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Formation of ordered dolomite in anaerobic photosynthetic biofilms

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8 ABSTRACT

9 Dolomite enabled the preservation of fine microbial textures in some Archean and Proterozoic 10 marine microbialites, but has rarely done so during the Phanerozoic. Here, we report precipitation of dolomite in anoxygenic photosynthetic biofilms grown under chemical conditions relevant for 11 Archean seawater. Ordered dolomite nucleates primarily on the surfaces of photosynthetic cells 12 when manganese(II) is also present, nanocrystals of disordered dolomite form on exopolymeric 13 substances (EPS) in microbial cultures grown either in the dark or without manganese. Dolomite 14 15 nucleation and maturation on different surfaces in photosynthetically active cultures amended with 0.1-1 mM manganese(II) enables the preservation of biofilm textures at scales larger than 16 individual microbial cells. This provides a new model for the preservation of microbial textures 17 by dolomite before the oxygenation of the oceanic photic zone. 18

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20 INTRODUCTION

21 Microbes, exopolymeric substances (EPS) and organic surfaces can mediate the nucleation of protodolomite in some shallow hypersaline and supratidal, evaporation-prone environments 22 (Bontognali et al., 2010; Sanchez-Roman et al., 2008; Shinn et al., 1965) and deep organic-rich 23 sediments colonized by methanogenic Archaea and sulfate reducing bacteria (Kenward et al., 24 2009; Krause et al., 2012). High-Mg phases that form in this manner typically does not preserve 25 26 the primary textures of photosynthetic microbial mats or fine ooid laminae (Bontognali et al., 2010; Spadafora et al., 2009). Thus, these models do not adequately account for the origin of dolomite 27 with a high content of manganese or iron that preserved fine microbial textures in some Archean 28 and Proterozoic microbialites (e.g., Veizer et al., 1990; Wright, 2000; Wright and Tucker, 1990). 29

Here, we use experiments to investigate the preservation potential of benthic 30 photosynthetic communities under chemical conditions relevant for carbonate-depositing marine 31 environments during the Archean and the Proterozoic Eons. Benthic communities before the rise 32 33 of atmospheric oxygen are thought to have relied on sulfide, Fe(II), Mn(II) or hydrogen as electron donors, and the oceans contained ~0.1 mM of iron and manganese (Anbar and Holland, 1992; 34 Beukes, 1987), although more recent estimates suggest lower iron concentrations in the deep ocean 35 (Eroglu et al., 2018). This motivates our use of microbial communities driven by sulfide-based 36 photosynthesis in low-sulfate and 0.1-1 mM manganese solutions that are in equilibrium with a 37 high pCO₂. The culture medium contained less than 0.9 µM sulfate, 8 µM Fe(II) and less than 4 38 39 nM O₂ (Pajusalu et al., 2018). The medium was amended by 0.05 mM H₂S and 0.1-1 mM MnCl₂, and was in equilibrium with an atmosphere of 80% N₂ and 15% CO₂ at pH 7. We describe the 40 mineral composition, nucleation and maturation as a function of the medium composition, 41 microbial activity and types of microbial surfaces. The detection of ordered dolomite as the 42 primary precipitate in our cultures inspires a new model for the formation of early, fabric-retentive 43 44 high-Mg carbonate phases in manganese-rich sediments.

45 **METHODS**

Samples of sediments were retrieved from Fayetteville Green Lake (FGL), New York, by a 46 metallic gravity scoop and photosynthetic biofilms were enriched from these samples. All batch 47 enrichment cultures were grown anaerobically in modified basal FGL (MFGL) medium for two 48 weeks (supplementary information section 1 in the GSA Data Repository¹). The medium was 49 reduced by the addition of 20-50 µM sulfide. Sulfide was also the main electron donor for 50 photosynthesis. The composition of microbial communities in enrichment cultures was analyzed 51 by high-throughput Illumina sequencing. The biofilms did not contain abundant sulfate reducing 52 bacteria. All photosynthesizing cultures, dark culture controls, and sterile controls (without 53 microbes) were incubated for two weeks or longer. The medium was supersaturated with respect 54 to the precipitation of both calcite ($SI_{Calcite} = 0.85$) and dolomite ($SI_{dolomite} = 2.42$) (Supplementary 55 data section 5), but sterile controls did not contain precipitates. Precipitates from all microbial 56 cultures were characterized by X-ray powder diffraction (XRD), scanning electron microscope 57 (SEM) and transmission electron microscopy (TEM) (supplementary information section 2, 3 and 58

59 4). Each experiment was performed at least three times.

