in a range of applications such as electrical insulation (e.g., SF_6) or semiconductor manufacture (e.g., NF₃ and C₂F₆) [Weiss et al., 2008; Mühle et al., 2010; Rigby et al., 2010; Arnold et al., 2013]. Some gases, such as HFC-23 (CHF₃), have little practical use but are released to the atmosphere as unwanted by-products during certain industrial processes [Miller et al., 2010; Miller and Kuijpers, 2011]. Despite being present in the atmosphere at levels of only a few hundred parts per trillion or less, synthetic greenhouse gases (SGHGs) have a significant climate impact because of their very high radiative efficiencies and, in many cases, very long lifetimes (tens to thousands of years) [Ravishankara et al., 1993; Forster et al., 2007].

Here we examine recent trends in 25 of the most abundant SGHGs measured by the Advanced Global Atmospheric Gases Experiment (AGAGE) [*Prinn et al.*, 2000]: CFC-11, CFC-12, CFC-113, CFC-114, CFC-115, CCl₄, CH₃CCl₃, HCFC-22, HCFC-141b, HCFC-142b, HFC-23, HFC-32, HFC-125, HFC-134a, HFC-143a, HFC-152a, HFC-227ea, HFC-245fa, HFC-365mfc, CF₄, C_2F_6 , C_3F_8 , SF₆, NF₃, and SO₂F₂. There are some SGHG with an RF known to be higher than some of the more minor members of this list, which are not included here because AGAGE measurements are not yet available for all or part of the time period investigated. Examples of these compounds include $c-C_4F_8$ [*Oram et al.*, 2012], some halons [*Fraser et al.*, 1999], and some minor CFCs and HCFCs [e.g., *Culbertson et al.*, 2004; *Maione et al.*, 2013]. Measurements have also been made of SGHGs

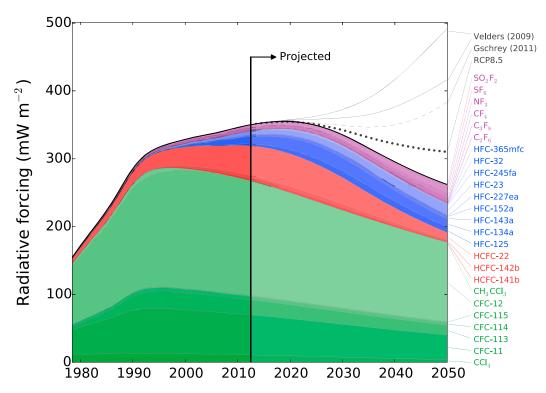


Figure 1. Global radiative forcing due to long-lived synthetic greenhouse gases, derived from AGAGE observations from 1980 to 2012 (darkly shaded areas). The more lightly shaded areas show the predicted radiative forcing due to each SGHG from 2013 until 2050, assuming that recent plans to phase down HFCs are implemented, existing Montreal Protocol obligations are met, and emissions of FFGHG show moderate growth. The solid, dotted, and dashed lines show SGHG radiative forcing under three previously published HFC scenarios that do not involve major emissions abatement measures.

that exert a smaller influence on global RF than the 25 investigated here [e.g., Straume et al., 1998; Sturges et al., 2000; Simmonds et al., 2002; Vollmer et al., 2011; Ivy et al., 2012a, 2012b]. We estimate that the combined RF due to these omitted species was of the order of 1% of the overall SGHG RF in 2012.

Since 1978, AGAGE and its previous incarnations as the Atmospheric Lifetimes Experiment and Global Atmospheric Gases Experiment have charted the trends in SGHG from remote stations, chosen to observe "background" air masses, using gas chromatography-electron capture detection and gas chromatography-mass spectrometry techniques [Simmonds et al., 1995; Prinn et al., 2000; Miller et al., 2008; Arnold et al., 2012]. The data used in this paper have been compiled from both in situ AGAGE observations and new and previously published measurements of archived air samples from Cape Grim, Tasmania and several northern hemispheric locations [O'Doherty et al., 2004, 2009; Miller et al., 2010; Mühle et al., 2010; Rigby et al., 2010; Vollmer et al., 2011; Saikawa et al., 2012; Arnold et al., 2013]. We have assimilated these observations into a model of atmospheric chemistry and transport to determine global trends and infer emissions of the entire suite of SGHG, using the methodology outlined in Rigby et al. [2011, 2013] and in the supporting information. Atmospheric lifetimes in the model are based on the recent evaluation carried out under the Stratospheric-troposphere Processes and their Role in Climate (SPARC) Lifetime of Halogen Source Gases report [SPARC, 2013].

