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Cell patterning using a dielectrophoretic–hydrodynamic trap

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Abstract The paper presents a dielectrophoretic method for cell patterning using dielectrophoretic–hydrodynamic trap. A distinctive characteristic of the device is that the dielectrophoretic (DEP) force is generated using a structure that combines conventional electrode-based DEP (eDEP) with insulator-based DEP method (iDEP). The conventional eDEP force is generated across the microfluidic channel between a top plate indium tin oxide electrode and a thin CrAu electrode. Meantime, an isolating cage built from SU8 photoresist around the thin electrode modifies the electric field generating an iDEP force. The cells that are flowing through a microfluidic channel are trapped in the SU8 cage by the total DEP force. As a result, according to the cell dimension and the thickness of the SU8 layer, different cell patterns can be achieved. If the cell's size is sensitively smaller than the dimensions of the hydrodynamic trap, due to the dipole–dipole interaction, the cell can be organized in 3D structures. The trapping method can be used for conducting genetic, biochemical or physiological studies on cells.

Keywords Dielectrophoresis · Cell patterning · Microfluidic device · Dielectrophoretic–hydrodynamic trap

1 Introduction

The use of microfluidic systems as an analytical tool in chemistry, biochemistry and life science is now well known. Microfluidic systems present significant advantages

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compared to bulk analytical techniques such as high throughput and efficiency; moreover, a small amount of sample can be processed (Arora et al. 2010). Microfluidic devices start to play an important role in cell and tissue engineering (Choudhury et al. 2011; Ni et al. 2009; Yeo et al. 2011) creating an "in vivo-like" environment for cellular behavior studies (Bhadriraju and Chen 2002; Huang et al. 2012) or for in vitro models for drug testing (Dittrich and Manz 2006; Haeberle and Zengerle 2007; Neužil et al. 2012). The necessity of conducting genetic, biochemical or physiological cell studies requires, firstly, a microsystem capable of trapping single cell (Rosenthal and Voldman 2005; Sun and Morgan 2010), 2D (Wu et al. 2011; Zhang et al. 2011) or 3D cell structures (Di Carlo et al. 2006; Morimoto and Takeuchi 2013) and, secondly, a configuration of the microsystem that assures the environmental conditions for maintaining the cell functionalities. According to Johann (2006) and Nilsson et al. (2009), the methods used for cell trapping can be divided into two major groups: chemical and physical. A detailed analysis of the methods used for cell trapping in tissue engineering is presented by Choudhury et al. (2011). Chemical trapping is based on chemical modification of the surface (Wong and Ho 2009), gel-based system (Khademhosseini et al. 2004) or cell aggregation mediated by transient intercellular linker (Mo et al. 2010). Physical trapping methods are hydrodynamic (Yang et al. 2002), dielectrophoretic (Iliescu et al. 2005), magnetic (Ino et al. 2008), acoustic (Manneberg et al. 2008) and laser based/optical (Birkbeck et al. 2003). Microfluidic chips having obstacles/barriers with dimensions comparable with the cell's size had been used for hydrodynamic trapping. Few methods of hydrodynamic trapping were studied: front trapping (Skelley et al. 2009), side trapping (Tan and Takeuchi 2007), gravity trapping (Khademhosseini et al. 2005). Hydrodynamic trapping methods are targeting mainly single cell trapping. Meanwhile, the process is simple, the main issue being related to the uniformity of the trapping process. Magnetic fields can be used for cell trapping/manipulation in microfluidic devices. In this case, the manipulation of the cells is performed using a magnetic force in the range of 2–1000 pN (Choudhury et al. 2011). The method requires paramagnetic properties of cells to be trapped (Iliescu et al. 2009a) or surface chemistry for cell attachment to magnetic beads (Ito et al. 2007). Mechanical force for cell trapping can be generated using ultrasonic waves. The generated force depends on the cell volume and wave frequency, the cells being concentrated either in the nodes or in the antinodes of the periodic wave pattern (Yeo et al. 2006). Optical trapping of cell or particles uses a focused laser beam to trap and handle particles. Optical tweezers can manipulate objects ranging from 10 Å to 10 μm and employ forces up to few hundreds of pN (Piggee 2009). Such forces are suitable for single cell handling, and

as a result, optical trapping is a low-throughput method for tissue engineering applications.

Among all the other techniques, cell patterning using dielectrophoresis (DEP) has the advantage of being a relatively fast process, precise, easy in operation and with a low degree of cell damage (Lin et al. 2006). Different DEP methods for cell trapping have been developed (Čemažar et al. 2013; Li et al. 2014; Martinez-Duarte 2012; Pethig 2010). The DEP techniques can be classified according to the method used to generate a gradient of electric field (key role in achieving an effective DEP force) as follows: travel wave, optical, conductivity gradient, insulating and electrode-based DEP. In the traveling wave DEP technique (Higginbotham and Sweatman 2008; Pethig et al. 2003), the gradient of electric field is induced by the phase change of the applied voltage. Optical DEP (Hwang et al. 2009) is a method where the gradient of electric field is generated using an optical image on a photodiode surface. Spatial/temporal conductivity gradient of media (Markx et al. 1997) can be used to generate a gradient of electric field. For insulating DEP (iDEP) (Iliescu et al. 2007; Jen et al. 2010; Lewpiriyawong and Yang 2014), the gradient is generated using a non-homogenous dielectric medium in a capacitor-like structure. In electrode-based DEP (eDEP), dissimilar size or shape of the electrodes is used to generate gradients of the electric field. These electrodes can be 2D (Masuda et al. 1989), 3D (Iliescu et al. 2006a; Li et al. 2013; Nasabi et al. 2013; Wang et al. 2009; Xing et al. 2013), combination of thin and bulk electrodes (Iliescu et al. 2009b), as well as 3D gates generated by placing the electrodes on top and bottom of the microfluidic channel (Choi et al. 2010). According to the response of the cells to the electric field, the DEP phenomenon can be described as either positive (pDEP)—cells are moving toward high electric field strengths—or negative (nDEP)—the cells are moving to the lower values of the electric field.