60

61 **RESULTS**

The green sulfur bacterium *Chlorobium limicola* was the most abundant microbe and the only photosynthetic organism in the red-brown biofilms. Various anaerobic non-photosynthetic bacteria such as *Geobacter* sp. and *Acholeplasma* sp. were also present (Daye et al., submitted). *Chlorobium* sp. that was incubated in the presence of light was photosynthetically active, and approximately 0.5-1 mm thick biofilms grew in two weeks. The pH in the active cultures increased from 7.24 to 7.33 (0.6%) and the alkalinity decreased from 3958.3 mg/L to 3858.5 mg/L (2%) (DR Section 6).

69 Sterile controls did not contain any detectable precipitates after two weeks, but an ordered dolomite phase was abundant in photosynthetic cultures amended with MnCl₂ (Fig. 1; Fig. DR2; 70 71 Table DR1). This mineral was identified by the $2\Theta^{\circ} = 30.9$ peak corresponding to the basal reflections of (104) crystal plane, the presence of superstructure reflections (hkl), and the match 72 with the dolomite standard (Fig. DR2). The 101 ordering reflection was not visible, but 015, 113 73 and 021 ordering reflections were present (Gregg et al., 2015; Reeder, 1983) (Fig. DR2, FigDR3A, 74 75 B). Dolomite in these cultures encrusted the surfaces of cells (Figure 1, Fig. DR4 A-C) and nucleated on EPS (Fig. DR5). Calcium carbonate and elemental sulfur, which is the expected 76 77 product of sulfide-based photosynthesis by Chlorobi, were the main precipitates in photosynthetic cultures that were not amended with $MnCl_2$ (Fig. DR2; Table DR1). Only small amounts (per ~ 10 78 mg of the analyzed biofilm) of disordered dolomite formed in the dark cultures amended with 79 MnCl₂ or in photosynthesizing cultures that were not amended with MnCl₂ (Figure DR1, Table 80 DR1, Fig. DR4D, DR4E). These minerals nucleated on EPS, and did not encrust cell surfaces 81 (Figs. DR6, DR7). Dark cultures without Mn(II) did not contain any detectable minerals (Fig. 82 DR8). Thus, anaerobic photosynthetic biofilms promoted the precipitation of ordered dolomite 83 only in the presence of light and Mn(II). 84

Dolomite in photosynthesizing, Mn-amended cultures nucleated after one week as an amorphous Ca-Mg carbonate phase with < 5 nm domains (Fig. 1A-D). This phase nucleated on the surfaces of cells and contained manganese (Fig. 1B). After an additional week, the amorphous nuclei matured into < 5 nm rounded nanocrystals in polycrystalline aggregates. The more mature nanocrystals had a uniform lattice fringe and interplanar spacing of 2.19Å corresponding to (11-3) crystal plane of dolomite (Fig. 1E-H). Both the amorphous phase and the more mature 91 nanocrystals were detected primarily on the surfaces of *Chlorobium* sp. cells, recognized by both 92 spinae (Fig. 1A, E), and chlorosomes, i.e., complexes of photosynthetic antennae (Fig. DR4 A-C). 93 Some dolomite grains < 200 nm in diameter were also present on the fibrous EPS (Fig. DR5). Cell 94 surfaces in photosynthesizing cultures that were incubated without Mn(II) were not encrusted by 95 any minerals (Fig. DR4D, E), but < 200 nm wide dolomite grains in these cultures nucleated on 96 EPS and matured into 1 μ m aggregates (Fig. DR6). Calcite nucleated on EPS in dark cultures (Fig. 97 DR7).

