Controlled release of paclitaxel from a self-assembling peptide hydrogel formed in situ and antitumor study in vitro

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Controlled release of paclitaxel from a self-assembling peptide hydrogel formed in situ and antitumor study in vitro

Background: A nanoscale injectable in situ-forming hydrogel drug delivery system was developed in this study. The system was based on a self-assembling peptide RADA16 solution, which can spontaneously form a hydrogel rapidly under physiological conditions. We used the RADA16 hydrogel for the controlled release of paclitaxel (PTX), a hydrophobic antitumor drug.

Methods: The RADA16-PTX suspension was prepared simply by magnetic stirring, followed by atomic force microscopy, circular dichroism analysis, dynamic light scattering, rheological analysis, an in vitro release assay, and a cell viability test.

Results: The results indicated that RADA16 and PTX can interact with each other and that the amphiphilic peptide was able to stabilize hydrophobic drugs in aqueous solution. The particle size of PTX was markedly decreased in the RADA16 solution compared with its size in water. The RADA16-PTX suspension could form a hydrogel in culture medium, and the elasticity of the hydrogel showed a positive correlation with peptide concentration. In vitro release measurements indicated that hydrogels with a higher peptide concentration had a longer half-release time. The RADA16-PTX hydrogel could effectively inhibit the growth of the breast cancer cell line, MDA-MB-435S, in vitro, and hydrogels with higher peptide concentrations were more effective at inhibiting tumor cell proliferation. The RADA16-PTX hydrogel was effective at controlling the release of PTX and inhibiting tumor cell growth in vitro.

Conclusion: Self-assembling peptide hydrogels may work well as a system for drug delivery.

Keywords: self-assembling peptide, hydrogel, paclitaxel, drug delivery, antitumor

Introduction
Paclitaxel (PTX) is one of the most effective antitumor drugs found in nature in the past few decades.1-2 It promotes polymerization of tubulin dimers, which results in the formation of highly stable nonfunctional microtubules in the absence of guanosine triphosphate and microtubule-associated proteins, ultimately preventing cell division.3,4 Because of its unique mechanism of action against tumor cells, PTX has demonstrated excellent therapeutic efficacy for a wide range of cancers, although it works especially well for treatment of ovarian and breast cancers.1

Paclitaxel is hydrophobic and is poorly soluble in water. Currently, the PTX used clinically is in a vehicle composed of Cremophor® EL and ethanol at a 50:50 (v/v) ratio. Cremophor EL is a nonionic polyethoxylated castor oil solubilizer that can enhance PTX solubility. However, this solvent causes severe hypersensitive reactions and cytotoxicity and is incompatible with polyvinyl chloride, which is commonly used in injection systems.2,5 In addition, PTX can cause serious side effects at excessive
doses, such as bone marrow suppression and cardiac rhythm disturbances. Therefore, researchers have developed a variety of vehicles, including microspheres, nanoparticles, surgical pastes, polymeric micelles, and hydrogels, to control PTX release and to eliminate the toxicity of Cremophor EL to the body.

Hydrogels have been commonly used in drug delivery, and various natural polymers and synthetic copolymer hydrogels, such as chitosan, hyaluronic acid, alginate, poly(methyl methacrylate), and polyethylene glycol, have been studied. These compounds usually form hydrogel-drug complexes outside of the body and are then delivered into the body. However, the formation of hydrogels frequently involves ultraviolet photopolymerization and various chemical cross-linking techniques, and there is a potential risk that toxic reagents may fail to be removed completely prior to hydrogel implantation. Furthermore, some hydrogels have high elasticity, which generally prevents their extrusion through a syringe needle. Therefore, an alternative vehicle is necessary.