Dielectrophoresis was previously used for single cell patterning (Mittal et al. 2007; Taff et al. 2009). Early work in DEP cell patterning for tissue engineering applications—reported in (Albrecht et al. 2005)—proves that DEP electropatterning provides high resolution in cell localization in order to enhance cell–cell interaction. This study was further developed in (Albrecht et al. 2006), reporting 3D micropatterning of chondrocytes within hydrogels using DEP and in (Albrecht et al. 2007) where cells packaged in hydrogel microcapsules were trapped using pDEP generating a "tissue-like" structure. A uniform patterning of HepG2 cells in a 2D structure using a planar interdigitated ring electrode array is presented by Hsiung et al. (2008). Ho et al. (2006) demonstrate a 2D arrangement of cells in a structure that mimics the morphology of the liver tissue. Also, 2D cell structure patterned using nDEP is presented by Suzuki et al. (2008). One of the problems of the existing

cell patterning techniques is the low uniformity of the patterning process with possible consequences in undesirable cellular responses (Bhatia et al. 1999; Iwasa et al. 2003). For in vitro cellular studies, uniform and dense patterning techniques that assure cell–cell interactions are desired.

Here, we report a device architecture and a method for the precise and uniform cell patterning of single cell, 2D or 3D cell sheets. The method involves a DEP–hydrodynamic trap placed in a microfluidic structure. A unique characteristic of the DEP device is the merging of the iDEP and eDEP techniques. The gradient of electric field, generated in the cross section of the microfluidic structure, between thin electrodes (with a desired shape) placed on the bottom of the microchannel and a top plate (eDEP effect), is enhanced by positioning, on top of the thin electrode, an insulating layer (iDEP effect) with openings that follow the thin electrode shape. These openings in the insulating layers define, at the same time, hydrodynamic traps where the velocity of the fluid is sensitively reduced. The cells that are flown through a microfluidic channel are trapped in a hydrodynamic cage using cumulative iDEP–eDEP force. 2D and 3D structures were achieved using C3A cell line and red blood cells (RBC).

2 DEP device, materials and methods

2.1 DEP microfluidic device

The architecture of the dielectrophoretic trap is illustrated in Fig. 1a, while a detailed picture with the structure of the device is presented in Fig. 1b. The top electrode consists of a thin indium tin oxide (ITO) electrode plate deposited on glass. The bottom electrode is a thin CrAu layer patterned on an insulated substrate. On the surface of this electrode, a SU8 layer is placed with a pattern that follows the electrode shape. A PDMS spacer, adhesive bonded on the top of the ITO electrode, defines the depth and the shape of the microfluidic channel. The difference in electrode dimensions assures a gradient of electric field across the microfluidic channel, generating a DEP force (eDEP effect). This DEP force is enhanced by the presence of the insulating structure—SU8 layer—(iDEP effect). These "double DEP" configurations assure an effective cell trapping with a low applied voltage, and as a consequence, a low Joule heating effect (Tay et al. 2007).

The cell patterning method is shown in Fig. 1c. The cells suspended in the DEP buffer are flown through the
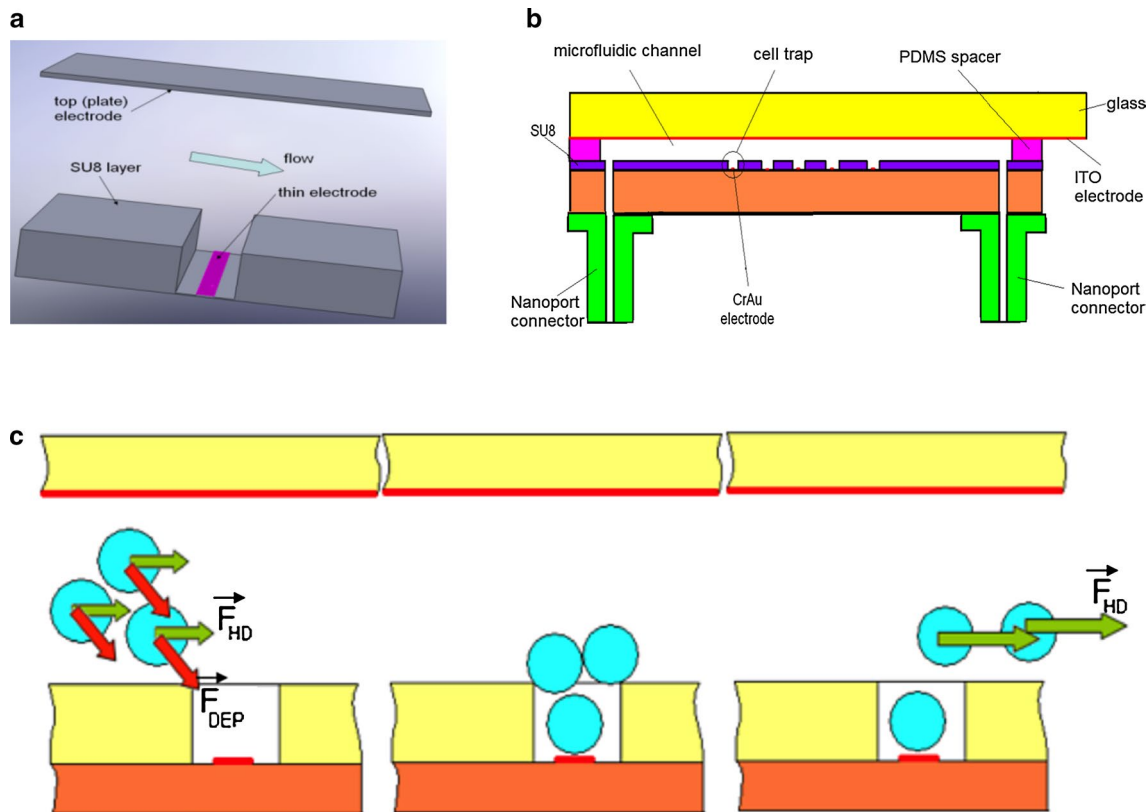

