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Citation: Roussel, Anaïs et al. "Biomarker stratigraphy in the Athel Trough of the South Oman Salt Basin at the Ediacaran-Cambrian Boundary." *Geobiology* 18, 6 (July 2020): 663– 681. © 2020 John Wiley & Sons Ltd

As Published: <http://dx.doi.org/10.1111/gbi.12407>

Publisher: Wiley

Persistent URL: <https://hdl.handle.net/1721.1/133138>

Version: Author's final manuscript: final author's manuscript post peer review, without publisher's formatting or copy editing

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1 **Biomarker stratigraphy in the Athel Trough of the South Oman Salt Basin at the**
2 **Ediacaran-Cambrian Boundary**

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18 **Highlights:**

19

20 1) High resolution biomarker chemostratigraphy reveals microbial community and redox
21 variability through the Ediacaran-Cambrian transition at the Athel Trough of the South Oman
22 Salt Basin

23 2) Low concentrations of aromatic carotenoids from GSB and PSB suggest weak or episodic
24 photic zone euxinia

25 3) Unusual patterns of aromatic carotenoids, distinct from those most commonly seen in most
26 Phanerozoic sediments and oils, are consistent with contributions from cyanobacteria

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35 **Abstract**

36 The South Oman Salt Basin (SOSB) has been studied extensively for knowledge concerning
37 the habitat of the enigmatic Ediacaran-Cambrian oils that are produced from that region.
38 Geological, geochemical, geophysical and geochronological investigations have all
39 contributed to improved understanding of the range of late Neoproterozoic depositional
40 environments recorded there. Of particular interest has been the deep Athel depocenter within
41 the SOSB that features a silica rich interval known as the Al Shomou Member or Athel
42 Silicilyte and the co-eval A4 carbonate evaporite sequence that straddles the Ediacaran-
43 Cambrian boundary. The deep basin has been suggested to be anoxic and euxinic based on
44 studies of sulfur isotopes, trace metal distributions and other proxies. Organic geochemistry
45 has provided some clues concerning aspects of the depositional environments and microbial
46 communities prevailing during this interval. However, ambiguities remain including a paucity
47 of convincing molecular evidence for euxinia in the photic zone of the basin. Here, we present
48 a comprehensive study of biomarker hydrocarbons, including steroids, triterpenoids and
49 carotenoids. Among the compounds detected is a distinctive array of aromatic carotenoids.
50 Relatively low abundances of monoaromatic carotenoids, such as chlorobactane, okenane and
51 β -isorenieratane, suggest the possibility of transient photic zone euxinia with a shallow
52 chemocline or, perhaps, exogenous inputs from microbial mats. However, it is the dominance
53 of renieratane and renierapurpurane over isorenieratane in diaromatic carotenoids and their
54 association with abundant C₃₈ and C₃₉ carotenoids that identifies cyanobacteria as major
55 contributors to the inventory of carotenoids. Our results, based on multiple lines of molecular
56 evidence and statistical analysis, also suggest that the Athel Silicilyte was biogeochemically
57 distinct from the other units of the Ara Group. Overall, our study has important implications
58 for understanding other late Neoproterozoic depositional environments.

59

60 **Introduction**

61 Over recent decades, the South Oman Salt Basin (SOSB) has attracted attention owing to the
62 presence of commercial oil deposits derived from Neoproterozoic-Cambrian sedimentary
63 sequences. The very unusual biomarker characteristics of the early SOSB oil discoveries
64 enabled their attribution to units within the Huqf Supergroup (Grantham et al., 1988a; Terken
65 et al., 2001), a thick sequence of rocks that covers most of the ~635 to ~541 Ma Ediacaran
66 Period. A sequence within the Huqf Supergroup, the Ara Group is of special biogeochemical
67 and paleontological interest as these sediments span the Ediacaran-Cambrian boundary.
68 Where represented by a shallow carbonate platform, the Ara sequence is comprised of up to

69 seven carbonate-evaporite cycles denoted A0 to A6 (Grotzinger and Al-Rawahi, 2014).
70 Rifting at ~543 Ma during the deposition of the fourth of these cycles, denoted as the A4,
71 resulted in a package of sediments including more clastic material deposited in the deeper
72 Athel sub-basin. It is composed of the Al Shomou Member of the Athel Formation or Athel
73 Silicilyte (Al Rajaibi et al., 2015; Amthor et al., 2015) and is bracketed by the U Shale
74 Member below and the Thuleilat Shale Member above. All three sedimentary units were
75 deposited under anoxic conditions. The sequences and composition of evaporites show that
76 the waters of the SOSB were rich in sulfate while the prevalence of pyrite combined with
77 trace metal and biomarker data suggest that euxinic conditions prevailed in the water column
78 during deposition of the deep basinal facies (Grantham et al., 1988a; Grosjean et al., 2009;
79 Schroder and Grotzinger, 2007; Stolper et al., 2017). However, none of these previous studies
80 reported convincing evidence for photic zone euxinia (PZE) during deposition of the Athel
81 Formation.

82

83 Photic zone euxinia is the phenomenon where sulfidic waters protrude into the euphotic zone
84 thereby permitting anoxygenic photosynthesis by the green sulfur bacteria (GSB) and purple
85 sulfur bacteria (PSB) and giving rise to an array of distinctive aromatic carotenoid and
86 chlorophyll pigments. The aromatic carotenoids, which are widely observed in sediments
87 deposited over the Phanerozoic Eon, and especially during oceanic anoxic events (French et
88 al., 2015; Whiteside and Grice, 2016), are rarely reported in Pre-Cambrian strata.

89 Nevertheless, the oldest aromatic carotenoids were discovered in the a lagoonal facies of the
90 Barney Creek Formation at 1.6 Ga in the Paleoproterozoic Era (Brocks et al., 2005).

91 Additionally, a few studies have reported aromatic carotenoids in the Neoproterozoic Era
92 (Bhattacharya et al., 2017; French et al., 2015). Recently, two aromatic carotenoids (i.e.,
93 chlorobactane, okenane) were detected for the first time in the SOSB (French et al., 2015),
94 thereby prompting further detailed investigation on the evidence for PZE in this depositional
95 system.