Self-assembling peptides are a type of designed biomaterial and have been used in a number of biomedical applications, such as cell culture, tissue engineering, and drug delivery. Self-assembling peptides can spontaneously assemble to cross-linked nanofibers of 10–20 nm in diameter and form a stable second structure of β-sheet in aqueous solution. Under physiological conditions, the peptide solution can rapidly self-assemble into hydrogels, which contain more than 99% (w/v) water. Previous studies have used RADA16 peptide to slow the release of proteins and localized therapies through injection into a particular tissue. Because the process of hydrogel formation is ionic-dependent and is irrelevant with ultraviolet or chemical cross-linking, it was supposed that RADA16 may serve well as a potential drug delivery vehicle.

The purpose of this study was to develop an injectable in situ gel drug delivery system based on a self-assembling peptide. The incorporation of peptide and drug was achieved simply by mix and magnetic stirring. To investigate the interaction of RADA16-PTX, the release properties of hydrogel, and antitumor efficiency in the current study, we performed a morphology analysis, circular dichroism, dynamic light scattering, rheological test, in vitro release test, and a cell viability test.

**Materials and methods**

**Incorporation of peptide and PTX**

Self-assembling peptide RADA16 (n-RADARADARA-DARADA-c) was commercially synthesized and purified by Shanghai Biotech Bioscience and Technology Co Ltd (Shanghai, China). Paclitaxel was purchased from the National Institute for the Control of Pharmaceutical and Biological Products (Beijing, China). Peptide solutions at concentrations of 5 and 10 mg/mL were prepared by dissolving RADA16 peptide powder in Milli-Q water with 10 minutes of sonication. The peptide solution incorporated with PTX was prepared by adding peptide solution (5 or 10 mg/mL) into a glass vial with PTX at a final concentration of 1 mg/mL, followed by magnetic stirring for 48 hours, whereas the control sample was formed by adding Milli-Q water instead of peptide solution. All of the vials and solvents were sterilized, and the samples were prepared in a biological safety cabinet to avoid possible contamination for cell culture experiments.

**Atomic force microscopy**

The morphology of samples was observed using an atomic force microscope (SPI4000 Probe Station, Seiko Instruments Inc, Chiba, Japan) with a tapping mode. For the preparation of the samples, RADA16 solution or RADA16-PTX suspension was diluted 50 times in Milli-Q water, and then 10 µL of diluted solution was evenly placed on a freshly cleaved mica substrate. The mica surface was then rinsed with Milli-Q water to remove unattached peptide and air-dried. The measurement was performed at room temperature and all images were recorded with a resolution of 512 × 512 pixels. The typically scan parameters were set as follows: scan speed 1.00 Hz, amplitude 1–1.2 V, and integral and proportional gains of 0.2–0.3 and 0.03–0.04, respectively.

**Dynamic light scattering**

The particle size distribution of the RADA16 solution and RADA16-PTX suspension was investigated on a Nano-Particle Size Analyzer (LB-550, Horiba Ltd, Kyoto, Japan) with appropriate viscosity and refractive index settings, and the temperature was maintained at 25°C during measurement. The RADA16 solution and RADA16-PTX suspension was diluted 20 times with Milli-Q water, and each sample was tested at least three times to generate the intensity-based size distribution plot report.

**Circular dichroism spectroscopy**

The circular dichroism measurement was performed at 25°C on an Aviv Model 400 (Aviv Biomedical Inc, Lakewood, NJ) with a 2 mm path length quartz cuvette. All samples of RADA16 solution and RADA16-PTX suspension were
In vitro antitumor cell study

The human breast cancer cell line MDA-MB-435S was used for the in vitro cell viability study. The cells were cultured in L-15 medium containing 10% fetal bovine serum at 5% CO₂. When cells reached 85% confluence, they were detached from cell culture flasks with trypsin and resuspended in cell culture medium at a concentration of 10⁵ cells/mL. After 200 µL of the cell suspension was added to each well of a flat-bottom 96-well plate and incubated overnight, the culture medium was removed. Twenty microliters of diluted PTX (Taiji Pharmaceutical Co Ltd, Sichuan, China), RADA16, or RADA16-PTX suspension was then pipetted into the central bottom of the well and equilibrated with 200 µL of culture medium to initiate peptide gel formation. The culture medium for all groups was changed daily. The cell viability assay was carried out at prearranged times.

