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Estimates of New and Total Productivity in Central Long Island Sound from In Situ Measurements of Nitrate and Dissolved Oxygen

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1 **Authors:** James R. Collins^{1,2*}, Peter A. Raymond¹, W. Frank Bohlen³, and Mary M. Howard-
2 Strobel³

3
4 **Title:** Estimates of new and total productivity in central Long Island Sound from *in situ*
5 measurements of nitrate and dissolved oxygen

6
7 **Affiliations:** ¹ School of Forestry & Environmental Studies, Yale University, New Haven, CT
8 06511

9 ² MIT/WHOI Joint Program in Oceanography/Applied Ocean Science and Engineering,
10 Department of Marine Chemistry & Geochemistry, Woods Hole, MA 02543

11 ³ Department of Marine Sciences, University of Connecticut, Groton, CT 06340

12
13 ***Corresponding author:** phone: (508) 289-3808; fax: (508) 457-2164; e-mail: jrco@mit.edu

14
15 **Abstract:** Biogeochemical cycles in estuaries are regulated by a diverse set of physical and
16 biological variables that operate over a variety of time scales. Using *in situ* optical sensors, we
17 conducted a high-frequency time-series study of several biogeochemical parameters at a mooring
18 in central Long Island Sound from May to August 2010. During this period, we documented
19 well-defined diel cycles in nitrate concentration that were correlated to dissolved oxygen, wind
20 stress, tidal mixing, and irradiance. By filtering the data to separate the nitrate time series into
21 various signal components, we estimated the amount of variation that could be ascribed to each
22 process. Primary production and surface wind stress explained 59% and 19%, respectively, of the
23 variation in nitrate concentrations. Less frequent physical forcings, including large-magnitude

1 wind events and spring tides, served to decouple the relationship between oxygen, nitrate, and
2 sunlight on about one-quarter of study days. Daytime nitrate minima and dissolved oxygen
3 maxima occurred nearly simultaneously on the majority (> 80%) of days during the study period;
4 both were strongly correlated with the daily peak in irradiance. Nighttime nitrate maxima
5 reflected a pattern in which surface-layer stocks were depleted each afternoon and recharged the
6 following night. Changes in nitrate concentrations were used to generate daily estimates of new
7 primary production ($182 \pm 37 \text{ mg C m}^{-2} \text{ d}^{-1}$) and the *f*-ratio (0.25), i.e., the ratio of production
8 based on nitrate to total production. These estimates, the first of their kind in Long Island Sound,
9 were compared to values of community respiration, primary productivity, and net ecosystem
10 metabolism, which were derived from *in situ* measurements of oxygen concentration. Daily
11 averages of the three metabolic parameters were 1660 ± 431 , 2080 ± 419 , and $429 \pm 203 \text{ mg C}$
12 $\text{m}^{-2} \text{ d}^{-1}$, respectively. While the system remained weakly autotrophic over the duration of the
13 study period, we observed very large day-to-day differences in the *f*-ratio and in the various
14 metabolic parameters.

15

16

17 **Keywords:** Long Island Sound, nitrate, nitrogen, new production, *in situ* measurements,
18 eutrophication, dissolved oxygen, net ecosystem metabolism, wind forcing, wind stress, *f*-ratio

19

20

1 **1 Introduction**

2 Nearly 100,500 tons of nitrogen is exported annually to Long Island Sound (LIS), a large
3 estuary in the densely populated and highly developed northeastern United States (NYSDEC and
4 CTDEP 2000). Most of this nitrogen has an anthropogenic origin: By the most recent estimate,
5 roughly 66 percent of the LIS nitrogen load comes from human activities such as wastewater
6 transport and treatment, agriculture (fertilizers and animal wastes), and the burning of fossil fuels.
7 Nearly a quarter of the anthropogenic nitrogen load is injected by tributary rivers and streams or
8 deposited directly from the atmosphere onto the water surface. The majority of the anthropogenic
9 load (58.7%) comes from point sources, including wastewater treatment plants and combined
10 sewer overflows, that feed directly into the Sound (NYSDEC and CTDEP 2000).

11 The large inputs of nitrogen to LIS have led to eutrophication, particularly in the western
12 part of the basin. Eutrophication has profound and adverse effects on the health of estuaries
13 worldwide (Breitburg et al. 2009; Diaz and Rosenberg 2008; Ryther and Dunstan 1971; Zhang et
14 al. 2010). In LIS specifically, eutrophication is responsible for recurring patterns of seasonal
15 hypoxia (Anderson and Taylor 2001; O'Donnell et al. 2008; Welsh and Eller 1991), blooms of
16 harmful “brown tide” algae and neurotoxin-producing cyanobacteria (Bushaw-Newton and
17 Sellner 1999), and the alteration of basic ecosystem trophic structures (Capriulo et al. 2002).

18 Combined measurements of primary production, respiration, and net ecosystem
19 metabolism (NEM) have been elegantly applied to assess the health and trophic status of a wide
20 variety of coastal marine ecosystems and estuaries for at least the last half-century (Caffrey
21 2003; Howarth et al. 1992; Kemp and Boynton 1980; Odum 1956; Smith 2007; Swaney et al.
22 1999). There continues to be considerable attention to quantifying the relationship between
23 nitrogen loading and primary production in many estuaries (Caffrey 2004; D'Avanzo et al. 1996;

1 Nixon 1992). By comparing rates of primary production and nitrogen loading as we do here, one
2 may obtain an index of eutrophication that can be used to make comparisons across ecosystems
3 and through time (Smith 2007).

4 Most of Long Island Sound's nitrogen load is in the forms of nitrate (NO_3^-) and
5 ammonium (NH_4^+) (Anderson and Taylor 2001). In Long Island Sound as in other eutrophic
6 embayments, nitrate is of particular concern because it is the form most readily assimilated by
7 primary producers, entering the surface layer from either below the pycnocline or from
8 allochthonous inputs such as freshwater runoff and sewage. Once allochthonous nutrients are
9 delivered to the surface layer, they become the basis for what Dugdale and Goering (1967)
10 termed "new" production. This new production, much of it often based on anthropogenic sources,
11 is of particular concern in temperate coastal ecosystems, where it frequently accounts for a very
12 large share of total productivity (Carpenter and Dunham 1985) and net oxygen and metabolism
13 changes.

14 Various biological and physical processes act over multiple timescales to determine the
15 delivery of nitrate to the surface layer and its subsequent uptake by phytoplankton (Collos 1998;
16 Garside 1985; Johnson 2010; Probyn et al. 1996). The apportioning of variation among these
17 various processes, particularly those that define nitrate concentrations on hourly and sub-hourly
18 timescales, demands further research in the coastal ocean. Prior to the development of *in situ*
19 spectrophotometric methods that measure concentrations at very high sampling intervals
20 (Johnson 2010; Johnson et al. 2006), studies examining the impact of wind forcing and other
21 controls *in situ* were largely restricted to theory and to anecdotal evidence obtained through
22 fortuitous circumstances at sea (Andersen and Prieur 2000; Eppley and Renger 1988).

1 Bearing in mind the relative importance of anthropogenic nitrate inputs to the overall
2 state of eutrophication in Long Island Sound, we utilize high-frequency measurements to couple
3 nitrate-based estimates of new production with estimates of gross primary productivity and
4 ecosystem metabolism based on free-water dissolved oxygen concentrations.

5 **2 Methods**

6 **2.1 Study Site**

7 Automated, high-resolution *in situ* observations of nitrate, dissolved oxygen, water
8 temperature, conductivity, chlorophyll fluorescence, turbidity, and incident photosynthetically
9 active radiation (PAR) were made at 15-minute intervals from the Central Sound Buoy, a
10 mooring in central Long Island Sound, from May to August 2010. The buoy (position 41° 8.25'
11 N, 072° 39.30' W, approx. 14.3 km south of Guilford, Connecticut) is one of four permanent
12 moorings that, together with a number of fixed stations, comprise the University of
13 Connecticut's LISICOS monitoring network (formerly MYSound; Tedesco et al. 2003). As such,
14 the buoy has a suite of calibrated meteorological instruments that provided us with additional
15 high-frequency observations of wind velocity, wave height, air temperature, and barometric
16 pressure. The site lies roughly along the major east-west axis of the estuary and is coincident
17 with Connecticut Department of Energy and Environmental Protection (CTDEEP; formerly
18 CTDEP) station I2, one of 17 designated stations at which the agency has made monthly
19 observations of water quality since 1994 (Fig. 1). Depth at the Central Sound Buoy station is 27
20 m, which is slightly deeper than the LIS average of approx. 20 m (Bowman 1977).

21 The site is nearly equidistant from the Connecticut coastline (14 km) and the north shore
22 of Long Island (17 km). The site's distance from land, considerable water depth, and immediate
23 bathymetry distinguish it from both the coastal ocean at the eastern mouth of the Sound and from

1 the shallower, more heavily eutrophied waters of the Sound's western reaches. While the central
2 Sound does not experience the recurring low oxygen conditions typical at the westernmost sites,
3 it is still somewhat sheltered from the full energy of the Atlantic Ocean. Reduced tidal forcing
4 and the considerable lee provided by Long Island to the site's south mean it does not experience
5 the degree of water-mass advection that is observed at more easterly sites.

6 Additional data reported in this study were collected by CTDEEP at monitoring stations
7 elsewhere in the Sound (Fig. 1). Rainfall observations are reported for KHAVN (at Tweed-New
8 Haven Airport), the closest weather station maintained by the U.S. National Weather Service; the
9 airport is 24.6 km northwest of the study site. Collection of discrete samples and instrument
10 maintenance at the site were performed aboard R/V Catch the Joy, a 23-foot Parker owned and
11 operated for research on Long Island Sound by the Yale School of Forestry & Environmental
12 Studies. The vessel was moored at the Yale-Peabody Marine Station in Guilford, Connecticut.

13 **2.2 Automated Observations**

14 Data from *in situ* instrumentation were recorded at 15-minute intervals by a Campbell
15 Scientific data logger onboard the mooring; this data were transmitted to shore via cellular
16 telemetry, where they were downloaded to a server at the University of Connecticut.

17 **2.2.1 Nitrate**

18 Nitrate was estimated at ~2 m depth using data from a Submersible Ultraviolet Nitrate
19 Analyzer (SUNA), a commercially-available, 256-channel *in situ* ultraviolet spectrophotometer
20 (Satlantic Inc., Halifax, Nova Scotia). The SUNA is an "off the shelf" version of the In Situ
21 Ultraviolet Spectrophotometer (ISUS) developed at the Monterey Bay Aquarium Research
22 Institute (Johnson and Coletti 2002; Johnson et al. 2006). The SUNA uses factory-installed
23 software to determine nitrate concentrations in real time from an absorbance spectrum between

1 217.5 to 240 nm; these onboard estimates were useful for remote monitoring of the instrument's
2 fouling condition during peak periods of productivity, but proved not to be as accurate at
3 estimating nitrate as the post-hoc calculations we performed using the raw spectral data. The
4 onboard algorithm often overestimated nitrate concentrations by 2-3 μM in the moderately
5 brackish (26-28 psu) waters at the study site.

6 Nitrate concentrations reported in this study were instead calculated using the method
7 described in Sakamoto et al. (2009), in which simultaneously collected temperature and
8 conductivity data are applied to correct for absorbance of the bromide ion. Bromide absorbs light
9 at many of the same wavelengths as nitrate. CTD data are used to predict bromide concentration
10 using the known bromide-to-chlorinity ratio; the instantaneous UV spectrum due to bromide is
11 then computed from temperature-corrected molar absorptivity (Sakamoto et al. 2009). This
12 spectrum is subtracted from the observed spectrum, which, when combined with a baseline
13 absorbance, yields the concentration of nitrate. Calibrations were performed in the laboratory at a
14 variety of temperatures and salinities that bracketed the range of temperatures and salinities
15 observed during the study period.

16 **2.2.2 SUNA Calibration and Validation**

17 Final, temperature-corrected, salinity-subtracted (TCSS) nitrate concentrations were
18 validated against discrete samples collected in triplicate at the study site. Filtration to 0.7 μm
19 with a combusted Whatman GF/F filter was performed in the field using acid-washed tubing and
20 filter holder directly into acid-washed high-density polyethylene bottles; samples were
21 immediately frozen. Determination of nitrate was performed by automated colorimetric titration
22 at two independent laboratories. Samples were analyzed by segmented flow analysis (Astoria-
23 Pacific Inc., Clackamas, Oregon; method detection limit 0.007 μM) at the Yale Analytical

1 Laboratory and by flow injection analysis (SmartChem discrete analyzer, Unity Scientific, LLC,
2 Brookfield, Connecticut; method detection limit 0.005 μM) at the University of New Hampshire
3 Water Quality Analysis Lab. Orthophosphate was also determined by segmented flow analysis at
4 the Yale facility.

5 **2.2.3 Biofouling**

6 Biofouling of the instrument was monitored by both visual inspection during regular
7 cruises to the site and by examining drift in the baseline of the UV absorption spectrum in the
8 real-time data. A perforated copper guard was installed over the optical chamber window to
9 inhibit marine growth and preventative manual cleaning of the optic was performed during each
10 site visit with a soft cloth and 10% HCl solution. However, approximately one month of data
11 from the SUNA was discarded due to evidence of fouling, which occurred during a period from
12 30 June to 25 July when visits could not be made to the site (Fig. 2).

13 **2.2.4 Other Automated Measurements**

14 Dissolved oxygen (DO), water temperature, and conductivity were measured using a YSI
15 6600 series submersible sonde (YSI Inc., Yellow Springs, Ohio). The DO, temperature, and
16 conductivity sensors onboard the mooring were calibrated according to manufacturer
17 specifications and LISICOS (formerly MYSound) operating procedures (Tedesco et al. 2003)
18 prior to the study period on 28 April 2010, once during the study period, and then upon
19 completion of the study period in September 2010. Chlorophyll fluorescence was measured at 1
20 m depth using a YSI 6025 series chlorophyll fluorometer. Turbidity in units of NTU was
21 measured at 2 m depth using a Campbell Scientific OBS-3+ turbidity sensor (Campbell
22 Scientific Inc., Logan, Utah). Calibration was performed according to manufacturer
23 specifications using a series of AmcoClear SDVB nephelometric turbidity standards. The

1 concentration of total suspended solids (TSS) was determined from turbidity using a regression
2 (not shown) based on data from a nearby site in Long Island Sound (OSI 2009).

3 Over the course of the study period, drift was noted in two of these instruments. First,
4 after heavy biofouling was discovered, we discarded segments of the conductivity data from 4 to
5 21 of July and 6 to 10 of August. Given the small and predictable linear increase in salinity that
6 historically occurs over the course of the summer months in the central Sound (Riley 1956), we
7 interpolated the missing values from historical averages and CTDEEP data (Matthew Lyman,
8 CTDEEP, unpublished data).