Fig. 1 Device structure and cell patterning method using eDEP and iDEP: **a** configuration of the microfluidic channel with electrodes and dielectric layer; **b** cross section with the cell patterning device; **c** cell patterning method: The cells are trapped using a strong DEP force, and the hydrodynamic force removed the excess cells

Fig. 2 Main steps of the fabrication process of the microfluidic chip with dielectrophoretic–hydrodynamic traps: **a** growth of dry SiO_2 (150 nm thick) on a silicon wafer, **b** deposition and patterning of the CrAu layer; **c** deposition and patterning of the SU8 layer and dicing of the Si wafer; **d** dicing of the ITO glass wafer; **e** cutting of the PDMS spacer and deposition of a thin PDMS adhesive layer on one surface; **f** adhesive bonding of the PDMS spacer on ITO glass; **g** contact imprinting of a PDMS layer on the PDMS spacer; **h** adhesive bonding between the ITO glass die (with the PDMS spacer) and silicon die; **i** photograph of the chip
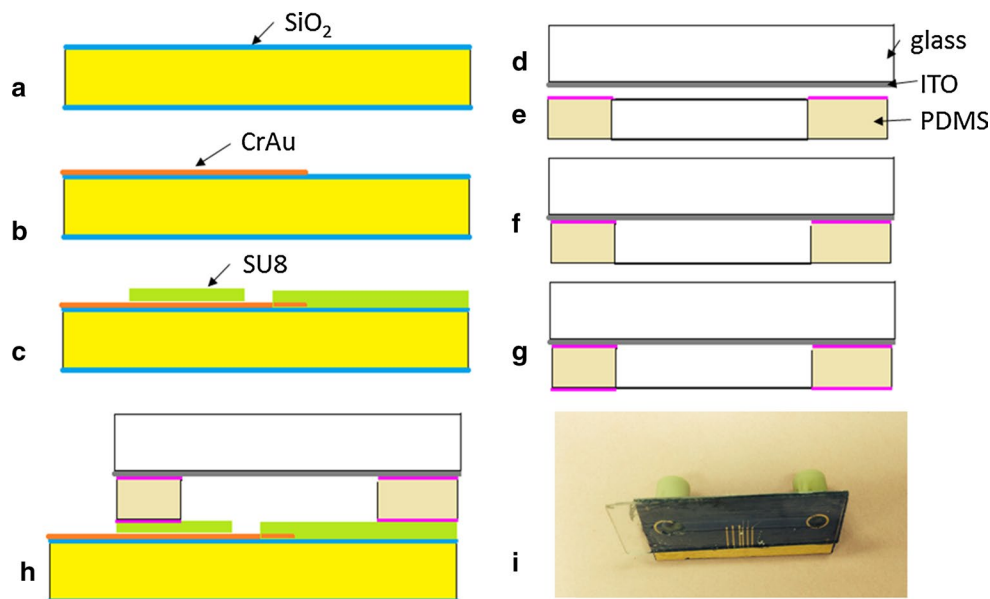
microfluidic channel. Due to the strong DEP force, the cells are moving to the thin electrode placed in the hydrodynamic trap (positive DEP). It results in a 2D or 3D aggregation of the cells in the trap (according to the cell size and trap dimension). By increasing the flow rate, the hydrodynamic force removes the cells placed on the top of the trap or on the top of the SU8 layer. The role of this insulating layer is not only to enhance the gradient of electric field, but also to act as a hydrodynamic cell trap, generating a "dead volume" where the velocity of the fluid is reduced.

2.2 Device fabrication

The device consists of two dies that are assembled together:

- a silicon die with the thin electrode and the SU8 hydrodynamic trap;
- a glass die having embedded the large ITO electrode.

The fabrication of the silicon die starts with a 4″ double-polish wafer, <100> crystallographic orientation, with a resistivity of 1–100 Ω/cm. The wafer was cleaned in piranha (H_2SO_4/H_2O_2 in ratio 2/1) at 120 °C for 20 min, then washed in DI (deionized) water and spun-rinsed dried. A dry 150-nm-thick SiO_2 layer was grown in a Tystar furnace at 1050 °C, for 3 h (Fig. 2a). A metal layer (Cr/Au 30/400 nm) was deposited with an e-beam evaporator (CHA). A 2-μm-thick positive photoresist (AZ7218 from Clarian) was deposited with a EVG 101 spin coater at 3000 rpm for 30 s. The film was exposed on a EVG 620 mask aligner and further developed. The metal layer was etched through the photoresist mask using commercially available Au and Cr etchants. In the next step,