Cell viability was estimated using a modified MTT assay in this study. Briefly, at different time points, 200 µL of serum-free medium and 20 µL of MTT solution (5 mg/mL in phosphate-buffered saline) were added to each sample, followed by incubation at 37°C for four hours to allow MTT formazan formation. The medium was then removed, and the converted dye was dissolved with 200 µL of 10% sodium dodecyl sulfate in 0.01 M HCl. After all crystals were solubilized, the optical density of the solution was determined at 570 nm against an sodium dodecyl sulfate solution blank using an enzyme-linked immunosorbent assay plate reader. Four parallel replicates were read for all samples. Cell viability was calculated by the following equation:

\[
\text{Cell viability (\%)} = \left( \frac{\text{Abs}_{\text{test cells}}}{\text{Abs}_{\text{control cells}}} \right) \times 100\%
\]

in which Abs_{\text{test cells}} and Abs_{\text{control cells}} were the absorbencies of cells with different treatments and cells incubated with medium only, respectively.

Cell viability was also assessed by a Live/Dead assay. A work solution containing both 1 µM calcein AM and 2 µM ethidium homodimer in phosphate-buffered saline were prepared according to the Live-Dead assay (Molecular Probes®, Invitrogen, Carlsbad, CA) package instructions, and 200 µL of work solution was added to each well. All cells were imaged using a fluorescence microscope.

Statistical analysis

All quantitative data were shown as the mean ± standard deviation. Statistical analysis was performed using SPSS...
Results and discussion
Interaction of RADA16 and PTX

The self-assembling RADA16 peptide is a designed nano-scale biomaterial, the sequence of which contains alternating hydrophilic and hydrophobic amino acid residues. Paclitaxel is a hydrophobic antitumor drug that is poorly soluble in water. After stirring for 48 hours, PTX in RADA16 solution was changed to a colloidal suspension, while the same component in pure water was still clear, and a significant amount of PTX powder was visible (Figure 1). The data suggested that RADA16 interacted with PTX in aqueous solution. Figure 2A and C showed the typical morphology of RADA16, which consisted of multiple cross-linked nanofibers that were $21.5 \pm 3.3$ nm in width and $2.2 \pm 0.2$ nm in height. After stirring with PTX for 48 hours, the morphology of RADA16 changed. Some bulged nanofibers and aggregates could be seen (Figure 2B and D), and the width and height of those nanofibers were $45.3 \pm 10.6$ nm and $5.6 \pm 2.3$ nm, respectively. Some PTX particles coated by peptide nanofibers were also observed (Figure 2B and D). From the circular dichroism analysis, the RADA16 peptide had a typical $\beta$-sheet structure (minimum molar ellipticity at 215–217 nm and maximum at 195–206 nm), with 50.6% $\beta$-sheet content (Figure 3 and Table 1, $S(r) + S[d]$) according to the CDPro software analysis. However, when RADA16 was stirred with PTX for 48 hours, there was an approximately 18.4% decrease in the $\beta$-sheet content and a 19.8% increase in the unordered structure content. Dynamic light scattering measurements showed that the diameter of nanofibers from a pure RADA16 solution was in the range of 10–50 nm. When PTX was stirred with 0.5% and 1% RADA16 solution, the mean particle sizes were $2744.7 \pm 551.3$ nm and $2569.4 \pm 434.9$ nm, respectively, but there was no difference between the 0.5% and 1% RADA16 solution groups (Figure 4). When PTX was stirred with Milli-Q water, the mean particle size was over 6 $\mu$m, and dynamic light scattering measurements could not be conducted.

