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Progressive Saturation Improves the Encapsulation of Functional Proteins in Nanoscale Polymer Vesicles

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ABSTRACT

Purpose: To develop a technique that maximizes the encapsulation of functional proteins within neutrally charged, fully PEGylated and nanoscale polymer vesicles (i.e. polymersomes).

Methods: Three conventional vesicle formation methods were utilized for encapsulation of myoglobin (Mb) in polymersomes of varying size, PEG length, and membrane thickness. Mb concentrations were monitored by UV-Vis spectroscopy, inductively coupled plasma optical emission spectroscopy (ICP-OES) and bicinchoninic acid (BCA) assay. Suspensions were subject to protease treatment to differentiate the amounts of surface-associated vs. encapsulated Mb. Polymersome sizes and morphologies were monitored by dynamic light scattering (DLS) and by cryogenic transmission electron microscopy (cryo-TEM), respectively. Binding and release of oxygen were measured using a Hemeox analyzer.

Results: Using the established “thin-film rehydration” and “direct hydration” methods, Mb was found to be largely surface-associated with negligible aqueous encapsulation within polymersome suspensions.

Through iterative optimization, a novel “progressive saturation” technique was developed that greatly increased the final concentrations of Mb (from < 0.5 to > 2.0 mg/mL in solution), the final weight ratio of Mb-to-polymer that could be reproducibly obtained (from < 1 to > 4 w/w% Mb/polymer), as well as the overall efficiency of Mb encapsulation (from < 5 to > ~90%). Stable vesicle morphologies were verified by cryo-TEM; the suspensions also displayed no signs of aggregate formation for > 2 weeks as assessed by DLS. “Progressive saturation” was further utilized for the encapsulation of a variety of other proteins, ranging in size from 17 to 450 kDa.

Conclusions: Compared to established vesicle formation methods, “progressive saturation” increases the quantities of functional proteins that may be encapsulated in nanoscale polymersomes.

KEY WORDS: Protein, Encapsulation, Myoglobin, Polymersome, Vesicle

ABBREVIATIONS

BSA: bovine serum albumin
BCA: bicinchoninic acid
Cryo-TEM: cryogenic transmission electron microscopy
DLS: dynamic light scattering
Hb: hemoglobin
ICP-OES: inductively coupled plasma optical emission spectroscopy
IgG: immunoglobulin G
mBlue: methylene blue
Mb: myoglobin
%Mb EE: myoglobin encapsulation efficiency
Mb:PBS: volume ratio of myoglobin solution to phosphate-buffered saline
metMb: metmyoglobin
oxyMb: oxygenated (ferrous) myoglobin
PAA: poly(L-amino acid)
PAAc: poly(acrylic acid)
PCL: poly(ϵ -caprolactone)
PBD: polybutadiene
PDEAEMA: poly(2-(diethylamino)ethyl methacrylate)
PDPA: poly(2-(diisopropylamino)ethyl methacrylate)

PEG: polyethylene glycol
PEO: polyethylene oxide
PEO-b-PBD: poly(ethylene oxide)-block-poly(1,2 butadiene)
PEM: polymersome-encapsulated myoglobin
PIAT: poly(lisocyanalanine(2-thiophen-3-yl-ethyl)amide)₅₀
PLA: poly(lactic acid)
PMPC: poly(2-(methacryloyloxy)ethyl phosphorylcholine)
PNIPAM: poly(N-isopropylacrylamide)
PPS: polypropylene sulfide
PSM: polymersome

INTRODUCTION

Proteins have recently shown outstanding pharmaceutical potential due to their ability to overcome significant limitations associated with traditional small molecule therapies (1, 2). When compared to the interaction of a small-molecule with its biological target, protein therapeutics have higher folding energies (typically ~7-20 kcal/mol) that allow for the adoption of larger and more precise three-dimensional configurations, which are often required for efficient binding and/or control of complex biological function (3). As such, proteins can achieve superior binding selectivity and more potent on-target activity. Currently, the modest number of macromolecular therapies in use, including the ~200 protein drugs available worldwide, has demonstrated the remarkable potential of functional proteins as new leads in drug development (4-6). Nevertheless, several barriers have hindered the ready development of macromolecules as human therapeutics, including: (i) the difficulty and/or expense of commercial scale production; (ii) biochemical instability that occurs in pathophysiologic environments or with prolonged storage; (iii) short circulatory half-lives; (iv) large steric hindrance that prevents effective tissue penetration; and, (v) risks associated with their potential to promote severe adverse effects, such as immunologic and anaphylactic reactions (5).

To overcome some of these limitations, most pharmaceutical compounds either employ biocompatible polymers (e.g. polyethylene glycol (PEG) or hyaluronic acid) or liposomes (i.e. lipid based vesicles) for protein complexation and *in vivo* delivery. A vast and expanding body of literature, however, has

demonstrated that synthetic nanoparticles exhibit superior properties and can enhance drug delivery as compared to these more conventional formulations, which are limited by incomplete PEGylation and/or unstable encapsulation. Among the classes of nanoparticle vehicles, polymersomes, or self-assembled polymer vesicles comprised of amphiphilic block copolymers (7-12), are an emerging nanoscale delivery platform. Possessing thick polymeric membranes, polymersomes enable: (i) facile and stable loading of diverse therapeutic payloads through non-covalent interactions; (ii) improved mechanical stability that is 5-50 greater than that of liposomes or micellar structures constructed from similar molecular weight copolymers; (iii) a rich diversity in biochemical properties, which are imparted by their construction from a variety of copolymer compositions and which include: fully PEGylated surfaces and tunable *in vivo* circulation times, site-specific targeting, environmental responsiveness, as well as complete biodegradation; and, (iv) economic and large scale production that obviates the need for costly post-manufacturing purification (7, 9, 11, 13-15).

While the incorporation of proteins in long-circulating nanoparticles may enhance their pharmacologic performance and improve their on-target activity, most encapsulation efforts have utilized electrostatic interactions to incorporate a handful of highly anionic proteins (16); others have relied on chemical or genetic modification of the original protein for efficient and reproducible nanoparticle formation (17-21). A major unaddressed challenge, however, is to establish and validate a generalizable method for the incorporation of large quantities of native protein in neutrally charged and/or fully PEGylated nanoparticles, which could be employed to prevent protein deactivation, improve circulatory half-life, achieve tissue localization, and minimize systemic activity (22). Most nanoparticle fabrication techniques also involve the input of thermal, electric, ultrasonic or mechanical energy for particle formation (8, 23-25); alternative methods use organic co-solvents (26, 27). In most cases, these fabrication protocols damage the structure and/or function of the protein, making nanoparticle encapsulation more challenging and obviating its utility (28).

In the case of polymersomes, several methods have been developed for protein encapsulation, including: 1) rehydration of dry polymer (e.g. the “thin-film rehydration” technique) (11, 29-31) or the extrusion of preformed vesicles (e.g. by “hollow-fiber extrusion”) in the presence of protein solution (32); 2) “direct hydration” (22, 33); and, 3) “electroporation” (34, 35). Adopted from the liposome literature and widely employed for the encapsulation of organic small molecules, these methods result in low yields of polymersome-encapsulated protein (e.g. “thin-film rehydration”), are limited to small-scale preparations (< 1 mL; e.g. “direct hydration”), and/or are best employed for application with a small number of highly charged proteins (e.g. those amenable to vesicle encapsulation by “electroporation”). While the “hollow-fiber extrusion” technique has been used for large-scale preparations of liposome-encapsulated protein (32), elevated temperatures and pressures are required for polymersome formation, which have limited its widespread applicability.

Here, we sought to perform a comparative and quantitative study so as to establish a generalizable method for producing scalable quantities of neutrally charged, fully PEGylated, and nanoscale polymersomes that encapsulate functional proteins. We first examined differences in small molecule and protein encapsulation by employing two related and extensively utilized polymersomes formulations, which were both comprised of diblock copolymers of poly(ethylene oxide)-block-poly(1,2 butadiene) (PEO-b-PBD) but that differed with respect to molecular weight and, ultimately, vesicle membrane thickness (i.e. OB18 and OB2-9) (**Table I**; (36, 37)). Methylene blue (mBlue; Mw = 319.85 g/mol) was used as a model small molecule and myoglobin (Mb; Mw = 17,600 Da) as a model protein with unique biological function (i.e. oxygen storage and release) (**Scheme 1A**). Two conventional vesicle formation techniques were preferentially employed for the incorporation of these water-soluble agents within PEO-b-PBD polymersomes: “thin-film rehydration” and “direct-hydration” (**Scheme 1B and 1C**); a third-technique (“electroporation”) produced uniformly poor results and was excluded from further comparisons (**Scheme S1, Supplemental Information**).

We compared the encapsulation of mBlue and Mb in PEO-*b*-PBD polymersomes, which were generated by each of these different methods, in order to examine differences in the fidelity of small molecule versus protein encapsulation. We also monitored the concentrations and redox states of iron in the heme group of Mb (by inductively coupled plasma optical emission spectroscopy (ICP-OES) and by ultraviolet-visible (UV-Vis) absorption spectroscopy, respectively) in order to verify the encapsulation of intact protein by independent methods; the functional capabilities of the encapsulated Mb to bind and release oxygen were measured using a Hemox analyzer. Finally, we describe the development of a highly reproducible and scalable “progressive saturation” technique for the efficient generation of polymersome-encapsulated Mb (PEM) suspensions. The generalizability of this method is further established by utilizing a variety of different proteins, ranging from 17-450 kDa in size.

MATERIALS AND METHODS

Materials

PEO(3900)-*b*-PBD(6500) (OB18) and PEO(1300)-*b*-PBD(2500) (OB29) were purchased from Polymer Source (Dorval, Quebec, Canada; **Table I**). Horse skeletal muscle Mb, bovine blood hemoglobin (Hb), bovine serum albumin (BSA), catalase (C), fibrinogen (F), sodium hydrosulfite, poly(ethylene glycol) dimethyl ether (PEG500; Mw = ~500 Da), protease from *Streptomyces griseus* (“pronase”), and dichloromethane (DCM) were purchased from Sigma-Aldrich (St. Louis, USA). Horse spleen apoferritin (aFr) was purchased from Alfa Aesar (Ward Hill, USA). Immunoglobulin G (IgG) was purchased from LEE Biosolutions (St. Louis, USA). Dialysis tubing and vials were purchased from Spectrum Laboratories (Rancho Dominguez, USA). Sodium chloride, potassium chloride, sodium phosphate dibasic, potassium phosphate monobasic, mBlue, and Triton X-100 were purchased from Fisher Scientific (Suwanee, USA). All chemicals were of reagent grade unless otherwise stated.