96

97 A variety of other hydrocarbon biomarkers have been commonly applied in the study of
98 microbial communities in modern and ancient settings (Peters et al., 2005). This is based on
99 the idea that a certain subset of compounds is synthesized by a limited number of specific
100 organisms and under particular environmental conditions (Table 1). Most commonly, steranes
101 and hopanes are the fossilized remains of sterols and hopanoids synthesized by eukaryotes
102 and bacteria, respectively. Another common sedimentary biomarker is gammacerane,

103 regarded as a diagenetic product of tetrahymanol derived from protists living at redox
104 boundaries in stratified water columns (Harvey and Mcmanus, 1991). In this study, we
105 measured patterns of biomarkers aiming to shed further light on the dynamic evolution of
106 environments across the Ediacaran-Cambrian boundary in deep basinal sediments of the
107 SOSB and on the depositional conditions and microbial processes taking place during
108 deposition of the enigmatic Athel Silicilyte.

109

110 The Athel Silicilyte, which was deposited at the same time as the A4C unit on the carbonate
111 platform, provides an archive to investigate the characteristics of deep-water environments in
112 relation to shallow water analogues. Distinct from the carbonate platform, which can be
113 biased by benthic microbial mats or terrigenous inputs, the Silicilyte deposits in the Athel
114 deep basin record the signal best representative of the microbial communities in the water
115 column and is ideal for testifying the presence of anoxygenic phototrophs at the Ediacaran-
116 Cambrian transition. While this study only focuses on one well (Athel-1), our regional data
117 extrapolation is based on comparison with samples from shallow carbonate platforms,
118 previously published oils and rocks (Grosjean et al., 2009), and another section further west
119 (Marmul Northwest-7; Figure 1) that were studied in earlier research (Stolper et al., 2017).

120

121 2. Regional Geological Setting

122 The stratigraphic sequence of South Oman is characterized by an early Neoproterozoic non-
123 sedimentary basement overlain unconformably by the Huqf Supergroup dated as
124 Neoproterozoic to earliest Cambrian. The Huqf Supergroup is distinguished from the
125 overlying Haima Supergroup of middle Cambrian to Silurian age by a regional unconformity.
126 It is now recognized that the Huqf Supergroup holds valuable information on petroleum
127 formation, basin development, climate conditions and evolution of the biosphere over the
128 Neoproterozoic-early Cambrian (Allen, 2007; Bowring et al., 2007).

129

130 The Huqf Supergroup is subdivided into three distinct sedimentary units. In an ascending
131 order, they are the Abu Mahara Group, the Nafun Group, and the Ara Group (Figure 1). The
132 lowermost Abu Mahara Group, time equivalent to the Sturtian and Marinoan Snowball Earth
133 events, features glacial clastics capped by a carbonate layer that signals post-glacial
134 transgression. The Nafun group above the cap carbonate consists of two siliciclastic to
135 carbonate cycles (Aitken, 1966). The Nafun Group witnessed the broadening of the subsiding
136 basin, possibly related to the post-rift thermal contraction of the lithosphere (Kapellos et al.,

137 1992) or the subduction of oceanic lithosphere into the mantle (Grotzinger et al., 2002;
138 Gurnis, 1988, 1992). The Nafun Group-Ara Group boundary marks the uplift of basin
139 basement blocks and the subsequent segmentation of the broader Nafun basin into three fault-
140 bounded sub-basins, namely as the South Oman Salt Basin (SOSB), the Ghaba Salt Basin,
141 and the Fahud Salt Basin.

142

143 The ensuing Ara Group records further segmentation of each sub-basin with polarized
144 sedimentation, a shift toward arid climatic conditions and accumulation of evaporative salt
145 sequences. Within the SOSB, further segmentation subdivided the SOSB into three domains,
146 namely as the Northern Carbonate Domain, the Athel deep sub-sub-basin (a.k.a. Athel
147 Trough), and the Southern Carbonate Domain (Figure 1), related to uneven basement
148 subsidence rates (Grotzinger et al., 2002). The stratigraphy of the two carbonate domains
149 comprises up to seven carbonate-evaporite cycles designated as A0 to A6 (Grotzinger and Al-
150 Rawahi, 2014). Each cycle starts with deposition of evaporites at lowstand and is topped by
151 carbonates formed during highstand consequent from a renewed infusion of seawater. Such
152 water depth and salinity cycles are explained by a tectonic-eustatic model with periodic
153 connections to the open ocean (Allen, 2007; Amthor et al., 2002). The stratigraphy of the Ara
154 Group in the Athel trough displays the majority of carbonate-evaporite cycles present on the
155 carbonate platforms and, additionally, includes a thick and unique unit rich in clays and silica
156 (Figure 1). This layer is equivalent in time to the A4 carbonate deposition (A4C) and is
157 bracketed by the A4 and A5 evaporite cycles (A4E and A5E). Since the lithologies of this
158 layer are heterogeneous, it is further subdivided into the U Shale Member, the Al Shomou
159 Silicilyte Member, and the Thuleilat Shale Member (Forbes et al., 2010), representing a
160 transgression-regression sequence with sea level peaked during the Al Shomou Silicilyte
161 Member. Microstructural studies have revealed micron scaled lamination in the Silicilyte
162 matrix intercalated with organic matter-rich and finely crystalline quartz-rich layers. Other
163 minerals of diagenetic origin include pyrite, apatite, dolomite, magnesite and barite (Amthor
164 et al., 2015). All three units are estimated to be deposited under a minimum water depth of
165 ~200m (Amthor et al., 1998). This implies that the Athel deep basin continued subsiding
166 while the carbonate platforms remained uplifted during A4 time. Although the mechanisms
167 for such disparate eustatic rebound are not well understood, it is clear that the depositional
168 environment of the SOSB was not uniform throughout the Ara Group.

169

170 Pertinent to the molecular analyses discussed below, carbon and sulfur isotopic analyses of
171 SOSB sediments speak to the ocean becoming increasingly oxygenated during deposition of
172 the Huqf Supergroup (Canfield et al., 2010; Fike et al., 2006; Leavitt et al., 2013; Sim et al.,
173 2011). Sulfur isotopic data, in particular, show a progressive increase in $\Delta\delta^{34}\text{S}$ throughout
174 deposition of the Nafun Group to values sometimes in excess of 46‰ by the end of Buah
175 time. This value approaches the maximal fractionation expressed by bacterial sulfate
176 reduction, indicates the presence of a significant marine sulfate inventory and implies a
177 substantial pool of H_2S persistent in the SOSB (Fike and Grotzinger, 2008, 2010). Above the
178 Nafun Group, paired sulfate–pyrite $\delta^{34}\text{S}$ data for sediments of the Ara evaporite cycles
179 deposited just prior to, and through the Ediacaran-Cambrian Boundary, show anomalous
180 enrichments in both $\delta^{34}\text{S}_{\text{sulfate}}$ and $\delta^{34}\text{S}_{\text{pyrite}}$ interpreted as an increase in pyrite burial
181 potentially driven by a substantial iron supply, supporting possible ferruginous ocean
182 conditions at the Ediacaran-Cambrian transition (Fike & Grotzinger, 2010). Thus, the sulfur
183 isotopic data for the SOSB indicate a very active biogeochemical sulfur cycle with substantial
184 pyrite deposition but also the possibility of supporting the activities of anoxygenic
185 phototrophs that utilize sulfide as an electron donor for photosynthetic carbon fixation.