98 To further characterize the organic surfaces and microbial textures in anoxygenic photosynthetic biofilms, we imaged horizontal cryo-thin sections of a 400 µm thick biofilm by 99 SEM/EDS (Fig. 2, Figs. DR9,10). The crystal sizes and the extent of mineral coverage depended 100 both on the type and the depth (age) of organic surfaces. At 25 µm below the surface of the biofilm 101 (Fig. 2A), dense aggregates of cells contained $>2 \mu m$ wide aggregates of dolomite crystals. These 102 crystals nucleated on cell surfaces, did not replicate the shapes of individual cells, and were 103 distributed sporadically across the horizontal surface (Fig. 2B). Five to ten µm wide globular 104 dolomite covered more extensive areas in cell-rich zones deeper in the biofilm (Fig. 2E, F) and in 105 older biofilms (Figs. DR9,10). Spatially discontinuous minerals smaller than $< 1 \mu m$ were directly 106 associated with fibrous exopolymeric substances 50 µm below the surface of the two-week old 107 biofilm (Fig. 2C, D). Thus, differences in the shape and size of dolomite crystals reflected the 108 relative densities of cells and EPS. SEM imaging of two-month old biofilms confirmed the 109 persistence of these trends over time (Figs. DR9, 10), and the dolomitizing biofilms did not exhibit 110 111 evidence of extensive degradation or mineral dissolution. Rather, small minerals that nucleated on EPS matured into $\sim 2 \,\mu m$ wide grains. 112

113

114 **DISCUSSION**

115 Active anoxygenic photosynthetic microbes, exopolymeric substances and dissolved manganese stimulate the nucleation and early maturation of ordered dolomite. This process occurs in a well-116 buffered bulk medium and does not depend on large, metabolically induced changes in pH and 117 alkalinity (Bosak and Newman, 2003), but requires the surfaces of photosynthetically active 118 Chlorobium cells as well as EPS (Fig. 1, Figs. DR5-7). Our observations are consistent with a 119 report of dolomite in the zones of some modern microbialites where manganese oxyhydroxydes 120 are brought in close contact with intermediate and reduced sulfur species (Petrash et al., 2016). 121 Differences between the biofilms of various sulfur-cycling or oxygen-producing photosynthetic 122 microbes, such as purple or green sulfur bacteria, purple non-sulfur bacteria and cyanobacteria, 123 are an open question. 124

Previous studies proposed roles for organic compounds in the desolvation of Mg²⁺-water 125 complexes on the growing surfaces of carbonate crystals to form dolomite (Roberts et al., 2013; 126 Zhang et al., 2015). This process, mediated by EPS, carboxylic groups or dead cells under aerobic 127 or anaerobic conditions, likely accounts for the nucleation of poorly crystalline, less ordered high-128 Mg calcite and protodolomite (Bontognali et al., 2014; Roberts et al., 2013). In contrast to many 129 of these studies, we observe ordered dolomite. Given that the intensities of the 015 and 110 130 ordering reflections were not equal, the mineral formed under our experimental conditions was not 131 perfectly ordered (Gregg et al., 2015). This can be attributed primarily to the incorporation of 132 Mn(II) into the dolomite crystal structure. This process should cause the disappearance of the 101 133 ordering reflection and a shift toward larger d-values. 134

The mechanism by which manganese helps nucleate dolomite in systems where the concentration of magnesium exceeds that of calcium is currently not known. Previous studies

showed that Mn(II) inhibits the formation of synthetic dolomite at high temperatures (Lumsden et 137 al., 1989). However, the addition of 0.1-1 mM Mn(II) markedly increased both the abundance and 138 the ordering of dolomite in photosynthetic biofilms (Fig. 1B, F, Fig. DR2). This is consistent with 139 140 the previously noted requirement for Mg(II) in the formation of high-Mn dolomite-type mineral phases in inorganic ternary systems of CaCO₃-MnCO₃-MgCO₃ carbonates (Mucci, 2004). The 141 phases formed in these inorganic experiments did not contain more than 10% Mg (Mucci, 1988), 142 but our observations suggest that the formation of authigenic high-Mg carbonate phases in Mn-143 rich anoxic, microbially colonized marine sediments is also possible. 144