the photoresist mask was removed in a NMP photoresist remover using ultrasonic agitation (Fig. 2b). In order to improve the adhesion of the SU8 layer to the substrate, a 50-nm-thick Teflon layer was deposited on the wafer in an ICP reactor (Alcatel) using C_4F_8. A 20-μm-thick SU8 2025 negative photoresist (for the definition of the hydrodynamic traps) was spin-coated on top of the Teflon layer at a speed of 4000 rpm and followed by a soft bake on a hot plate at 65 °C for 3 min and 95 °C for 6 min. Then, the SU8 was exposed on the EVG 620 mask aligner with constant exposure energy of 150 mJ/cm². After that, a postexposure bake at 65 °C for 1 min and 95 °C for 6 min was conducted, and the patterned SU8 was developed in SU8 developer and rinsed with isopropanol IPA. A short cleaning process of 1 min in O_2 plasma on a RIE (SPTS) was used for removing the Teflon layer and also to assure the clean surface of the electrode (Fig. 2c). The thickness of the SU8 layer was 20 μm. For the protection of the SU8 layer, the wafer was covered with a thick photoresist layer AZ4620 (~20 μm) before dicing of the individual chips on a DISCO 3350 equipment. Inlet and outlet holes were performed using a drilling machine with diamond beads.

For the second die, a glass wafer cover with ITO layer was diced on the same DISCO 3350 equipment (Fig. 2d). PDMS spacer was prepared by cutting the shape of the spacer from 75-μm-thick PDMS sheet (AAA ACME Rubber Co.). Curable PDMS elastomer kit: Sylgard 184 (Dow Corning) prepared in the ratio of 10/1 was spun on a dummy silicon wafer at 1000 rpm. A thin layer of curable PDMS was imprinted on the PDMS spacer (Fig. 2e) for the adhesive bonding between the ITO glass die and PDMS spacer (Fig. 2f). The curing of the soft PDMS was performed at 110 °C for 30 min.

The final bonding between the silicon and the glass dies was performed also using curable PDMS as adhesion layer. Similar to the method presented before, the curable PDMS was spun on a dummy wafer, and from this dummy wafer, a thin PDMS layer was imprinted on a Teflon cylinder and further transferred to the glass die (Fig. 2g). A similar method used for SU8 adhesive bonding was described in details by (Yu et al. 2006). The glass and the silicon dies were placed in contact and bonded in the same conditions as the bonding of the PDMS spacer on ITO glass die (Fig. 2h). Nanoport microfluidic connectors (Upchurch) were attached on the chip using UV curable glue (NOA63 form Norland Products). An image with the fabricated chip is presented in Fig. 2i.

2.3 Cell culture and DEP buffer

C3A cells (ATCC, USA) were cultured in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10 % FCS, 1.5 g/L sodium bicarbonate, 1 mM sodium pyruvate, 100 units/mL penicillin and 100 g/mL streptomycin. Red blood cells were isolated from whole blood by centrifugation and were resuspended in DEP buffer prior to experiment.

The dielectrophoretic buffer plays an important role in achieving the desired dielectrophoretic force. The DEP buffer was prepared by diluting 10 × TBE buffer (Promega Singapore) in deionized water and by adding 8.6 % (w/w) sucrose and 0.3 % (w/w) dextrose. The conductivity of the solution was 550 μS/m (measured with Oakton meter), while the pH was 7.

C3A cells were cultured in a 24-well plate until confluence. On the experiment day, the cells were removed from the incubator and culture medium was changed once prior to the start of the experiment in order to remove dead cells. After trypsinization and centrifugation, the culture medium was replaced by the DEP buffer at their corresponding time point, so that the cells were incubated in the DEP buffer at room temperature for 6, 4, 3, 2, 1 and 0.5 h, respectively. At the end of the experiment, the cell viability was measured using MTS assay and compared against C3A cells cultured in medium at room temperature (i.e., 0 h of incubation in DEP buffer).

2.4 Testing setup

A digital microscope (Keyence VHX 500) was used to monitor the cell trapping inside the DEP device. The flow of the cells suspended in the DEP buffer was performed using a syringe pump (New Era Pump Systems Inc.). A function generator (TG210 from TTi) and a linear amplifier (A-303 from A.A Lab Systems Ltd) were used to generate the drive signal applied to the DEP device. In order

to prevent the generation of bubbles inside the channel by electrolysis, prior flowing of the cell suspension into the microfluidic structure, the function generator and amplifier were powered on, but their outputs were set to the minimum.

3 Simulations and analytical model

3.1 Simulation of the electric field

A simulation of the electric field is presented in Fig. 3 (using QuickField form Tera Analysis Ltd). For comparison, variation of the applied voltage (10 V), the electric field and the gradient of the electric field were presented for the proposed structure (Fig. 3a, c) and also in the absence of SU8 layer (Fig. 3b, d). As it can be noticed from Fig. 3c, a very strong variation of the electric field is displayed around the SU8 corners, which proves the enhancement of the DEP force due to the insulating structure. The presence of the SU8 layer makes possible the trapping of the cells that are flowing at a certain distance from the thin electrode plane.

3.2 Simulation of the flow

For the flow inside the microfluidic channel, 2D simulations using the FLUENT™ CFD commercial package were performed in order to compute the isothermal flow of the Newtonian viscous fluids through the analyzed microgeometry. All the simulations considered steady laminar flows, double precision with convergence criteria of 10^{-10} for the unknown functions and the continuity equation being used.