186

187 **2. Materials and Methods**

188 2.1. Samples and Preparation

189 We focus primarily on the A4-equivalent units of the Athel-1 well drilled in the deep basin
190 near the Eastern Flank Platform in order to better constrain depositional environment of the
191 Silicilyte and bounding shale formations in the deep basin (Figure 1). The Silicilyte Member
192 here is ~300m thick, finely laminated, rich in organic carbon and composed of 80-90%
193 microcrystalline quartz. In this well it is bounded by the Thuleilat Shale member above
194 (~1125m) and the U Shale Member below (~1425m). It was proposed that the water depth of
195 the deep basin increased to the maximum of ~100200m (Schröder et al., 2003) or >200m
196 (Amthor et al., 1998; Amthor et al., 2002) during deposition of the Silicilyte. The base of the
197 Silicilyte was deposited during the early Cambrian ($541\pm 0.13\text{Ma}$), when the major
198 diversification of the metazoan phyla took place (Erwin, 2015).

199

200 In this study, we examined a total of twenty-one cleaned cuttings samples from the Athel-1
201 well covering a depth from 1051m to 1602m, with three samples from the Thuleilat Shale,
202 eleven from the Silicilyte and seven from the U Shale. For comparison, we also included eight
203 additional samples from the Shuram and Buah formations from Athel-1 well and the A4

204 carbonate from Birba-3 well (Figure 1). Specifically, samples from the Silicilyte were
205 selected to ensure thorough coverage of the unit. Furthermore, the Silicilyte Member is
206 divided into upper and lower units, based on gamma ray data. Cuttings samples (1–2 g) were
207 placed in a puck mill and powdered in a shatterbox. The mill was cleaned sequentially with
208 fired sand, DI water, methanol and DCM between samples, to avoid cross-contamination.

209

210 2.2. Bulk OC and $\delta^{13}\text{C}$ Measurements

211 About 0.3g of each ground sample was decarbonated using 1N HCl. The supernatant was
212 decanted after centrifuging. The precipitated particles were further rinsed several times with
213 Millipore water to remove residual acid. The decarbonated residuals were dried at 60°C,
214 followed by grinding by hand with a mortar and pestle. The samples were weighted into tin
215 capsules and introduced to an autosampler. Total organic carbon, carbon isotopes and
216 nitrogen isotopes were measured on an elemental analyzer (EA) coupled to an isotope ratio
217 mass spectrometer (IRMS). Instrumental precision based on duplicate measurements of
218 USGS 40 standard was 0.11‰ for $\delta^{13}\text{C}$ values.

219

220 2.3. Biomarker Analysis

221 About 0.7g of each ground sample was extracted with DCM:MeOH (9:1), using a Dionex
222 Accelerator Solvent Extractor 350 (ASE), operated at 100°C. Total solvent extracts were re-
223 dissolved in ~1ml pentane to induce asphaltene precipitation, which was removed after
224 centrifugation. Elemental sulfur was removed by the addition of activated copper shot which
225 was then removed by centrifugation. The resultant extracts were dried to <100ul and
226 fractionated on a silica gel column. Hexane : dichloromethane (4:1; v:v) and dichloromethane
227 : methanol (4:1;v:v) were used to elute the non-polar and polar fractions, sequentially.

228

229 The non-polar fraction was analyzed on an Agilent gas chromatograph (GC, 7890B) coupled
230 to an Agilent triple quadrupole MS (QQQ, 7010A) operated in multiple reaction monitoring
231 (MRM) mode (Table S1). A multi-mode injector with an initial injection temperature of 45°C
232 was ramped at a rate of 720°/min to 340°C. A DB-5MS column (60m×250µm×0.25µm) was
233 installed with the GC oven temperature held isothermally at 40° for 2 mins, ramped to 320°C
234 at a rate of 4°/min, and then held at this temperature for 22 mins. The transfer line and source
235 temperatures were set at 300° and 250°, respectively. The electron energy was set at 50eV to
236 ensure a stronger signal for the precursor-product transitions. All biomarker data were
237 processed using MassHunter QQQ software. Each compound was identified and integrated

238 under MRM mode within a narrow retention time window (0.5min). For quality control, four
239 samples were randomly selected for duplicate measurements and error bars indicated in
240 figures represent standard deviation.

241

242 2.4. Statistical Analysis

243 Two-tailed unequal variance *t*-tests were performed to determine significant differences
244 between independent groups of variables, while one-tailed unequal variance *t*-tests were
245 applied when comparing for higher or lower values (e.g., Thuleilat vs Silicilyte). A *p* value of
246 < 0.05 was considered as significant, while $0.05 < p < 0.1$ was considered as weakly
247 significant, however, caution was exercised when a group contains limited number of
248 samples. Principal components analysis (PCA) was performed to discriminate controlling
249 variables linked with bulk and biomarker patterns. Cluster Analysis was performed to
250 categorize individual samples into groups based on proxy similarities. The Thuleilat Shale,
251 upper Silicilyte Member, lower Silicilyte Member and U Shale were treated separately in the
252 comparison.

253

254 3. Results and Discussion

255 3.1. Maturity and the potential for oil migration

256 The thermal maturity of these three units (the U Shale Member, the Al Shomou Silicilyte
257 Member, and the Thuleilat Shale Member) from Athel-1 well were assessed earlier using
258 composite cuttings samples, where Rock-Eval T_{max} values range from 419 to 430 °C
259 (Grosjean et al., 2009). The molecularly-based thermal maturity parameters presented here
260 confirm a relatively mild thermal history.