2. Recent Trends

Global RF due to SGHGs from 1978 to 2012, calculated using the assimilated suite of AGAGE observations, is shown in Figure 1. Overall RF has been dominated by the CFCs from the beginning of this record. However, since the international ratification of the Montreal Protocol in 1987 and subsequent strengthening through amendments during the 1990s, the reported use of CFCs has declined dramatically. This has led to a 7% reduction in overall global CFC and other ODS RF since the peak in 1996, making the Montreal Protocol a relatively effective piece of global climate legislation

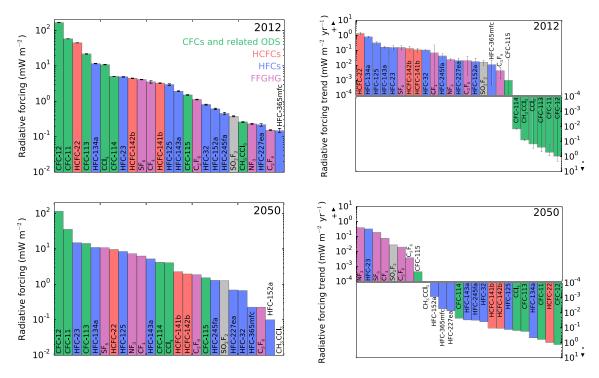


Figure 2. (left) Radiative forcing and (right) 5 year averaged trend observed in the (top) years preceding 2012 and (bottom) predicted for 2050, assuming moderate FFGHG emissions growth trajectories and Montreal Protocol HFC, CFC, and HCFC phase-down. One-sigma uncertainties in radiative forcing due to the AGAGE assimilation are shown in Figure 2, top as error bars.

(asnoted by *Molina et al.* [2009], *Velders et al.* [2012], and *Estrada et al.* [2013]). In response to the phaseout of CFCs, use of HCFCs and HFCs has increased in similar applications. AGAGE observations show a rapid growth in concentrations of HCFCs and HFCs over the last two decades, such that in 2012, HCFCs and HFCs were responsible for 15% and 6% of the SGHG radiative forcing, respectively. The FFGHGs have made a more minor contribution to 2012 radiative forcing, currently at 3% of overall SGHG RF. However, their long lifetimes mean that their radiative impact will only increase in the coming decades, even if emissions drop to relatively low levels.

Analysis of the contributions of individual species to global RF reveals that CFC-12 remains the most important SGHG (Figure 2), with a RF of 168 ± 2 mW m⁻² in 2012, making it responsible for $48\pm3\%$ of the SGHG RF. However, it is also the species whose RF is declining the most quickly, with a trend of -0.9 ± 1 mW m⁻² yr⁻¹ (averaged over the 5 years up to 2012). In fact, all of the CFCs and ODSs examined here, except for CFC-115 (which has a lifetime of around 1020 years) [*SPARC*, 2013], show significantly negative RF growth trends in recent years. In contrast, every other species is increasing its contribution to global RF, with rates that vary by almost 4 orders of magnitude. The major non-CFC contributor to both SGHG RF and RF growth is HCFC-22, which was adding 1.3 ± 0.3 mW m⁻² yr⁻¹ to its contribution of 45 ± 2 mW m⁻² in recent years. Of the HFCs, HFC-134a currently makes a smaller contribution to overall RF than HCFC-22 (12 ± 0.2 mW m⁻²), but its RF trend is of a similar order of magnitude, at 0.8 ± 0.2 mW m⁻² yr⁻¹. Despite being relatively minor contributors to global RF, some HFCs are growing particularly quickly. For example, HFC-125 and -143a are the thirteenth and fourteenth most important contributors to absolute SGHG RF, but are growing at the third and fourth highest rates, respectively. Of the FFGHGs, SF₆ is currently the most significant, both in terms of RF and RF growth.

3. Global HFC and PFC Emissions Under-Reporting

To investigate the drivers of these trends, we compared the emissions inferred from AGAGE observations to available reports to the United Nations Framework Convention on Climate Change

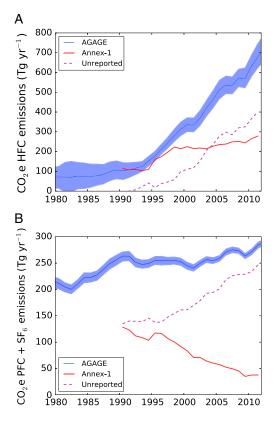


Figure 3. (a) HFC and (b) PFC and SF₆ CO₂-equivalent emissions derived from AGAGE observations (blue line and shading, representing 1-sigma uncertainties) compared to reported emissions from Annex 1 countries (red). Unreported emissions (AGAGE minus reported emissions) are shown in purple.