The FLUENT™ code solves the Cauchy equation of motion in which the extra-stress tensor is expressed as a generalized Newtonian model:

$$\rho\left[\frac{\partial \mathbf{u}}{\partial t} + (\mathbf{u} \cdot \nabla)\mathbf{u}\right] = \rho\mathbf{b} - \nabla\rho + 2\nabla(\eta(\dot{\gamma})\mathbf{D}) \tag{1}$$

where ρ is the fluid density (assumed constant), $\boldsymbol{b}$ the specific mass force, $\mathbf{D}$ is the stretching, t is the time, $\mathbf{u}$ is the velocity vector, p is the pressure, and $\eta(\dot{\gamma})$ is the viscosity function, dependent on the shear rate $\dot{\gamma}$. For a Newtonian fluid, $\eta(\dot{\gamma}) = \eta_0$ and Eq. (1) becomes the Navier–Stokes equation, as the viscous term has the simplified expression $\eta_0 \Delta \mathbf{u}$.

The numerical code complements the equation of motion with the mass conservation equation, which, for incompressible fluids, is reduced to $\nabla \cdot \mathbf{u} = 0$. The computational mesh has a total number of cells (NC) of 91,200. The validity of fluid homogeneity and the no-slip boundary condition were assumed for the tested fluid. The governing equations above are solved numerically using a finite
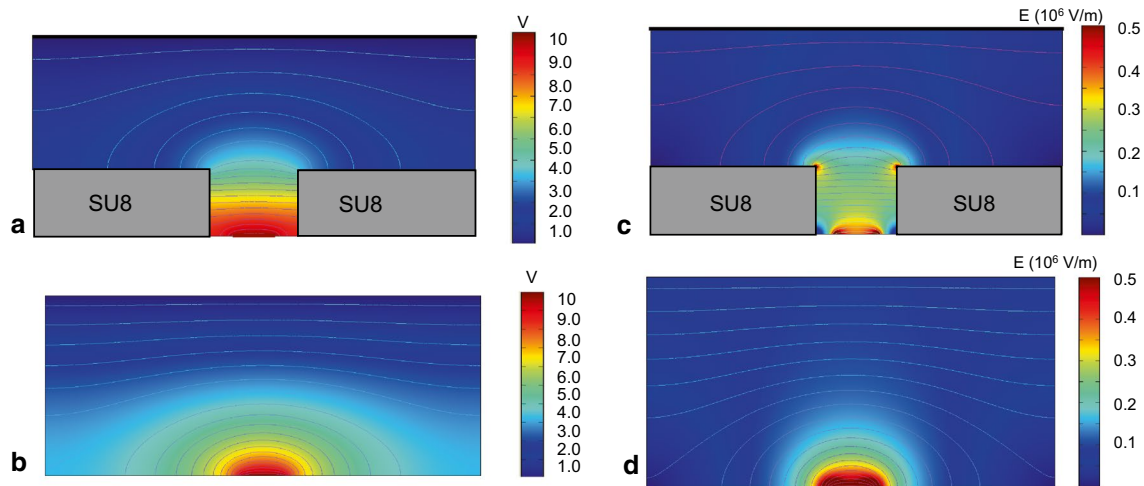
Fig. 3 Variation of the applied voltage (**a**, **b**) electric field (**c**, **d**) across the vertical cross section of the microfluidic channel for the structure with the isolating trap (**a**, **c**) and for a eDEP structure (**b**, **d**)

volume method. In this methodology, the resulting algebraic equations relate the dependent variables (**p**, **u**), which are calculated at the center of the cells forming the computational mesh, to the values in the nearby surrounding cells. Because the interest in this work is for steady-state calculations, the time derivative is discretized with an implicit first-order Euler scheme. The physical properties of the fluid used in the numerical calculations were selected to match water ($\eta_0 = 10^{-3}$ Pa s). The initial conditions to be imposed were as follows: (1) no-slip conditions at the solid walls, (2) at the inlet boundary, located far upstream of the interest regions, a uniform velocity profile and a null stress components were imposed and (3) the imposed outflow boundary conditions involved vanishing stress components, respectively, a constant of gradient pressure (atmospheric pressure) was imposed at the channel outlet.

The numerical results for two different flow regimes ($Q = 1$ μL/min and $Q = 10$ μL/min) are presented in Fig. 4. For this range of velocity of the fluid, no difference in flow behavior and vortex structures can be observed.

3.3 Analytical model

The DEP buffer plays an important role in achieving the desired patterning using electric fields. The DEP force can be defined according to Jones (2005) as:

$$F_{\text{DEP}} = 2\pi a^3 \text{Re}[k]\nabla E^2 \qquad (2)$$

where Re[K] is the real part of the Clausius–Mossotti (CM) factor (also called "polarization factor"):

$$K = \frac{\varepsilon_p^* - \varepsilon_m^*}{\varepsilon_p^* + 2\varepsilon_m^*}, \quad \varepsilon^* = \varepsilon - j\frac{\sigma}{\omega} \qquad (3)$$

where ε_p^* and ε_p^* are the complex permittivity of the particle and medium, respectively. The complex permittivity is a function of permittivity (particle or medium) ε, electrical conductivity σ and the angle frequency ω of the applied electrical field E. Depending on the sign of Re(K) (Albrecht et al. 2005), the particles can be attracted to the regions with high electric field (so-called positive DEP–pDEP) or repelled—negative DEP (nDEP). In our case, the device structure requires the handling of pDEP effect.