261

262 The C_{27} hopane T_m , with its $17\beta(H)$ configuration $17\beta(H)$ -22,29,30-trisnorhopane, is
263 thermodynamically less stable than the neohopane T_s $18\alpha(H)$ -22,29,30-trisnorneohopane,
264 hence, the $T_s/(T_s+T_m)$ ratio is widely observed to gradually increase with increasing burial
265 depth and maturation process (Peters et al., 2005). The ratios of $T_s/(T_s+T_m)$ varied between
266 0.18 and 0.29 (Table 2; Figure S1), lower and narrower compared to those reported
267 previously for the Marmul Northwest-7 well (Stolper et al., 2017). Other maturity indices
268 further constrain the samples being placed in the early oil generation window. For example,
269 the $C_{31}H$ $S/(S+R)$ ratios are based on the thermally-driven isomerization at C-22 position of
270 hopane side-chain during maturation. The $C_{31}H$ $S/(S+R)$ ratios from this study are broadly
271 constant and fell in a range between 0.51 and 0.57 (Figures S1, S3), placing samples in the

272 early (0.50 – 0.54) and the earliest stage of main oil generation (0.57 – 0.64) windows (Seifert
273 and Moldowan, 1980). Additionally, the comparable isomer ratios for the steranes,
274 exemplified by the $C_{29}S$ $\alpha\alpha\alpha S/(S+R)$, which is based on the isomerization of C_{29} $5\alpha,14\alpha,17\alpha$
275 steranes at C-20 (R, S), ranged between 0.46 and 0.52 (Table 2; Figures S2, S3), lower than
276 the maximum equilibrium value of 0.52-0.55 (Seifert and Moldowan, 1986). So, together,
277 these parameters suggest that the entire section is in the earliest stages of oil generation.

278
279 In contrast, some parameters display distinct ranges that do not match with the earliest oil
280 generation window or expected maturation trend down section (e.g., $Ts/(Ts+Tm)$), suggesting
281 that other factors, including organic facies and sediment lithologies (Moldowan et al., 1986;
282 Seifert and Moldowan, 1986), are playing a prominent role in modulating thermal-maturity-
283 related proxies. For example, the $C_{30}H$ $\beta\alpha/(\alpha\beta+\beta\alpha)$ ratio, otherwise known as
284 moretane/(moretane+hopane) ratio, involves conversion of $17\beta,21\alpha$ -hopanes (moretanes) to
285 $17\alpha,21\beta$ -hopanes at elevated temperatures, resulting in an equilibrium mixture of $\beta\alpha$ -hopanes
286 over $\alpha\beta$ -hopanes at a ratio ~ 0.05 (Peters et al., 2005). Therefore, samples pre-equilibrium are
287 expected to have $C_{30}H$ $\beta\alpha/(\alpha\beta+\beta\alpha)$ ratios >0.05 . In the present study, $C_{30}H$ $\beta\alpha/(\alpha\beta+\beta\alpha)$ ratios
288 exhibited a narrow range between 0.03 and 0.04, even lower than the suggested equilibrium
289 ratio of 0.05 (Figure S3). Additionally, remarkably low proportions of diasteranes (0.02 –
290 0.07), defined as C_{29} dia/(dia+reg), are observed in the Silicilyte Member as compared to the
291 bounding clastic units (Figure S3) and affirm the idea that lithological features, and especially
292 low clay contents (French et al., 2012; van Kaam-Peters et al., 1998), as well as redox
293 conditions experienced during diagenesis, influence processes involved in the epimerization
294 and rearrangement reactions that underpin these maturity proxies (Peters et al., 2005).

295
296 Overall, stratigraphic variability in thermal maturity proxies is observed down section.
297 Notably, the much lower $C_{31}H$ $S/(S+R)$ values (0.51-0.54) are observed in the Thuleilat Shale
298 than the underlying units (0.55-0.57; $p < 0.05$), while the lowest $Ts/(Ts+Tm)$ values are found
299 in the U Shale (Figures S1, S3). In addition to the lowest C_{29} dia/(dia+reg) sterane ratios in the
300 Silicilyte Member, $C_{29}S$ $\alpha\alpha\alpha S/(S+R)$ ratios decrease down section, opposite to the normal
301 expectation. Stratigraphically, stratigraphic heterogeneity in thermal maturity indices and
302 biomarker distributions (discussed below), in concert with the presence of impermeable
303 evaporite layers (A4 and A5) strongly argues against oil migration and post-depositional
304 overprinting. Furthermore, mild Rock-Eval T_{max} values (419 – 430 °C) indicate immature
305 source rocks in the Athel-1 well (Grosjean et al., 2009). Overall, it is very likely that oils

306 found in the Silicilyte Member were self-sources and experienced minimal migration,
307 validating paleoenvironmental reconstruction based on organic proxies.

308

309 3.2. Microbial community changes during basin development

310 Organic carbon contents ranged between 1.53% and 8.89%. Other than some higher values
311 found in the Thuleilat Shale and U Shale, no obvious trend was observed down-core, which
312 precludes the dependency of biomarker trends on organic carbon contents. Carbon isotopic
313 values ($\delta^{13}\text{C}$) for bulk organic carbon spanned a large range from -34.2‰ to -40.6‰. Such
314 low $\delta^{13}\text{C}$ values are typical for Huqf sediments and oils (Grantham et al., 1988b; Grosjean et
315 al., 2009), nevertheless, significant differences in $\delta^{13}\text{C}$ values were not observed between any
316 two groups of samples (Table 2). Consistent with previous studies, biomarker parameters
317 display significant variations of microbial communities (Grosjean et al., 2009; Love et al.,
318 2009; Stolper et al., 2017). Hopanes and steranes are the two major groups of compounds
319 typically attributed to bacteria and eukaryotes, respectively (Damsté and Koopmans, 1997;
320 Peters et al., 2005), and therefore the sterane/hopane ratio is indicative of relative community
321 density of eukaryotes versus bacteria. At the early stage of U Shale deposition, algae (i.e.
322 eukaryotes) were a significant component of the microbiota as indicated by sterane/hopane
323 ratios in excess of 1 (Figure 2). The deposition of the U Shale records a period of
324 transgression, coupled with fresh intrusions of oxygenated seawater and tectonic subsidence
325 in the deep basin (Grotzinger and Al-Rawahi, 2014). Sterane/hopane ratios >1 are typical of
326 crude oils and source rocks deposited in open marine environments (Moldowan et al., 1985;
327 Schiefelbein et al., 1999). However, this contrasts with sterane/hopane ratios as low as 0.01 in
328 contemporaneous oligotrophic settings deposited in Baltica (Pehr et al., 2018), suggesting that
329 other factors including the availability of nutrients may also have been important in
330 determining the relative proportion of eukaryotes and bacteria in marine environments at this
331 time.