A 1510 Branson[®] ultrasonic bath sonicator with a maximal energy of 70 watts was utilized for all sonication procedures (Branson Ultrasonics, Danbury, USA). Electroporation was performed using a Gene Pulser[®] (Bio-Rad, Hercules, CA, USA). Particle sizes were measured using Delsa[™] Nano, a dynamic light scattering (DLS) instrument (Beckman Coulter, Indianapolis, USA). Mb and mBlue concentrations were determined by absorption spectroscopy using a Genesys[™] 10S UV-Vis spectrophotometer (Thermo Scientific, Suwanee, USA). The concentrations of all proteins in polymersome suspensions were further measured using a BCA Protein Assay Kit, utilizing UV-Vis spectrophotometry and by following the manufacturer's protocols (Pierce Biotechnology, Inc; Rockford, USA). Iron concentrations in polymersome-encapsulated Mb suspensions were determined using a Vista-PRO[™] CCD ICP-OES (Varian, USA). Oxygen equilibrium binding was studied using a Hemox[™]-Analyzer (TCS Scientific Corp, New Hope, USA).

“Thin-film rehydration” method

10 mg of OB18 polymer was dissolved in 200 μ L of DCM. The polymer solution was deposited on Teflon wafers (15 \times 15 mm) that were subsequently dried for 30 min at room temperature (RT). The films were further kept under vacuum overnight at RT to ensure DCM evaporation. For mBlue encapsulation, polymer films were then hydrated with mBlue solution (21 mg/mL) in phosphate buffered saline (PBS; 10 mM, pH 7.4) for 24-48 h at 23, 40 or 60 $^{\circ}$ C. The samples were sonicated for 30 min at RT, followed by (x10) freeze-thaw cycles using liquid nitrogen. The samples were dialyzed for 40 h at RT (MW cutoff = 100 kDa). For Mb encapsulation, polymer films were hydrated with Mb solution (150 mg/mL) in PBS (10 mM, pH 7.4) for 60 h at 23, 40, and 60 $^{\circ}$ C. The samples were then sonicated for 30 min at RT followed by dialysis for 40 h at 4 $^{\circ}$ C (MW cutoff = 1000 kDa). The dialysis buffer was exchanged with 1 Liter of fresh buffer every after 7-8 h; free mBlue or Mb solutions were also subject to dialysis and served as controls to verify their full removal in the dialysate under these conditions. Quantification of mBlue and Mb encapsulation in polymersomes was determined *vide infra* (see “Quantification of mBlue/Mb”).

“Direct hydration” method

The “direct hydration” method was adopted from O’Neil *et al.* with minor modification (22). In brief, the protocol consists of 4 steps: 1) addition of 10 mg of OB18 to 10 mg of PEG500 followed by heating a 1.5 mL centrifuge tube for 20 min at 95 °C; 2) mixing by vortexing and cooling to RT followed by the addition of 10 µL of mBlue solution (21 mg/mL) or Mb solution (150 mg/mL) in PBS (10 mM, pH 7.4); 3) diluting with 20, 70, and 900 µL of PBS with mixing after each step (via vortexing); and 4) dialysis for 30 h at RT or at 4 °C (MW cutoff = 1000 kDa) to remove unencapsulated mBlue or Mb, respectively. The dialysis buffer was exchanged with 1 Liter of fresh buffer every after 7-8 h. Quantification of mBlue and Mb encapsulation in polymersomes suspensions was determined *vide infra*. As control experiments, free mBlue and Mb suspensions were also subject to similar dialysis conditions; UV-Vis spectroscopy of both the dialysate and retained solution suspensions at the completion of the dialysis experiment demonstrated no retention of unencapsulated agent by the dialysis membrane.

Quantification of mBlue/Mb

The amounts of mBlue or Mb that were encapsulated in dialyzed polymersome suspensions were determined by measuring the solution absorbance at 665 nm (mBlue) or at 410 nm (Mb), using a UV-Vis spectrophotometer. In brief, calibration curves for mBlue and Mb were developed using serial dilutions of known concentrations. Before UV-vis measurements, polymersome-encapsulated mBlue or Mb samples were diluted with a 10-fold excess volume of THF or 1% triton, respectively. This step was necessary to destroy the vesicles and release the encapsulated mBlue or Mb agent, thereby reducing light scattering during UV-Vis measurements that would otherwise ensue due to intact particles. The UV-vis spectra were also baseline corrected to obtain the peak absorbance of mBlue or Mb in solution. The samples concentration of mBlue or Mb from the dialysis controls were similarly subtracted from the concentration of the mBlue or Mb in the destroyed polymersome samples to obtain the absolute concentration each agent in the original polymersome suspension. To measure the iron content in PEM suspensions (as a

corroboration of Mb concentration in that sample), 5-10% (v/v) of Triton X-100 was added; the mixture was digested by heating in aqua regia for 3 h at 98 °C and the sample was subsequently diluted with deionized water. ICP-OES measurements were then conducted and the iron content in the experimental samples was determined by comparison to the standard calibration curve. For each experimental sample, the concentrations of Mb (as calculated by UV-Vis) were compared to those obtained via ICP-OES and via the BCA assay, which was also conducted by utilizing UV-Vis spectrophotometry and by following the manufacturer's protocols (Pierce Biotechnology, Inc; Rockford, USA). Loading of aqueous encapsulants in polymersomes was quantified and expressed as the final weight percentages of encapsulant-to-polymer that comprised the vesicles in suspension (e.g. w/w% Mb/polymer).

Quantification of metMb

The amount of metmyoglobin (metMb, i.e. oxidized Mb with an Fe(III)-heme group) in polymersome suspensions was quantified using a modified UV-Vis absorption protocol that was previously established for the measurement of cyanomethemoglobin (38-43). In brief, the absorbance of Mb was measured at 630 nm (L_1) against a blank reference (deionized water). One drop of KCN solution (1 part 10% KCN and 1 part 50 mM phosphate, pH 7.6) was added and mixed with the treated Mb samples. This reaction step was necessary to convert metMb to cyanometmyoglobin (cyanoMb), which does not absorb at 630 nm. After 2 min, the absorbance was measured at 630 nm (L_2); the absorbance of deionized water served as the blank reference. The concentration of metMb was determined using **Equation 1**:

$$[\text{metMb}] \text{ (mM)} = \frac{L_1 - L_2}{1 \times E} \times D_1 \quad (1)$$

where $E = 3.7 \text{ (cm} \times \text{mM)}^{-1}$ and is the extinction coefficient of metMb at 630 nm; D_1 is the dilution factor in this experiment (cuvette length = 1 cm as denoted in the equation).

To determine the concentration of Mb, one drop of 20% $\text{K}_3(\text{Fe}(\text{CN})_6)$ was added and mixed with the treated Mb sample. The solution was allowed to react for 2 min and an additional drop of 10% KCN was

added and mixed. The absorbance of the sample was then measured at 540 nm (L3). The concentration of total Mb was determined using **Equation 2**:

$$[\text{total Mb}](\text{mM}) = \frac{L_3}{1 \times E} \times D_2 \quad (2)$$

where $E = 11.3 (\text{cm} \times \text{mM})^{-1}$ and is the extinction coefficient for cyanometMb at 540 nm; D_2 is the dilution factor (cuvette length = 1 cm as denoted in the equation).

The percentage of metMb in the original solution was determined using **Equation 3**:

$$[\text{metMb}](\%) = \frac{[\text{metMb}]}{[\text{metMb}] + [\text{total Mb}]} \times 100 \quad (3)$$

Structural characterization of polymersomes

Polymersome suspensions were diluted in PBS solution and their hydrodynamic diameters were measured by DLS using a standard 1.5 mL semi-micro Plastibrand polystyrene cuvette (VWR, Atlanta, USA). The morphologies of blank polymersomes and PEM were visualized by cryo-TEM (JEOL 2100F, USA). In brief, polymersome samples were suspended in a microperforated grid, rapidly vitrified using liquid ethane (-183 °C), and loaded onto a cryogenic sample holder for cryo-TEM imaging at 200 kV.

Optimization of various steps in the direct hydration protocol

Modification of each of the 4 steps in the original direct hydration protocol were conducted following protocols summarized below as well as in Table II.

Modification of Step 1: Effects of blending of polymer and PEG using an organic solvent vs. heat

Following the “direct hydration” protocol (*vide supra*), 10 mg of OB18 and 10 mg of PEG500 were blended either by: A) mixing by dissolution in DCM (50 μL), or B) heating to 95 °C for 1 h followed by cooling to room temperature. Samples that were blended by dissolution in DCM were further subject to drying under vacuum at RT (overnight) to ensure removal of all organic solvent prior to addition of

aqueous protein solutions. In the case of heating, the samples were subject to 95 °C for 1 h (instead of only 20 min as utilized in the original “direct hydration” protocol) to further ensure maximal polymer mixing prior to cooling. In all cases, Mb concentrations in the final suspensions were determined by UV-Vis, ICP-OES and compared.

Modification of Step 2: Effects of Mb oxidation state (i.e. utilizing metMb vs. oxymyoglobin)

The “direct hydration protocol” (*vide supra*) was modified to expose the initial mixture of polymer and PEG500 to 1 h (instead of 20 min) of heating at 95 °C. The effect of the oxidation state of iron in the heme group of Mb on the efficiency of polymersome encapsulation was studied by using metMb vs. oxymyoglobin (oxyMb; i.e. Fe(II)Mb) for polymersome encapsulation. MetMb solution was first prepared by dissolving lyophilized Mb in PBS; the same solution was treated with 1 wt% Na₂S₂O₄ to obtain the reduced Mb form (oxyMb). Mb concentrations, weight percentages of Mb-to-polymer, and the final efficiencies of polymersome encapsulation of Mb, using either metMb vs. oxyMb in the protocol, were measured by UV-Vis, ICP-OES and compared.