Figure 5 gives a schematic of the microfluidic channel with a trapped cell. For low Mach numbers, the fluid can be considered as incompressible and its density is therefore constant. Within the channel, and considering a no-slip boundary condition at the wall, the fluid velocity exhibits a parabolic profile that follows the Poiseuille equation:

$$v(z) = v_{\text{max}}\left(1 - \frac{4z^2}{H^2}\right) \qquad (4)$$

with $v(z)$ the fluid velocity across the channel and oriented along the x-axis, v_{max} its maximal value taken on the channel axis and H the height of the channel. The effect of the width of the channel is neglected. v_{max} is given by the viscosity η_0 and the pressure gradient dp/dx—assumed constant along the channel—as $v_{\text{max}} = (H^2/8\eta_0)dp/dx$. The effect of the traps at the bottom of the channel is difficult to take rigorously into account, especially because of the formation of vortices. Numerical simulations reveal that the laminar flow penetrates into the trap down to a depth δ. Below this penetration depth, the fluid is rotational, but the velocities are small compared to those in the channel. In the lower half of the channel above a trap, we consider that the Poiseuille flow still holds but extends up to $z = H/2 + \delta$. As
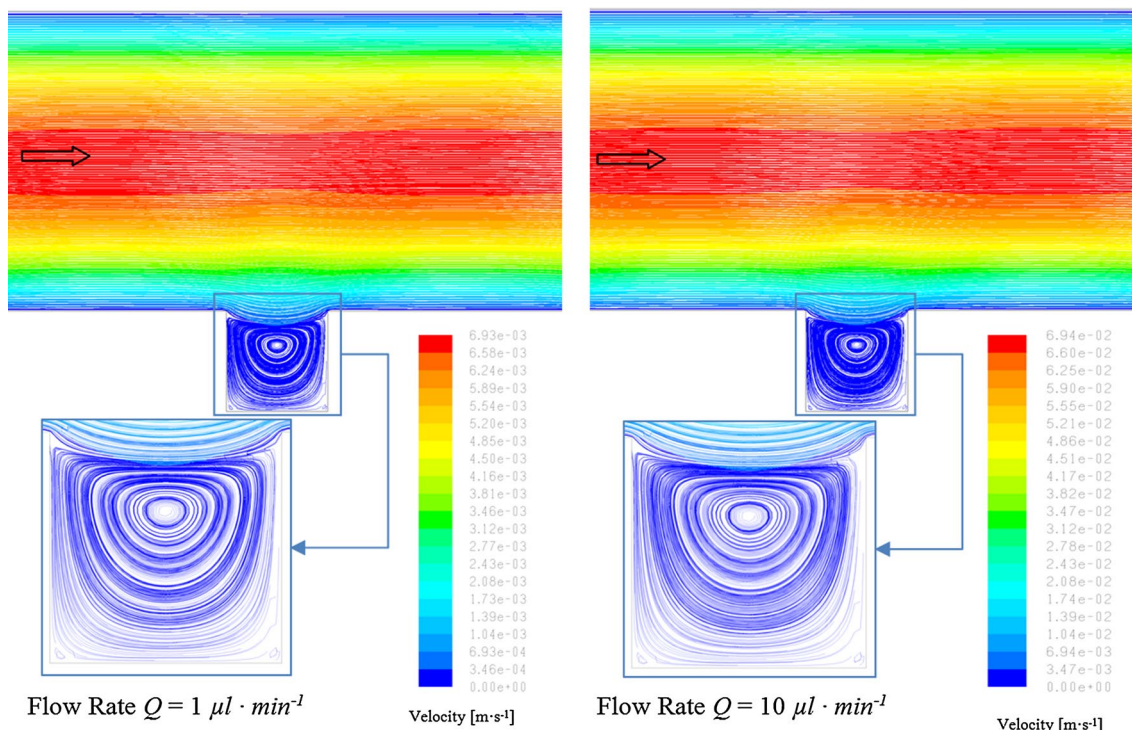
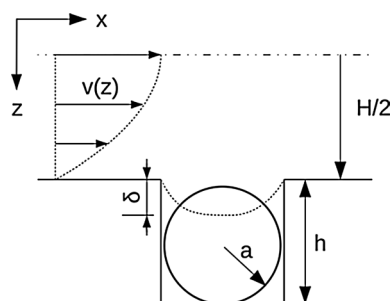**Fig. 4** Velocity in the microfluidic channel for flow rates of 1 and 10 μL/min

Fig. 5 Schematic of the microfluidic channel with a spherical cell placed in a trap. $v(z)$ is the fluid velocity, H the height of the channel, h the height of the trap, a the cell radius and δ the flow penetration depth

a result, using the general Eq. (4), the flow velocity at the top of the trap reads

$$v_{\text{trap}} = v_{\text{max}} \left(1 - \frac{H^2}{(H + 2\delta)^2} \right) \approx 4 v_{\text{max}} \frac{\delta}{H} \tag{5}$$

If the cells are not deeply buried into the trap, which is the case because we need an access to their surface for practical applications, they can feel the effect of the laminar flow in the channel. In particular, if δ is comparable

or greater than $h - 2a$, the difference of pressure between the interior of the trap and the laminar flow may untrap the cells yet maintained by the dielectrophoretic force. According to Bernoulli's principle, a heuristic upper limit for the lift force F_{L} exerted on the cells is:

$$F_L = 2\pi \rho a^2 v_{\text{trap}}^2 \approx 32\pi \rho a^2 \left(\frac{\delta}{H} \right)^2 v_{\text{max}}^2 \tag{6}$$

after replacing v_{trap} by its expression in (4). As we can see, the lift force scales as v_{max}^2 which means that the untrapping rate increases rapidly with the flow velocity in the channel. Equilibrating the lift force (6) with the expression of the dielectrophoretic force F_{DEP} yields a condition on v_{max}— and consequently on the flow rate—to keep the cells in the traps:

$$v_{\text{max}} < \frac{H}{4\delta} \sqrt{\frac{a}{\rho} \text{Re}[K] \nabla E^2} \tag{7}$$

With a typical dielectrophoretic force F_{DEP} of 1 nN for a cell radius of 7.5 μm, a fluid density $\rho = 1000$ kg/m^3, a channel height $H = 60$ μm and a penetration depth $\delta = 5$ μm, the maximal fluid velocity should not exceed 0.16 m/s[1], which corresponds to a flow rate of 15.3 μL/min by integrating (4) over the cross section of the channel. Above this flow rate, the cells start to get untrapped.