332

333 Consistent with elevated sterane/hopane ratios during transgression, the highest levels of C_{30}
334 steranes, including 24-*n*-propylcholestane ($\text{C}_{30}\text{-npc}$) and 24-*i*-propylcholestane ($\text{C}_{30}\text{-ipc}$), were
335 also observed in the U Shale (2.0 – 3.4%) (Figure 3). The presence of $\text{C}_{30}\text{-ipc}$ has been
336 attributed to the emergence of demosponges in the Neoproterozoic (Love et al., 2009).
337 Although $\text{C}_{30}\text{-npc}$ found in Devonian and younger samples has been attributed to the
338 emergence of marine pelagophyte algae since mid-Paleozoic (Moldowan et al., 1984), its
339 occurrence in Neoproterozoic and early Paleozoic strata has been ascribed to other sources,

340 possibly including foraminifera and sponges (Gold et al., 2016; Grabenstatter et al., 2013).
341 More recently, another rare C₃₀ steroid 26-methylstigmastane, and its co-occurrence with C₃₀-
342 ipc and C₃₀-npc in suites of Neoproterozoic samples attribute all three C₃₀ steranes to
343 demosponges (Zumberge et al., 2018), however, is a topic of continuing debate (Hallmann et
344 al., 2019; Love et al., 2019; Nettersheim et al., 2019). Consistent with a source from
345 demosponges, 26-methylstigmastane was consistently present, albeit in trace amounts, in all
346 samples analyzed (Figure S2).

347
348 The transition from the U Shale to the Silicilyte in the Athel-1 well records a subtle increase
349 in the relative abundance of prokaryotic biomass, as indicated by lower sterane/hopane ratios
350 above the U Shale/Silicilyte boundary (Figure 2) and a clear shift based on cluster analysis
351 and PCA results, possibly driven by nutrient availability (Figure 4). Concurrently, we
352 observed that the percentages of C₂₉/(C₂₆₋₃₀) steranes, representing the proportions of green
353 algae relative to all eukaryotes, increase significantly ($p < 0.01$) from the U Shale (41.9 –
354 60.8%) to the lower Silicilyte Member (58.3 – 66.5%), where the highest percentages of C₂₉
355 steranes were observed (Figure 3). The prevalence of green algae over other types of algae is
356 a common observation in the Neoproterozoic and early Cambrian in the SOSB and elsewhere,
357 which is further corroborated by our study. A recent study proposed that the rise to
358 prominence of green algae in post-Cryogenian times was triggered by enhanced nutrient
359 supplies generated during deglaciation (Brocks et al., 2017). If this is true, it implies that
360 higher green algae contribution in the lower Silicilyte may also be a consequence of nutrient
361 availability, possibly attributed to terrestrial inputs from the surrounding land (e.g., Eastern
362 Flank Platform) due to basin subsidence, instead of fresh seawater intrusion. 2-Methylhopane
363 index (2-MHI) has been applied as an indicator for cyanobacterial nitrogen fixation, since
364 cyanobacteria fix nitrogen from the atmosphere and many taxa synthesize 2-methylhopanes
365 (Table 2). Therefore, high values of 2-MHI together with low $\delta^{15}\text{N}$ values (~0‰) are
366 consequently indicative of strong nitrogen fixation. However, we observed relatively lower 2-
367 MHI (7.0 – 10.3%) in the U Shale and lower Silicilyte when compared with the upper
368 Silicilyte and Thuleilat Shale, and the data from the Marmul Northwest-7 well (Stolper et al.,
369 2017). Additionally, our samples show constantly positive $\delta^{15}\text{N}$ values, negating strong
370 atmospheric nitrogen fixation (Pehr et al., 2018). Thus, our data do not support a strong
371 preference for microbial nitrogen fixation, but instead indicate nitrate concentrations
372 sufficient to support primary productivity in the oxygenated nutrient rich surface waters
373 across the U Shale/Silicilyte boundary.

374

375 Overall, one of the clearest trends that can be observed in the Athel-1 well is the apparent
376 correlation between the proportions of C₃₀ steranes (C₃₀-npc + C₃₀-ipc; C₃₀/C₂₆₋₃₀) and C₂₆
377 steranes (27-norcholestanes; C₂₆/(C₂₆₋₃₀)), of which the latter ones are pronounced but have so
378 far unknown sources in Ediacaran samples. Both parameters decrease to minima just above
379 the U Shale/lower Silicilyte transition, and being anti-correlated with the increasing trend of
380 C₂₉/(C₂₆₋₃₀) steranes (Figures 3, 7). Further, the ratio of C₃₀-ipc/C₃₀-npc is anti-correlated with
381 C₃₀/(C₂₆₋₃₀) steranes, indicating that C₃₀-ipc, C₃₀-npc and the 27-norcholestanes have discrete
382 sources. Lastly, relative abundances of 3β-methyl-24-ethylcholestanes (3β-Me C₂₉/C₂₉) do
383 not follow those of the 24-ethylcholestanes (C₂₉ steranes) but display a trend opposite to that
384 of the sterane/hopane ratios (Figure 2). Therefore, the production of 3β-methyl-24-
385 ethylcholestanes appears to be the consequence of a distinct process and not directly
386 controlled by the relative abundance of green algal sources of the C₂₉ steranes. This is
387 consistent with these compounds having a specific biological origin, possibly via bacterial
388 modification of algal sterols.

389

390 3.3. Redox conditions and corresponding biological indicators

391 Pristane/phytane ratios were low throughout the sequence and ranged between 0.14 and 0.73,
392 with significantly lower values in the Silicilyte Member (0.16 – 0.39) than the U Shale (0.24 –
393 0.44; $p < 0.02$) and the Thuleilat Shale (0.43 – 0.73; $p < 0.04$; Table 2). It points to the
394 existence of water column anoxia throughout the succession, with extremely reducing
395 conditions in the Athel Silicilyte, consistent with previous results from the SOSB (Grosjean et
396 al., 2009; Love et al., 2009; Stolper et al., 2017) and contrasting to the observations made in
397 the contemporaneous oligotrophic setting in Baltica studied by Pehr et al. (2018). Under
398 anoxic conditions, nitrate, sulfate, and CO₂ are sequentially consumed as electron acceptors,
399 with progressively less energy generation, respectively (Amend and Shock, 2001). When
400 nitrate is preferentially consumed, nitrogen fixation from the atmosphere is required to sustain
401 photosynthesis in the photic zone. In this study, relatively high 2-MHI values up to 13.7% in
402 the upper Silicilyte and Thuleilat Shale are comparable to cases of oceanic anoxic events
403 (Kuypers et al., 2004), suggestive of some degree of nitrogen fixation by cyanobacteria but
404 also the prevalence of nitrate reduction (Figures 2, S1).