Modification of Step 2: Effects of sonication and temperature on Mb oxidation

Lyophilized Mb was dissolved in PBS (at 150 mg/mL) and reduced with Na₂S₂O₄ to obtain an oxyMb solution that was further subjected to various environmental conditions, including heating at 40 °C or at 50 °C (with or without sonication) for 2-5 h. The extent of Mb oxidation was determined by measuring the percentages of metMb in the total polymersome-encapsulated (oxy)Mb suspensions, using the cyanomethemoglobin method (*vide supra*).

Modification of Step 3: Effects of sonication after each dilution step

Following the “direct hydration protocol” (*vide supra*), upon addition of PEG500 to OB18 polymer, the sample was heated, mixed by vortexing, cooled to RT, and 10 µL of Mb solution (150 mg/mL) in PBS (10 mM, pH 7.4) were added. The sample was then further combined with 10, 20, 50, and, finally, 100 µL

of PBS, followed by sonication for either: A) 0 min, B) 15 min, C) 30 min, D) 45 min, or E) 60 min after each dilution step. All samples were then dialyzed for 40 h at 4 °C (MW cutoff = 1000 kDa). The final Mb concentrations, weight percentages of Mb-to-polymer, and the efficiencies of polymersome encapsulation of Mb were measured by UV-Vis, ICP-OES and compared.

Modification of Step 4: Effects of proteolysis to distinguish non-specifically bound vs. encapsulated protein in the final polymersome suspensions

PEM suspensions were generated by the “direct hydration” method (*vide supra*), using OB18 polymer and oxyMb solution (150 mg/mL) with subsequent dialysis at 4 °C for at least 40 h (MW cutoff = 1000 kDa). The samples were then treated with 0.4 wt% pronase solution for 1, 2, 3, 6, 12, 18, or 24 h at RT and again dialyzed overnight at 4 °C. Free oxyMb solutions were also subjected to similar pronase treatment in order to serve as a positive control (for the rate and extent of proteolysis). Mb encapsulation in polymersomes (before and after proteolysis) was measured by UV-Vis, ICP-OES and compared.

Improvement of Mb encapsulation efficiency (%Mb EE)

Following the “direct hydration” protocol, 10 mg polymer and 10 mg of PEG500 were initially heated in 1.5 mL microcentrifuge tubes for 1 h at 95 °C and subsequently cooled to RT. Different volumes of: A) pure PBS, or B) 5 different Mb solutions (varying in concentration from 8 to 150 mg/mL oxyMb in PBS) were then combined with this mixture in 5 addition steps: i.e. 10, 10, 20, 50, and, finally, 100 µL of solution were added to the polymer mixture. When introducing additional amounts of Mb to the polymer mixture, the final (v/v) ratio of Mb to PBS (i.e. “Mb:PBS”) used in each of the 5 addition steps were 10:180 (i.e. 8 mg/mL Mb), 20:170 (i.e. 15 mg/mL Mb), 40:150 (i.e. 31 mg/mL Mb), 90:100 (i.e. 71 mg/mL Mb), and 190:0 (i.e. 150 mg/mL Mb). The samples were then proteolyzed using 0.4 wt% pronase and again dialyzed overnight at 4 °C (MW cutoff = 1000 kDa). Mb encapsulation was measured using UV-Vis spectrophotometry. Mb encapsulation efficiencies were calculated using **Equation 4**:

$$\text{Mb Encapsulation Efficiency} = \left[1 - \frac{v_1 c_1 - v_2 c_2}{v_1 c_1} \right] \times 100 \quad (4)$$

Where v_1 = Initial volume of the unencapsulated Mb (mL); c_1 = Initial concentration of unencapsulated Mb (mg/mL); v_2 = volume of polymersome-encapsulated Mb obtained after dialysis and proteolysis (mL); c_2 = concentration of encapsulated Mb obtained after dialysis and proteolysis (mg/mL).

Development of the “progressive saturation” technique

In combining the various experimental conditions that yielded the most favorable results in each of the comparative experiments (*vide supra*), a “progressive saturation” technique was developed that optimized the generation of polymersome-encapsulated protein suspensions. In brief, 10 mg of polymer and 10 mg of PEG500 were heated to 95 °C for 1 h. The mixture was then centrifuged and cooled to RT. Lyophilized Mb protein was dissolved in PBS, reduced with Na₂S₂O₄ (1 wt%) to generate an oxyMb solution (at 150 mg/mL in PBS), and used for stepwise addition to the polymer mixture: 10, 10, 20, 50 and 100 μL of oxyMb solution were introduced, followed by thorough mixing and sonication for 30 min at RT after each step. The samples were then sonicated for an additional 30 min at RT followed by dialysis at 4 °C for at least 40 h (MW cutoff = 1000 kDa). The dialysate was exchanged with 1 Liter of fresh buffer every after 8-10 h and UV-Vis measurements of the dialysis control solution (e.g. free Mb solution subject to identical dialysis conditions) verified that all free protein was successfully removed. Surface-associated protein was further eliminated by proteolysis via treatment with 0.4 wt % pronase solution followed by a second overnight dialysis at 4 °C (MW cutoff = 1000 kDa). Mb encapsulation (before and after proteolysis) in polymersome suspensions was determined via UV-vis, ICP-OES and compared.

Stability of polymersome-encapsulated protein suspensions

PEM suspensions were prepared from OB18 polymer and oxyMb using the optimized “progressive saturation” technique (*vide supra*). The samples were stored at 4, 23, and 37 °C for 2 weeks. At predetermined time points, Mb concentrations were quantified using UV-Vis spectroscopy (*vide supra*);

weights of polymer were measured after lyophilization; and, particle sizes were determined by DLS for PEM suspensions under each condition.

Equilibrium binding of oxygen in polymersome-encapsulated Mb suspensions

The equilibrium binding and dissociation curves for oxygen in suspensions of free Mb and PEM were obtained at 37 °C using a HemoxTM- Analyzer. Samples were allowed to saturate to a pO₂ of 147 mmHg (using compressed air) and then deoxygenated (using a compressed nitrogen stream). The absorbance of oxygenated and deoxygenated free Mb and PEM suspensions were recorded as a function of pO₂ via dual wavelength spectroscopy. Oxygen equilibrium curves were fit to a four-parameter (A₀, A_∞, P₅₀, n) Hill model (**Equation 5**). In this model, A₀ and A_∞ represent the absorbance at 0 mmHg and at 147 mmHg, respectively. The pO₂ represents the partial pressure of oxygen; P₅₀ represents the partial pressure where the sample is 50% saturated with O₂; and, *n* represents the cooperativity coefficient for the sample.

$$Y = \frac{Abs - A_0}{A_\infty - A_0} = \frac{pO_2^n}{pO_2^n + P_{50}^n} \quad (5)$$

Polymersome encapsulation of different proteins using the “progressive saturation” technique

The generalizability of the “progressive saturation” technique was tested using proteins of varying sizes: i.e. Mb (17 kDa), hemoglobin (Hb; 64 kDa), bovine serum albumin (BSA; 66 kDa), immunoglobulin G (IgG: 150 kDa), catalase (250 kDa), fibrinogen (340 kDa), and apoferritin (450 kDa). Each protein was dissolved in PBS (10 mM, pH 7.4) at its maximal solubility, corresponding to final suspension concentrations of 150, 150, 40, 20, 50, 50, and 25 mg/mL, respectively. The “progressive saturation” protocol was followed to encapsulate these proteins in OB29 polymersomes. Free proteins were separated by dialysis for at least 40 h at 4 °C (MW cutoff = 1000 kDa). Surface-associated protein was removed by proteolysis via treatment with 0.4 wt% pronase solution followed by overnight dialysis at 4 °C (MW cutoff = 1000 kDa). Protein encapsulation (before and after proteolysis) in polymersome suspensions was quantified via the BCA assay, utilizing UV-Vis spectrophotometry and by following the manufacturer’s

protocols (Pierce Biotechnology, Inc; Rockford, USA). The final concentrations of protein were divided by those of polymer and expressed as the final weight ratios of protein-to-polymer that comprised the polymersomes in suspension (i.e. w/w% Mb/polymer).

Statistical analysis

Data are presented as the mean \pm the standard deviation of the mean (SD). A minimum of three experimental replicates was used for each condition. One-way analysis of variance (ANOVA) was conducted using GraphPad software (San Diego, USA). A *p value* of < 0.05 was considered statistically significant.

RESULTS

Encapsulation of mBlue and Mb in polymersome suspensions using conventional methods

In order to establish a baseline for comparisons of small molecule and protein encapsulation in polymersome suspensions, the final concentrations, weight percentages (i.e. weight of encapsulated agent compared to the weight of the polymer that comprises the nanoparticle), and the efficiencies of encapsulation for the small molecule mBlue were first determined with OB18-based polymersomes. Using the “thin-film rehydration” method, no significant differences in mBlue encapsulation were observed in polymersomes formed at 40 vs. 60 °C (e.g. final polymersome composition consisted of 4.1 vs. 5.0 w/w% mBlue/polymer, respectively; **Fig. 1A**). When thin-film rehydration was attempted at 23 °C (room temperature), the encapsulation of mBlue was found to be negligible (results not shown), possibly due to the observation that the polymer films did not swell after 48-72 h of hydration; PEO-b-PBD-based polymersomes require the input of energy for vesicle formation, which is usually supported by using elevated temperatures (e.g. > 45 °C) (8, 44). To improve the efficiency of encapsulation at lower temperatures, which would be necessary when employing labile proteins, encapsulation of mBlue was

also studied by the “direct hydration” method; the final weight ratio of mBlue-to-polymer in these polymersomes was, however, again found to be very low (~ 1.2 w/w% mBlue/polymer; **Fig 1B**).