9 A smaller segment of the turbidity data (from 7 June to 14 June) was discarded as a result
10 of similar fouling. Since turbidity remained extremely low throughout the study period,
11 consistent with historical observations for this region of the Sound (Matthew Lyman, CTDEEP,
12 unpublished data), we averaged the values over two days on either side of the gap and then
13 interpolated using that figure.

14 Irradiance was measured at approx. 3 m above the sea surface using a LI-COR LI190SB
15 photosynthetic photon flux sensor (LI-COR Biosciences Inc., Lincoln, Nebraska). A TRIAXYS
16 Directional wave sensor (AXYS Technologies Inc., Sidney, British Columbia) provided real-
17 time data on wave height, direction, and period. Anemometer height was 3.5 m above the sea
18 surface.

19 **2.3 Collection of Discrete Samples**

20 **2.3.1 Chlorophyll *a***

21 Volumetric filtrations of seawater collected at the study site were performed for
22 chlorophyll *a* at the Yale-Peabody Marine Station following each site cruise. Triplicate 25 mm
23 Whatman GF/F filters (0.7 μm pore size) were frozen immediately pending analysis;

1 determination of chlorophyll was performed in the laboratory according to U.S. Environmental
2 Protection Agency (EPA) method 445.0 (Arar and Collins 1997) on a Turner Designs TD-700
3 fluorometer (Turner Designs, Sunnyvale, California). A calibration curve based on these values
4 (not shown) was then used to obtain a high-resolution, pheophytin-corrected chlorophyll *a* time
5 series (Fig. 2d; 4c) from the *in situ* chlorophyll fluorescence data.

6 **2.3.2 DOC and TDN**

7 Triplicate samples were collected for analysis of dissolved organic carbon (DOC) and
8 total dissolved nitrogen (TDN) in acid-washed polycarbonate bottles. Filtration was performed in
9 the field directly into sample bottles using a Cole-Parmer peristaltic field pump (Cole-Parmer
10 Inc., Vernon Hills, Illinois), combusted Whatman GF/F filter (0.7 μm pore size), and acid-
11 washed tubing. Samples were immediately frozen. Upon thawing, quantification of DOC and
12 TDN was performed immediately by high-temperature catalytic oxidation on a TOC-V_{CSH} total
13 organic carbon/total nitrogen analyzer (Shimadzu Inc., Columbia, Maryland; detection limit of 4
14 $\mu\text{g L}^{-1}$ for TOC and 0.1 mg L^{-1} for TDN).

15 **2.4 High-Pass and Low-Pass Signal Processing**

16 To isolate diel, photosynthetically-driven cycles of nitrate and DO concentrations from
17 longer-term trends in the same variables, we applied fast Fourier transforms using the “TSA”
18 (Chan 2008) and “mFilter” (Balcilar 2009) libraries for R (version 2.10.1), a statistical
19 computing package, according to the methods described in Cryer and Chan (2008). We applied
20 transforms only to the nitrate and dissolved oxygen data, yielding two complementary datasets
21 that we used for different purposes. First, we created a nitrate-dissolved oxygen dataset that
22 excluded all cycles with periods > 25 hours (Fig. 3c); from this first dataset, we also excluded
23 sources of very high frequency variation (i.e., signals with periods < 2.5 hours). This second step

1 eliminated any stochastic noise in the data that would have interfered with the amplitude analysis
2 described below. We then produced a second, complementary dataset using only a low-pass
3 filter; this dataset included only long-term trend components of the nitrate and dissolved oxygen
4 signals. We used the first dataset to estimate nitrate-based primary production (see section 2.5.2
5 below), and to examine correlations at sub-daily timescales between dissolved oxygen, nitrate
6 concentration, photosynthetically active radiation, and other variables. We used the second
7 dataset to examine long-term trends in dissolved oxygen and nitrate concentration. Time series
8 analysis techniques have been applied successfully in several other recent studies to separate
9 different signal components in high-resolution biogeochemical datasets, including those based on
10 *in situ* spectrophotometric data (Johnson 2010; Johnson et al. 2006).

11 **2.5 Data Analysis: Determination of Community Metabolic Parameters**

12 We report community metabolic parameters as depth-integrated values for the euphotic
13 zone at the study site. The vast majority of values reported in the literature from similar studies
14 are in areal units, i.e., integrated over the depth of the euphotic zone. While there are some
15 disadvantages to the use of depth-integrated estimates in assessments of coastal eutrophication
16 (see, e.g., Smith 2007), we concluded that our results would be most useful to others in this form.

17 **2.5.1 Depth of Euphotic Zone (Z_{eu})**

18 The depth of the euphotic zone was estimated from our automated, *in situ* observations of
19 PAR, chlorophyll *a* fluorescence, conductivity, and TSS, using a parameterization developed for
20 a large temperate estuary (Xu et al. 2005):

$$21 \quad K_d(\text{PAR}) = 1.17 + 0.024 \times [\text{chl-}a] + 0.006 \times [\text{TSS}] - 0.0225 \times S \quad (1)$$

22 and the relationship (Kirk 2011)

1
$$Z_{eu} = \frac{4.6}{K_d} \quad (2)$$

2 where K_d (PAR) is the light attenuation for photosynthetically active radiation, [chl- a] is the
3 concentration of chlorophyll a in $\mu\text{g L}^{-1}$, [TSS] is the concentration of total suspended solids in
4 mg L^{-1} , S is the salinity, and Z_{eu} is the depth of the euphotic zone. This method uses the sum of
5 three parameters to approximate the individual contributions of each constituent to the
6 attenuation and scattering of each wavelength of photosynthetically active light; it thus involves
7 considerable assumptions about the inherent optical properties of the water at our study site (Kirk
8 1984; Kirk 2003). Z_{eu} was estimated from the data at 15-minute intervals; the values were then
9 averaged over the course of each 24-hour day.

10 While this parameterization was developed for the lower reaches of the Chesapeake Bay,
11 it functioned as a reasonable, if not ideal, means of estimating euphotic zone depth in Long
12 Island Sound during our study period. To validate the applicability of the model, we compared
13 the depth estimates to both Secchi disk measurements we made at the site during each of our
14 sampling cruises ($n = 10$), and to PAR profiles from monthly CTDEEP CTD casts at Station I2
15 ($n = 3$). While these two independent measures generally confirmed the applicability of the
16 model to conditions at our study site, the model underestimated Z_{eu} by 2 meters relative to the
17 May CTD profile (data not shown).

18 **2.5.2 New Production Based on Nitrate**

19 To determine the quantity of new biomass produced through uptake of allochthonous
20 nitrate over the course of each daylight period, we calculated changes in daily nitrate
21 concentrations ($-\Delta\text{NO}_3$; $\mu\text{mol NO}_3^- \text{L}^{-1} \text{h}^{-1}$) for each daylight hour using the high-pass-filtered
22 nitrate dataset. The daily sum of these changes represents the rate of new production (in $-\mu\text{mol}$
23 $\text{NO}_3^- \text{L}^{-1} \text{d}^{-1}$).

1 Volumetric estimates of new production were normalized to a 12-hour day and then
2 converted to estimates in units of $\text{mg C m}^{-3} \text{ d}^{-1}$ as described in Johnson et al. (2006) by invoking
3 the ratio identified by Redfield (1934), i.e., $-\Delta\text{NO}_3 (\mu\text{mol N L}^{-1} \text{ d}^{-1}) \times 6.6 \mu\text{mol C}/\mu\text{mol N} \times 12$
4 $\mu\text{g C}/\mu\text{mol C}$. The volumetric production rate was then integrated over the depth of the euphotic
5 zone. Rates of new production were not calculated for the period of study during which fouling
6 of the SUNA occurred.

7 **2.5.3 Respiration and Primary Production**

8 We estimated rates of respiration and primary production from observed changes in
9 dissolved oxygen concentration in the bulk free water mass at the study site. Variations of this
10 method have been successfully applied to estimate ecosystem metabolism in a number of other
11 estuaries (Caffrey 2004; Caffrey 2003; D'Avanzo et al. 1996; Johnson et al. 1981) and in
12 offshore systems (DeGrandpre et al. 1997; Johnson et al. 1981; Johnson 2010). During daylight
13 hours, two processes occur. Oxygen is evolved as an end-product of photosynthesis in proportion
14 to available light, nutrients, and other factors; and some amount of oxygen is consumed through
15 respiration, generally as a terminal electron acceptor. In the absence of light, only respiration
16 occurs. Staehr et al. (2012) provide a useful synthesis of the diel oxygen flux method as it
17 currently used in several ecosystem types.

18 We estimated rates of ecosystem metabolism in $\text{g O}_2 \text{ m}^{-3} \text{ hr}^{-1}$ from observed changes in
19 dissolved oxygen over each one-hour time interval during the study. For these estimates, we used
20 unfiltered dissolved oxygen data (i.e., data without any signal processing). First, we estimated
21 the mean DO concentration at the top of each hour by averaging the 15-minute observations over
22 the previous time interval. We then calculated a depth-integrated rate of change in units of g O_2
23 $\text{m}^{-2} \text{ hr}^{-1}$ according to the formula (after Caffrey 2004)

1
$$\text{Oxygen flux} = (\text{DO}_{i2} - \text{DO}_{i1}) \times Z_{eu} + F \quad (3)$$

2 where DO_{i1} and DO_{i2} are the concentrations at the beginning and end of each hour, Z_{eu} is the
3 depth of the euphotic zone and F is the air-sea flux over the given time period. Because we did
4 not have a bottom-layer oxygen sensor or benthic flux chamber, our method does not directly
5 capture any contribution from the benthos. While our study focuses exclusively on water-column
6 processes and we make no claims to account for the benthos, we do assume based on the site's
7 27 m water depth that our free water measurements in the pelagic zone provide a reasonable
8 approximation for overall ecosystem metabolism. In estuaries, the relative contribution of
9 benthic respiration to total ecosystem metabolism generally decreases with depth when the depth
10 exceeds a threshold of approximately 10 m (Hopkinson and Smith 2005).

11 To obtain daily rates of community respiration, R_c , we first calculated a mean hourly rate
12 of respiration, R_h , by adding the estimated hourly fluxes over each nighttime period and dividing
13 by the number of hours of darkness. We then extrapolated this rate to a 24-hour day, assuming
14 that respiration was not significantly different during the daytime, to obtain a rate in $\text{g O}_2 \text{ m}^{-2} \text{ d}^{-1}$.
15 Because respiration is conventionally a positive quantity, we multiplied this value by -1 .

16 We then calculated a rate of daytime net community production, NCP_d , by adding the
17 oxygen fluxes over each daylight hour. We estimated daily rates of gross primary production,
18 GPP, and net ecosystem metabolism, NEM, according to the formulae

19
$$\text{GPP} = \text{NCP}_d + R_c \quad (4)$$

20 and

21
$$\text{NEM} = \text{GPP} - R_c \quad (5)$$

22 We converted daily rates from units of $\text{g O}_2 \text{ m}^{-2} \text{ d}^{-1}$ to $\text{mg C m}^{-2} \text{ d}^{-1}$ assuming a
23 respiratory quotient ($\text{mol O}_2 : \text{mol C}$) of 1.0 and, based on previous work specific to Long Island

1 Sound, a photosynthetic quotient (PQ) of 1.2, i.e., $1 \text{ g O}_2 \times 1 \text{ mol O}_2 / 32 \text{ g O}_2 \times 1 \text{ mol C} / 1.2 \text{ mol}$
2 $\text{O}_2 \times 12 \text{ g C} / 1 \text{ mol C} \times 1000 \text{ mg C} / 1 \text{ g C}$ (Goebel and Kremer 2007).

3 **2.5.4 Tidal Advection**

4 As in most studies employing fixed moorings, tidal advection in Long Island Sound
5 induces the movement of distinct water masses past our study site with different physical and
6 chemical properties; these water masses may, for example, have different concentrations of
7 dissolved oxygen and nitrate. In considering a time series such as the one presented here, one
8 must therefore employ some method to correct for or exclude this additional source of variation.
9 Tidal circulation in LIS is dominated by the lunar semidiurnal, or M_2 , component, which has an
10 approximate period of 12.4 hours (Bennett et al. 2010; Kenefick 1985). By basing our estimates
11 on hourly flux measurements (and not, as we might have done, on a single measurement of $\Delta[\text{O}_2]$
12 or $\Delta[\text{NO}_3^-]$ over the course of the 24-hour production day), we attempt to exclude any signals
13 with a period > 1 hour, including the 12.4-hour tidal component.

14 While we are confident our method excludes much of this unwanted variation, our
15 estimates likely still reflect some artifact of tidal advection. By calculating the maximum
16 possible tidal excursion of a given water mass in any direction from the study site, we can
17 roughly estimate the surface area that would be integrated by our method if we were
18 unsuccessful at excluding unwanted variation. This figure is a basic estimate of the advective
19 scale of our study. Using the maximum tidal current velocities predicted during our study period
20 (ebb and flood of 0.67 and 0.41 m s^{-1} , respectively, occurring on 8 August 2010, for a National
21 Ocean Service tidal current station 4.2 km west of our study site), we estimate that our technique
22 could integrate variation in water mass properties over a total distance of 11.0 km, i.e., up to 5.5
23 km on either side of the mooring location, along the primary east-west axis of tidal flow.

1 **2.5.5 Correction for Air-Sea Gas Exchange**

2 Because free water mass estimates of ecosystem metabolism are based on changes in
3 concentration of a dissolved gas (i.e., O₂ or CO₂), the method also requires that one account for
4 gain from or loss to the atmosphere across the air-water interface. We estimated this gain or loss
5 in terms of an hourly flux, F , according to:

$$6 \quad F = k_{O_2} \left[\frac{([O_2]_{\text{sat}, t1} - [O_2]_{\text{obs}, t1}) + ([O_2]_{\text{sat}, t2} - [O_2]_{\text{obs}, t2})}{2} \right] \quad (6)$$

7 where F is in units of g O₂ m⁻² hr⁻¹ and k_{O_2} is the gas transfer velocity in m hr⁻¹. The bracketed
8 term describes the average concentration gradient driving the flux over the course of each hour,
9 where $[O_2]_{\text{sat}, t1} - [O_2]_{\text{obs}, t1}$ and $[O_2]_{\text{sat}, t2} - [O_2]_{\text{obs}, t2}$ are the differences between concentration at
10 saturation and observed concentration (both in g m⁻³) at the beginning and end of each hour,
11 respectively. We determined $[O_2]_{\text{sat}}$ according to Weiss (1970), based on our observations of
12 salinity and temperature.

13 A negative value of F in this case results when the water is supersaturated with respect to
14 DO (i.e., $[O_2]_{\text{obs}} > [O_2]_{\text{sat}}$); the negative flux is net evasion, or diffusion of oxygen from the
15 water to the atmosphere.