405

406 Under conditions where all other electron acceptors are sufficiently depleted, CO₂ may be
407 consumed through the formation of methane in the water column or in surface sediments and

408 subsequently oxidized by microaerobic methanotrophs at redox transitions. Aerobic
409 methanotrophy, being an energetically favored process (Megonigal et al., 2004), takes
410 precedence over anaerobic oxidation on methane when sulfate and, potentially nitrate, are
411 able to act as alternative electron acceptors. Aerobic methanotrophs synthesize 3-
412 methylhopanoids in addition to regular hopanoids, thus, the former one has been extensively
413 applied as a biomarker for methanotrophy. In this study, low values of 3-methylhopane index
414 (3-MHI) (≤ 0.01), indicative of insignificant aerobic methanotrophy in the Athel deep sub-
415 basin of the SOSB and implying an impoverished methane cycle (Table 2, Figure S1). These
416 results are in contrast to previous work of Grosjean et al. (2009) and Stolper et al. (2017),
417 where C₃₁ 3-methylhopane was identified as a significant peak. Based on a re-evaluation of
418 previously published data and a comparison with standards, we attribute this to the superior
419 separation capability of the DB-5MS GC column and the specificity of the QQQ detection
420 that combine to minimize co-elution of the 3 β -methylhopane with $\beta\alpha$ -homohopane isomers
421 and methylgammacerane. Such a phenomenon is commonly observed in other marine
422 environments, especially when sulfate is preferred as the electron acceptor. This is in
423 agreement with previous work suggesting that SOSB was rich in sulfate (Grotzinger and Al-
424 Rawahi, 2014). Therefore, we conclude that sulfate reduction was prevalent as a redox
425 process in the SOSB.

426
427 The existence of sulfidic waters during deposition of SOSB sediments is supported by trace
428 element data from Schroder and Grotzinger (2007) and Wille et al. (2008), through sulfur
429 isotopic analyses of sulfur species by Fike and Grotzinger (2008 and 2010), and the
430 biomarker studies of Grosjean et al. (2009) and Stolper et al. (2017). The hydrocarbon
431 biomarker patterns evident in our analysis are entirely consistent with this, further suggesting
432 sulfate reduction was pervasive in the Athel deep basin. For example, 28,30-dinorhopane
433 (DNH) has been attributed to chemoautotrophic bacteria, possibly H₂S oxidizers in the
434 chemocline (Schoell et al., 1992). Further, 25,28,30-trisnorhopane (TNH) has been proposed
435 as a demethylation product of DNH due to the relatedness of their molecular structures.
436 Although 25-norhopanes, in general, are often prominent in biodegraded oils (Peters et al.,
437 2005; Schoell et al., 1992), this and prior studies have afforded no evidence for this process
438 from the deep basinal facies of the SOSB (Grosjean et al., 2009). The ratio of
439 (TNH+DNH)/C₃₀H ranged between 0.21 and 1.10, comparable to values reported in Stolper et
440 al. (2017) and Grosjean et al. (2009). It displays a general pattern of increase from the U
441 Shale to the lower Silicilyte Member and remained stable in the upper Silicilyte Member and

442 Thuleilat Shale (Figure 2). Although this index is highly dependent on maturity (Moldowan et
443 al., 1984; Peters et al., 2005), other low-maturity samples deposited under anoxic/euxinic
444 conditions display comparable values to those found in this study (Peters et al., 1994), further
445 supporting sulfidic conditions in the water column during deposition of the SOSB deep basin
446 (Figure 2).

447
448 In addition, a high C₃₅ homohopane index (HHI; C₃₅/C₃₁₋₃₅) is commonly attributed to
449 preservation of C₃₅ bacteriohopanepolyols (BHP) by sulfurization as opposed to conditions
450 where free oxygen in the water column or surface sediments can lead to oxidation of BHP to
451 C₃₁ or C₃₂ homologs (Peters et al., 1995; Peters et al., 2005). Gammacerane, with its typical
452 precursor molecule tetrahymanol being synthesized by bacterivorous ciliates, is an indicator
453 of water column stratification (Damste et al., 1995). The homohopane index (HHI; %) and the
454 gammacerane index ranged from 7% to 15%, and from 0.07 to 0.26, respectively (Figure 2),
455 with an overall increasing trend upwards through the sediment column. Large variations, with
456 no clear overall trend, were observed within the upper Silicilyte Member and Thuleilat Shale,
457 possibly due to the limitations of sample numbers and the use of cuttings rather than core. The
458 U Shale displayed significant differences to the Silicilyte Member ($p < 0.001$) and Thuleilat
459 Shale ($p < 0.05$) for both parameters. Additionally, the lower and upper Silicilyte samples
460 were remarkably different in gammacerane index ($p < 0.04$). High values of HHI, high ratios
461 of C₃₅/C₃₄ (> 1), and moderate values of gammacerane index in this study are comparable to
462 many anoxic systems and are much higher than observed in oxidizing environments (Peters
463 and Moldowan, 1991; Peters et al., 1994; Peters et al., 2005) (Figures 2, S1).

464 465 3.4. Euxinic photic zone or not?