We next sought to compare the results obtained for mBlue encapsulation to those that could be achieved for Mb using the same polymersome formation techniques. Polymersome-encapsulated Mb (PEM) suspensions formed at 23 °C by “thin-film rehydration” were initially found to be comprised of ~ 2.7 w/w% Mb/polymer. After the addition of a proteolysis step to remove any surface-associated Mb (i.e. free protein that was nonspecifically bound), the final composition of the polymersomes was found to be only ~ 0.5 w/w% Mb/polymer, indicating that only very small amounts of protein were being encapsulated within polymersomes via this method (**Fig 1C**). In order to improve the concentrations and the final weight percentages of Mb in PEM suspensions, polymersome generation at higher temperatures was again attempted (e.g. utilizing “thin-film rehydration” at 40 and 60 °C); these experiments, however, only resulted in protein denaturation and aggregation (results not shown). In contrast, PEM suspensions prepared by “direct hydration” at 23 °C displayed good colloidal properties and the characteristic absorption spectra of intact Mb; the final loading ratio of Mb-to-polymer in these PEM suspensions, however, was again very low (i.e. ~ 0.3 and 0.1 w/w% Mb/polymer before and after proteolysis, respectively; **Fig 1D**). Thus, while “thin-film rehydration” and “direct hydration” may be used to reproducibly encapsulate mBlue at reasonable efficiencies, their utility for Mb encapsulation was found to be uniformly poor.

Optimization of the “direct hydration” protocol to improve protein encapsulation

In order to develop an improved method for the generation of polymersome-encapsulated protein suspensions, the contributions of each of the various steps in the “direct hydration” protocol on the final concentrations, weight percentages, and encapsulation efficiencies of Mb in OB18-based polymersomes were systematically evaluated (**Table II**). We sought to combine features of both the “direct hydration”

and “thin-film rehydration” techniques in order to improve polymersome encapsulation of functional protein. As such, changes were implemented iteratively in order to optimize experimental conditions.

Step 1: Effects of blending technique (i.e. dissolving polymer in organic solvent vs. adding heat)

Improving polymer dissolution is critical to efficient encapsulation of protein. As such, we sought to first compare the effects of utilizing an organic solvent versus adding heat in order to blend OB18 with PEG500 homopolymer in the first step of the “direct hydration” protocol; we subsequently compared both strategies with respect to the final yield of polymersomes and, ultimately, to the concentrations and efficiencies of protein encapsulation that could be obtained by each method. If the two polymers were first mixed by dissolution in DCM (followed by polymersome formation after organic solvent evaporation), the final weight ratio of Mb-to-polymer in PEM suspensions was ~2 w/w% Mb/polymer. In comparison, initial heating of dry OB18 and PEG500 to 95 °C for 1 h improved mixing and promoted more efficient polymersome generation, yielding a significantly higher final weight ratio of Mb-to-polymer in the PEM suspensions (~5 w/w% Mb/polymer; **Fig 2A**), corresponding to a greater amount of encapsulated protein.

Step 2: Effects of Mb oxidation state (utilizing metMb vs. oxyMb for polymersome encapsulation)

In addition to improving polymeric blending, Mb encapsulation in polymersomes was found to be further augmented when the starting Mb stock solution was first reduced with sodium dithionite to convert all metMb to the oxyMb form. OxyMb contains a central heme group with iron in the ferrous state (i.e. Fe(II)), which improves the solubility of the protein when compared to its metMb form that contains Fe(III). This oxyMb solution was further desalted via dialysis prior to utilization in the initial addition step of the “direct hydration” protocol, which was found to be necessary to increase the loading of Mb in PEM suspensions (i.e. the final weight ratio of Mb-to-polymer). When oxyMb was used in the initial protocol step, PEM suspensions comprised of ~6 w/w% Mb/polymer were formed, which was a

statistically significant improvement over the 4 w/w% Mb/polymer that could be obtained if metMb was utilized instead (**Fig 2B**).

Step 2: Effects of sonication and temperature on Mb oxidation

More than 40% of the oxyMb that was used in the initial step for polymersome encapsulation was found to be reoxidized to metMb within 2 h at 50 °C; by contrast, only ~15% metMb was generated from the initial oxyMb solution if lower temperatures were employed for polymersome formation (e.g. heating for 2 h at 40 °C). The rate of Mb oxidation at 50 °C was also significantly higher than that observed at 40 °C, regardless of the addition of sonication or the power that was utilized (**Fig 2C**). As such, it was determined that sonication had no effect on Mb oxidation; and, it was, thus, preferentially employed to both promote polymer mixing and to provide interfacial energy to augment polymersome formation.

Step 3: Effects of sonication on improving the encapsulation of Mb in polymersomes

Despite improvements in polymer blending that could be afforded by heating the polymer and PEG500 solution in the first step of the “direct hydration protocol”, as well as increases in the yield of myoglobin in the final vesicles suspensions afforded by utilizing oxyMb for polymersome encapsulation, the final weight ratios of Mb-to-polymer that were reproducibly obtained in PEM suspensions after pronase treatment were found to again be very low (~0.2 w/w% Mb/polymer), indicating that most of the Mb was only surface associated as opposed to encapsulated within the vesicle suspensions. The encapsulation efficiency, however, could be increased by >30x if the samples were sonicated for 30 min at RT after each dilution step (i.e. sonicating after introducing additional volumes of aqueous solution to dilute the concentrations of polymer in suspension). As seen from the results depicted in **Fig 2D**, the relative amounts of Mb in PEM suspensions could be increased to ~5.5 w/w% Mb/polymer, supporting the addition of this sonication step to the original “direct hydration” protocol.

Step 4: Effects of proteolysis on differentiating surface-associated vs. encapsulated Mb

Upon formation, PEM suspensions were treated with 0.4% pronase solution for up to 24 h at RT in order to examine the duration of time required for the complete digestion of any surface-associated (i.e. non-specifically bound) Mb. It was observed that all free protein in solution, which was utilized as a positive control for proteolysis, and all surface-associated Mb in PEM suspensions were digested in 2 h; neither increasing pronase exposure time nor concentration further augmented Mb loss. These results would indicate that only encapsulated Mb was thereafter retained (**Fig 2E**).

Optimization of Mb encapsulation efficiency

Five sets of experiments were done with various Mb-to-PBS volume ratios (Mb:PBS) in order to establish the optimal Mb concentration to use in each subsequent addition step in our modification of the original “direct hydration” protocol. Notably, when Mb:PBS increased, the final w/w% Mb/polymer in the PEM suspensions also increased; but, the Mb encapsulation efficiency (%Mb EE) decreased as a result. In other words, the final Mb-to-polymer mass ratio was maximized when all addition steps were conducted using a maximally concentrated Mb solution (i.e. Mb:PBS = 190:0, corresponding to 150 mg/mL oxyMb) (**Fig 2F**). %Mb EE (as defined in *Materials and Methods*), however, was largest when Mb:PBS was minimal (i.e. 10:180). As the amount of protein in the final polymersome suspension is ultimately the metric that must be optimized for therapeutic administration (in order to minimize the amount of associated carrier polymer that is introduced to a subject), it was determined that a pure Mb solution (i.e. 150 mg oxyMb/mL) would be used for each addition step in the ultimate encapsulation protocol, maximizing the final loading of myoglobin (i.e. w/w% Mb/polymer) in PEM suspensions.

Maximizing protein encapsulation through “progressive saturation”

By incorporating each of the aforementioned steps, an optimized “progressive saturation” technique was established that improved upon the results of the original “direct hydration” method (**Scheme 1D**). Following the “progressive saturation” protocol, the final content of Mb in OB-18-based PEM

suspensions was found to be 6.1 and 3.2 w/w% Mb/polymer before and after proteolysis, respectively. Quantification of the iron content (and hence the numbers of intact heme groups in each of the polymersome suspensions) by ICP-OES corroborated UV-Vis measurements of protein concentration; the final loading percentages of Mb in the polymersomes were found to be 7.9 and 5.1 w/w% Mb/polymer before and after proteolysis, respectively (**Table III**). Notably, the percentage of metMb (with respect to the total Mb content in these suspensions) was determined by UV-Vis spectrophotometry and found to be ~8 and 6% in non-proteolyzed (PEM-SE) and proteolyzed (PEM-E) samples, respectively (**Table III**).

Polymersome encapsulation using block copolymers and proteins of varying molecular weight

The “progressive saturation” technique was also utilized to encapsulate Mb in a second PEO-b-PBD-based polymersome system comprised of the OB29 diblock copolymer. When compared to OB18, OB29 has a smaller molecular weight ($M_w = 3800$ g/mol vs. 10400 g/mol) and generates polymersomes with a shorter PEG brush border (1.3 vs. 3.9 kDa), thinner bilayer membrane (9.6 nm vs. 14.8 nm), and smaller average hydrodynamic diameter (130 vs. 200 nm) (**Table I**) (36, 37). When compared to the initial experiments with OB18, similar encapsulation results were obtained with Mb in OB29-based polymersomes (**Table IV**). The generalizability of the “progressive saturation” technique was further established when analogous results were obtained with several larger proteins, including hemoglobin (Hb; 64 kDa), bovine serum albumin (BSA; 66 kDa), immunoglobulin G (IgG; 150 kDa), catalase (250 kDa), fibrinogen (340 kDa), and apoferritin (450 kDa) (**Table IV**).

Characterization of the final PEM suspensions

The size distributions of the final OB18- and OB29-based PEM suspensions were measured by both DLS (**Fig. 3A**) and by cryo-TEM (**Fig. 3B** and **Fig. 3C**). These results confirmed a mean particle diameter of approximately 200 and 130 nm for OB18 vs. OB29-based polymersomes, respectively. The stability of the OB18-based PEM suspensions were further examined over 2 weeks and at three different temperatures (i.e. 4, 23, and 37 °C); the particles demonstrated no aggregation as evidence by their

consistent particle numbers and stable size distributions in suspension (**Fig. 3D**). The extent of Mb encapsulation in these PEM suspensions was also found to be unaffected after two weeks of exposure to these three different temperatures (**Fig. 3E**). Finally, the functional status of encapsulated Mb in the PEM suspensions (i.e. retention of Mb's ability to bind and release oxygen) was verified by dual wavelength spectroscopy (**Fig. 3F**). The oxygen equilibrium curve of PEM after proteolysis (PEM-E) was similar to that of free Mb but demonstrated a slightly hyperbolic shape at lower oxygen tensions, indicating the encapsulation of functional protein within polymersomes but with subtle changes in oxygen binding and release; the P_{50} (i.e. the partial pressure that corresponds to 50% saturation with O_2) and n (the cooperativity coefficient) for PEM-E were also larger than that of free Mb (**Table V**). The oxygen equilibrium curve and the values of P_{50} and n for PEM before proteolysis (PEM-SE) were intermediate between those of free Mb and PEM-E, supporting the incorporation of both surface-associated and encapsulated Mb in these vesicle suspensions.