16 **2.5.6 Estimation of Gas Transfer Velocity**

17 The model used to calculate gas transfer velocity across the air-water interface (k_{O_2}) can
18 significantly affect estimates of ecosystem metabolism in estuaries (see, e.g., sensitivity analyses
19 in Caffrey 2004; Howarth et al. 1992). One must therefore carefully select an appropriate
20 boundary-layer model and choose an appropriate method for estimating the velocity (Raymond
21 and Cole 2001). Because our site draws aspects of its physical and biogeochemical character

1 from both the coastal ocean and the shallower, more traditional estuary to its west, we estimated
2 k_{O_2} as a function of wind velocity using three different parameterizations. The metabolic rates we
3 report below are the mean values of the three independent estimates we obtained using the
4 different models.

5 We first estimated k_{O_2} according to the cubic relationship of Wanninkhof and McGillis
6 (1999)

$$7 \quad k_{660} = 0.0283 \times U_{10}^3 \quad (7)$$

8 where U_{10} is the wind velocity normalized to a height of 10 m above the sea surface and k_{660} is
9 the transfer velocity for carbon dioxide in seawater at 20 °C, which corresponds to a Schmidt
10 number (Sc) of 660. We estimated U_{10} from our observations of wind speed at 3.5 m according
11 to Panofsky and Dutton (1984). The parameterization in Equation 7 is based on carbon dioxide
12 covariance flux measurements in the open ocean.

13 We converted the velocity at Sc = 660 to k_{O_2} at the Schmidt number for oxygen gas at the
14 observed temperature and salinity, Sc_{O_2} , according to:

$$15 \quad Sc_{O_2} = \frac{\eta}{\rho_{SW} D_{O_2}} \quad (8)$$

16 and

$$17 \quad k_{O_2} = k_{660} \left(\frac{Sc_{O_2}}{Sc_{660}} \right)^n \quad (9)$$

18 where η is the dynamic viscosity in $kg\ m^{-1}\ s^{-1}$ (Sharqawy et al. 2010), ρ_{SW} is the density of
19 seawater at the observed salinity and temperature in $kg\ m^{-3}$, D_{O_2} is the mass diffusivity of oxygen

1 gas in water in $\text{m}^2 \text{s}^{-1}$ according to Wanninkhof (1992) using the association factor of Hayduk
2 and Laudie (1974), and n is the Schmidt number exponent. We applied an exponent of $-\frac{2}{3}$ for
3 $U_{10} \leq 4.2 \text{ m s}^{-1}$ and $-\frac{1}{2}$ for $U_{10} > 4.2 \text{ m s}^{-1}$.

4 We also computed the gas transfer velocity according to the best-fit model of Raymond
5 and Cole (2001)

$$6 \quad k_{600} = 1.91e^{0.35(U_{10})} \quad (10)$$

7 and the floating dome model of Marino and Howarth (1993) for oxygen gas in fresh water at
8 20°C :

$$9 \quad k_{530} = 1.09 + 0.249 \times U_{10} \quad (11)$$

10 Using the form of Equation 9 with the appropriate Schmidt numbers, we converted the velocities
11 obtained from these two models, $k_{CO_2, 600}$ and $k_{O_2, 530}$, respectively, to our desired k_{O_2} for
12 measured salinity and water temperature.

13 **3 Results**

14 **3.1 Nitrate**

15 From our *in situ* spectrophotometric data (Fig. 2b), we obtained for the study period a
16 mean nitrate concentration of $1.7 \mu\text{mol L}^{-1}$ and daytime mean concentration of $1.4 \mu\text{mol L}^{-1}$.
17 These observations were consistent with the low concentrations we measured in our discrete
18 samples, with annual data for central Long Island Sound dating back to 1992 (CTDEEP
19 unpublished data; Fig. 4), and with previous observations of summer nutrient concentrations in
20 the central Sound (Anderson and Taylor 2001; Bowman 1977; Chen et al. 1988; Riley and
21 Conover 1956). Nitrate in the central Sound did not exceed a concentration of $3.9 \mu\text{mol L}^{-1}$ at
22 any time during the study period.

1 The nitrate results give us confidence that the SUNA is accurate to within $\pm 0.5 \mu\text{mol L}^{-1}$
2 in coastal/estuarine waters characterized by minimal turbidity (calibration data not shown;
3 discrete samples are plotted in in Fig. 2b as open red circles). The SUNA proved capable of
4 detecting relative changes on the order of $0.1 \mu\text{mol L}^{-1}$ on scales > 12 hours. While higher
5 turbidity can decrease the accuracy of *in situ* optical nitrate measurements by up to $1.5 \mu\text{mol L}^{-1}$
6 (e.g., Zielinski et al. 2011), our results are consistent with findings from other studies that have
7 employed similar instruments in relatively clear waters (Christensen and Melling 2009; Johnson
8 2010; Johnson et al. 2006; Pidcock et al. 2010). Time-series analysis allowed us to isolate and
9 quantify clear diel fluctuations in nitrate and DO (e.g., Fig. 3c), which were often partly obscured
10 in the raw data by other dynamics that act on longer time scales, such as seasonal variation and
11 mixing of multiple water masses (Fig. 2a; 2b).

12 **3.2 Nitrogen Speciation**

13 Nitrate comprised, on average, between 10 and 15% of total dissolved nitrogen (TDN)
14 within the water column at the site. This value, calculated from SUNA-derived nitrate
15 concentrations and TDN measured in discrete samples at 11 dates over the course of the study
16 period, was consistent with trends in speciation observed over the previous two decades (Fig. 4).
17 By contrast, only a very low fraction of water-column TDN ($< 5\%$ on average) was present as
18 total ammonia ($\text{NH}_3 + \text{NH}_4^+$). We did not consider any potential ammonia fluxes from
19 underlying sediments at the site.

20 Because each species of algae is necessarily adapted to exploit different proportions of
21 the various forms of nitrogen, the nature and severity of eutrophication often hinges not merely
22 on the total quantity of anthropogenic nitrogen export, but on the form in which it is delivered to
23 a particular ecosystem (Paerl and Piehler 2007). The low abundance of both dissolved inorganic

1 species relative to TDN suggests that in central Long Island Sound, as elsewhere, most dissolved
2 nitrogen was present as dissolved organic nitrogen (DON). While certain species of
3 cyanobacteria, diatoms, and heterotrophic dinoflagellates can in some instances assimilate DON
4 directly during cell growth (Antia et al. 1991), most DON present in the oceans is in complex
5 macromolecules not readily accessible to phytoplankton engaged in the fixation of organic
6 matter (McCarthy et al. 1997). In considering here the potential enrichment of phytoplankton
7 populations, we thus focus our attention on the dissolved inorganic forms, with particular
8 emphasis on partitioning between the two species.

9 The inorganic nitrogen speciation we observed at the study site is consistent with the
10 general profile of anthropogenic nitrogen input to Long Island Sound, which is comprised
11 predominately of discharges from wastewater treatment facilities. Various estimates put the
12 relative magnitude of these WWTP inputs at between 40.1% (NYSDEC and CTDEP 2000) and
13 56.7% (Castro et al. 2003) of the total Long Island Sound nitrogen load. Generally, in urbanized
14 watersheds such as those responsible for much of the drainage to the Sound, nitrate often
15 comprises a higher percentage of the total nitrogen load relative to ammonia. Nitrate
16 predominates in the non-point sources (e.g., impervious surface runoff) and wastewater treatment
17 discharges that are the primary inputs to these systems. Conversely, in watersheds dominated by
18 intensive livestock agriculture, a greater proportion of the total nitrogen load often comes in
19 reduced forms such as ammonium and organic nitrogen (Paerl and Piehler 2007). Notably, only
20 15.4% of the total Long Island Sound nitrogen load originates in agriculture; by comparison,
21 agricultural activities in the Chesapeake Bay watershed, including intensive poultry feeding
22 operations, are responsible for more than 53% of the total nitrogen load (Castro et al. 2003).

1 The observed speciation is also supported by direct evidence and $\delta^{15}\text{N}$ – $\delta^{18}\text{O}$ isotopic
2 analysis from examination of several Long Island Sound wastewater treatment discharges to the
3 Naugatuck and Quinnipiac rivers (Anisfeld et al. 2007). The vast majority of plants that
4 discharge to the Sound are secondary treatment facilities that accomplish only nitrification, the
5 conversion of ammonia, which is generally toxic, to nitrate. A very limited number of facilities
6 that discharge to the Sound are equipped to accomplish any significant denitrification, a
7 microbially-catalyzed tertiary treatment process that converts nitrate to nonreactive dinitrogen
8 gas (N_2); only this process reduces the total amount of nitrogen that ends up in receiving waters.
9 This distribution of treatment capacity produces an aggregate end-member nitrogen discharge
10 stream composed overwhelmingly (i.e., > 80% of TDN) of nitrate (Anisfeld et al. 2007).

11 **3.3 Production and Respiration**

12 **3.3.1 Net Community Productivity**

13 Daily rates of all five measures of ecosystem metabolism ranged considerably, often
14 changing from one day to the next by an order of magnitude (Table 2; Fig. 5a; Fig. 6a). Daily,
15 depth-integrated rates of daytime net community production ranged from -2880 to 4460 mg C m^{-2}
16 d^{-1} ; the mean rate of NCP_d ($\pm 95\%$ CI) during the study period was $1100 \pm 208 \text{ mg C m}^{-2} \text{ d}^{-1}$
17 (Table 1). That the two daily rate extremes occurred on adjacent dates (10 and 9 June,
18 respectively) suggests large metabolic variation can occur over very short time scales. The 9
19 June NCP_d maximum coincides with a maximum in observed chlorophyll fluorescence
20 corresponding to a peak chlorophyll *a* concentration of $6.8 \mu\text{g L}^{-1}$ (Fig. 5c). Day to day variations
21 of similar magnitude have been observed in daytime ecosystem production rates in Long Island
22 Sound (Anderson and Taylor 2001), in other temperate estuaries (D'Avanzo et al. 1996) and in
23 offshore ecosystems (DeGrandpre et al. 1997).

1 The rate of community productivity increased steadily over the course of the spring and
2 summer study period from a mean of $672 \pm 531 \text{ mg C m}^{-2} \text{ d}^{-1}$ in May to a mean of 1650 ± 478
3 $\text{mg C m}^{-2} \text{ d}^{-1}$ in August. (The average rates $\pm 95\%$ CI for June and July were 1200 ± 444 and
4 $1010 \pm 214 \text{ mg C m}^{-2} \text{ d}^{-1}$, respectively.) The difference between the May and August monthly
5 mean values was significant ($p < 0.05$; paired t -test). The average hourly net community
6 production rate was weakly correlated with average temperature, suggesting at least some of the
7 increase in NCP_d was driven by the steady rise in sea surface temperature (Fig. 2e) typical of
8 Long Island Sound and other temperate estuaries in summertime ($R^2 = 0.12$; $p = 0.002$).

9 **3.3.2 Community Respiration**

10 Rates of community respiration, R_c , also varied widely over daily timescales (Fig. 6a). A
11 maximum, depth-integrated daily respiration rate of $6240 \text{ mg C m}^{-2} \text{ d}^{-1}$ was observed during the
12 final weeks of the study period on 6 August, when surface temperatures were highest. The mean
13 community respiration rate for the study period was $1660 \pm 431 \text{ mg C m}^{-2} \text{ d}^{-1}$. As in the case of
14 net community production, the respiration rate increased over the course the study period from a
15 monthly mean of $257 \pm 1630 \text{ mg C m}^{-2} \text{ d}^{-1}$ in May to a mean of $2650 \pm 1340 \text{ mg C m}^{-2} \text{ d}^{-1}$ in
16 August. Much of this increase appeared to be driven by seasonal warming (Fig. 2e); the average
17 hourly respiration rate was weakly correlated with sea surface temperature ($R^2 = 0.13$; $p < 0.001$).

18 The respiration rates we present fall well within the range of previous measurements of
19 ecosystem metabolism in Long Island Sound (Anderson and Taylor 2001; Goebel et al. 2006)
20 and in other temperate estuaries (Caffrey 2004; Kemp et al. 1997); our findings agree generally
21 with these previous measurements despite significant differences in the methods used (Table 3).
22 As in the case of daytime net production, estuaries exhibit significant day-to-day variation in
23 rates of community respiration. This is particularly true at elevated summertime temperatures of

1 systems that experience large anthropogenic inputs of nutrients and organic matter (Caffrey
2 2003; D'Avanzo et al. 1996; Zhang et al. 2010). Goebel and Kremer (2007) ascribed variation in
3 respiration rates in Long Island Sound to factors such as light availability (particularly in late
4 summer) and standing chlorophyll concentration.

5 **3.3.3 Gross Primary Productivity**

6 Rates of gross primary productivity ranged from near zero at the outset of the study
7 period (two days in May) to a maximum of $6390 \text{ mg C m}^{-2} \text{ d}^{-1}$ on 22 July. The mean rate of GPP
8 ($\pm 95\%$ CI) for the study period was $2080 \pm 419 \text{ mg C m}^{-2} \text{ d}^{-1}$. These rates agree with previous
9 summertime and annual estimates of GPP in Long Island Sound and in other temperate estuaries
10 of similar size (Table 3). Monthly mean rates for May, June, July and August were, respectively,
11 563 ± 1450 , 2160 ± 718 , 2210 ± 494 , and $3150 \pm 1140 \text{ mg C m}^{-2} \text{ d}^{-1}$. In spite of these general
12 trends, rates of GPP, like rates of NCP_d and R_c , varied considerably from day to day (Fig. 6a).

13 **3.3.4 Net Ecosystem Metabolism**

14 Large day-to-day variations in the rate of ecosystem metabolism, a fundamental measure
15 of the balance between respiration and photosynthesis, reflected the significant variations evident
16 in rates of R_c , NCP_d , and GPP (Fig. 6b). Rates of NEM ranged from $-2940 \text{ mg C m}^{-2} \text{ d}^{-1}$ on 21
17 May to a maximum of $3870 \text{ mg C m}^{-2} \text{ d}^{-1}$. While the mean rate for the study period (426 ± 203
18 $\text{mg C m}^{-2} \text{ d}^{-1}$) indicated the system remained weakly autotrophic at seasonal timescales, the mean
19 rate for July ($292 \pm 235 \text{ mg C m}^{-2} \text{ d}^{-1}$) was effectively zero, suggesting conditions at the site
20 trended toward near trophic balance or heterotrophy as the summer progressed.

21 **4 Discussion**

22 **4.1 Controls on Nitrate Concentration and Uptake**

23 **4.1.1 Timescales of Hours to Days: Cyclical and Recurring Controls**

1 As is true in most natural systems, nitrate concentrations in central Long Island Sound
2 were correlated to a variety of physical and biological forces that acted over different timescales.
3 On timescales of hours to days, variation in surface-layer nitrate concentrations was dominated
4 by the central biological processes of photosynthesis and respiration. The high-frequency data
5 show a daily, cyclical rise and fall that is strongly coupled to the diel photosynthetic cycle (Fig.
6 3); the cyclical trend in the nitrate signal is enhanced by applying a high-pass filter to the data
7 (Fig. 3c). Quantitatively, the daily average dissolved oxygen concentration, a proxy for the
8 relative balance on any given day between photosynthesis and respiration, was the single most
9 significant predictor of nitrate over the course of the study period, explaining 59% of the
10 variation in observed concentrations of the nutrient (Fig. 7a; $p < 0.001$).