Microbial textures preserved by microcrystalline dolomite or high-Mg calcite record 145 information about microbial processes and interactions with sediments at the sediment-water 146 interface and/or during early burial. Micrometer-size dolomite grains that form in anoxygenic 147 photosynthetic biofilms, the expected products of their maturation/diagenesis, and the reduction 148 of sediment porosity due to the presence of microbial mats may account for the exceptional 149 preservation of textures in some Archean and Proterozoic Mn- or Fe-rich dolomitic microbialites, 150 coated microbial filaments and grains. In all these instances, the fabric-retentive, Mn- or Fe-rich 151 dolomite that contains organic inclusions is interpreted as a product of early diagenesis of a high-152 153 Mg precursor under anaerobic conditions (Simonson and Jarvis, 1993; Wright, 2000; Wright and Tucker, 1990). If high-Mg phases are already abundant at the sediment-water interface, different 154 compositions of Ca and Mg isotopes can be expected (Blättler et al., 2015; Higgins et al., 2018). 155 We hypothesize that fabric-retentive early dolomite that nucleates in benthic communities will 156 appear sediment-buffered (sensu Higgins et al., 2018), perhaps due to the sealing properties of 157 mineral-rich microbial mats. These predictions can be tested by studies of fabric-retentive 158 dolomites. 159

The new model for microbial dolomite formation and the reports of well-preserved textures 160 of Archean dolomitic microbialites inspire the following hypothesis: the distribution of texture-161 preserving dolomite in time and space may reflect the changing abundances of anoxygenic or 162 oxygenic phototrophs as well as concentrations of reduced ions in the pore fluids due to the 163 progressive oxygenation of the marine realm. For example, the abundant limestone in the 164 shallower stromatolitic and oolitic strata of the large carbonate platform in the Neoarchean Ghaap 165 Group of the Campbellrand-Malmani is consistent with the inferred redoxcline on the shelf of the 166 C-M platform based on iron minerals and isotopes (Eroglu et al., 2018). By the same token, the 167 fine dolomitic laminae from the older, Mesoarchean microbialites (Fig. 1, Bosak et al., 2013; Siahi 168 169 et al., 2016) may record active anoxygenic photosynthesis in those benthic environments. These hypotheses can be explored by comparing the preservation potential of different microbial 170 communities under a range of redox and other chemical conditions, the composition of minerals 171 in these communities, as well as the associated calcium and magnesium isotope signatures. 172

174 CONCLUSION

173

175 Nano- and microcrystalline ordered dolomite forms in anaerobic photosynthetic biofilms that use sulfide as the primary electron donor and grow in the presence of 0.1-1 mM Mn(II). Dolomite 176 nucleates on the surfaces of microbial cells or EPS. In cell-rich areas, this mineral matures into 177 178 microcrystalline globular minerals. In contrast, nanocrystalline minerals that nucleate on fibrous extracellular polymeric substances remain finer grained and preserve the porosity of EPS-rich 179 areas. These observations provide a new model by which to account for the nucleation of Mn-rich 180 181 dolomite in pore fluids and for the preservation of microbial textures by fabric-retentive dolomite during the Archean and Proterozoic. 182