DISCUSSION

While lipid-based vesicles (i.e. liposomes) have been extensively utilized in biomedical research, examples of their widespread clinical adoption are few. These translational shortcomings may be attributable to material limitations inherent to phospholipid-based drug delivery vehicles, including compromised suspension stability, premature drug release, and limited product shelf-life. In contrast, polymersomes are formed from higher molecular weight amphiphilic block copolymers that impart a broad and tunable range of carrier properties. To date, however, a generalized method for the efficient and reproducible encapsulation of functional proteins in neutrally charged, fully PEGylated, and nanoscale polymer vesicles has not been established. While adoption of various liposome encapsulation techniques has enabled facile incorporation of small molecules within polymersomes, these methods cannot directly be applied for scalable encapsulation of functional proteins. Often, there is a trade-off in the maximal

concentration of aqueous protein that may be encapsulated (i.e. mg protein/mL solution), the final loading ratio of protein-to-polymer that comprises the polymersome structure (i.e. w/w% protein/polymer), and/or the protein encapsulation efficiency (i.e. the percentage of the initial protein suspension that is retained). Further, the value of each of these parameters often varies upon experimental run and is highly dependent on the nature of the protein, the exact block copolymer formulation, and the encapsulation method that is utilized (**Table VI**). Here, we have performed a comparative and quantitative study of several conventional liposome encapsulation techniques in order to generate an optimized and reproducible method to improve the encapsulation of functional proteins in nanoscale polymer vesicles.

PEO-b-PBD copolymers were used to form polymersomes that possessed fully PEGylated surfaces, that were uncharged, and that were nondegradable; thus, they were an ideal system for ensuring vesicle integrity and minimizing unwanted protein interactions or modifications. Two different molecular weight PEO-b-PBD polymers (i.e. OB18 and OB29) were further employed to determine the generalizability of the results as they pertain to polymersomes of different PEG lengths, membrane core thicknesses, and minimal sizes. Methylene blue (mBlue; $M_w = 319.85$ g/mol) was first used as a model small molecule in order to establish the baseline encapsulation parameters (i.e. aqueous suspension concentrations, final weight percentages, and encapsulation efficiencies) that could be obtained by each method. mBlue is highly stable in aqueous suspension and has a strong near-infrared absorbance that enables ready spectrophotometric detection; it was thus employed to confirm the fidelity of the results in reference to previously published studies on the encapsulation of other organic small molecules. Mb was used as a model protein for polymersome encapsulation as it possesses a size ($M_w = 17,600$ Da) and thermal stability (denaturation above 60 °C) that is comparable to other small proteins with therapeutic potential; Mb also has a strong UV-Vis absorbance that enables ready identification of its functional status (as determined by the redox state of its Fe-containing heme group, which mediates its oxygen binding capabilities). Mb has additionally been used in other investigations and its employment in this study enabled ready comparisons of our results (**Table IV**) to baseline values from the literature (**Table VI**).

mBlue was easily encapsulated in PEO-b-PBD polymersomes formed by “thin-film rehydration” (at elevated temperatures), yielding final weight ratios of mBlue-to-polymer of 4.1 and 5.0 w/w% when formed at 40 and 60 °C, respectively; for Mb encapsulation, however, similar conditions only led to protein degradation. Notably, when vesicles are formed by “thin-film rehydration”, hydrophilic lamellae (a.k.a. sponge-like structures) are first formed (22, 45); this is followed by transformation into hexagonally packed vesicles and finally into fully dispersed polymersomes (22, 45, 46). When “thin-film rehydration” is attempted in solutions of soluble small molecules (or proteins), these water-soluble species adsorb to the surfaces of the budding lamellae, which subsequently adopt a spontaneous (or preferred) curvature. During formation, these membranes preferentially bend away from the aqueous compartment that contains the higher concentration of adsorbing species, thereby excluding the water-soluble agents from vesicle encapsulation (45). Ultimately, the input of energy can overcome this spontaneous surface tension in order to promote vesicle encapsulation. The amount of energy that is required scales with the size of the adsorbed molecule and the membrane thickness of the vesicle (36, 37, 47, 48). Thus, while it is easy to disrupt liposomes and enable effective small molecule and protein loading by “thin-film rehydration”, the input of thermal (and/or sonic) energy only enables effective encapsulation of small molecules into polymersome suspensions.

The “direct hydration” method, by contrast, was developed as a hybrid of two other preparation methods, namely “solvent dispersion” and “homopolymer addition” (22). In the “direct hydration” protocol, the hydrophilic polymer PEG500 is used to disrupt the interactions of hydrophobic chains in the forming polymer lamellae (22). With subsequent additions of aqueous solution, self-assembly of vesicles from budding lamellae that have dispersed protein is promoted and results in improvements in aqueous encapsulation; encapsulation efficiencies as high as 37% have been observed (22). Supporting these initial studies, we generated PEM suspensions by “direct hydration” at 23 °C and measured protein encapsulation efficiencies of >10%; the encapsulated Mb exhibited good suspension properties and the characteristic absorption spectra of the intact protein. The final loading of Mb in these PEM suspensions,

however, was found to be very low (~0.3 w/w% Mb/polymer), which was consistent with results calculated from the original paper (**Table VI**) (22). Upon addition of a protease solution to induce proteolysis of all surface-associated (i.e. non-specifically bound) protein, the final Mb composition in PEM suspensions was found to be even lower (< 0.1 w/w% Mb/polymer). As the loading of therapeutic proteins within the aqueous cavities of polymersomes is ultimately the metric that must be maximized for translational applications (in order to minimize the amount of associated carrier that is introduced to a subject), it was evident that the development of a more robust encapsulation protocol was required.

By optimizing and combining various steps from multiple vesicle formation methods, a new “progressive saturation” technique was developed that enabled efficient encapsulation of functional proteins within polymersomes. Factors influencing the final concentration of protein, the relative loading level that could be achieved within the carrier (i.e. w/w% protein/polymer), and the efficiency of protein encapsulation were systematically evaluated using Mb as a model protein and OB18 as the polymersome system. We found that factors such as the molecular weight of the polymer, the oxidation state and concentration of the protein, the pH and nature of the buffered solution, the exact polymer hydration conditions (i.e. time, temperature, and blending technique), and the number and duration of sonication steps all had effects on the concentration and the fidelity of the final polymersome-encapsulated protein product. For the optimized “progressive saturation” protocol, 5 subsequent additions of Mb solution were also made to the PEG/polymer mixture, in lieu of serially dilutions with PBS (as described in the original “direct hydration” paper (22)). The significant improvements in the final Mb-to-polymer weight ratios that were obtained (i.e. 4-6 w/w% Mb:polymer with “progressive saturation” vs. 0.1-0.3 w/w% by “direct hydration”) suggest that the polymersome formation process is not complete during the initial addition step; further encapsulation is accomplished with each subsequent introduction of protein solution.

Notably, we found that there was a direct tradeoff between Mb encapsulation efficiency and the final weight ratios of Mb-to-polymer that could be achieved based on the concentrations of free Mb that were used for each addition step. While other studies have reported achieving both high encapsulation

efficiencies and large loading capacities (i.e. high final weight ratios of Mb-to-polymer), it is important to underscore that these investigations did not attempt to differentiate the extent of surface-associated vs. encapsulated protein in the final nanoparticle suspensions. Aqueous encapsulation of protein is, of course, preferred in order to assure that the final product meets the original objectives for utilizing a polymersome delivery vehicle: i.e. to avoid biochemical instability, to increase circulatory half-life, to minimize adverse side effects, and to achieve controlled release of the associated protein. Larger scale batches of PEM, which were generated from 500 mg of polymer using the “progressive saturation” method, yielded similar encapsulation results to those obtained with these small-scale batch preparations, supporting the scalability of the technique. Moreover, by employing different proteins that varied over a large range of molecular weights and sizes, including those associated with therapeutically relevant antibodies and enzymes, the robustness and the versatility of the “progressive saturation” method were established.

Using nondegradable PEO-b-PBD-based polymersomes as a model system, we demonstrated the colloidal stability of PEM suspensions at 4, 23, and 37 °C for over two weeks; the oxygen binding and release capabilities of Mb in these polymersome suspensions were further examined and verified to be preserved. Notably, the oxygen equilibrium curve of PEM demonstrated a slightly hyperbolic shape at low oxygen tensions as compared to that of free Mb; parametric fitting also demonstrated a larger P_{50} and n (cooperativity coefficient) for PEM as compared to the unencapsulated protein. Together, these results indicate a slight reduction in oxygen binding affinity but also a mild decrease in the off-loading of oxygen at low oxygen tensions for PEM as compared to free Mb. Mechanistically this may be mediated by decreased oxygen transport across the thick polymersome membranes. Notably, the P_{50} and n of polymersome suspensions that contained both surface-associated and encapsulated Mb (PEM-SE) were of intermediate values to those of free Mb and pure polymersome-encapsulated Mb (PEM-E) obtained by proteolysis of all surface-associated protein. The explanation that membrane transport affects are most likely determining changes in oxygen binding is further supported by the results of kinetic studies that measured the oxygen binding and release of hemoglobin (Hb) in polymersomes (49); they observed a

decrease in the rate of O₂ offloading in polymersome-encapsulated Hb suspensions, in RBCs, as well as with PEG-conjugated liposome-encapsulated Hbs (LEHs) as compared to acellular (free) Hb.