(UNFCCC). So-called "Annex 1" countries (including the USA, EU members, and many other developed countries) have been required to report emissions of HFCs, perfluorocarbons (PFCs), and SF₆ based on "bottom-up" accounting methodologies from 1990 onward. For the HFCs, these reports initially showed a rapid rise in emissions, resulting in more than a doubling of the global release rate in the 1990s, before emissions growth slowed significantly from around 1998 onward (Figure 3). Global emissions inferred from AGAGE observations show relatively good agreement with those reported in the early 1990s. However, emissions have continued to rapidly increase after reported emissions growth slowed, to such an extent that less than half of the world's HFC emissions were unaccounted for in 2011, the last year for which emissions reports were available (680 ± 70 Tg-CO₂e yr^{-1} compared to 279 Tg-CO₂e yr^{-1} reported).

Compared to the dramatic growth in HFC emissions, aggregated emissions of the PFCs and SF₆ have grown more moderately between 1980 and 2012. Recent studies have noted that SF₆ emissions increased significantly during this period [*Rigby et al.*, 2010], whereas emissions of the major PFC, CF₄, have declined [*Mühle et al.*, 2010]. In contrast to the HFCs, emissions of PFCs and SF₆ have never been fully accounted for by reported release rates from Annex 1 countries. In 1990, more than half of the CO₂-equivalent global emissions were unreported and this

discrepancy has increased significantly in recent years, with inferred global emissions growing at the same time as reported emissions have declined. As a result, over 90% of combined PFC and SF₆ emissions were unreported in 2012.

These comparisons highlight at least two potential issues with current emissions reporting. First, incomplete global reporting is likely to leave much of the world's emissions unaccounted for. For example, recent observations from East Asia indicate significant emissions of HFCs and FFGHGs from non-Annex 1 countries in the region, notably China [Kim et al., 2010; Yao et al., 2012]. Additionally, the growing discrepancy between reported and inferred emissions for both groups of compounds could also be indicative of growth in emissions from growing nonreporting economies. Second, the emissions reports may themselves be subject to inaccuracies. For example, recent regional emissions estimates inferred from atmospheric measurements suggest that SF₆ may be under-reported by Annex 1 countries [Levin et al., 2010; Rigby et al., 2010]. Similar discrepancies in other gases could be partly responsible for the overall difference between the reported and derived emissions.

In the following section, we explore possible future RF trajectories for each group of SGHG, particularly in light of potential changes to the Montreal Protocol. The emissions derived above serve as a 2012 "baseline" from which projections to 2050 can now be made.

4. Future RF Trajectories and Possible HFC Phase-Down

Owing to their potential to deplete stratospheric ozone, the HCFCs are currently being phased-down in developed countries under the Montreal Protocol, and will begin phase-down in developing countries in 2015. By 2030, almost all HCFC consumption should be halted. HCFCs, which have lifetimes of around 10 years, are removed from the atmosphere much more quickly than CFCs, which have lifetimes of around 50

Table 1. Emissions Scenarios for Each Group of SGHG and Resulting Range in Predicted Radiative Forcing (Including the 1-Sigma Range in Atmospheric Lifetimes as Estimated by *SPARC* [2013])^a

Group	Scenarios	2050 RF Range (mWm^{-2})
CFCs	Daniel et al. [2011]	169–187
	RCP3-PD, 4.5, 6, and 8.5 ^b	
HCFCs	Daniel et al. [2011]	13–41
	RCP3-PD	
	RCP4.5, 6, and 8.5 ^b	
HFCs (excluding HFC-23)	This work (US EPA phase-down)	25–281
	RCP3-PD	
	RCP4.5	
	RCP6	
	RCP8.5	
	Velders et al. [2009] ^c	
	Gschrey et al. [2011]	
HFC-23	Miller and Kuijpers [2011] ^d	4–15
	RCP3-PD	
	RCP4.5	
	RCP6	
	RCP8.5	
FFGHGs	RCP3-PD	20–28
	RCP4.5	
	RCP6	
	RCP8.5	
	Gschrey et al. [2011]	

^aProjections in bold were used in the "HFC phase-down" scenario.

to hundreds of years. Therefore, the phase-down of these compounds has the potential to provide relatively rapid global climate benefits. Using emissions projections of CFC/ODS and HCFC from 2012 to 2050 that are consistent with the atmospheric mole fractions in *Daniel et al.* [2011], we estimated their future RF. We find that RF due to CFCs/ODSs should decline steadily in the coming decades as lingering emissions gradually decline and the gases are photochemically removed from the atmosphere (Figure 1). By 2050, CFC/ODS RF could be 34% lower than in 2012 and 41% lower than the maximum (which occurred in 1996). Despite their comparatively delayed phase-down schedule, the HCFCs show a more dramatic 73% reduction in RF by 2050 (compared to 2012), reflecting their more rapid removal from the atmosphere than CFCs. Under this scenario, HCFC RF is projected to peak in 2020 at 57 mW m⁻².