11 The lowest nitrate concentrations, often reaching levels below the detection limit of the
12 SUNA, were typically observed in midafternoon, coinciding with daily peaks in irradiance and
13 dissolved oxygen production, indicators of maximum photosynthetic activity. The highest
14 concentrations were typically observed in the very early morning hours after the surface-layer
15 nitrate inventory was replenished overnight from a nitrate reservoir of slightly higher
16 concentration below the mixed layer. This pattern, in which daytime photosynthetic assimilation
17 of nitrate by phytoplankton results in a sharp drawdown in concentration that is counterbalanced
18 by nighttime recharge through tidal advection and diffusion, has been observed in previous time-
19 series studies using similar instruments in waters with higher absolute nitrate concentrations
20 (Johnson 2010; Johnson et al. 2006; Zielinski et al. 2011).

21 The data presented here suggest nitrate concentrations at the site often averaged between
22 1 and 2 $\mu\text{mol L}^{-1}$ over the course of a 24-hour day. We do not believe these observations are
23 necessarily incompatible with the zero or near-zero summer nitrate concentrations recorded for

1 the central Sound in previous studies (Bowman 1977; Riley and Conover 1956) and by the
2 CTDEEP monthly sampling program (CTDEEP unpublished data), which have analyzed discrete
3 samples using traditional colorimetric methods. Sample collection in these previous studies has
4 traditionally been performed during daylight hours when nitrate concentrations are lowest; few
5 samples have been collected during the early morning, when concentrations appeared to hit their
6 maximum on many days during this study. We suggest that these previous studies may have
7 underestimated daily average nitrate concentrations in the central Sound by 1 or 2 $\mu\text{mol L}^{-1}$.

8 The horizontal advection of water masses due to tidal transport also appeared to produce
9 some variation in nitrate over daily and hourly timescales. Decoupling of the nitrate signal from
10 the photosynthetic cycle (i.e., distortion of the signal so that the diurnal pattern of photosynthesis
11 is not evident) often coincided with stronger than average tidal currents during spring tides at
12 Sachem Head, the nearest station for which predictions are made (data not shown). Elsewhere in
13 Long Island Sound, tidal advection promotes the horizontal movement of water masses with
14 significantly different chemical properties (O'Donnell et al. 2008).

15 **4.1.2 Seasonal-Scale Controls**

16 On monthly to seasonal timescales, nitrate concentrations were controlled by changes in
17 the structure and physical properties of the estuarine water column. A typical water-column
18 stratification regime developed in the central Sound over the course of the study period (data not
19 shown), though storm events frequently served to temporarily weaken the stratification regime
20 and permit mixing across the thermocline. Conservative profiles of both temperature and salinity
21 on 27 May suggest stratification did not begin to develop at the site until early June. A deepening
22 of the thermocline in mid-June was followed by establishment of a deep, well-formed
23 thermocline by early July. The mean summertime nitrate concentrations of less than 2 $\mu\text{mol L}^{-1}$

1 during this period are typical of surface waters throughout much of Long Island Sound following
2 the stratification onset and intensification that occurs in the late spring (Anderson and Taylor
3 2001; Bowman 1977). Only at stations in the extreme western reaches of the Sound (i.e.,
4 approaching New York City and its immediate environs) are surface concentrations in excess of
5 $5 \mu\text{mol L}^{-1}$ routinely observed in summertime (Bowman 1977).

6 **4.1.3 Intermittent Control by Wind Forcing**

7 While the photosynthetic cycle and other recurring controls appeared to explain the
8 majority of variation in nitrate concentration, wind stress was also a powerful intermittent
9 physical forcing that acted to modify the principal nitrate trend component significantly and
10 often unpredictably. In the absence of significant storm activity or changes in prevailing weather,
11 winds were generally weakest in early morning and strongest and most variable in the late
12 afternoon, when a sea-breeze predominated. Winds varied in direction, with minor storm events
13 occurring on 7, 10, and 17 June (Fig. 5), on 14-15 and 22 July, and on 4-5 August. These storms
14 were accompanied by moderate rainfall (Fig. 2). Although no large storms occurred during the
15 study period, wind forcing had a significant effect on surface-layer nitrate concentrations through
16 both its variable daily behavior and during more pronounced intermittent storm events.

17 The large dataset we acquired through sustained, high-frequency sampling allowed us to
18 quantitatively assess the effect of wind stress on nitrate uptake at the site over an extended period.
19 As expected, on a daily basis, more pronounced wind stress generally increased surface layer
20 nitrate stocks through vertical mixing. Wind stress explained a significant ($p < 0.05$) albeit
21 modest proportion of the overall variation in nitrate concentration. Wind alone explained 19% of
22 the overall variation, while significant wave height, measured independently though largely a
23 function of surface wind speeds, explained 18% of the variation (Fig. 7b; 7c). On days with

1 higher winds, more nitrate was presumably entrained from below the mixed layer, consistent
2 with the classical conception of wind-driven nutrient entrainment (see, e.g., Klein and Coste
3 1984). The rapid response of surface-layer nitrate concentrations to wind forcing has been
4 previously observed in oligotrophic zones of the open ocean: In several observational studies,
5 sustained winds entrained nitrate and other nutrients from below a heavily depleted mixed layer,
6 causing measureable increases in nutrient concentrations followed by drawdown as the nutrients
7 were assimilated by phytoplankton (Eppley and Renger 1988; Klein and Coste 1984; Marra et al.
8 1990). In LIS, wind-induced vertical mixing is the primary mechanism by which dissolved
9 oxygen is exchanged in summer between surface and subpycnocline waters (O'Donnell et al.
10 2008). This same vertical mixing explains the exchange of nitrate and other species across the
11 pycnocline.

12 While wind stress generally had an effect consistent with that observed in the open ocean,
13 we observed during a nighttime storm event on 7 June an intriguing ecosystem response
14 inconsistent with the general model (Fig. 8). During the event, wind-induced deepening of the
15 mixed layer resulted in rapid loss of surface-layer nitrate stocks that had accumulated during a
16 period of relatively calm winds over the previous week. On the days leading up to the storm
17 event, we observed the characteristic, photosynthetically-driven daily oscillations in both nitrate
18 and dissolved oxygen concentration that were evident throughout the study period. This pattern
19 of diel rise and fall (solid red trace in Fig. 8) was accompanied by a gradual accumulation of
20 both dissolved oxygen and nitrate that extended over a time period of several days (broken red
21 trace). Following the rapid decrease in surface-layer concentration caused by the event,
22 continued wind forcing appeared to perturb or mask the regular photosynthetic cycles in both
23 species for several days.

1 In reconciling this event with our statistical analysis, we note that the classic wind stress-
2 entrainment response model is not universally applicable. Nutrient entrainment through surface
3 wind forcing occurs only if winds are sufficiently strong and long in duration, and only if the
4 water column is not so strongly stratified that it resists mixing (Bigg et al. 1989). Additionally,
5 because bottom-layer nitrate concentrations in the central Sound are not generally any
6 appreciably higher than concentrations at the surface during summer months (Matthew Lyman,
7 CTDEEP, unpublished data), nitrate is not necessarily entrained into the mixed layer even when
8 stratification is weakened by wind stress.

9 **4.1.4 No Correlation with Rainfall**

10 Unexpectedly, we did not find any significant statistical relationship between surface-
11 layer nitrate concentrations and rainfall. Evidence suggests atmospheric deposition can be
12 significant source of anthropogenic ammonium and nitrate species ($\text{NO}_3^- + \text{HNO}_3$) to the overall
13 LIS DIN pool (Hu et al. 1998; Luo et al. 2002). Deposition has been previously linked to
14 surface-layer nutrient increases over short time scales in western Long Island Sound (Anderson
15 and Taylor 2001). Relative to other inputs, however, the atmospheric input of all forms of
16 nitrogen to Long Island Sound is estimated to be small (< 13% of the total load; NYSDEC and
17 CTDEP 2000). Furthermore, many of the observed increases in ambient nitrate concentrations
18 associated with rainfall in these previous studies were attributed not to direct wet deposition, but
19 largely to the ensuing pulse in stormwater runoff and (untreated) sewage discharge that usually
20 accompany large storm events in urbanized areas (Anderson and Taylor 2001). Because of our
21 study site's relative distance from land, we hypothesize that, at least over short time scales, it
22 should be somewhat more buffered with respect to these discharges than is the western Sound.

23 **4.2 Estimates of New Production and the *f*-ratio**

1 Given the apparent nitrogen speciation at our site, the relatively small magnitude of the
2 atmospheric nitrogen input, and evidence that very little of the primary anthropogenic input to
3 Long Island Sound (i.e., wastewater) was as ammonium, we approximate total new production
4 here as the rate of fixation of new biomass based on allochthonous nitrate. This is the simple,
5 original definition of Dugdale (1967) as modified by Eppley and Peterson (1979). In the case of
6 the mesotrophic and oligotrophic open-ocean systems in which the concept of new production
7 was conceived, nitrate is assumed to be allochthonous because it is delivered to the surface layer
8 from below the thermocline. In the case of estuaries such as Long Island Sound, a significant
9 fraction of the allochthonous nitrate may also be delivered from lateral terrestrial inputs that are
10 either natural or anthropogenic.

11 For the subset of the study period for which we had reliable nitrate data, rates of new
12 production in the central Sound ranged from near-zero ($\sim 10 \text{ mg C m}^{-2} \text{ d}^{-1}$ on 23 May) to a
13 maximum of $546 \text{ mg C m}^{-2} \text{ d}^{-1}$ on 15 June. The mean new production rate was $182 \pm 37 \text{ mg C m}^{-2} \text{ d}^{-1}$,
14 well above the $102 \text{ mg C m}^{-2} \text{ d}^{-1}$ estimated by Eppley and Peterson (1979) for the world's
15 inshore waters but lower than their accompanying estimate for neritic waters of $460 \text{ mg C m}^{-2} \text{ d}^{-1}$.

16 Statistically significant changes in amplitude of the nitrate signal were not always evident
17 in the high-pass dataset. On approximately 18% of days, nitrate appeared to be decoupled from
18 the photosynthetic cycle and a distinct rise and fall could not be discerned; on these days, rates of
19 nitrate-based production were not calculated. The degree of decoupling, i.e., the extent to which
20 the nitrate signal diverged from the regular daily pattern, appeared to be a function of wind and
21 tidal strength. Strong winds and stronger than usual tides seemed to promote divergence, while
22 the nitrate signal was more evident on days when meteorological and physical forces were weak.

1 Daily values of the f -ratio, i.e. the ratio of new production to production based on all
2 forms of fixed nitrogen

$$3 \quad f = \frac{P_{\text{NO}_3}}{\text{NCP}_d} \quad (12)$$

4 ranged from near zero on several dates to a maximum of 0.50 on 28 May. The study period
5 average f -ratio of 0.25 is of the same order as values obtained for other temperate estuaries, and
6 approximately equal to the Eppley and Peterson (1979) global estimate for inshore waters (Table
7 4). The only other published f -ratio estimate using nitrate data from an *in situ* spectrophotometer
8 (0.68 ± 0.12) is for a coastal ocean station 20 km west of Monterey Bay, California, which is not
9 biogeochemically or physically similar to our study site (Johnson et al. 2006). F -ratio values in
10 the literature are traditionally defined as a fraction of ^{14}C primary productivity, as is the estimate
11 we cite here for the station west of Monterey Bay. In recognition of the fact that rates obtained
12 from the ^{14}C method are generally thought to represent something between net and gross primary
13 production (Marra 2009), we defined our f -ratio as a fraction of net community production
14 (NCP_d) rather than GPP.

15 We note here that the classic definition of new production neglects the processes of
16 nitrogen fixation (Paerl 1997), direct uptake of DON (Bronk et al. 1994), and surface-layer
17 nitrification (Yool et al. 2007). Depending on their magnitude and significance to the particular
18 ecosystem, the failure to include these processes may result in over- or underestimation of the
19 rate of new production and thus the f -ratio. The definition we adopt also neglects the fact that in
20 estuaries specifically, ammonium may come from both allochthonous and autochthonous sources,
21 making it a possible additional basis for new production (Mortazavi et al. 2000a; Paerl 1997).

22 The apparent biogeochemistry at our study site, and the general nature of nitrogen
23 enrichment in Long Island Sound, suggest that the Eppley and Peterson definition should provide

1 an adequate approximation of the rate of new production in central Long Island Sound. In other
2 estuaries for which estimates of new production have been made, such as Apalachicola Bay, a
3 shallow estuary in Florida that receives more than 72% of its nitrogen load from agricultural
4 activities (Castro et al. 2003), an approximation neglecting the ammonia species would very
5 likely underestimate the rate of new production. In Apalachicola Bay, average annual new
6 production was estimated to be $143 \text{ mg C m}^{-2} \text{ d}^{-1}$ using a mass-balance nutrient model and
7 primary productivity estimates from ^{14}C incubations (Mortazavi et al. 2000a). The f -ratio we
8 present here of 0.25 suggests that even at low concentrations, nitrate drives a very significant
9 fraction of total production in central Long Island Sound.

10 **4.3 Effect of Wind Forcing on New Production and the f -ratio**

11 Wind forcing had a significant effect on both the rate of new production ($R^2 = 0.49$; $p <$
12 0.005) and on the f -ratio ($R^2 = 0.30$; $p < 0.05$). Even as higher wind velocities increased surface-
13 layer nitrate stocks, the forcing appeared to drive down the rate of new production and the value
14 of the f -ratio (Fig. 9). These relationships suggest that the delivery of nitrate to the surface layer
15 is not alone sufficient to support an increase in primary production: Winds must be of
16 appropriate strength and duration for the particular stratification regime, and in an open ocean
17 setting, also be accompanied by sufficient Ekman pumping. In two instances, a measureable
18 bloom in primary production was observed only after a succession of separate wind events
19 occurred over span of multiple days (Andersen and Prieur 2000; Wu et al. 2008). In Long Island
20 Sound specifically, Chen et al. (1988) have suggested that nutrient entrainment and any resultant
21 increase in primary production may result only from a combination of surface wind stress and
22 tidal advection.