CONCLUSIONS

Conventional vesicle formation methods may be used to encapsulate small molecules within polymersomes (as demonstrated in this study using mblue); but, they do not yield consistent, reproducible, or scalable encapsulation of functional proteins in neutrally charged, fully PEGylated and nanoscale polymersomes. By optimizing and combining various steps from several of these established protocols, we introduce a novel “progressive saturation” technique that achieves increased encapsulation of functional protein in polymersomes of varying size, PEG length, and membrane thickness. We demonstrate a tradeoff between the degree of polymersome loading (i.e. weight percentage of protein-to-polymer) and the encapsulation efficiency of protein (with respect to the initial quantity) that may be achieved. Moreover, we introduce a novel proteolysis step that enables accurate quantification of the amounts of both encapsulated protein (i.e. the desired outcome) as well as that of surface-associated (i.e. non-specifically bound) product that is obtained in polymersome suspensions. Previous studies have reported that large amounts of protein may be loaded within polymersomes and at high efficiencies, using conventional liposome encapsulation techniques. They had not, however, sought to differentiate between encapsulated and surface-associated protein, which may account for discrepancies in their observations as compared to the results presented here. Importantly, we have also demonstrated that “progressive saturation” enables stable encapsulation of functional protein; it may be further employed to maximize the aqueous encapsulation of many different proteins that vary over a large range of molecular weights and sizes, including those associated with therapeutically relevant antibodies and enzymes. In summary, “progressive saturation” is a robust, scalable, and generalizable technique for generating polymersome-encapsulated proteins in quantities and at efficiencies that may enable further translational development.

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CONFLICTS OF INTEREST

Dr. Ghoroghchian is a co-founder and the previous Chief Scientific Officer of Vindico Pharmaceuticals, Inc, where he directed research efforts. He currently serves as a consultant and as the head of the company's scientific advisory board. The authors declare that they have no conflicts of interest.

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TABLE LEGEND

Table I. Properties of poly(ethylene oxide)-block-poly(butadiene) (PEO-b-PBD) diblock copolymers and their subsequent polymersome formulations used for small molecule and protein encapsulation

Table II: Summary of the factors studied for optimization of the direct hydration protocol and to develop the progressive saturation method for encapsulation of proteins in polymersomes.

Table III. Properties of OB18-based PEM suspensions prepared using the “progressive saturation” technique.

Table IV. Encapsulation of proteins in OB29-based polymersomes using the “progressive saturation” technique.

Table V. P_{50} values for free myoglobin (Mb) as well as polymersome-encapsulated Mb prior to proteolysis (PEM-SE) and after pronase treatment (PEM-E), as obtained from O_2 equilibrium curves. PEM-SE and PEM-E samples were prepared by the “progressive saturation” technique.

Table VI. Summary of reported studies that have examined encapsulation of proteins in polymersomes.

FIGURE LEGEND

Scheme 1. Schematic depicting: A) amphiphilic diblock copolymers, water-insoluble agents, and water-soluble species that may be encapsulated within polymersomes. The B) “thin-film rehydration”, C) “direct hydration”, and D) modified “progressive saturation” methods for encapsulation of water-soluble species within the aqueous cavities of polymersomes.

Figure 1. Incorporation of mBlue and Mb in polymersome suspensions, using conventional liposome formation techniques. (A) Encapsulation of mBlue in OB18 polymersomes by “thin-film rehydration” at 40 °C and 60 °C; (B) Encapsulation of mBlue in OB18 polymersomes by “direct hydration” at 23 °C. (C) Encapsulation of Mb in OB18 polymersomes by “thin-film rehydration” at 23 °C. (D) Encapsulation of Mb in OB18 polymersomes by “direct hydration” at 23 °C. Polymersomes were subjected to proteolysis to remove any surface-associated (free) Mb in suspension. Mb concentrations in each sample were analyzed using UV-Vis spectrophotometry. Data represent mean values and error bars denote standard deviations of the mean for an n = 5 experimental replicates. (***) $p < 0.001$.

Figure 2. Optimization of various steps in the “direct hydration” protocol in order to improve the encapsulation of Mb in OB18 polymersomes (as determined by the final weight percentage of Mb-to-polymer (i.e. w/w% Mb/polymer) that could be obtained in PEM suspensions). A) The effects of utilizing an organic solvent vs. the addition of heat during the first step of the “direct hydration” protocol. B) The effects of the oxidation state of Mb used for polymersome encapsulation and for each subsequent addition step. C) The rate of Mb oxidation (as expressed by the percentage of metmyoglobin (i.e. %metMb) formed over time) as a function of Mb exposure to different solution conditions. D) The effects of sonication time after each dilution step. E) Optimization of the proteolysis time to remove all surface-

associated Mb from polymersome suspensions; note: “% Mb loss” denotes surface-associated Mb fraction and not total Mb; samples were proteolyzed for 18 h to remove surface-associated Mb followed by UV-Vis quantification. F) Final weight percentage of Mb-to-polymer (i.e. “w/w% Mb/polymer”) vs. Mb encapsulation efficiency (i.e. “%Mb EE”) obtained by tuning the relative volume of Mb suspension (150 mg/mL Mb) to PBS used in each addition step. Data represent mean values and error bars denote standard deviations of the mean for an $n = 5$ experimental replicates. (* $p < 0.05$ and ** $p < 0.01$).

Figure 3. Characterizations of the final polymersome-encapsulated protein suspensions formed by using the “progressive saturation” technique. A) The average hydrodynamic diameter of particles in OB18- and OB29-based PEM suspensions as assessed by DLS. Cryo-TEM images of vesicles in B) OB18- and C) OB29-based PEM suspensions; scale bar = 50 and 20 nm in Fig. 3B and 3C, respectively. D) Average hydrodynamic diameters of particles (as determined by DLS) in OB18-based PEM suspensions as a function of temperature (4, 23, and 37 °C) and time. E) Final weight percentage of Mb-to-polymer (i.e. “w/w% Mb/polymer”) of particles in OB18-based PEM suspensions as a function of temperature (4, 23, and 37 °C) and time. F) Oxygen equilibrium curves for free oxyMb and oxygenated OB18-based PEM suspensions. Data represent mean values and error bars denote standard deviations of the mean for an $n = 5$ experimental replicates.

Table I. Properties of poly(ethylene oxide)-block-poly(butadiene) (PEO-b-PBD) diblock copolymers and their polymersome formulations used for small molecule and protein encapsulation (36, 37).

Copolymer Name	Mn* x 10 ³ PEO-b-PBD	Mw (kg/mol)	P.D.*	Membrane Thickness (nm)	Vesicle Diameter (nm)
OB18	3.9-b-6.5	10400	1.10	14.8	205
OB29	1.3-b-2.5	3800	1.04	9.6	132

*Mn denotes number-average molecular weight and P.D. is the polydispersity index of the polymer.

Table II: Summary of the factors studied for optimization of the direct hydration protocol and to develop the progressive saturation method for encapsulation of proteins in polymersomes.

PROTOCOL	CONDITIONS				RESULTS	
Original Direct Hydration Protocol	STEP 1: Heating of polymer and PEG500 to 95 °C followed by cooling to RT STEP 2: Addition of protein solution followed by vortexing mix sample STEP 3: Encapsulation driven by serial dilution of protein solution with PBS STEP 4: Removal of unencapsulated protein by dialysis				Figure 1D	
Optimization of Direct Hydration Protocol	Steps of Original Direct Hydration Protocol that were Modified					
	1	2	3	4		
Blending Method	Heating of polymer and PEG for 1h at 95 °C followed by cooling to room temperature	X			Figure 2A	
	Dissolving polymer and PEG in DCM followed by solvent evaporation					
Oxidation State of Mb Solution that is Added	Utilizing oxyMb for encapsulation Utilizing metMb for encapsulation		X		Figure 2B	
Effect of Sonication and Temperature on Mb Oxidation	Heating the oxyMb solution at 40 °C Heating the oxyMb solution at 50 °C Sonication the oxyMb solution at 40 °C		X		Figure 2C	
Mixing Technique for Addition of Protein Solution	Sonication for 0 min after each dilution step Sonication for 15 min after each dilution step Sonication for 30 min after each dilution step Sonication for 45 min after each dilution step Sonication for 60 min after each dilution step			X	Figure 2D	
Removal of Unencapsulated Protein	Dialysis only; no proteolysis treatment Dialysis and treatment with pronase for 2h			X	Figure 2E	
Mb Encapsulation Efficiency vs. Ratio of Mb:PBS that is Added	Mb:PBS = 10:180 Mb:PBS = 20:170 Mb:PBS = 40:150 Mb:PBS = 90:100 Mb:PBS = 190:0		X	X	X	Figure 2F

Table III. Properties of OB18-based PEM suspensions prepared using the “progressive saturation” technique.

PEM Properties	Before Proteolysis	After Proteolysis
Final w/w% Mb/Polymer (by UV-vis)	6.08 ± 0.75	3.17 ± 0.54
Final w/w% Mb/Polymer (by ICP-OES)	7.96 ± 0.91	5.10 ± 0.62
% metMb	7.98 ± 1.90	5.55 ± 0.97
Size (d.nm)	205	205

Table IV. Encapsulation of proteins in OB29-based polymersomes using the “progressive saturation” technique.