The circular dichroism data showed that the $\beta$-sheet structure of RADA16 was impaired to some extent by the addition of PTX, a result that was in accordance with the change in morphology on atomic force microscopy. The particle size of PTX was obviously decreased in the RADA16 solution compared with that in the water. All of the data indicated that RADA16 did not simply mix with PTX but rather interacted with it. As illustrated in Figure 5, RADA16 consists of repeated sequences of Arg-Ala-Asp-Ala, in which Arg and Asp are hydrophilic amino acid residues and Ala is a hydrophobic amino acid residue; the $\beta$-sheet structure of the peptide therefore had both hydrophobic and hydrophilic regions. Because PTX is a typical hydrophobic compound, the hydrophobic regions of peptide would be adsorbed to the hydrophobic surface of separated PTX particles under...
stirring, a phenomenon that resulted in the impaired β-sheet structure of RADA16 shown in the circular dichroism data. The atomic force microscopy images showed that there were many bulged RADA16 nanofibers, which may have resulted from PTX molecules that were incorporated with nanofibers or from the aggregation of nanofibers induced by hydrophobic interactions. The data indicate that the self-assembling peptide and PTX could interact with each other and that the amphiphilic peptide was able to stabilize hydrophobic drugs in aqueous solution, a finding that was in agreement with previous reports.20–22 This type of interaction also helps to increase the solubility of PTX and decrease the mean particle size of PTX in RADA16 solution under stirring. Therefore, the RADA16-PTX suspension could potentially be injected into the body with a syringe needle. These results indicated that PTX molecules and particles can coexist in RADA16-PTX suspension, and stabilized PTX can be obtained in this system. However, further investigation is necessary to determine the percentage of each state in the complexes.

**Characterization of RADA16-PTX hydrogel formation**

The gelation of RADA16-PTX suspension was characterized by rheometry. As in previous reports, the RADA16 solution was able to form a hydrogel in ionic environments, such as phosphate-buffered saline, physiological saline, and body fluid. However, could the RADA16-PTX suspension gel under physiological conditions? The test was performed at 37°C, and phosphate-buffered saline was used to simulate body fluid. The storage modulus (G') and loss modulus (G'') responded to the elasticity and viscosity of materials, respectively. As shown in Figure 6, the frequency sweep indicated that the value of the storage modulus (G') in the RADA16-PTX suspension was larger than that of the loss modulus (G'') both in 0.5% and 1% RADA16, and all of the low values (<10 Pa) for the storage and loss moduli were typical for low viscosity systems with a low degree of elasticity. With the addition of medium, significant changes occurred, ie, the storage and loss moduli increased. The storage modulus of 0.5% RADA16 was over 300 Pa and that of 1% RADA16 was over 1000 Pa, whereas the original values for both 0.5% and 1% RADA16 were under 10 Pa. The great increase in the storage modulus suggested the formation of a strong hydrogel. At the mean time, it was observed that the storage modulus of 0.5% RADA16 hydrogel was much lower than that of the 1% RADA16 hydrogel, which indicated that the concentration

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**Table 1** The calculated secondary structure fractions of peptide

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<th>Items</th>
<th>Secondary structure fractions (%)</th>
<th>H (r)</th>
<th>H (d)</th>
<th>S (r)</th>
<th>S (d)</th>
<th>Turn</th>
<th>Unrd</th>
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<td>RADA16</td>
<td></td>
<td>0.4</td>
<td>1.9</td>
<td>33.8</td>
<td>16.8</td>
<td>20.4</td>
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<tr>
<td>RADA16-PTX</td>
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<td>27.1</td>
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Abbreviations: H (r), regular α-helix; H (d), distorted α-helix; S (r), regular β-strand; S (d), distorted β-strand; Turn, β-turn structure; Unrd, unordered structure.
of peptide had a positive correlation with the elasticity of the hydrogel. The results demonstrated that the RADA16-PTX suspension could form a highly elastic hydrogel under physiological conditions. As an in situ forming gel drug delivery system, the ideal vehicle should be an injectable solution in vitro and change to a gel in vivo. RADA16 was able to form a hydrogel rapidly under physiological conditions without heating or chemical reaction, making it a favorable vehicle for drug delivery.