1 The fact that different physical mechanisms govern nutrient delivery to the mixed layer in
2 the open ocean and in coastal environments and estuaries (Klein and Coste 1984) may explain
3 why we did not consistently observe the expected response at our study site. In estuaries and
4 other shallow waters, for example, the wind forcing that accompanies storm events can intensify
5 turbidity and reduce available light even as mixed-layer deepening delivers nutrients to the
6 surface layer (Cloern 1987). This mechanism has been recently hypothesized in shallow systems
7 where wind forcing penetrates into the sediment layer (Brand et al. 2010). Despite the central
8 Sound's greater water depth and generally low turbidity, the data suggest wind-induced sediment
9 suspension might still have played some role in inhibiting photosynthetic uptake of newly-
10 delivered nitrate at our site.

11 **4.4 Variations in Net Ecosystem Metabolism**

12 While the differences between monthly mean rates of ecosystem metabolism were not
13 significant, average values suggest the system transitioned from net autotrophy to trophic
14 balance or net heterotrophy over the course of the study period. A June average of 585 ± 389 mg
15 C m⁻² d⁻¹ reflects a state of net autotrophy that dominated during the first half of the summer. By
16 comparison, July and August means of 292 ± 235 and 494 ± 456 mg C m⁻² d⁻¹, respectively,
17 suggest the system trended toward even trophic status as sea surface temperatures at the site
18 increased to nearly 25 °C (Fig. 2e) and encouraged higher rates of respiration.

19 The transition to even trophic status or slight net heterotrophy reflected a steady decline
20 in dissolved oxygen over the course of the study period (Fig. 2a). While this seasonal drawdown
21 was not of the order of magnitude that accompanies the severe benthic hypoxia typical of
22 westerly sites in the Sound (O'Donnell et al. 2008; Welsh and Eller 1991), it nevertheless
23 reflects a unifying seasonal trend within the estuary.

1 The state of metabolic balance (or weak net autotrophy) we report for our study site
2 suggests the central Sound is more similar in many ways to the deeper, more evenly
3 metabolically-balanced coastal ecosystems that lie to its east than it is to the heavily eutrophic
4 western Sound. Like many shallower temperate estuaries, western Long Island Sound is
5 generally (and at times strongly) heterotrophic. In a study of 42 sites within 22 U.S. estuaries,
6 Caffrey (2004) found that just two systems — both of them in the northeastern U.S. — were net
7 autotrophic in summertime. All 42 systems had average depths much shallower than that of Long
8 Island Sound, and none of the sites had a depth approaching that of our study site. Waquoit Bay,
9 a small estuary in Massachusetts, was significantly autotrophic in summer and exhibited weak
10 net autotrophy throughout the year; the larger Narragansett Bay was markedly net heterotrophic
11 in all seasons except winter (Table 3). Central Long Island sound appears in many ways similar
12 to the lower, polyhaline reaches of the Chesapeake Bay, for which similar estimates of GPP
13 ($1670 \text{ mg C m}^{-2} \text{ d}^{-1}$), community respiration ($1460 \text{ mg C m}^{-2} \text{ d}^{-1}$), and ecosystem metabolism
14 ($210 \text{ mg C m}^{-2} \text{ d}^{-1}$) have been reported (Kemp et al. 1997). The shallower, more eutrophic upper
15 reaches of the Bay were generally similar in trophic status to the western third of the Sound.

16 **4.5 Challenges Inherent in Free-Water Mass Measurements of Ecosystem Metabolism**

17 **4.5.1 Dissolved Oxygen Anomalies**

18 In this study, minimum rates of community respiration approached zero on eight separate
19 dates, seven of which occurred at the outset of the study period in May. These were days on
20 which we observed apparent nighttime increases in dissolved oxygen; because these anomalous
21 nighttime increases would have yielded negative community respiration rates according to our
22 method, we manually assigned to these days respiration rates of near zero. These dates
23 encompassed some of the lowest temperatures observed during our study (a minimum daily

1 average surface temperature of 12.9 °C was noted on 22 May). While near-zero rates of
2 community respiration are more commonly observed in winter months when Long Island Sound
3 surface waters reach temperatures near freezing, near-zero rates have been previously observed
4 in the Sound in May (Goebel and Kremer 2007).

5 Apparent nighttime production is evident on these dates even in the high-pass filtered
6 dissolved oxygen dataset (Fig. 3c). On several days, two dissolved oxygen maxima are evident.
7 The first is the expected, biologically-driven midafternoon maximum that occurs on all study
8 days, corresponding to both peak irradiance (Fig. 3d) and the daily nitrate minimum (red trace in
9 Fig. 3c); the other dissolved oxygen maximum, however, occurs at nighttime, suggesting an
10 abiotic process is responsible. Though unlikely, it is possible these instances of apparent dark
11 oxygen production were the result of abiotic or biologically mediated hydrogen peroxide
12 decomposition in surface waters during the night (Pamatmat 1997; Vermilyea et al. 2010).
13 However, concentrations of H₂O₂ typical of coastal waters in the northeastern U.S. (Moffett and
14 Zafiriou 1990) would not support an O₂ input of the magnitude necessary to explain the increases
15 we observed.

16 The magnitude of these nighttime anomalies (e.g., an apparent negative respiration rate of
17 $-5000 \text{ mg C m}^{-2} \text{ d}^{-1}$ on 22 May) suggests we either overestimated evasion of O₂ to the
18 atmosphere on these dates, or that the hourly interval we used for our flux calculations was not
19 sufficiently short to exclude variations in water mass properties caused by tidal advection. That
20 seven of the eight instances occurred during the earliest phase of the study period strengthens our
21 belief that the signal was masked on these dates by one or both of these physical processes.
22 Water mass heterogeneity can be particularly confounding during periods of lower metabolic
23 activity, when a weak biologically-driven dissolved oxygen signal is easily overwhelmed by

1 physical processes (Caffrey 2003; Swaney et al. 1999). The complex partitioning of variation in
2 dissolved oxygen concentrations between physical and biological factors was noted in some of
3 the earliest *in situ* estimates of ecosystem metabolism (Kemp and Boynton 1980).

4 With the exception of these instances, hourly flux calculations ably captured the changing
5 balance between photosynthetic oxygen production on the one hand, and respiration (oxygen
6 consumption) on the other (Fig. 10). The relative strength of these two opposing yet
7 complementary processes yields the core biological signal typical of oxygen fluxes in natural
8 systems: Oxygen production rates rise steadily throughout the day to a peak in midafternoon as
9 photosynthesis gradually outpaces oxygen consumption from respiration; as available light then
10 decreases with the setting sun, respiration increasingly dominates. Net oxygen consumption is
11 evident in the negative flux values that dominate evening and overnight hours.

12 While this common biological signal was evident in dissolved oxygen concentrations as
13 they changed over the course of each individual study day, the many other physical and
14 biological processes that regulate community metabolism in natural systems produced wide
15 variations in hourly flux rates (see, e.g., the many individual flux measurements that lie outside
16 the 5 and 95% confidence intervals in Fig. 10).

17 Our methods allowed us to isolate biogeochemical processes from physical interferences
18 to a very significant degree. However, as our results indicate, signal processing is not a panacea.
19 One might alternatively have attempted to capture the impact of physical forcings, particularly
20 the movement of different water masses, by deploying an array of multiple moorings over a
21 relatively small spatial extent. This approach is not dissimilar to the many studies in rivers and
22 other flowing waters (beginning with Odum 1956) that have attempted to follow the same water
23 mass over its transit within a confined water body. Implementing this sort of experimental design

1 at a site such as the one in this study would require significant investments in additional
2 instrumentation and field time; a simple, two-station, upstream-downstream study design would
3 not be adequate.

4 **4.5.2 Air-Sea Exchange**

5 The central Sound remained supersaturated with respect to dissolved oxygen throughout
6 the study period, consistent with our overall finding of net autotrophy over the course of the
7 study period. The magnitude of the air-sea flux varied significantly from day to day (see, e.g.,
8 Table 2), comprising, on average, 11% of the total carbon flux. Both the degree of variability and
9 the apparent magnitude of the flux relative to net ecosystem metabolism are consistent with
10 estimates of air-sea exchange in other temperate estuaries (Caffrey 2004; D'Avanzo et al. 1996;
11 Howarth et al. 1992; Kemp and Boynton 1980; Marino and Howarth 1993).

12 The fact that the air-sea flux accounts for so significant a percentage of the total organic
13 matter processed with the ecosystem emphasizes the need for great care in selecting a model of
14 air-sea gas exchange. A sensitivity analysis comparing the independent estimates based on each
15 of the three gas exchange models to the mean values on which we based our analysis (Table 5)
16 illustrates the significant impact assumptions about gas exchange can have on estimates of
17 biogeochemical indicators. On a percentage change basis, our estimates of NEM were the most
18 affected by modification of the gas exchange model. Our estimates of the f -ratio also changed
19 significantly under the various scenarios.

20 **5 Conclusion**

21 From continuous *in situ* measurements of dissolved oxygen and nitrate concentrations
22 and several other ancillary biogeochemical parameters, we were able to discern significant
23 hourly, daily, and seasonal variations in rates of GPP, new production, and NEM. While the

1 balance between photosynthesis and respiration explained the majority of variation in nitrate
2 concentration, wind stress modified the nitrate signal in ways both consistent and inconsistent
3 with patterns of nitrate entrainment previously observed in the open ocean.

4 Coupled with measurements of primary production, estimates of new production and the
5 f -ratio may be used as diagnostic tools to evaluate the degree of eutrophication in estuaries over
6 very short time scales, giving an oft-referenced and frequently re-examined model in chemical
7 oceanography new utility. While its effectiveness as a measurement of the strength of the
8 biological pump in the open ocean has been questioned (Yool et al. 2007), we suggest the f -ratio
9 can be a valuable tool in near-coastal eutrophic ecosystems — provided one makes accurate
10 assumptions about the nature and source of allochthonous nitrogen.

11 Because we derived our estimates almost exclusively from data gathered with *in situ*
12 devices, we confronted significant challenges that accompany the many benefits of these new
13 technologies. Investigators hoping to make the most of these instruments would do well to
14 incorporate into the very earliest stages of study design robust schemes for validation and
15 calibration of instrument data. Of equal necessity is the early and informed development of plans
16 and contingencies to deal with biofouling, a vexing obstacle that continues to represent a major
17 challenge in this exciting emerging field within the marine and aquatic sciences.

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2

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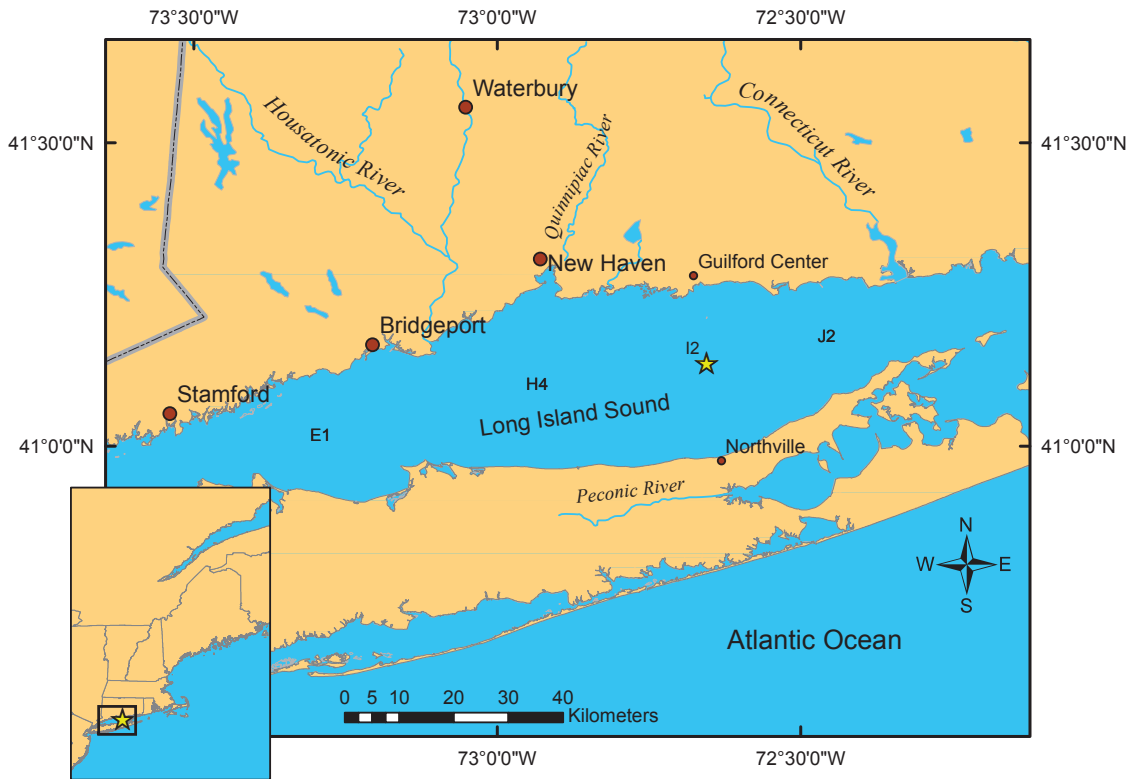
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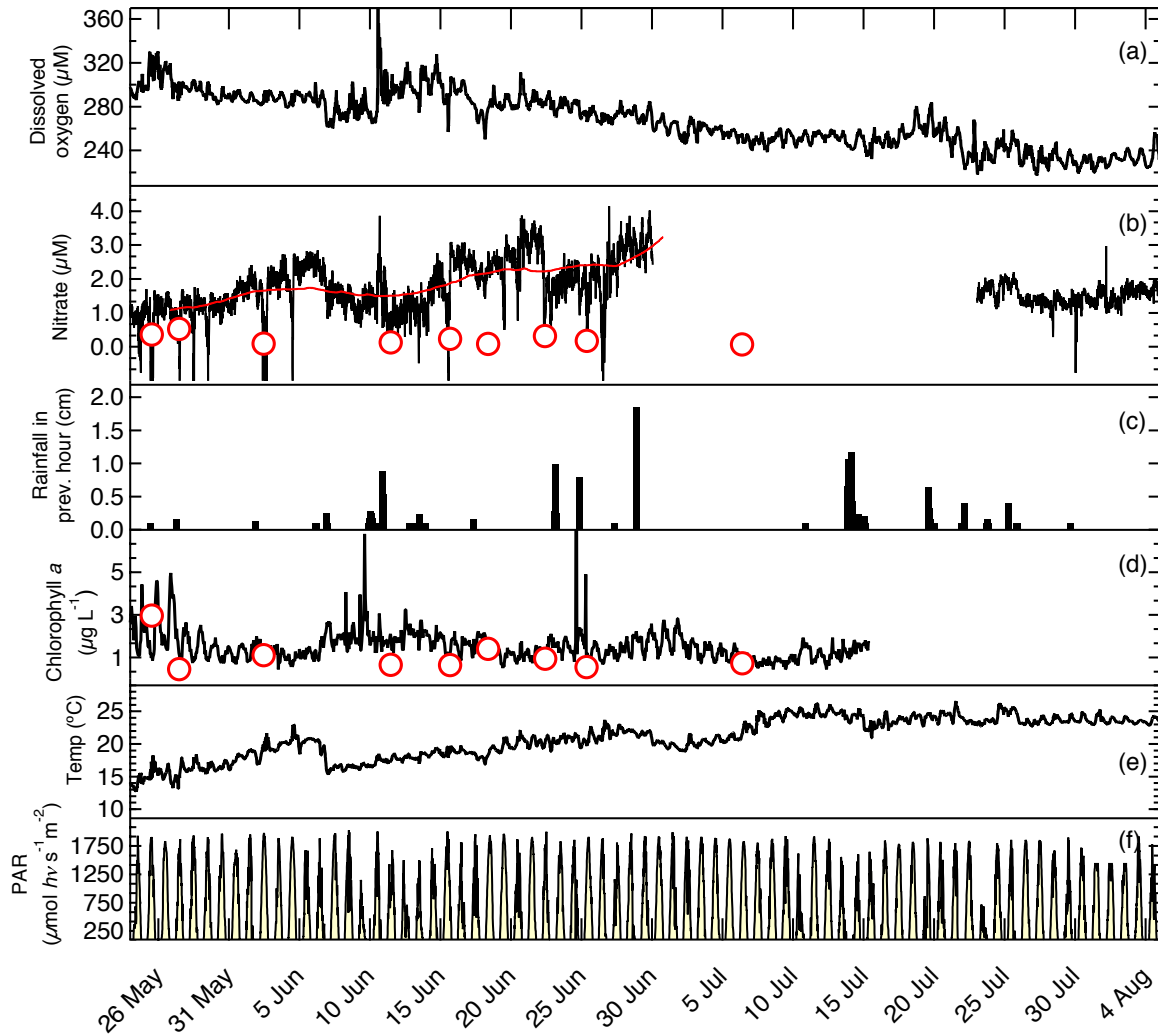
1 Zhang, J., D. Gilbert, A. J. Gooday, L. Levin, S. W. A. Naqvi, J. J. Middelburg, M. Scranton, W.
2 Ekau, A. Pena, B. Dewitte, T. Oguz, P. M. S. Monteiro, E. Urban, N. N. Rabalais, V.
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7 concentrations in turbid coastal waters using an in situ ultraviolet spectrophotometer.
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1 **Figures**