4 Results and discussion

4.1 DEP buffer

As previously mentioned, due to thermic considerations (Iliescu et al. 2006b), a low electrical conductivity of the DEP buffer is required, the incremental temperature rise (ΔT) being approximated with the formula (Ramos et al. 1998):

$$\Delta T \approx \frac{\sigma V^2}{k} \tag{8}$$

where V is the applied voltage, and k is the thermal conductivity of the solution. It is well known that the temperature modification can induce changes in the cell physiology (Cima et al. 2013; Lindquist 1986), and even if is not clear what is the range of temperature that induces a cell response, Joule heating effect must be minimized.

As it can be noticed from Eq. (3), the CM factor is a function of frequency. The polarizability of the cell is a cumulative effect of its internal structure (Voldman 2006). As a result, each type of cell presents a "dielectrophoretic signature" (Flanagan et al. 2008), signature that can be characterized using the method proposed by (Voldman et al. 2003). Overall, as a general trend, at low-frequency range (up to 100 kHz), the cells are less polarizable and experience mainly negative DEP (nDEP)—the cells move toward the regions with lower field strength. For high-frequency range (1–100 MHz), the CM factor is given by the difference between cell cytoplasm and media conductivities, resulting in nDEP for highly conductive solutions and positive (pDEP) in low conductive media (Voldman 2006). For the pDEP effect, the cells are moving to the regions with high field strength. In our case, for the trapping of the cells in the designed cage, we need a pDEP effect. As a result, low conductivity of the DEP buffer and an applied voltage in the MHz range are required in our application. In our case, the effect of frequency variation on C3A cells and RBC (suspended in the DEP buffer) is summarized in Table 1.

A second condition related to the buffer is its compatibility in regard to the cells. For this reason, we used the protocol previously described to verify the viability of the C3A cells in the DEP buffer up to 6 h. This duration is sensitively longer than the time frame required for DEP experiment (usually 1 h). The results presented in Fig. 6 shows that at least for the first 3 h, the viability of C3A cells due to the prolonged incubation in the DEP buffer will not be affected considerably. The testing method was described in Sect. 2.2.

4.2 Cell patterning

For the 3D cell patterning, RBCs and the previously mentioned DEP buffer were used. The average diameter

Table 1 Cell response to the frequency of applied electric field

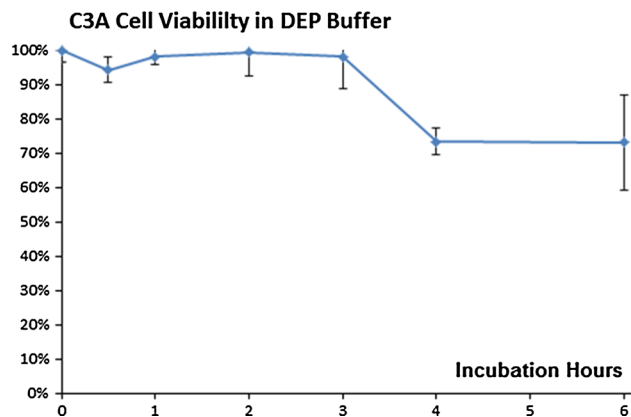| Cell type	DEP effect	Cutting frequency	DEP effect
C3A	nDEP	7.6–8.2 kHz	pDEP
RBC	nDEP	170–180 kHz	pDEP