No global legislation currently exists to regulate the use of HFCs or FFGHGs. However, recent proposals have been made to phase down global HFC consumption through the Montreal Protocol. Using the US Environmental Protection Agency's proposed consumption phase-down schedule [Environmental Protection Agency (EPA), 2013] and assumptions about release rates from HFC "banks" (following Velders et al. [2009]), we have developed a projection of HFC emissions to 2050 (see supporting information for detailed scenario development). The radiative forcing predicted using this emissions trajectory is shown in Figure 1. HFC-23, which is a potent greenhouse gas, and unwanted by-product of HCFC-22 production, is not included in the US EPA proposals. The RF projection in Figure 1 assumes that HFC-23 emissions follow the "reference" scenario of Miller and Kuijpers [2011]. Estimates of future radiative forcing due to FFGHGs are based on the Representative Concentration Pathway (RCP) scenarios [Fujino et al., 2006; Smith and Wigley, 2006; Clarke et al., 2007; Riahi et al., 2007; van Vuuren et al., 2007; Meinshausen et al., 2011] and the projections of EPA [2012], as detailed in the supporting information.

Under this scenario, we find that HFC RF will plateau around 2035 and decline slightly by 2050 (Figure 1). By 2050, despite the global phase-down, HFC RF would be around 125% higher than 2012. However, when combined with the coincident reduction in CFC and HCFC RF, the proposed amendments to the Montreal

^bEmissions in each of these scenarios are identical.

^cBoth "high" and "low" emissions growth scenarios included.

^dThe *Miller* and *Kuijpers* [2011] "Reference case" scenario was used in our HFC phase-down projections, and their "less mitigation" and "best practices" scenarios were included in the sensitivity study.



Protocol could result in a very significant climate impact, with a 26% decline in overall SGHG RF by 2050, compared to 2012, reaching 260 mW m^{-2} by 2050, and despite continued growth in FFGHG and HFC-23 RF.

The actual trajectory of future SGHG RF will depend critically on future emissions scenarios and their rate of removal from the atmosphere [e.g., Velders and Daniel, 2013]. To investigate the sensitivity of our projections to both of these factors, we ran the model forward using a suite of available projections and the currently accepted range of atmospheric lifetimes [SPARC, 2013]. We find that the single most important factor determining SGHG RF in 2050 will be the HFC emissions trajectory (Table 1). Therefore, we have examined the HFC phase-down schedule in contrast to a range of "no policy" HFC emissions trajectories (holding all other factors constant) to determine the possible influence of including HFCs in the Montreal Protocol (Figure 1).

The "HFC phase-down" trajectory leads to a 26% decline in RF by 2050, in contrast to an 11% to 42% increase in RF when we used scenarios from Gschrey et al. [2011] and Velders et al. [2009], or a more modest 13% decrease under RCP8.5. These differences equate to a 50–240 mW m⁻² reduction in global 2050 RF, equivalent to a cumulative emissions reduction of between 17 and 95 Pg CO₂-e emissions, or 0.5-2.8 years of global CO₂ emissions, at 2010 levels [Boden et al., 2013].

Our projections of SGHG emissions show that, if the proposed changes to the Montreal Protocol are implemented by 2050, the positive contributions to RF growth from SGHGs would be primarily due to the FFGHGs, SO₂F₂, and HFC-23 (Figure 2). CFC-12 will still remain the dominant forcing agent and will still be responsible for most of the year-to-year reduction in RF. However, all HCFCs and all but the longest-lived HFCs (particularly HFC-23) could also contribute significantly to the reduction in SGHG RF.

Provided that suitable low-global warming potential alternatives to the HFCs can be found [e.g., Molina et al., 2009; Daniel et al., 2011; Velders et al., 2012], which do not dramatically increase the energy intensity of the applications they are used in, the proposed changes to the Montreal Protocol would continue to make it a remarkably effective piece of global legislation for limiting the radiative impact of synthetic greenhouse gases.

Acknowledgments

We are particularly indebted to the insights and input from Archie McCulloch, who provided guidance on the generation of emissions scenarios. We are very thankful to the staff at the AGAGE and NOAA sites for their continuing dedication to the production of high-quality measurements of atmospheric trace gases. Matthew Rigby is supported by an advanced research fellowship from the UK Natural Environment Research Council (NERC) NE/I021365/1. The AGAGE research program is supported by the NASA Upper Atmospheric Research Program in the U.S. with grants NNX11AF17G to MIT and NNX11AF15G and NNX11AF16G to SIO, by NOAA, the UK Department of Food and Rural Affairs (DEFRA) and Department for Energy and Climate Change (DECC) in the UK, and by CSIRO and the Australian Government Bureau of Meteorology in Australia

The Editor thanks two anonymous reviewers for their assistance in evaluating this paper.