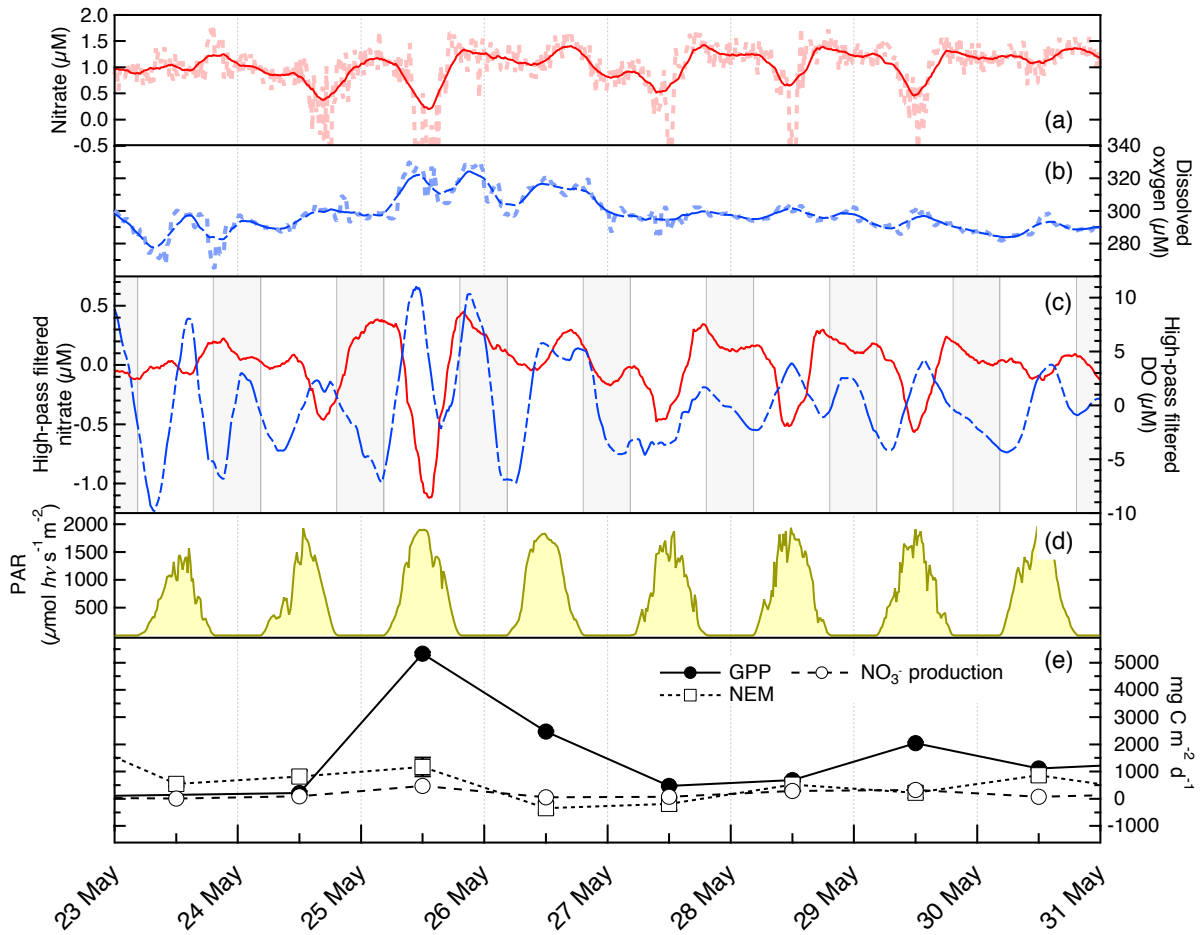


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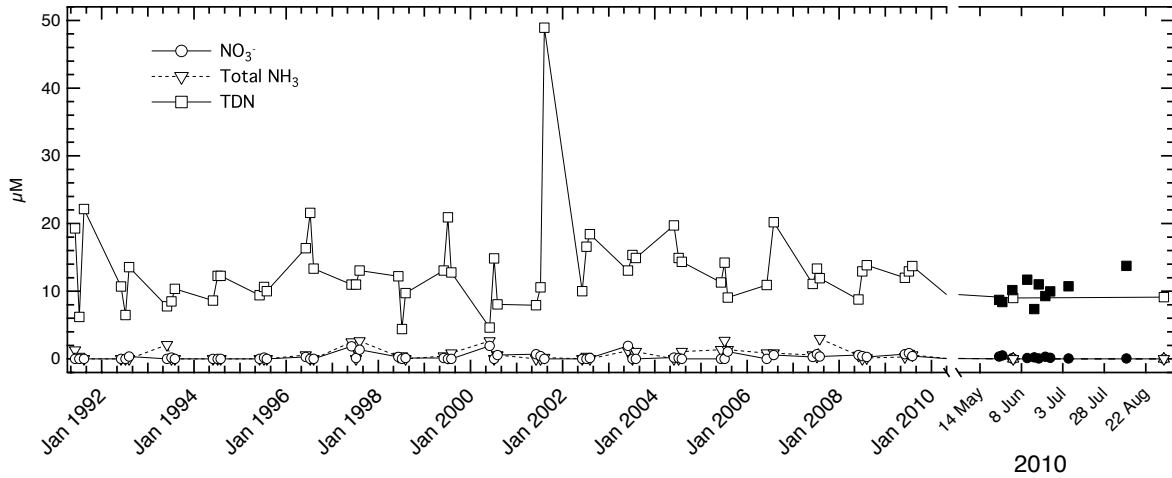
Fig. 1 Mooring location (☆; 41° 8.25' N, 072° 39.30' W; 27 m depth) in central Long Island Sound and selected CTDEEP water quality monitoring stations (E1, H4, I2, and J2) referenced in this study. Station I2 is coincident with the study site



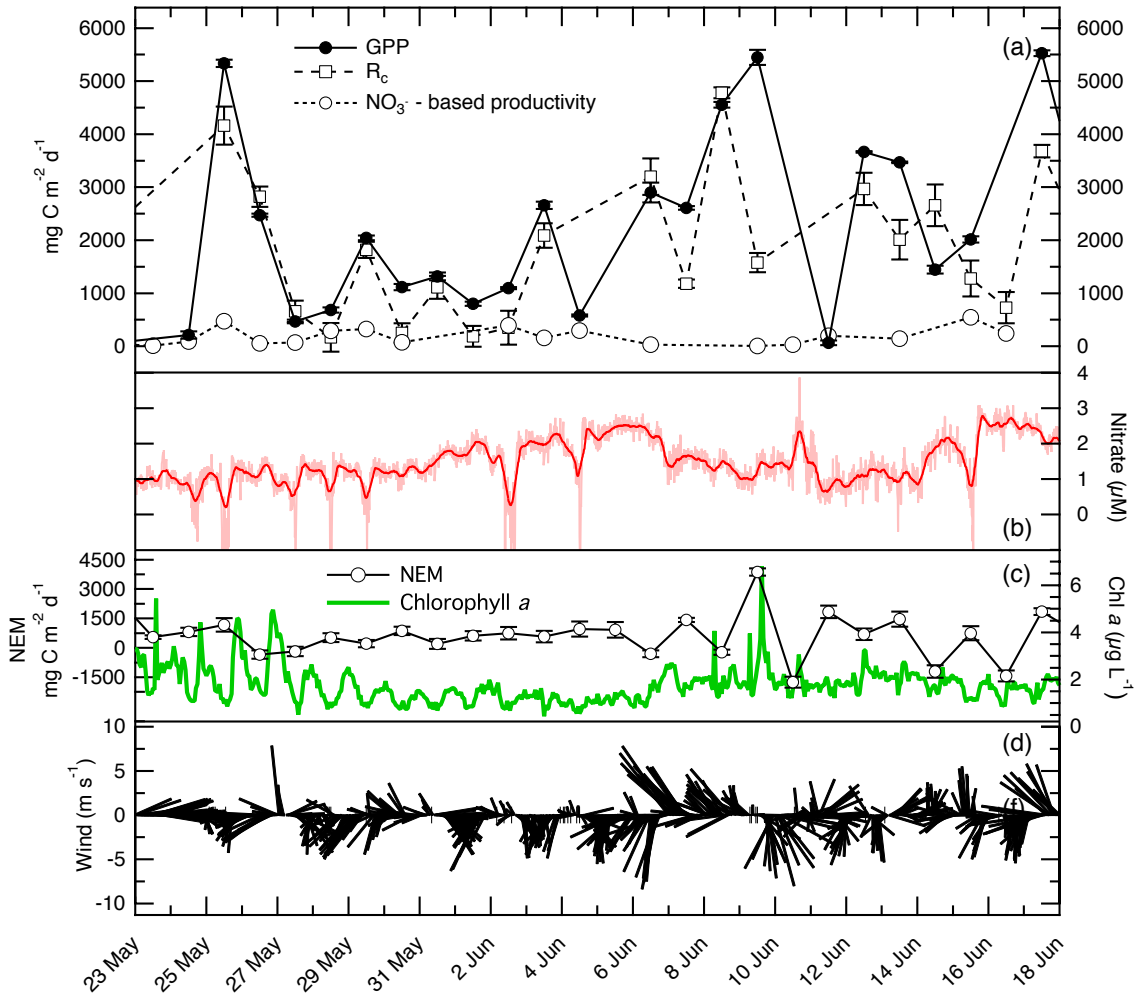
1
 2 **Fig. 2** Time series observations over the course of the study period at the study site in central
 3 Long Island Sound: (a) dissolved oxygen, (b) nitrate, (c) rainfall for the previous hour at Tweed-
 4 New Haven Airport, (d) chlorophyll *a*, (e) sea surface temperature, and (f) photosynthetically
 5 active radiation. The open (red) circles are concentrations of nitrate and chlorophyll *a* in discrete
 6 samples. The red trace in (b) is a low-pass Fourier transform, showing the long-term trend in the
 7 nitrate data set. A segment of the nitrate data is omitted from the period during which biofouling
 8 was discovered on the SUNA.



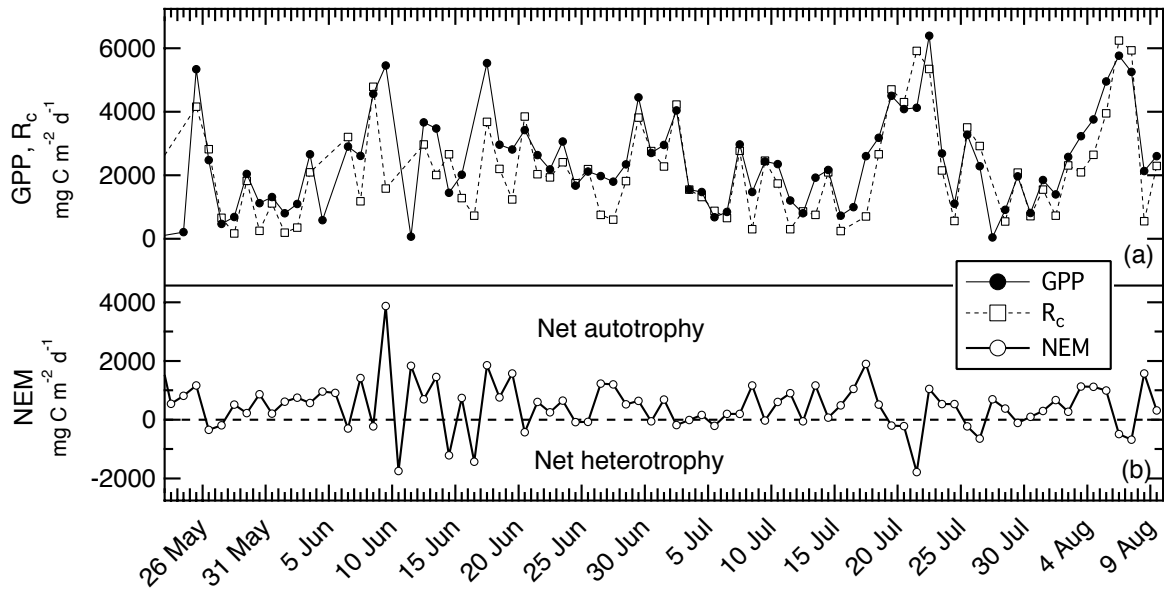
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 2 **Fig. 3** Observations of (a) nitrate (solid line), (b) dissolved oxygen (dashed line), (c) high-pass
 3 filtered nitrate and dissolved oxygen, and (d) photosynthetically active radiation during a
 4 representative nine day period (23-31 May 2010) at the mooring in central Long Island Sound. A
 5 high-pass Fourier transformation of the nitrate and DO datasets (c) emphasizes the inverse
 6 daytime relationship between nitrate assimilation and DO production, which we exploited to
 7 estimate rates of ecosystem metabolism and new production in (e). The diel oscillations in DO,
 8 nitrate, and PAR evident in (a)-(d) plots are coupled through the central processes of
 9 photosynthesis and respiration.