183

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189 **REFERENCES CITED**

- Anbar, A. D., and Holland, H. D., 1992, The photochemistry of manganese and the origin of banded iron
 formations: Geochimica and Cosmochimica Acta, v. 56, p. 2595-2603.
- 192 Beukes, N. J., 1987, Facies relations, depositional environments and diagenesis in a major early Proterozoic
- 193 stromatolitic carbonate paltform to basinal sequence, Campbellrand Subgroup, Transvaal Supergroup,
- 194 Southern Africa: Sedimentary Geology, v. 54, p. 1-46.
- Blättler, C. L., Miller, N. R., and Higgins, J. A., 2015, Mg and Ca isotope signatures of authigenic dolomite
 in siliceous deep-sea sediments: Earth and Planetary Science Letters, v. 419, p. 32-42.
- 197 Bontognali, T. R. R., McKenzie, J. A., Warthmann, R. J., and Vasconcelos, C., 2014, Microbially
- influenced formation of Mg-calcite and Ca-dolomite in the presence of exopolymeric substances produced
 by sulphate-reducing bacteria: Terra Nova, v. 26, p. 72–77.
- 200 Bontognali, T. R. R., Vasconcelos, C., Warthmann, R. J., Bernasconi, S. M., Dupraz, C., and McKenzie, J.
- A., 2010, Dolomite formation within microbial mats in the coastal sabkha of Abu Dhabi (United Arab
 Emirates): Sedimentology, v. 57, p. 824-844.
- Bosak, T., Knoll, A. H., and Petroff, A. P., 2013, The meaning of stromatolites: Annual Review of Earth
 and Planetary Sciences, v. 41, p. 21-44.
- Bosak, T., and Newman, D. K., 2003, Microbial nucleation of calcium carbonate in the Precambrian:
 Geology, v. 31, p. 577–580.
- Daye, M., Klepac-Ceraj, V., Pajusalu, M., Rowland, S., Beukes, N., Tamura, N., Fournier, C., and Bosak,
 T., submitted, Light-driven anaerobic microbial oxidation of manganese.
- 209 Eroglu, S., Schoenberg, R., Pascarelli, S., Beukes, N. J., and Swanner, E. D., 2018, Open ocean vs.
- continentally-derived iron cycles along the Neoarchean Campbellrand-Malmani Carbonate platform, South
- 211 Africa: American Journal of Science, v. 318, p. 367-408.
- 212 Gregg, J. M., Bish, D. L., Kaczmarek, S. E., and Machel, H. G., 2015, Mineralogy, nucleation and growth
- of dolomite in the laboratory and sedimentary environment: a review: Sedimentology, v. 62, no. 6, p. 17491769.
- Higgins, J., Blättler, C., Lundstrom, E., Santiago-Ramos, D., Akhtar, A., Ahm, A. C., Bialik, O., Holmden,
- C., Bradbury, H., and Murray, S., 2018, Mineralogy, early marine diagenesis, and the chemistry of shallow-
- 217 water carbonate sediments: Geochimica et Cosmochimica Acta, v. 220, p. 512-534.
- Kenward, P. A., Goldstein, R. H., Gonzalez, L. A., and Roberts, J. A., 2009, Precipitation of low-temperature dolomite from an anaerobic microbial consortium: The role of methanogenic Archaea:
- 220 Geobiology v. 7, p. 556–565.
- 221 Krause, S., Liebetrau, V., Gorb, S., Sanchez-Roman, M., and McKenzie, J. A., 2012, Microbial nucleation
- of Mg-rich dolomite in exopolymeric substances under anoxic modern seawater salinity: New insight into
- 223 an old enigma: Geology, v. 40, p. 587-590.
- Lumsden, D. N., Shipe, L. G., and Lloyd, R. V., 1989, Mineralogy and Mn geochemistry of laboratory-
- synthesized dolomite: Geochimica and Cosmochimica Acta, v. 53, p. 2325-2329.
- 226 Mucci, A., 1988, Manganese uptake during calcite precipitation from seawater: Conditions leading
- to the formation of a pseudokutnahorite: Geochimica et Cosmochimica Acta, v. 52, p. 1859-1868.
- Mucci, A., 2004, The behavior of mixed Ca–Mn carbonates in water and seawater: controls of manganese concentrations in marine porewaters: Aquatic Geochemistry, v. 10, no. 1-2, p. 139-169.
- Pajusalu, M., Borlina, C. S., Seager, S., Ono, S., and Bosak, T., 2018, Open-source sensor for measuring
- oxygen partial pressures below 100 microbars: PloS one, v. 13, no. 11, p. e0206678.