**In vitro PTX release from RADA16-PTX hydrogel**

Paclitaxel is a hydrophobic drug that is poorly soluble in water. To maintain sink conditions throughout the release
process, sodium dodecyl sulfate was added to the release medium to increase the solubility of PTX. The release was observed over a period of five days. As shown in Figure 7B, the RADA16-PTX suspension could form a milky-white gel in phosphate-buffered saline, and the volume of hydrogel was obviously reduced over the release time. The resulting release profiles of PTX from the RADA16-PTX hydrogels are shown in Figure 7A. When a hydrogel was prepared with 1% RADA16, approximately 40%-45% of incorporated PTX was released from the hydrogel within the first day, with a half-release time of 24-30 hours. In contrast, more than 60% of the PTX was released from the hydrogel within one day for

**Figure 6** Frequency sweep of RADA16-PTX hydrogel in rheological analysis. (A) 0.5% RADA16-PTX and (B) 1% RADA16-PTX. The results showed that the storage modulus of 0.5% RADA16-PTX hydrogel was much lower than that of 1% RADA16-PTX hydrogel.

*Abbreviation:* PTX, paclitaxel.
the 0.5% RADA16 hydrogel, and the half-release time was 8–10 hours. These results clearly demonstrated that a higher concentration of peptide in the hydrogel produced slower drug release rates. The initial burst effect was lower for the 1% RADA16 hydrogel compared with the 0.5% RADA16 hydrogel, and an increase in peptide concentration slowed the release rates of the loaded drug, resulting in a longer half-release time for the PTX. The atomic force microscopic image showed that many nanoparticles were present in the supernatant of the release medium (Figure 8A), and the width and height of the nanoparticles were 83.1 ± 30.3 nm and 4.9 ± 2.9 nm, respectively. However, sodium dodecyl sulfate was also an amphiphilic molecule, which might form assemblies in aqueous solution. Thus, an additional atomic force microscopy test with only sodium dodecyl sulfate in the release medium was performed to verify this result. The resulting images showed that sodium dodecyl sulfate was obviously accumulated on the surface of mica, and no similar size of nanoparticles formed (data not shown). These data suggest that PTX was encapsulated in peptide nanofibers and that RADA16-PTX nanoparticles were formed in the release process. A schematic representation of a possible model for PTX release from the PTX-RADA16 hydrogel is shown in Figure 8B. Considering the data from morphology and rheology analysis, it was found that the higher concentration of peptide solution had more cross-linked nanofibers and led to the formation of a more elastic hydrogel. Because hydrogel formed from a higher concentration of peptide showed increased stability with a higher density of nanofibers and smaller sizes of pores, the probability of specific interactions

![Figure 7](image1.png)  
*Figure 7* In vitro release of PTX from RADA16-PTX hydrogel with different peptide concentrations. (A) The rates of released PTX at different points and (B) photographs of 1% RADA16-PTX hydrogel in phosphate-buffered saline at 0.5 hours and 48 hours after the start of the release test.  
*Abbreviation:* PTX, paclitaxel.

![Figure 8](image2.png)  
*Figure 8* In vitro release of PTX from RADA16-PTX hydrogel. (A) Atomic force microscopy image of supernatant in release medium, in which many nanoparticles could be seen. (B) Schematic representation of modeling for PTX release from PTX-RADA16 hydrogel.  
*Abbreviation:* PTX, paclitaxel.
between PTX and peptide nanofibers and diffusion hindrance was increased, yielding a decrease in the diffusivity of PTX and finally resulting in slower drug release rates. These results suggested that the release rates of PTX can be controlled by changing the peptide concentration, so the release time could be extended by adding a greater amount of peptide in future studies. In addition, the addition of sodium dodecyl sulfate as a surfactant can increase the release rates of PTX from the hydrogel, an effect that would be caused by the increased degradation rates of the RADA16 peptide.