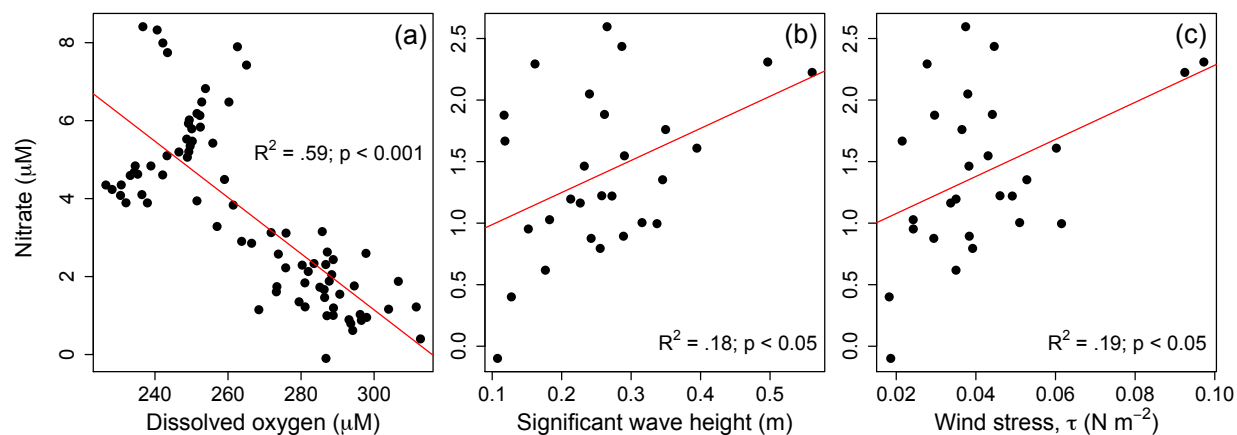
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 2 **Fig. 4** Interannual trends in the concentrations of bioavailable nitrogen species as a fraction of
 3 total dissolved nitrogen (TDN) at the study site in central Long Island Sound from 1992 to 2010,
 4 as measured in discrete samples. Open markers are from the CTDEEP (formerly CTDEP)
 5 monthly sampling program at station I2 (Matthew Lyman, unpublished data). The closed
 6 markers in the rightmost section of the plot are results from this study; note the expanded x-axis.
 7 All samples were taken during daylight hours.



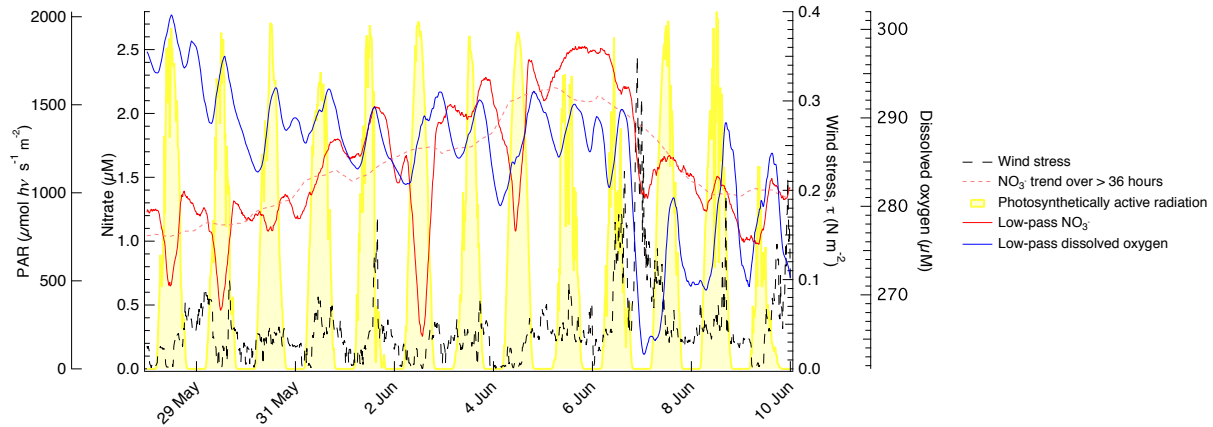
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 2 **Fig. 5** Daily rates of (a) community respiration, R_c , and new production based on nitrate, as a
 3 fraction of gross primary productivity over a three week segment of the study period in 2010.
 4 Variations in these three parameters reflect daily and longer-term changes in (b) nitrate
 5 concentration, (c) chlorophyll *a* (a proxy for the standing crop of biomass), and (d) wind speed
 6 and direction. Error bars in (a) and (c) represent 95% confidence intervals.



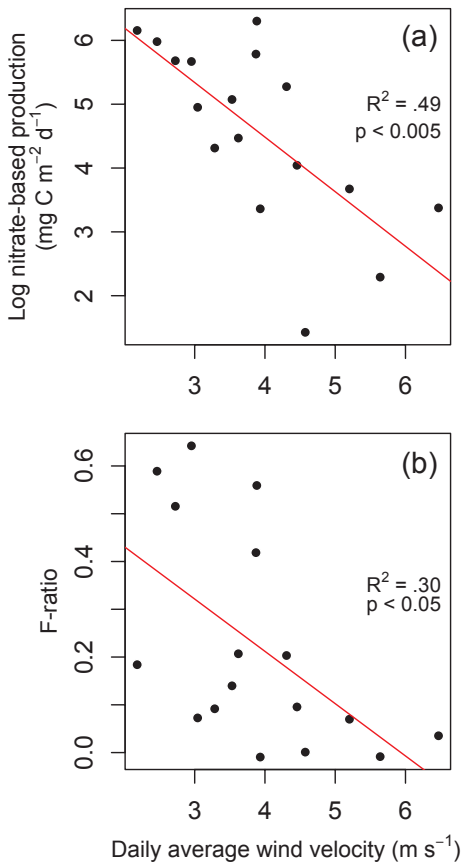
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 2 **Fig. 6** Significant variation in daily rates of (a) gross primary productivity and community
 3 respiration and (b) net ecosystem metabolism in central Long Island Sound during the summer
 4 and spring 2010. NEM is the balance between the two other variables. A negative value reflects
 5 net heterotrophy, i.e., the respiration of more organic matter than is being produced by
 6 autochthonous sources. A positive value indicates the system is net autotrophic.



1
 2 **Fig. 7** Relative contributions of three parameters to overall variation in the nitrate signal during a
 3 segment of the study period from 23 May to 30 June 2010. In (a), nitrate concentrations averaged
 4 over each 12-hour period are plotted versus average dissolved oxygen concentrations. In (b) and
 5 (c), nitrate is plotted versus significant wave height and wind stress. Red lines were fitted by
 6 linear least-squares regression. The relationship in each pairing was significant ($p < 0.05$).

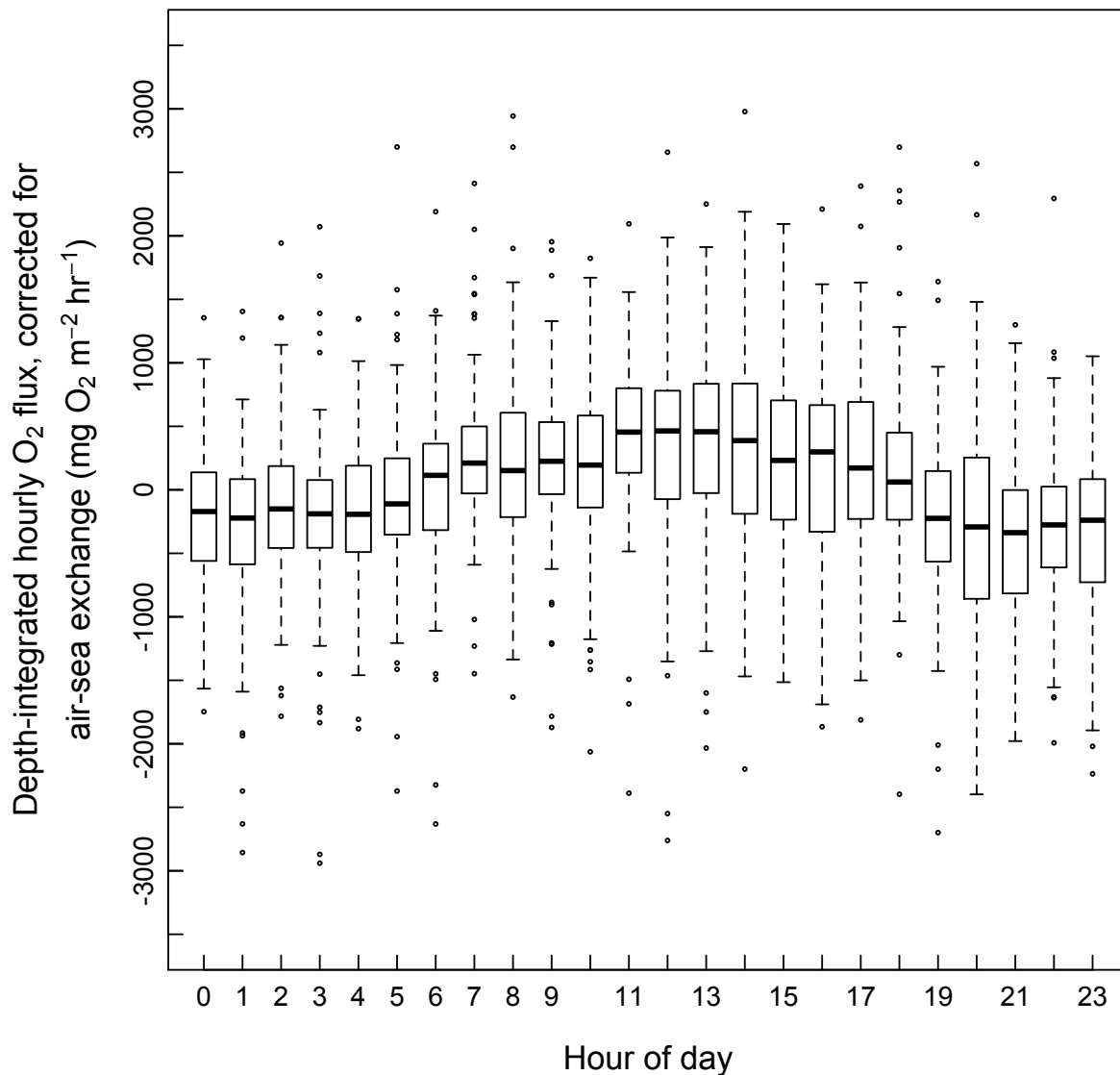


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 2 **Fig. 8** Dissipation of accumulated nitrate stock through deepening of mixed layer, immediately
 3 following the onset of a wind event during the night of 8 June 2010. A similar drop in surface
 4 DO concentration is also evident during the event, as oxygen in the well-ventilated surface water
 5 is exchanged with the oxygen-depleted bottom layer. Wind velocities observed during this storm
 6 event were the highest observed during the study period. The solid red line includes only
 7 components of the nitrate signal with periods > 2 hours and < 25 hours; very short term
 8 variability and the longest period signals were eliminated from this spectral subset using a band-
 9 pass Fourier transform. The broken red line is the product of a low-pass filter; this subset
 10 includes only cycles with periods > 25 hours.



1

2 **Fig. 9** Influence of wind velocity on (a) the daily NO_3^- based production rate and (b) daily
 3 estimates of the f -ratio at the Central Sound Buoy. Subset of data from May and June 2010.



1
 2 **Fig. 10** Box and whisker plot of the individual hourly oxygen hourly fluxes calculated during the
 3 study period (n = 1896), displayed by hour of day. During daylight hours, net oxygen production
 4 rises to a peak in midafternoon and then falls, in proportion to the amount of incident sunlight.
 5 During overnight hours, when only respiration takes place, net consumption of oxygen is evident.
 6 Fluxes were calculated according to Equation 3. The upper and lower members of the boxes are

- 1 the third and first quartiles of data in each category, respectively; whiskers roughly represent 5
- 2 and 95 percent confidence intervals.

Table 1 Representative daily variation in primary study parameters over two periods in central Long Island Sound. Presented are rates of gross primary production (GPP), NO₃⁻ based new production, daytime net community production (NCP_d), community respiration (R_c), net ecosystem metabolism (NEM), average surface wind speed, and the total daily air-sea gas flux (F) calculated using each of the three gas exchange models. Differences between the three air-sea gas exchange models were used to generate confidence intervals for the ecosystem metabolism figures. The *f*-ratio is calculated for the May and June dates.