466 The detection of a diverse assemblage of aromatic carotenoids (Appendix), biomarkers
467 typically attributed to anoxygenic phototrophic sulfur bacteria, signals the possible presence
468 of photic zone euxinia (Figure 5). Carotenoids with one or both end-groups aromatized are
469 commonly proposed to be largely derived from the GSB and PSB (Brocks et al., 2005;
470 Damsté and Koopmans, 1997; French et al., 2015; Liaaen-Jensen, 1978; van Gemerden and
471 Mas, 1995). Okenane (IV), as the diagenetic product of okenone (III) with a single aromatic
472 χ -ring, is a biomarker for purple sulfur bacteria (PSB) of the family Chromatiaceae that
473 proliferate as plankton and in benthic microbial mats where the chemocline is $< 24\text{m}$ and, in
474 many cases, $< 12\text{m}$ (Blankenship et al., 2006; Brocks and Schaeffer, 2008; French et al.,
475 2015; Liaaen-Jensen and Andrewes, 1972). Similarly, chlorobactene (V), a pigment with a

476 single aromatic ϕ -ring characteristic of the ‘green strains’ of the green sulfur bacteria (GSB)
477 family Chlorobiaceae, is the only known precursor of chlorobactane (VI), of which the
478 detection is indicative of euxinia as shallow as 15m in the water column (Kuypers et al.,
479 2002). Isorenieratane (XIV), with two ϕ -rings, is the diagenetic products of isorenieratene
480 (XIII) characteristic of ‘brown strains’ of the GSB and is the most common carotenoid
481 encountered in Phanerozoic marine settings (French et al., 2015). Isorenieratene-producing
482 ‘brown strains’ GSB are typically observed under low light conditions and are found as deep
483 as 100m in euxinic water columns (Overmann et al., 1992). When the chemocline is shallow,
484 either the ‘green strains’ of GSB are more favored or their carotenoids are tuned differently
485 (Bryant et al., 2012; Maresca et al., 2008b). The diaromatic carotenoid renierapurpurin (XIX),
486 originally isolated from marine sponges and the presumed precursor of renierapurpurane
487 (XX), possesses dual χ -rings and is the diaromatic counterpart of okenone, suggestive of an
488 origin from PSB although direct evidence of its occurrence in cultured PSB, to our
489 knowledge, has not been reported. Rather, the occurrence of their hydrogenated derivatives
490 along with renieratane (XVI) and okenane (IV) in sediments has led several researchers to
491 attribute all three compounds to sources within the Chromatiaceae (Behrens et al., 2000;
492 Brocks et al., 2005; Brocks and Schaeffer, 2008). As with chlorobactene, the carotenoid
493 precursor (IX) of β -isorenieratane (X) with one β -ring and one ϕ -ring is associated with
494 communities adapted to higher light intensities than those producing isorenieratene, and
495 therefore likely accommodate intermediate water depth (15 – 100 m) (Frigaard and Bryant,
496 2004; Overmann et al., 1992; Repeta, 1993).

497

498 Persistent signals of β -isorenieratane are commonly accompanied by β -paleorenieratane
499 (VIII) (Pers. Commun. with GeoMark Research Inc.) in Neoproterozoic and Paleozoic strata,
500 leading us to attribute them to ‘brown strains’ of the GSB, partially based on their structural
501 similarities to isorenieratane and paleorenieratane, respectively (Hartgers et al., 1993).

502 Similarly, β -renierapurpurane (XII), with its potential precursor being β -renierapurpurin (XI),
503 have been tentatively assigned with PSB (Behrens et al., 2000; Brocks & Schaeffer, 2008).

504 These observations, however, require further verification and are beyond the scope of the
505 present study.

506

507 The determination of absolute carotenoid concentrations is challenging. However, because β -
508 carotene (I) is a universal carotenoid present in the vast majority of algae and cyanobacteria,

509 its fossilized saturated counterpart β -carotane (II) can be used for normalization of aromatic
510 carotenoids. Okenane, chlorobactane, β -isorenieratane, and β -paleorenieratane are present at
511 low levels in relative to β -carotane throughout the succession (Figures 5, 6), suggestive of
512 episodic or transient sulfide intrusion into the shallow photic zone in the Athel deep basin,
513 especially during deposition of the U Shale and the lower Silicilyte. Comparatively, samples
514 from A4 carbonate stringer displayed the only case where okenane is in relatively higher
515 abundance than chlorobactane (Figure 5). So far, okenone has only been reported in modern
516 lake settings (Brocks and Schaeffer, 2008; Meyer et al., 2011) and, as okenane, in a few
517 ancient restricted basins (Bhattacharya et al., 2017; Brocks et al., 2005; French et al., 2015)
518 but not in open ocean settings. Therefore, we posit that the detection and stratigraphic
519 persistence of low levels of okenane, chlorobactane, β -isorenieratane and β -paleorenieratane
520 in this study is consistent with restricted water circulation and transient or episodic euxinia in
521 the shallow photic zone during deposition of the SOSB sedimentary sequence. Transient
522 euxinia is possibly disturbed by periodic seawater incursion into the restricted basin and is
523 further corroborated by the detection of biomarkers indicative of sponges, which survive in
524 oxic and suboxic conditions (Mills et al., 2014).

525

526 Signals of isorenieratane were near the detection limit in our samples from the Athel deep
527 basin. This is in contrast to the carbonate platform, exemplified by A4 carbonate stringer from
528 Birba-3 well, where isorenieratane was detected in minute traces. Since the carbonate
529 platform was presumably deposited under relatively shallower water than the Silicilyte, it
530 seems possible that the isorenieratane found in the platform could be benthic microbial mats
531 derived when the sediment-water interface was sulfidic under reducing conditions.

532 Comparatively, barely detectable isorenieratane in the deep basin deposits might be attributed
533 to sharply diminishing light intensity to the deep depth where ‘brown strains’ of GSB are
534 competitive (Connock et al., 2018), possibly due to light absorption by shallow water sulfur
535 bacteria and shading by a dense oxygenic phototrophic community, as supported by the
536 detection of steranes and hopanes. Another possibility is that the deep water was ferruginous
537 that sulfide in the deep water has been scavenged by excessive iron (Fike & Grotzinger,
538 2010), suppressing the productivity of low-light-adapted GSB. Alternatively, minimal
539 isorenieratane may not be interpreted as an absence of ‘brown strains’ of GSB inhabiting
540 deeper water depths. Indeed, in studies on modern lake systems where there is rDNA
541 evidence for the presence of GSB in the photic zone, isorenieratane (XIII) is either not
542 detected (Meyer et al., 2011) or detected at trace levels (Fulton et al., 2018) when the

543 chemocline is shallow (e.g., 6-20m) (Glaeser and Overmann, 2003; Meyer et al., 2011). Well-
544 studied examples of modern environments include Fayetteville Green Lake near Syracuse in
545 New York, which is a small, meromictic kettle lake with steep sides and limited wind fetch
546 such that seasonal overturn is precluded (Meyer et al., 2011).