Fig. 6 Viability of the C3A cells in a DEP buffer

of RBCs is around 6 μm that allowed a 3D organization in the hydrodynamic trap (with a cross section of 20 μm × 20 μm). In order to see the DEP effect, the cells were introduced in the microfluidic device, and at a flow rate of 7 μL/min, the cells were trapped using an applied voltage of 20 V at 1 MHz. The results are illustrated in Fig. 7a. As it can be observed, the cells tend to attach also to the edge of the dielectric layer (where the gradient of electric field is strong). Increasing the flow rate in the microfluidic channel (at 30 μL/min) reduces sensitively the efficiency of trapping (Fig. 7b), only the cell in the vicinity of the SU8 layer is trapped. It is interesting to notice that the cells try to align near the walls of the hydrodynamic trap. This can be explained through the increasing gradient of the electric field on the corner of the CrAu electrode inside the hydrodynamic trap (Fig. 3e) and also due to the adhesion force (Mittal et al. 2007) between cell and SU8. It seems that the cell–substrate (SU8) adhesion force makes difficult a perfect removing of the cells from the SU8 surface by hydrodynamic force (Fig. 7a). Figure 7c shows the trapping of the RBC in a static regime. It can be noticed that in this case, the dipole–dipole interaction can play an important role. The polarization of the cells in the presence of the electric field induces an alignment of the cells in a row. Practically, in the absence of the flow, there is no "packaging" of the cells in the hydrodynamic trap. Figure 7d illustrates the evolution of RBC's trapping in time, suggesting the organization of cells in a 3D structure. This "packaging" of cells can be explained
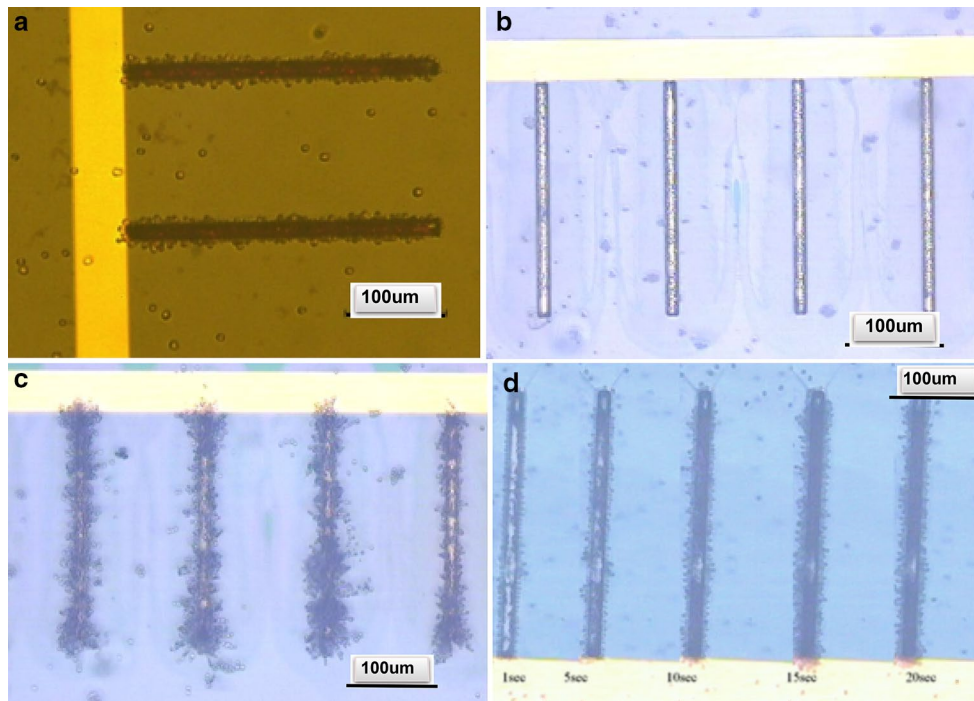
Fig. 7 Cell patterning (RBC's) under different flow conditions (**a**, **b**) and in a static regime (**c**). Evolution of RBC trapping in time (**d**). The width of the hydrodynamic trapped fabricated in the SU8 layer was 20 μm
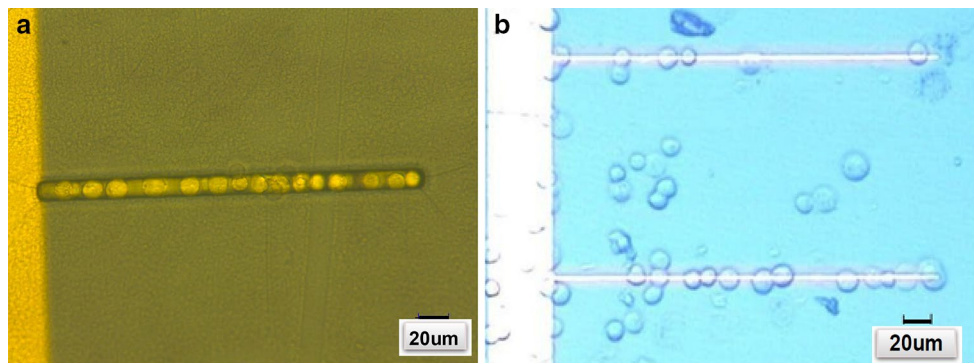

Fig. 8 Cell patterning using C3A cell line: **a** chip with the hydrodynamic trap; **b** trapping in the absence of the hydrodynamic trap (conventional DEP)

through the dipole–dipole interaction. A dielectric particle immersed in a medium and exposed to an electric field acquires a dipole moment. If a number of polarized particles come close enough to each other, they exhibit an attractive dipole–dipole interaction. This phenomenon gives rise to a "pearl chain" alignment of the particles and was previously used for aligning of different waterborne pathogen (Chow and Du 2011) or for electrofusion of lipid containers (Tresset and Iliescu 2007). In our case, confining the RBC's (having a diameter of 6 μm) in a trap having a cross section of 20×20 μm

will enhance the dipole–dipole interaction between cells contributing to their "packaging" in a 3D structure. It is interesting to observe that initially the RBCs are aligning along the edges (where high gradient of the electric field is expressed) and the walls of the trap.

Figure 8 illustrates cell patterning using C3A cell line. In this case, the diameter of the cell is almost identical with the width of the microfluidic trap. For the trapping of C3A cells, similar conditions were used. A very nice organization of the cells in the hydrodynamic trap can be observed (Fig. 8a). For comparison, conventional eDEP method was

used for trapping C3A cells in Fig. 8b. Although the cells align mainly along the electrode, there is a significant number of untrapped cells. Moreover, clustering of the cells due to dipole–dipole interaction is possible (and can be easily noted at the lower electrode). The removal of the untrapped cells and those attached on the bottom surface, using an increased flow rate, becomes more difficult, and the Stokes force also acts on the cells trapped using DEP. In this case, the adhesion force of the cell on the surface, estimated to be between 0.25 and 20 pN depending on the cell type and size (Mittal et al. 2007), is a strong competitor for the DEP force.

5 Conclusions

The paper proposes a DEP method for 2D–3D cell patterning. The architecture of the device consists of DEP–hydrodynamic traps placed in a microfluidic channel. A distinctive feature of the DEP structure is the combination between iDEP and eDEP effects for the generation of the dielectrophoretic force. The cells that are flown through a microfluidic channel are trapped in a hydrodynamic cage using cumulative iDEP–eDEP force, generating 2D or 3D cell patterns according to the ratio between cell dimension and the size of the cage. The trapping method was validated using C3A and red blood cells.

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