Protein	MW* (kDa)	Before or After Pronase	Final Protein (mg/mL) by BCA	w/w% Protein /Polymer by BCA	EE* (%)	Final Protein (mg/mL) by UV-Vis	w/w% Protein/ Polymer by UV-Vis	Final Protein (mg/mL) by ICP-OES	w/w% Protein/ Polymer by ICP-OES
Myoglobin	17	Before	2.68 ± 0.75	13.39 ± 3.75	4.70	2.21 ± 1.19	11.07 ± 5.97	3.82 ± 1.07	19.10 ± 5.36
		After	1.98 ± 0.43	9.19 ± 2.15	3.48	1.13 ± 0.44	5.64 ± 2.21	2.06 ± 0.80	10.31 ± 3.98
Hemoglobin	64	Before	5.26 ± 1.00	20.79 ± 6.87	7.30	3.14 ± 1.02	11.95 ± 2.21	8.00 ± 2.35	31.77 ± 12.50
		After	3.38 ± 0.46	13.30 ± 3.70	4.67	2.16 ± 0.02	8.40 ± 1.13	4.90 ± 1.20	19.37 ± 6.59
Bovine Serum Albumin	66	Before	3.53 ± 2.38	15.60 ± 11.02	13.69				
		After	3.49 ± 1.13	15.42 ± 5.15	13.53				
IgG	150	Before	5.54 ± 2.03	35.99 ± 13.21	94.72				
		After	3.20 ± 1.51	21.23 ± 9.39	55.88				
Catalase	250	Before	5.42 ± 3.10	30.79 ± 16.09	32.41				
		After	2.28 ± 0.70	13.09 ± 3.27	13.77				
Fibrinogen	340	Before	5.37 ± 0.76	26.10 ± 3.49	27.48				
		After	3.05 ± 0.39	14.82 ± 1.72	15.60				
Apo ferritin	450	Before	1.99 ± 1.02	8.51 ± 4.28	17.92				
		After	1.73 ± 1.23	7.57 ± 5.73	15.94				

*MW – molecular weight; EE – encapsulation efficiency

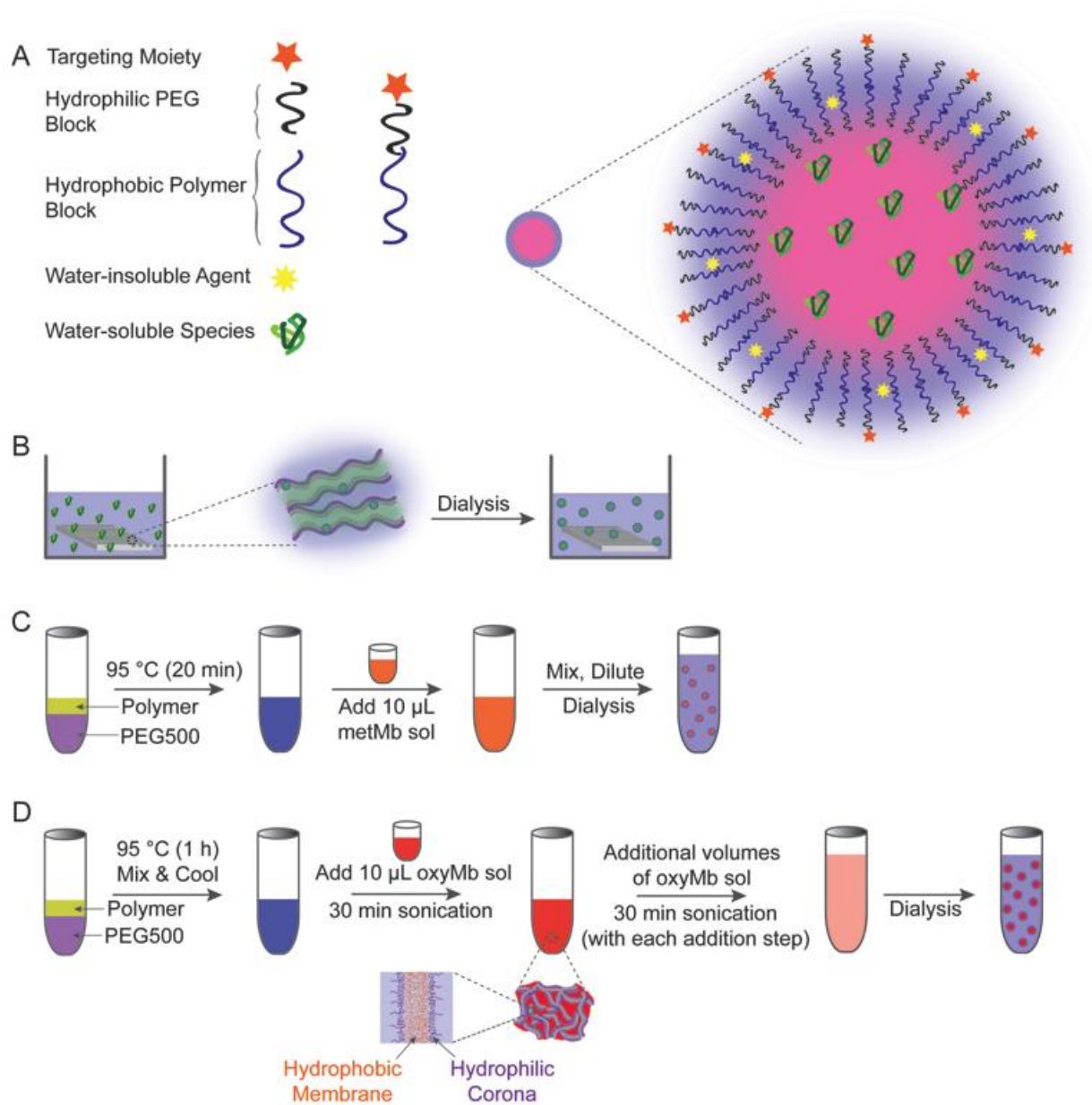
Table V. Values of P_{50} (binding affinity) and n (cooperativity coefficient) for free myoglobin (Mb) as well as polymersome-encapsulated Mb prior to proteolysis (PEM-SE) and after pronase treatment (PEM-E), as obtained from O_2 equilibrium curves. PEM-SE and PEM-E samples were prepared by the “progressive saturation” technique.

Sample	P_{50} (mmHg)	n
Mb	2.00 ± 0.01	0.79486 ± 0.00452
PEM-SE	7.95 ± 0.30	0.7951 ± 0.0281
PEM-E	17.14 ± 0.27	1.5683 ± 0.039

Table VI. Summary of reported studies that have examined encapsulation of proteins in polymersomes.

Block copolymer	Protein	PSM* size D [nm]	PSM* size after loading D [nm]	Protein Encapsulation [mg/mL]	Protein loading, w/w% protein/polymer	EE*, %	Ref.
PEG-SS-b-PDEAEMA	CC	55	59	0.010	5.0	100	(50)
	BSA	55	49	0.008	4.0	80.6	
PEG-B-PAAC-b-PNIPAM	BSA	162	150	2.025	40.5	81	(51)
	CC	162	155	2.000	40.0	80	
	Lys	162	149	1.615	32.3	64.6	
PEG-b-PCL-PDEAEMA	Ova	162	147	2.210	44.2	88.3	(16)
	BSA	154	146	0.098	19.6	78.5	
	CC	154	164	0.112	22.3	89.1	
	Lys	154	150	0.105	21.0	84.3	
PEO-b-PCL	Ova	154	113	0.106	21.2	84.7	(41)
	IgG	154	158	0.112	22.4	89.6	
PEO-b-PCL	Hb (Human)	100	110-140			2.0-12.0	(41)
PEO-b-PLA	Hb (Bovine)	100	110-140			4.0-20.0	(41)
PEO-b-PBD	Hb	100	100	0.081	8.1	2.7	(43)
	Hb	100	100	0.365	36.5	12	
PEO-b-PPS	Ova	200		0.006	0.06	9 ± 8	(22)
	Ova	200		0.024	0.24	37 ± 10	
	BSA	200		0.128	1.28	19 ± 5	
	γ-Globulin	200		0.050	0.50	15 ± 5	
PMPC-b-PDPA	BSA	110	110-120	0.309	3.09	12.37	(34)
	IgG	110	110-120	0.0005	0.0045	9.00	
	Mb	110	110-120	0.199	1.99	7.97	
	Lz	110	110-120	0.159	1.59	6.36	
PS-PIAT	GOX			0.625	125.0	25.00	(52)
	CalB			0.340	68.0	17.00	
	HRP			0.625	125.0	25.00	
PS-b-PAA	CC					66.00	(53)
	Poly-L-Lysine						
	GFP					35.00	

*PSM – polymersome; EE – encapsulation efficiency; the full names of all polymer formulations can be found in the *Abbreviations* section of the manuscript.



Scheme 1. Schematic depicting: A) amphiphilic diblock copolymers, water-insoluble agents, and water-soluble species that may be encapsulated within polymersomes. The B) “thin-film rehydration”, C) “direct hydration”, and D) modified “progressive saturation” methods for encapsulation of water-soluble species within the aqueous cavities of polymersomes.