References

- Arnold, T., J. Mühle, P. K. Salameh, C. M. Harth, D. J. Ivy, and R. F. Weiss (2012), Automated measurement of nitrogen trifluoride in ambient air., Anal. Chem., 84(11), 4798-804, doi:10.1021/ac300373e. [Available at http://www.ncbi.nlm.nih.gov/pubmed/22607353.]
- Arnold, T., et al. (2013), Nitrogen trifluoride global emissions estimated from updated atmospheric measurements., Proc. Natl. Acad. Sci. U.S.A., 110(6), 2029-2034, doi:10.1073/pnas.1212346110. [Available at http://www.pubmedcentral.nih.gov/articlerender.fcgi? artid=3568375&tool=pmcentrez&rendertype=abstract, Accessed 26 September 2013.]
- Boden, T. A., G. Marland, and R. J. Andres (2013), Global, regional, and national fossil-fuel CO₂ emissions, in Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, U.S. Department of Energy, Oak Ridge, Tenn., doi:10.3334/CDIAC/00001_V2013.
- Clarke, L. E., J. A. Edmonds, H. D. Jacoby, H. M. Pitcher, J. M. Reilly, and R. G. Richels (2007), Scenarios of greenhouse gas emissions and atmospheric concentrations. United States Climate Change Science Program (USCCSP), [Available at http://www.ncbi.nlm.nih.gov/pubmed/22275275.]
- Culbertson, J. A., J. M. Prins, E. P. Grimsrud, R. A. Rasmussen, M. A. K. Khalil, and M. J. Shearer (2004), Observed trends for CF3-containing compounds in background air at Cape Meares, Oregon, Point Barrow, Alaska, and Palmer Station, Antarctica., Chemosphere, 55(8), 1109-1119, doi:10.1016/j.chemosphere.2003.11.002. [Available at http://www.ncbi.nlm.nih.gov/pubmed/15050808, Accessed 29 January 2014.]
- Daniel, J. S., et al. (2011), A focus on information and options for policymakers, in Scientific Assessment of Ozone Depletion, World Meteorological Organization, Geneva, Switzerland,
- EPA (2012), Summary report: Global anthropogenic non-CO₂ greenhouse gas emissions: 1990 2030, Washington, D. C. [Available at http:// www.epa.gov/climatechange/EPAactivities/economics/nonco2projections.html.]
- EPA (2013), Benefits of addressing HFCs under the Montreal Protocol, Washingto, n D. C. [Available at http://www.epa.gov/ozone/downloads/Benefits_of_Addressing_HFCs_Under_the_Montreal_Protocol_6-21-2013.pdf.]
- Estrada, F., P. Perron, and B. Martínez-López (2013), Statistically derived contributions of diverse human influences to twentieth-century temperature changes, Nat. Geosci., 6(11), 1-6, doi:10.1038/ngeo1999. [Available from http://www.nature.com/doifinder/10.1038/ ngeo1999, Accessed 11 November 2013.]
- Forster, P., et al. (2007), Changes in atmospheric constituents and in radiative forcing, in Climate Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change, edited by S. Solomon et al., Cambridge Univ. Press, Cambridge, U. K. and New York.
- Fraser, P. J., D. E. Oram, C. E. Reeves, S. A. Penkett, and A. McCulloch (1999), Southern Hemispheric halon trends (1978–1998) and global halon emissions, J. Geophys. Res., 104(D13), 15,985–15,999, doi:10.1029/1999JD900113. [Available at http://www.agu.org/pubs/crossref/1999/ 1999JD900113.shtml.]
- Fujino, J., R. Nair, M. Kainuma, Y. Matsuoka, and T. Masui (2006), Multi-gas mitigation analysis on stabilization scenarios using AIM Global Model, Energy J., SI2006(01), doi:10.5547/ISSN0195-6574-EJ-VolSI2006-NoSI3-17.
- Gschrey, B., W. Schwarz, C. Elsner, and R. Engelhardt (2011), High increase of global F-gas emissions until 2050, Greenh. Gas Meas. Manage, 85-92, doi:10.1080/20430779.2011.579352. [Available at http://www.tandfonline.com/doi/ref/10.1080/20430779.2011.579352#tabModule.] Hall, B. D., E. J. Dlugokencky, S. A. Montzka, G. Dutton, J. Mühle, and J. W. Elkins (2012), Long-lived greenhouse gases, in State of the Climate in 2012, vol. 94, edited by J. Blunden and D. S. Arndt, pp. S31-S33, Bull. Am. Meteorol. Soc, Boston, Mass.