**Inhibition of RADA16-PTX hydrogel on the growth of tumor cells**

As an antiproliferation drug, PTX is a potent inhibitor of angiogenesis, cell migration, and collagenase production. Because it has been demonstrated that PTX has significant antitumor activity against various solid tumors, including breast cancer, the human breast cancer cell line MDA-MB-435S was used to study the antitumor efficiency of the RADA16-PTX hydrogel. As in previous reports, the antiproliferative activity of the complex was dependent on the concentration of PTX. To simulate the in vivo situation and optimize the antitumor test, PTX10 µg was administered to each well of the 96-well plate, and the culture medium was changed every day. In our study, the data showed that single use of RADA16 produced no inhibition of tumor cell proliferation, while the RADA16-PTX hydrogel inhibited growth of tumor cells effectively. As shown in Figure 9, there was no significant difference in cell viability rates between the PTX-treated group and both RADA16-PTX-treated groups at day 1. At day 3, there was a significant difference between the PTX group and the 1% RADA16-PTX group, but there was no significant difference between the 0.5% and 1% RADA16-PTX groups. At days 5 and 7, compared with the PTX group, both RADA16-PTX groups demonstrated lower cell viability rates, and the cell viability rate in the 1% RADA16-PTX group was significantly lower than that in the 0.5% RADA16-PTX group. These data suggest that PTX and 0.5% RADA16-PTX had poor ability to inhibit cell growth after days 3 and 5, respectively, whereas 1% RADA16-PTX hydrogel maintained relatively high inhibitory activity even at day 7. It was concluded that the RADA16 hydrogel could control release of PTX effectively and that a higher concentration of peptide hydrogel resulted in an increased release time for PTX. This result was in accordance with those from the in vitro release test. The fluorescence images showed that the tumor cells in the RADA16 group continued to proliferate over the course of the observed days (Figure 10). By contrast, in the PTX group and both RADA16-PTX groups, only a few cells could be seen at day 3, which indicated that cell proliferation was inhibited by PTX. However, the number of tumor cells in the PTX group was slightly higher than that in both groups.
RADA16-PTX groups. At day 5, in comparison with the 1% RADA16-PTX group, more tumor cells were observed in the PTX and 0.5% RADA16-PTX groups. At day 7, compared with previous days, an increased number of tumor cells was observed in the PTX group and both RADA16-PTX groups, but fewer cells were seen in the 1% RADA16-PTX group than those in the PTX group and the 0.5% RADA16-PTX group. It was concluded that the 1% RADA16-PTX hydrogel had a longer lasting inhibitory effect on proliferation of tumor cells than did the other tested groups.

These results indicate that the self-assembling peptide system could be used to control the release of PTX. The release time of PTX from the hydrogel showed increased correlation with the concentration of peptide, and a more elastic hydrogel led to stronger interactions between the drug and the nanofibers, which decreased the diffusion rates of PTX and resulted in longer drug release times. The spontaneously gelling ability of RADA16-PTX suspension in an ionic environment provided the controlled release properties. The RADA16 hydrogel prolonged the inhibitory effect of PTX on growth of breast cancer cells in vitro. This injectable, in situ-forming hydrogel system has great potential as a delivery vehicle for various hydrophobic drugs in future studies.

Conclusion

In conclusion, this study found that the RADA16-PTX complex could form colloidal suspensions after magnetic stirring for 48 hours and that PTX was stabilized in RADA16-containing aqueous solutions. RADA16-PTX suspensions were able to form hydrogel spontaneously in ionic solutions. The RADA16-PTX hydrogel formed was of high elasticity and demonstrated controlled drug release properties. The RADA16 hydrogel prolonged the inhibitory effect of PTX on growth of breast cancer cells in vitro. This injectable, in situ-forming hydrogel system has great potential as a delivery vehicle for various hydrophobic drugs in future studies.

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Disclosure

The authors report no conflicts of interest in this work.

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