- 232 Petrash, D. A., Lalonde, S. V., González-Arismendi, G., Gordon, R. A., Méndez, J. A., Gingras, M. K., and
- 233 Konhauser, K. O., 2016, Can Mn–S redox cycling drive sedimentary dolomite formation? A hypothesis:
- Chemical Geology, v. 404, p. 27–40. 234
- Reeder, R. J., 1983, Crystal chemistry of the rhombohedral carbonates: Reviews in Mineralogy and 235 Geochemistry, v. 11, p. 1-47. 236
- Roberts, J. A., Kenward, P. A., Fowle, D. A., Goldstein, R. H., Gonzalez, L. A., and Moore, D. S., 2013, 237
- Surface chemistry allows for abiotic precipitation of dolomite at low temperature: Proceedings of the 238 239 National Academy of Sciences of the United States of America, v. 110, p. 14540–14545.
- Sanchez-Roman, M., Vasconcelos, C., Zenobi, R., and Rivadeneyra, M. A., 2008, Aerobic microbial 240
- dolomite at the nanometer scale: Implications for the geologic record: Geology, v. 36, p. 879–882. 241
- Shinn, E. A., Ginsburg, R. N., and Lloyd, R. M., 1965, Recent supratidal dolomite from Andros Island 242 243 Bahamas.
- 244 Siahi, M., Hofmann, A., Hegner, E., and Master, S., 2016, Sedimentology and facies analysis of
- Mesoarchaean stromatolitic carbonate rocks of the Pongola Supergroup, South Africa: Precambrian 245 Research, v. 278, p. 244-264. 246
- 247 Simonson, B. M., and Jarvis, D. G., 1993, Microfabrics of oolites and pisolites in the early Precambrian
- Carawine Dolomite of Western Australia, in Rezak, R., and Lavoie, D., eds., Carbonate microfabrics: 248 249 Heidelberg, Springer-Verlag, p. 227–237.
- Spadafora, A., Perri, E., McKenzie, J. A., and Vasconcelos, C., 2009, Microbial biomineralization processes 250 251 forming modern Ca:Mg carbonate stromatolites: Sedimentology, v. 57, p. 27-40.
- Veizer, J., Clayton, R. N., Hinton, R. W., Brunn, V. V., Mason, T. R., Buck, S. G., and Hoefs, J., 1990, 252
- Geochemistry of Precambrian carbonates: 3-shelf seas and non-marine environments of the Archean: 253 254 Geochimica and Cosmochimica Acta, v. 54, p. 2717-2729.
- Wright, D. T., 2000, Benthic microbial communities and dolomite formation in marine and lacustrine 255 environments-a new dolomite model: Society for Sedimentary Geology Special Publication, v. 66, p. 7-256 257 20.
- 258 Wright, V. P., and Tucker, M. E., 1990, Carbonate Sedimentology, London, Blackwell Scientific 259 Publications, 482 p.:
- Zhang, F., Xu, H., Shelobolina, E. S., Konishi, H., Converse, B., Shen, Z., and Roden, E. E., 2015, The 260
- catalytic effect of bound extracellular polymeric substances excreted by anaerobic microorganisms on Ca-261
- 262 Mg carbonate precipitation: Implications for the dolomite problem: American Mineralogist, v. 100, p. 483-
- 494. 263

264 **Figure captions**

- Figure 1. Transmission electron micrographs (TEMs) of cells incubated in the light with 1 mM 265 Mn(II) for two weeks. Darker areas in A and E are covered by minerals, long protuberances are 266
- Chlorobium sp. spinae. A: TEM at 200 kV of a Chlorobium limicola cell. Dotted rectangles mark 267
- 268 the areas analyzed by EDS, high resolution TEM (HRTEM) and selected area X-ray diffraction (SAED). B: EDS spectrum of the area shown in A. C: HRTEM shows only an amorphous phase. 269
- D: Diffraction pattern of the same amorphous phase, scale bar: 5 1/nm. E: TEM at 200 kV of a cell 270
- with spinae. Dotted rectangle marks the area analyzed by EDS HRTEM and SAED. F: EDS 271
- spectrum of the area shown in E. G: HRTEM shows the lattice fringe of dolomite nanocrystals at 272
- the cell surface. H: Diffraction pattern corresponds to dolomite with (104) superstructure 273
- 274 diffraction, scale bar: 5 1/nm.
- 275
- 276 Figure 2. Cryo-SEM of horizontal sections through a two-week old biofilm. A: Cell-rich area 25 µm below the surface. Bright granules are dolomite. The dashed rectangle shows the area 277
- magnified in B. B: White arrows point to bacterial cells; the two bright globules are dolomite. C:
- 278 Porous EPS 50 µm below the surface. The dashed rectangle outlines the area magnified in D. D. 279
- EPS network with sporadically distributed brighter areas that contain minerals. E: Cell-rich area 280

- 80 µm below the surface. The area contains numerous dolomite grains. The dashed rectangle
 outlines the area magnified in F. F: White arrows point to bacterial cells; the very bright round
 aggregates are dolomite crystals.
- 284

¹GSA Data Repository item 201Xxxx, [materials and methods for media and enrichment

- protocol, X-ray diffraction, scanning electron microscopy, transmission electron microscopy and
 saturation indices, as well as figures (DR1, DR2, DR3, DR4, DR5, DR6) and table DR1] is
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