- lvy, D. J., et al. (2012a), Atmospheric histories and growth trends of C4F10, C5F12, C6F14, C7F16 and C8F18, Atmos. Chem. Phys., 12(9), 4313–4325, doi:10.5194/acp-12-4313-2012. [Available at http://www.atmos-chem-phys.net/12/4313/2012/, Accessed 11 July 2012.]
- lvy, D. J., M. Rigby, M. Baasandorj, J. B. Burkholder, and R. G. Prinn (2012b), Global emission estimates and radiative impact of C4F10, C5F12, C6F14, C7F16 and C8F18, Atmos. Chem. Phys., 12(16), 7635–7645, doi:10.5194/acp-12-7635-2012. [Available at http://www.atmos-chemphys.net/12/7635/2012/, Accessed 14 November 2012.]
- Kim, J., et al. (2010), Regional atmospheric emissions determined from measurements at Jeju Island, Korea: Halogenated compounds from China. J. Geophys. Res., 37, L12801, doi:10.1029/2010GL043263.
- Levin, I., et al. (2010), The global SF6 source inferred from long-term high precision atmospheric measurements and its comparison with emission inventories, *Atmos. Chem. Phys.*, 10, 2655–2662, doi:10.5194/acp-10-2655-2010. [Available at http://www.atmos-chem-phys. net/10/2655/2010/.]
- Maione, M., U. Giostra, J. Arduini, F. Furlani, F. Graziosi, E. Lo Vullo, and P. Bonasoni (2013), Ten years of continuous observations of stratospheric ozone depleting gases at Monte Cimone (Italy)—Comments on the effectiveness of the Montreal Protocol from a regional perspective., Sci. Total Environ., 445-446, 155–164, doi:10.1016/j.scitotenv.2012.12.056. [Available at http://www.ncbi.nlm.nih.gov/pubmed/23333511, Accessed 29 January 2014.]
- Meinshausen, M., et al. (2011), The RCP greenhouse gas concentrations and their extensions from 1765 to 2300, Clim. Change, 109(1-2), 213–241, doi:10.1007/s10584-011-0156-z. [Available at http://link.springer.com/10.1007/s10584-011-0156-z, Accessed 11 November 2013.]
- Miller, B. R., and L. J. M. Kuijpers (2011), Projecting future HFC-23 emissions, *Atmos. Chem. Phys.*, *11*(24), 13,259–13,267, doi:10.5194/acp-11-13259-2011. [Available at http://www.atmos-chem-phys.net/11/13259/2011/. Accessed 26 July 2013.]
- Miller, B. R., R. F. Weiss, P. K. Salameh, T. Tanhua, B. R. Greally, J. Muhle, and P. G. Simmonds (2008), Medusa: A sample preconcentration and GC/MS detector system for in situ measurements of atmospheric trace halocarbons, hydrocarbons, and sulfur compounds, *Anal. Chem.*, 80(5), 1536–1545, doi:10.1029/2004GL022228.(8).
- Miller, B. R., et al. (2010), HFC-23 (CHF₃) emission trend response to HCFC-22 (CHCIF₂) production and recent HFC-23 emission abatement measures, *Atmos. Chem. Phys.*, 10(16), 7875–7890, doi:10.5194/acp-10-7875-2010. [Available at http://www.atmos-chem-phys.net/10/7875/2010/, Accessed 15 September 2010.]
- Molina, M., D. Zaelke, K. M. Sarma, S. O. Andersen, V. Ramanathan, and D. Kaniaru (2009), Reducing abrupt climate change risk using the Montreal Protocol and other regulatory actions to complement cuts in CO₂ emissions., *Proc. Natl. Acad. Sci. U.S.A.*, 106(49), 20,616–20,621, doi:10.1073/pnas.0902568106. [Available at http://www.pubmedcentral.nih.gov/articlerender.fcgi? artid=2791591&tool=pmcentrez&rendertype=abstract.]
- Montzka, S. A., S. Reimann, A. Engel, K. Krüger, S. O'Doherty, and W. T. Sturges (2011), Ozone-Depleting Substances (ODSs) and related chemicals, in *Scientific Assessment of Ozone Depletion*, pp. 112, World Meteorological Organization, Geneva, Switzerland.
- Mühle, J., et al. (2010), Perfluorocarbons in the global atmosphere: Tetrafluoromethane, hexafluoroethane, and octafluoropropane, *Atmos. Chem. Phys.*, *10*(11), 5145–5164, doi:10.5194/acp-10-5145-2010. [Available at http://www.atmos-chem-phys.net/10/5145/2010/, Accessed 15 September 2010.]
- O'Doherty, S., et al. (2004), Rapid growth of hydrofluorocarbon 134a and hydrochlorofluorocarbons 141b, 142b, and 22 from Advanced Global Atmospheric Gases Experiment (AGAGE) observations at Cape Grim, Tasmania, and Mace Head, Ireland, J. Geophys. Res., 109, D06310, doi:10.1029/2003JD004277. [Available at http://www.agu.org/pubs/crossref/2004/2003JD004277.shtml, Accessed 30 August 2010.]