| Date (2010) | Productivity (mg C m ⁻² d ⁻¹) | | | R _c (mg C m ⁻² d ⁻¹ ± 95 CI) | NEM (mg C m ⁻² d ⁻¹ ± 95% CI) | Average surface wind speed (m s ⁻¹ ± SD) | Total <i>F</i> (mg C m ⁻² d ⁻¹ ; <i>F</i> < 0 indicates water is supersaturated and flux is thus from sea to air) | | | | <i>f</i> -ratio (NO ₃ ⁻ based new production : NCP _d) |
|-------------|--|--|----------------------------|---|---|---|---|-------------------------|---------------------------|---------------------------------|---|
| | GPP (± 95% CI) | NO ₃ ⁻ based new production ^a | NCP _d (±95% CI) | | | | Wanninkhof and McGillis (1999) | Raymond and Cole (2001) | Marino and Howarth (1993) | Mean of three models (± 95% CI) | |
| 24 May | 210 ± 69 | 87.3 | 580 ± 157 | -600 ± 143 | 810 ± 212 | 3.6 ± 1.3 | -100 | -370 | -340 | -271 ± 167 | 0.15 |
| 25 May | 5330 ± 67 | 471.1 | 2770 ± 211 | 4160 ± 356 | 1170 ± 344 | 2.2 ± 1.2 | -110 | -570 | -600 | -425 ± 312 | 0.17 |
| 26 May | 2470 ± 29 | 57.0 | 730 ± 134 | 2820 ± 190 | -350 ± 206 | 4.5 ± 1.0 | -490 | -1290 | -1080 | -952 ± 474 | 0.08 |
| 27 May | 470 ± 32 | – | 60 ± 158 | 660 ± 204 | -190 ± 235 | 3.6 ± 1.6 | -240 | -710 | -610 | -519 ± 284 | – |
| 28 May | 680 ± 48 | 289.7 | 580 ± 131 | 170 ± 270 | 520 ± 233 | 3.0 ± 1.8 | -230 | -730 | -660 | -541 ± 311 | 0.50 |
| 29 May | 2040 ± 43 | 325.3 | 910 ± 134 | 1820 ± 152 | 220 ± 191 | 3.9 ± 2.0 | -350 | -900 | -730 | -659 ± 323 | 0.36 |
| 30 May | 1120 ± 59 | 74.6 | 960 ± 149 | 250 ± 182 | 870 ± 215 | 3.3 ± 1.0 | -130 | -490 | -470 | -362 ± 230 | 0.08 |
| 31 May | 1320 ± 76 | – | 630 ± 178 | 1110 ± 216 | 210 ± 255 | 3.9 ± 1.7 | -350 | -950 | -800 | -702 ± 357 | – |
| 1 Jun | 800 ± 39 | – | 690 ± 161 | 190 ± 198 | 620 ± 235 | 3.6 ± 1.9 | -390 | -1060 | -860 | -772 ± 389 | – |
| 2 Jun | 1090 ± 20 | 395.6 | 880 ± 204 | 350 ± 321 | 750 ± 324 | 2.5 ± 1.4 | -160 | -650 | -640 | -485 ± 314 | 0.45 |
| 3 Jun | 2660 ± 67 | 159.8 | 1350 ± 205 | 2090 ± 233 | 570 ± 292 | 3.5 ± 1.4 | -310 | -990 | -900 | -734 ± 419 | 0.12 |
| 4 Jun | 580 ± 17 | 293.3 | 810 ± 244 | -370 ± 379 | 950 ± 387 | 2.7 ± 1.8 | -290 | -960 | -890 | -715 ± 416 | 0.36 |
| 6 Jul | 850 ± 36 | – | 440 ± 113 | 650 ± 125 | 190 ± 160 | 2.4 ± 0.5 | -40 | -280 | -290 | -203 ± 161 | – |
| 7 Jul | 2970 ± 46 | – | 1240 ± 137 | 2770 ± 155 | 200 ± 195 | 2.5 ± 1.1 | -90 | -400 | -400 | -293 ± 204 | – |

| | | | | | | | | | | | |
|--------|-----------|---|------------|------------|------------|-----------|------|-------|------|------------|---|
| 8 Jul | 1470 ± 28 | – | 1280 ± 171 | 300 ± 312 | 1170 ± 288 | 4.0 ± 0.9 | -250 | -750 | -660 | -554 ± 301 | – |
| 9 Jul | 2440 ± 26 | – | 900 ± 145 | 2460 ± 269 | -20 ± 246 | 3.6 ± 0.8 | -250 | -860 | -790 | -636 ± 378 | – |
| 10 Jul | 2350 ± 5 | – | 1260 ± 172 | 1740 ± 271 | 610 ± 275 | 2.6 ± 1.1 | -100 | -520 | -530 | -384 ± 278 | – |
| 11 Jul | 1210 ± 16 | – | 1020 ± 185 | 300 ± 290 | 910 ± 294 | 4.0 ± 1.1 | -340 | -1010 | -880 | -746 ± 401 | – |
| 12 Jul | 810 ± 92 | – | 270 ± 195 | 860 ± 294 | -50 ± 298 | 3.8 ± 1.3 | -350 | -1040 | -900 | -764 ± 413 | – |
| 13 Jul | 1920 ± 28 | – | 1460 ± 127 | 750 ± 160 | 1170 ± 188 | 4.3 ± 1.7 | -490 | -1280 | -950 | -907 ± 452 | – |
| 14 Jul | 2160 ± 62 | – | 870 ± 71 | 2090 ± 209 | 80 ± 150 | 3.3 ± 1.9 | -290 | -810 | -680 | -595 ± 306 | – |
| 15 Jul | 730 ± 45 | – | 580 ± 116 | 240 ± 168 | 490 ± 176 | 3.2 ± 1.7 | -80 | -300 | -280 | -221 ± 139 | – |

Parameters were calculated as described in Tables 1 and 2. All values are reported as depth-integrated values of mg C or mg O₂ m⁻² d⁻¹ for entire water column. Where necessary, to convert O₂-based measurements to units of C, we assumed a photosynthetic quotient of 1.2 (1.2 mol O₂ : 1 mol CO₂). To convert parameters based on ΔNO₃⁻, we assumed Redfield stoichiometry (6.6 mol C : mol N).

^a NO₃⁻ based production could not be calculated for days on which the diel NO₃⁻ flux was decoupled from the daily production cycle, i.e., the pairing between PAR and dissolved oxygen production; see text for explanation. On these days, we hypothesize that some other variable, e.g., tidal currents or wind forcing, caused perturbations in NO₃⁻ concentration that masked the underlying diel signal, making it impossible to accurately determine rates of new production.

Table 2 Mean depth-integrated values of community metabolic parameters at central Long Island Sound, May-August 2010

| Parameter (units) | Mean value (\pm 95% CI) | Basis |
|--|------------------------------|--|
| Hourly respiration, R_h (mg C m ⁻² h ⁻¹) | 69.1 \pm 17.9 ^a | High-frequency overnight measurements of ΔO_2 , corrected for air-sea exchange |
| Gross community respiration, R_c (mg C m ⁻² d ⁻¹) | 1660 \pm 431 ^a | $R_h \times 24$ (rate assumed to be constant over 24-hour period) |
| Daytime net community production, NCP_d (mg C m ⁻² d ⁻¹) | 1100 \pm 208 ^a | High-frequency daytime measurements of ΔO_2 , corrected for air-sea exchange |
| Gross primary production, GPP (mg C m ⁻² d ⁻¹) | 2080 \pm 419 ^a | $GPP = NCP_d + R_h \times \text{hours of daylight}$ |
| Net ecosystem metabolism, NEM (mg C m ⁻² d ⁻¹) | 426 \pm 203 ^a | $NEM = GPP - R_c$; a positive value is net autotrophy |
| New production, i.e., phytoplankton growth based on NO_3^- uptake, P_{NO_3} (mg C m ⁻² d ⁻¹) | 182 \pm 37 ^b | Hourly fluxes from high-pass filtered $[NO_3^-]$ |
| <i>f</i> -ratio, i.e., ratio of NO_3^- based production to production based on all forms of fixed N (Eppley and Peterson 1979) | 0.25 ^b | $P_{NO_3} : NCP_d$ |

For ease of comparison, all values are reported in units of mg C. To convert O₂-based measurements to units of C, we assumed a photosynthetic quotient of 1.2 (1.2 mol O₂ : 1 mol CO₂). To convert parameters based on ΔNO_3^- , we assumed Redfield stoichiometry (6.6 mol C : mol N).

^a mean for May-August

^b mean for May-June

Table 3 Estimates of community metabolic rates at selected locations in Long Island Sound, and in two other estuaries

| Study location | Productivity (mg C m ⁻² d ⁻¹ ± 95% CI) | | | Community respiration (mg C m ⁻² d ⁻¹ ± 95% CI) | Method | Reference |
|--|--|---|---------------------------------------|---|---|--------------------------|
| | Gross | Net community or net primary | Net ecosystem metabolism ^a | | | |
| Central Long Island Sound, spring-summer 2010 | 2080 ± 419 | 1100 ± 208 ^c | 426 ± 203 | 1660 ± 431 | High-frequency ΔO ₂ in free water mass, corrected for air-sea gas exchange | This study |
| Central-west Long Island Sound at station H4, approx. 14 km south of New Haven, spring-summer, 2002-2003 | 2620 ± 900 | — | — | — | O ₂ -based light/dark bottle incubations, depth integration using photosynthesis-irradiance curves | Goebel et al. 2006 |
| Central-west Long Island Sound, ranges of monthly averages at 8 stations to west of longitude 72° 40' W, 2002-2003 | 40 (± 110) to 17300 (± 6100) ^b | -12500 (± 2700) to 11600 (± 11000) ^{b,c} | — | -500 (± 300) to 16600 (± 2700) ^b | O ₂ -based light/dark bottle incubations, depth integration using photosynthesis-irradiance curves | Goebel et al. 2006 |
| Western Long Island Sound near mouth of East River at Throgs Neck, summer, 1992-1993 | — | 0 to 37000 mol C m ⁻² d ⁻¹ ^d | — | — | Dilution-culture incubations; rates extrapolated from change in chl <i>a</i> concentration | Anderson and Taylor 2001 |
| Narragansett Bay, RI, at T-Wharf, S end Prudence Island, annual average, 1995-2000 | 2500 ± 310 ^e | — | -410 ± 130 ^e | 2900 ± 380 ^e | High-frequency ΔO ₂ in free water mass, corrected for air-sea gas exchange | Caffrey 2004 |
| Waquoit Bay, MA, at Central Basin, annual average, 1995-2000 | 2100 ± 94 ^e | — | 94 ± 63 ^e | 2800 ± 130 ^e | High-frequency ΔO ₂ in free water mass, corrected for air-sea gas exchange | Caffrey 2004 |
| Chesapeake Bay, lower section (polyhaline; to south of latitude 38° N), annual average, 1990-1992 | 1670 ^f | — | 210 ^f | 1460 ^f | O ₂ -based light/dark bottle incubations, depth integration using photosynthesis-irradiance curves | Kemp et al. 1997 |

All estimates except those presented in the current study are reported as depth-integrated values of mg C m⁻² d⁻¹ for the water column; our estimates are integrated measurements for the euphotic zone. Where necessary, to convert O₂-based measurements to units of C, we assumed a photosynthetic quotient of 1.2 (1.2 mol O₂ : 1 mol C).

^a defined as the balance between gross primary (autotrophic) production and community respiration, i.e., $GPP - R_c$; a value < 0 indicates net heterotrophy while a positive value indicates net autotrophy

^b peak rates of GPP and NCP were observed in summer at the extreme western end of LIS; peak rate of R_c was observed in summer at station E1

^c reported as net community productivity (NCP)

^d reported as net primary productivity (NPP)

^e values reported as \pm SE

^f estimates of annual rates ($m^2 yr^{-1}$) were converted to daily rate estimates by dividing by 365; peak values of GPP ($\sim 5000 mg C m^{-2} d^{-1}$) were observed in in the Lower Bay in July

Table 4 Estimates of the *f*-ratio and gross versus new productivity for selected ecosystems and marine ecosystem types

| Water body | Productivity ^a (mg C m ⁻² d ⁻¹ , unless noted, ± 95% CI) | | <i>f</i> -ratio | Method | | Reference |
|--|---|---|-------------------|--|---|--|
| | Total | New | | Total production | New production | |
| Atlantic Ocean, modeled annual average ^b | 286 | 71 | 0.26 | ¹⁴ C uptake method | ¹⁵ N-labelled NO ₃ ⁻ and NH ₄ ⁺ | Eppley and Peterson 1979 |
| Neritic waters, estimated global annual average ^b | 1000 | 460 | 0.46 | ¹⁴ C uptake method | ¹⁵ N-labelled NO ₃ ⁻ and NH ₄ ⁺ | Eppley and Peterson 1979 |
| Inshore waters, estimated global annual average ^b | 340 | 102 | 0.30 | ¹⁴ C uptake method | ¹⁵ N-labelled NO ₃ ⁻ and NH ₄ ⁺ | Eppley and Peterson 1979 |
| Central Long Island Sound, May-June 2010 ^c | 719 ± 538 | 182 ± 37 | 0.25 | Daytime net community production calculated from high-frequency ΔO ₂ in free water mass, corrected for air-sea gas exchange | Diel change in high-pass filtered [NO ₃ ⁻] | This study |
| Northern California coastal waters, 20 km west of Monterey Bay, average for Johnson et al.'s station M1, 2004-2006 | 145 ± 21 mg C m ³ d ⁻¹ | 99 ± 11 mg C m ³ d ⁻¹ | 0.68 ± 0.12 | ¹⁴ C uptake method | Diel change in high-pass filtered [NO ₃ ⁻] | Johnson et al. 2006 |
| Apalachicola Bay, Florida, ecosystem average, 1994-1996 | 750 ± 68 ^d | — | 0.19 ^d | ¹⁴ C uptake method | Extrapolated from net DIN input (Σ NH ₄ ⁺ , NO ₃ ⁻ , NO ₂ ⁻ , urea) based on ecosystem-level mass balance model | Mortazavi et al. 2000a; Mortazavi et al. 2000b |
| Carmans River estuary, Long Island, New York, annual average | — | — | 0.37 | ¹⁴ C uptake method | ¹⁵ N-labelled NO ₃ ⁻ , NH ₄ ⁺ , and urea | Carpenter and Dunham 1985 |

All values are reported in units of mg C. Where necessary, to convert O₂-based measurements to units of C, we assumed a photosynthetic quotient of 1.2 (1.2 mol O₂ : 1 mol C). To convert parameters based on ΔNO₃⁻, we assumed Redfield stoichiometry (6.6 mol C : mol N).

^a reported as water column depth-integrated values of m⁻² d⁻¹, except for values from the current study, which are reported as integrated measurements for the euphotic zone; and Johnson et al.'s estimates for northern California coastal waters, which are reported in volumetric units of m⁻³ d⁻¹

^b estimates of annual productivity (m² yr⁻¹) were converted to estimated daily average rates by dividing by 365

^c reported for only the portion of study period for which we had good nitrate data

^d annual averages; peak GCP of 1812 mg C m⁻² d⁻¹ was observed in summer; *f*-ratio ranged from 0.11 in summer at low river flow to 0.74 in winter at peak DIN input

Table 5 Sensitivity analysis of air-sea gas exchange calculation, showing effect of chosen model on daily rates of GPP, R_c , and NEM, and on the value of the f -ratio. For GPP, R_c , and NEM, we evaluated the effect of model choice on the daily mean value for the entire study period (May-August). For the f -ratio, we were able to evaluate only the effect on the May-June mean.

| Metabolic parameter | Transfer velocity model | Daily mean value (mg C m ⁻² d ⁻¹ ± 95% CI) | Difference from 3-model mean (%) |
|---------------------|--|--|----------------------------------|
| GPP | 3-model mean (basis for metabolic values reported in this study) | 2080 ± 419 | 0 |
| | Wanninkhof and McGillis (1999) | 2070 ± 418 | -0.48 |
| | Raymond and Cole (2001) | 2100 ± 419 | +1.0 |
| | Marino and Howarth (1993) | 2080 ± 422 | 0 |
| R_c | 3-model mean | 1660 ± 431 | 0 |
| | Wanninkhof and McGillis (1999) | 1850 ± 430 | +11 |
| | Raymond and Cole (2001) | 1540 ± 430 | -7.2 |
| | Marino and Howarth (1993) | 1590 ± 433 | -4.2 |
| NEM | 3-model mean | 426 ± 203 | 0 |
| | Wanninkhof and McGillis (1999) | 219 ± 201 | -49 |
| | Raymond and Cole (2001) | 560 ± 205 | +31 |
| | Marino and Howarth (1993) | 498 ± 205 | +17 |
| f -ratio | 3-model mean | 0.25 | 0 |
| | Wanninkhof and McGillis (1999) | 0.32 | +28 |
| | Raymond and Cole (2001) | 0.22 | -12 |
| | Marino and Howarth (1993) | 0.24 | -4.0 |