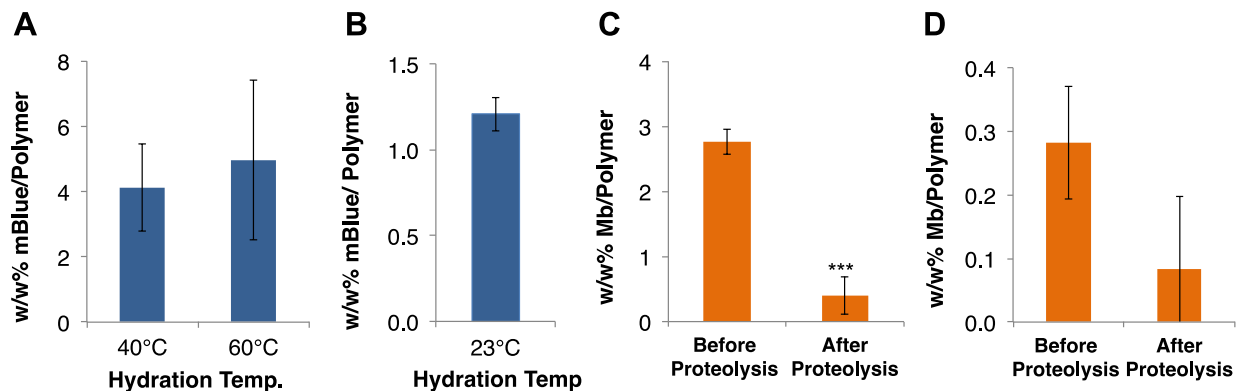


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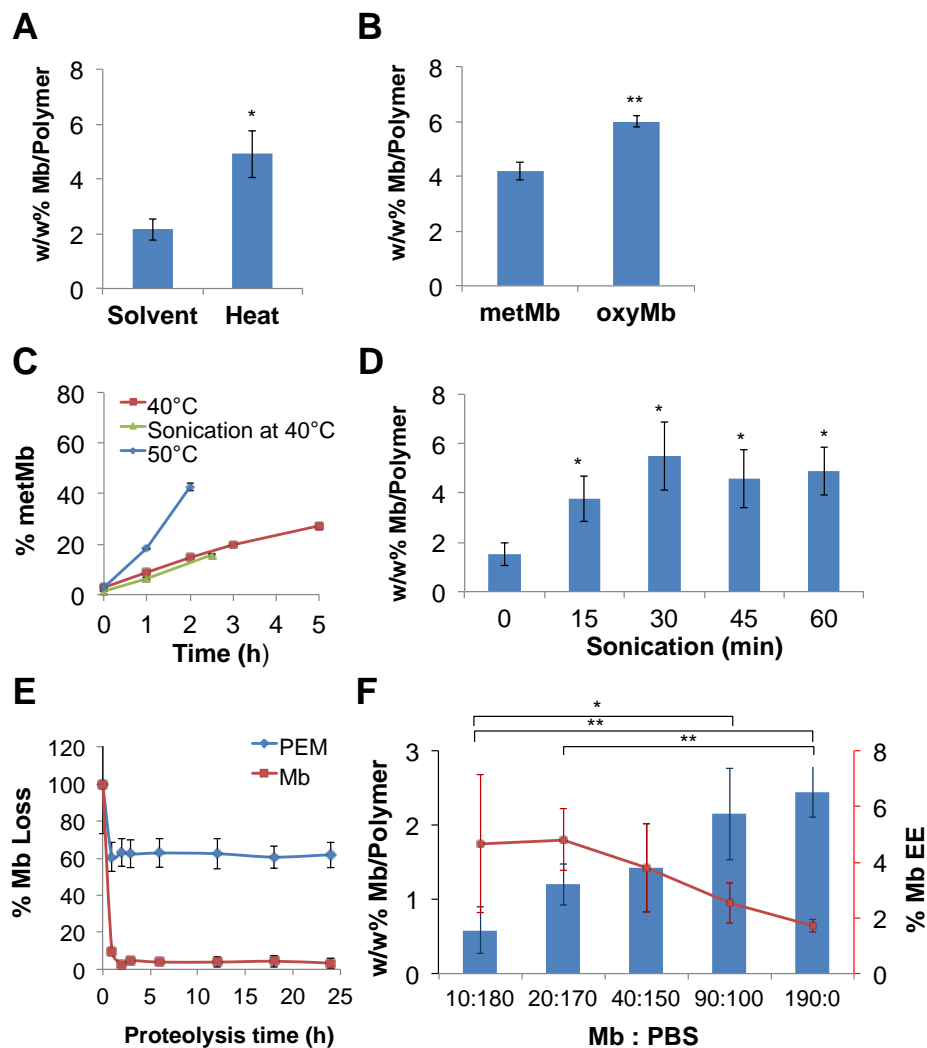


Figure 2. Optimization of various steps in the “direct hydration” protocol in order to improve the encapsulation of Mb in OB18 polymersomes (as determined by the final weight percentage of Mb-to-polymer (i.e. w/w% Mb/polymer) that could be obtained in PEM suspensions). A) The effects of utilizing an organic solvent vs. the addition of heat during the first step of the “direct hydration” protocol. B) The effects of the oxidation state of Mb used for polymersome encapsulation and for each subsequent addition step. C) The rate of Mb oxidation (as expressed by the percentage of metmyoglobin (i.e. %metMb) formed over time) as a function of Mb exposure to different solution conditions. D) The effects of sonication time after each dilution step. E) Optimization of the proteolysis time to remove all surface-associated Mb from polymersome suspensions; note: “% Mb loss” denotes surface-associated Mb fraction and not total Mb; samples were proteolyzed for 18 h to remove surface-associated Mb followed by UV-Vis quantification. F) Final weight percentage of Mb-to-polymer (i.e. “w/w% Mb/polymer”) vs. Mb encapsulation efficiency (i.e. “%Mb EE”) obtained by tuning the relative volume of Mb suspension (150 mg/mL Mb) to PBS used in each addition step. Data represent mean values and error bars denote standard deviations of the mean for an n = 5 experimental replicates. (* $p < 0.05$ and ** $p < 0.01$).

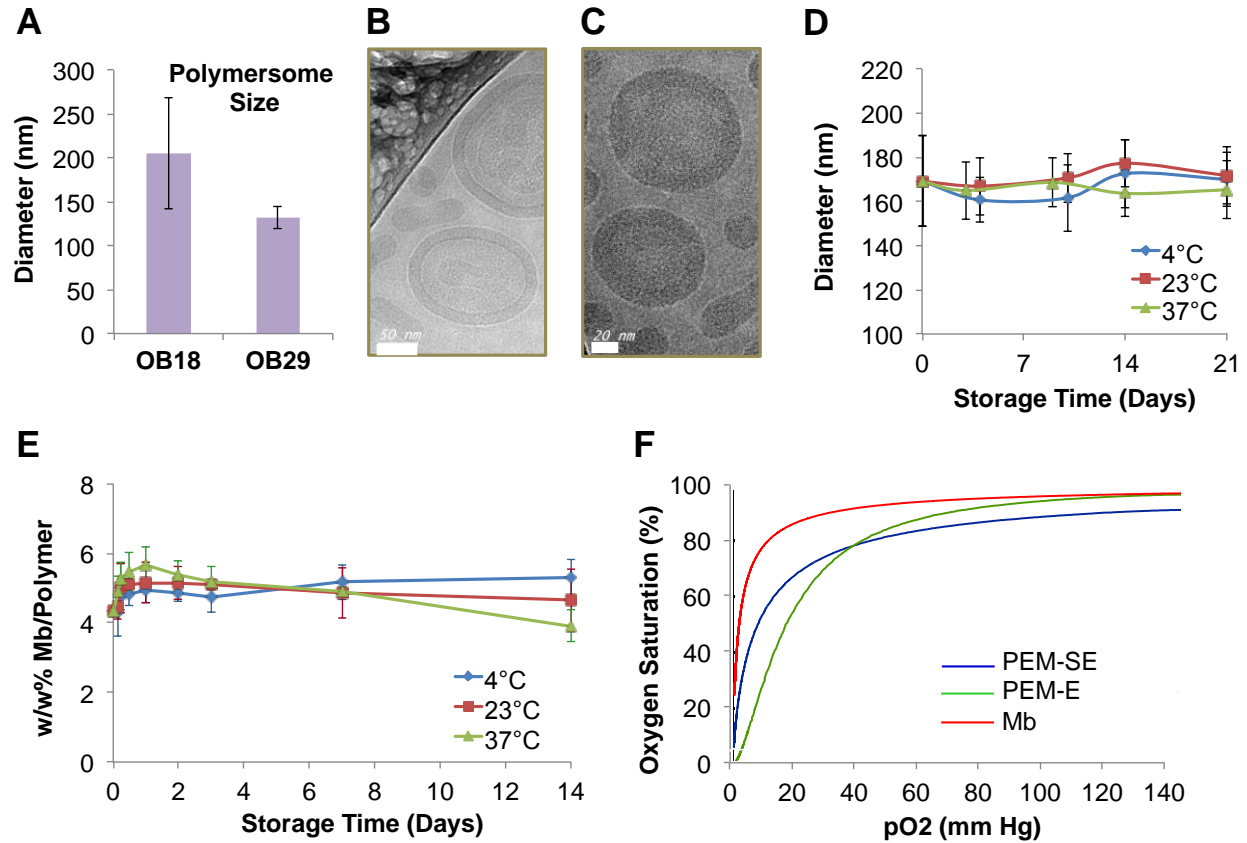
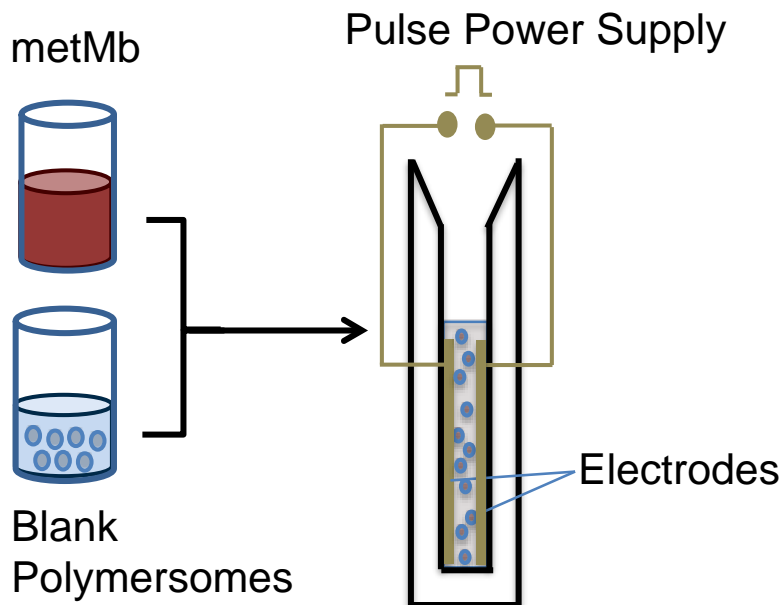


Figure 3. Characterizations of the final polymersome-encapsulated protein suspensions formed by using the “progressive saturation” technique. A) The average hydrodynamic diameter of particles in OB18- and OB29-based PEM suspensions as assessed by DLS. Cryo-TEM images of vesicles in B) OB18- and C) OB29-based PEM suspensions; scale bar = 50 and 20 nm in Fig. 3B and 3C, respectively. D) Average hydrodynamic diameters of particles (as determined by DLS) in OB18-based PEM suspensions as a function of temperature (4, 23, and 37 °C) and time. E) Final weight percentage of Mb-to-polymer (i.e. “w/w% Mb/polymer”) of particles in OB18-based PEM suspensions as a function of temperature (4, 23, and 37 °C) and time. F) Oxygen equilibrium curves for free oxyMb and oxygenated OB18-based PEM suspensions. Data represent mean values and error bars denote standard deviations of the mean for an n = 5 experimental replicates.

SUPPORTING INFORMATION:



Scheme S1. Aqueous encapsulation of water-soluble species in polymersomes by “electroporation”.

Encapsulation of Mb using “electroporation”

Preformed OB18-based polymersomes were prepared using “thin-film rehydration” and mixed with Mb solutions (150 mg/mL Mb). The suspensions were subsequently loaded into 400 μ L cuvettes and subjected to electromagnetic fields, using an electroporation apparatus (applied voltage: 1.0, 2.0, or 2.5 kV; output was a decaying exponential waveform with RC time constant of 5 ms; the number of pulses equaled 20). The samples were dialyzed for 40 h at 4 °C to remove free protein (MW cutoff = 1000 kDa). Mb encapsulation in polymersomes was measured using UV-Vis spectrophotometry and was found to be negligible (results not shown).