- O'Doherty, S., et al. (2009), Global and regional emissions of HFC-125 (CHF2CF₃) from in situ and air archive atmospheric observations at AGAGE and SOGE observatories, *J. Geophys. Res.*, 114, D23304, doi:10.1029/2009JD012184. [Available at http://www.agu.org/pubs/crossref/2009/2009JD012184.shtml, Accessed 15 September 2010.]
- Oram, D. E., F. S. Mani, J. C. Laube, M. J. Newland, C. E. Reeves, W. T. Sturges, S. A. Penkett, C. A. M. Brenninkmeijer, T. Röckmann, and P. J. Fraser (2012), Long-term tropospheric trend of octafluorocyclobutane (c-C4F8 or PFC-318), *Atmos. Chem. Phys.*, 12(1), 261–269, doi:10.5194/acp-12-261-2012. [Available at http://www.atmos-chem-phys.net/12/261/2012/, Accessed 18 December 2013.]
- Prinn, R. G., et al. (2000), A history of chemically and radiatively important gases in air deduced from ALE/GAGE/AGAGE, *J. Geophys. Res.*, 105(D14), 17,751–17,792, doi:10.1029/2000JD900141. [Available at http://www.agu.org/pubs/crossref/2000/2000JD900141.shtml.]
- Ravishankara, A. R., S. Solomon, A. A. Turnipseed, and R. F. Warren (1993), Atmospheric lifetimes of long-lived halogenated species., *Science*, 259(5092), 194–199, doi:10.1126/science.259.5092.194. [Available at http://www.ncbi.nlm.nih.gov/pubmed/17790983, Accessed 4 October 2013.]
- Riahi, K., A. Grübler, and N. Nakicenovic (2007), Scenarios of long-term socio-economic and environmental development under climate stabilization, *Technol. Forecast. Soc. Change, 74*(7), 887–935, doi:10.1016/j.techfore.2006.05.026. [Available at http://linkinghub.elsevier.com/retrieve/pii/S0040162506001387, Accessed 19 November 2013.]
- Rigby, M., et al. (2010), History of atmospheric SF6 from 1973 to 2008, Atmos. Chem. Phys., 10(21), 10,305–10,320, doi:10.5194/acp-10-10305-2010. [Available from http://www.atmos-chem-phys.net/10/10305/2010/, Accessed 23 November 2010.]
- Rigby, M., A. L. Ganesan, and R. G. Prinn (2011), Deriving emissions time series from sparse atmospheric mole fractions, *J. Geophys. Res., 116*, D08306, doi:10.1029/2010JD015401. [Available at http://www.agu.org/pubs/crossref/2011/2010JD015401.shtml, Accessed 1 September 2011.]
- Rigby, M., et al. (2013), Re-evaluation of the lifetimes of the major CFCs and CH3CCl3 using atmospheric trends, Atmos. Chem. Phys.,
- 13(5), 2691–2702, doi:10.5194/acp-13-2691-2013. [Available at http://www.atmos-chem-phys.net/13/2691/2013/, Accessed 7 March 2013.]
 Saikawa, E., et al. (2012), Global and regional emission estimates for HCFC-22, Atmos. Chem. Phys., 12(21), 10,033–10,050, doi:10.5194/acp-12-10033-2012. [Available at http://www.atmos-chem-phys.net/12/10033/2012/. Accessed 2 November 2012.]
- Simmonds, P. G., S. O'Doherty, G. Nickless, G. A. Sturrock, R. Swaby, P. Knight, J. Ricketts, G. Woffendin, and R. Smith (1995), Automated gas chromatograph/mass spectrometer for routine atmospheric field measurements of the CFC replacement compounds, the hydrofluorocarbons and hydrochlorofluorocarbons, *Anal. Chem.*, 67(4), 717–723, doi:10.1021/ac00100a005. [Available at http://pubs.acs.org/doi/abs/
- Simmonds, P. G., B. R. Greally, S. Olivier, G. Nickless, K. M. Cooke, and R. N. Dietz (2002), The background atmospheric concentrations of cyclic perfluorocarbon tracers determined by negative ion-chemical ionization mass spectrometry, *Atmos. Environ.*, *36*(13), 2147–2156, doi:10.1016/S1352-2310(02)00194-2. [Available at http://linkinghub.elsevier.com/retrieve/pii/S1352231002001942.]
- Smith, S. J., and T. M. L. Wigley (2006), Multi-gas forcing stabilization with MiniCAM, Energy J., Sl2006(01), doi:10.5547/ISSN0195-6574-EJ-VolSl2006-NoSl3-19. [Available at http://www.iaee.org/en/publications/ejarticle.aspx?id=2201, Accessed 20 November 2013.]
- SPARC (2013), SPARC Report No. 6: Report on the Lifetimes of Stratospheric Ozone-depleting Substances, Their Replacements, and Related Species, edited by M. Ko et al., World Climate Research Programme, Zurich, Switzerland.
- Straume, A. G., R. N. Dietz, E. N. Koffi, and K. Nodop (1998), Perfluorocarbon background concentrations in Europe, *Atmos. Environ.*, 32(24), 4109–4122, doi:10.1016/S1352-2310(98)00193-9. [Available at http://linkinghub.elsevier.com/retrieve/pii/S1352231098001939, Accessed 22 November 2013.]

10.1021/ac00100a005, Accessed 4 October 2013.1



- Sturges, W. T., T. J. Wallington, M. D. Hurley, K. P. Shine, K. Sihra, A. Engel, D. E. Oram, S. A. Penkett, R. Mulvaney, and C. A. M. Brenninkmeijer (2000), A potent greenhouse gas identified in the atmosphere: SF5CF3, *Science*, 289, 611–613. [Available at http://www.ncbi.nlm.nih.gov/pubmed/10915622.]
- Velders, G. J. M., and J. S. Daniel (2013), Uncertainty analysis of projections of ozone-depleting substances: Mixing ratios, EESC, ODPs, and GWPs, Atmos. Chem. Phys. Discuss., 13(10), 28,017–28,066, doi:10.5194/acpd-13-28017-2013. [Available at http://www.atmos-chem-phys-discuss.net/13/28017/2013/, Accessed 18 November 2013.]
- Velders, G. J. M., D. W. Fahey, J. S. Daniel, M. Mcfarland, and S. O. Andersen (2009), The large contribution of projected HFC emissions to future climate forcing, *Proc. Natl. Acad. Sci. U.S.A.*, 106(27), 10,949–10,954.
- Velders, G. J. M., A. R. Ravishankara, M. K. Miller, M. J. Molina, J. Alcamo, J. S. Daniel, D. W. Fahey, S. A. Montzka, and S. Reimann (2012), Preserving Montreal Protocol climate benefits by limiting HFCs., *Science*, 335(6071), 922–923, doi:10.1126/science.1216414. [Available from http://www.ncbi.nlm.nih.gov/pubmed/22362993, Accessed 23 July 2013.]
- Vollmer, M. K., et al. (2011), Atmospheric histories and global emissions of the anthropogenic hydrofluorocarbons HFC-365mfc, HFC-245fa, HFC-227ea, and HFC-236fa, *J. Geophys. Res.*, 116, D08304, doi:10.1029/2010JD015309. [Available from http://www.agu.org/pubs/crossref/2011/2010JD015309.shtml. Accessed 29 April 2011.]
- Van Vuuren, D. P., M. G. J. Elzen, P. L. Lucas, B. Eickhout, B. J. Strengers, B. Ruijven, S. Wonink, and R. Houdt (2007), Stabilizing greenhouse gas concentrations at low levels: An assessment of reduction strategies and costs, *Clim. Change*, *81*(2), 119–159, doi:10.1007/s10584-006-9172-9. [Available at http://link.springer.com/10.1007/s10584-006-9172-9, Accessed 20 November 2013.]
- Weiss, R. F., J. Mühle, P. K. Salameh, and C. M. Harth (2008), Nitrogen trifluoride in the global atmosphere, *Geophys. Res. Lett.*, 35, L20821, doi:10.1029/2008GL035913. [Available at http://doi.wiley.com/10.1029/2008GL035913, Accessed 26 September 2013.]
- Yao, B., M. K. Vollmer, L. X. Zhou, S. Henne, S. Reimann, P. C. Li, A. Wenger, and M. Hill (2012), In-situ measurements of atmospheric hydrofluorocarbons (HFCs) and perfluorocarbons (PFCs) at the Shangdianzi regional background station, China, *Atmos. Chem. Phys.*, 12(21), 10,181–10,193, doi:10.5194/acp-12-10181-2012. [Available at http://www.atmos-chem-phys.net/12/10181/2012/, Accessed 27 January 2014.]

Frratum

In the originally published version of this article, four instances of text were incorrectly typeset due to typographical errors. The following have since been corrected and this version may be considered the authoritative version of record.

In section 3, G_{2} -CO₂e has been changed to G_{2} -CO₂e. In Figure 3, G_{2} -CO₂e has been changed to G_{2} -CO₂e. In Figure 3, G_{2} -CO₂e has been changed to G_{2} -CO₂e.