547
548 Renierapurpurane (XX) and renieratane (XVI) show comparable abundances to β -
549 isorenieratane (X) and are much higher than isorenieratane (Figures 5, S4). Both compounds
550 have been proposed as being derived from PSB (Behrens et al., 2000; Brocks et al., 2005;
551 Brocks & Schaeffer, 2008). In this study, ratios of (renieratane+renierapurpurane) / β -
552 carotane, with ranges between 0.04 and 0.33, have a poor correlation with
553 (okenane+chlorobactene) / β -carotane ($R^2 < 0.2$) and a weak correlation with (β -
554 paleorenieratane+ β -isorenieratane) / β -carotane ($R^2 = 0.46$), suggestive of different sources.
555 Rather, the latter two proxies mimic each other and show strong correlation ($R^2 = 0.76$; Figure
556 7). Consequently, our results suggest multiple sources for the above-mentioned compounds
557 and question the sulfur bacterial origin of renierapurpurane and renieratane. Interestingly,
558 predominant renierapurpurane and renieratane have been detected in the Green River Fm. and
559 Barney Creek Fm. (Brocks et al., 2005; French et al., 2020). French et al. (2020) attributed
560 this pattern, and the abundant co-occurring 3β -methylhopanes, to these rocks having been
561 deposited in lacustrine settings, an explanation that does not apply to the restricted marine
562 environment of the sediments studied here. An alternative origin for renierapurpurane is from
563 cyanobacteria. Our alternative hypothesis is based on the distribution of lipids reported from
564 the euryhaline, unicellular taxon *Synechococcus* sp. strain PCC 7002 and related species,
565 which synthesize renierapurpurin (XIX) as an intermediate in the biosynthesis of the
566 cyanobacterial carotenoid χ,χ -caroten-18,18'-dioic acid (XXV), otherwise known as
567 synechoxanthin, via the intermediates XXI, XXII and XXIV (Graham and Bryant, 2008;
568 Graham et al., 2008). Further research has confirmed production of renieratene and traces of
569 isorenieratene from these strains (Cui et al., 2020). Studies of the presence of synechoxanthin
570 in *Synechococcus* sp. PCC7002 implicate CruH as the protein that is minimally responsible
571 for the hydroxylation/oxidation of the C-18 and C-18' methyl groups of renierapurpurin
572 (Graham and Bryant, 2008; Montero et al., 2011). Taking this logic further, we propose that
573 compounds XXI, XXII and XXIV, through diagenetic hydrogenation and loss of one or both
574 of these substituents from the structures and thermally-driven decarboxylation reactions,
575 could produce XXIII and XXVI, that is, C₃₉ and C₃₈ diagenetic products (Appendix).
576

577 C₃₉ and C₃₈ carotenoid diagenesis products are tentatively identified in this study. One of
578 them, as a major compound, was observed to elute between β -renierapurpurane and
579 isorenieratane and has a molecular ion of 532 Da with major fragment ions of 134, 133, and
580 120 Da. These features identify this compound as a C₃₉ diaryl isoprenoid with trimethyl
581 substitution in one ring and a dimethyl substitution in the other (Figure S4). In earlier work,
582 Schwark and Püttmann (1990) reported a C₃₉ diaryl isoprenoid at a similar relative retention
583 position and containing one less methylene carbon in the isoprenoid chain. However,
584 subsequent investigation deemed this unlikely and, instead, identified a set of C₄₀ tetra-
585 aromatic carotenoids with the same molecular weight of 532 Da (Koopmans et al., 1996).
586 None of these compounds were reported to have 120 Da as a major fragment ion (>50% of
587 134) making the compound reported here distinct. A second major compound identified in the
588 Athel-1 and Birba-3 sediments is a C₃₈ diaromatic aryl isoprenoid based on its relative
589 retention time and an MRM transition employing precursor and product ions of 518 and 120
590 Da (Figure S4). Knowing that decarboxylation is a facile, thermally-driven process during
591 catagenesis, we hypothesized that synechoxanthin (XXV) could be its potential biogenic
592 precursor. This hypothesis was reinforced by a report from Koopmans et al. (1997) who
593 identified the same C₃₈ dimethyl/dimethyl diaromatic carotenoid in the Green River Shale and
594 another by Zhang et al. (2011) who identified C₃₉ and C₃₈ aryl isoprenoids but speculated that
595 they resulted from the loss of a methyl group from one or both aromatic rings.

596

597 We sought additional support for our hypothesis by examining the relative abundance ratios
598 of different carotenoid and the respective diagenesis products. In this, we observed a strong
599 correlation between the ratios of the C₃₉ carotenoid / β -carotane and (renieratane +
600 renierapurpurane) / β -carotane which suggests that they may be biologically or diagenetically
601 related (Figure 7). Therefore, the detection of a distinctive biomarker assemblage in this study
602 comprising a predominance of renierapurpurane and renieratane together with putative C₃₉
603 and C₃₈ diagenetic products, and the strong correlation mentioned above, point to
604 cyanobacteria as an alternate and most likely source for these carotenoids (Graham and
605 Bryant, 2008; Graham et al., 2008).

606

607 3.5 Implications of the SOSB carotenoid data

608 It has been suggested that the presence of phototrophic sulfur bacteria in the water column, or
609 in microbial mats, and the types of carotenoids produced by them, are highly linked. For
610 example, the monoaromatic carotenoids are regarded as molecular indicators of high-light-

611 adapted PSB and GSB dependent upon shallow chemoclines or representing inputs from
612 microbial mats. Diaromatic carotenoids are typically attributed to low-light-adapted GSB
613 dwelling in chemoclines at up to 100 m depth (Overmann, 2008). In this study, no correlation
614 was found between (chlorobactene + okenane) / β -carotane and (renieratane +
615 reneirapurpurane) / β -carotane ($p > 0.3$), suggesting multiple independent sources and/or
616 multiple controlling factors. Notably, while reneirapurpurin does not appear to have been
617 identified in cultured PSB, it has been reported in a cultured cyanobacterium (Graham and
618 Bryant, 2008; Graham et al., 2008), which is further supported by this study. Further, the
619 biosynthesis of aromatic carotenoids has been observed in other cyanobacteria and,
620 accordingly, does not require sulfidic water columns nor PZE (Maresca et al., 2008a).

621
622 We also observed that (β -paleorenieratane + β -isorenieratane) / β -carotane displayed a weak
623 but significant correlation with (renieratane + reneirapurpurane) / β -carotane ($p < 0.001$), and a
624 stronger correlation with (chlorobactene+okenane) / β -carotane ($p < 0.001$; Figure 7).

625 Considering no correlation between (chlorobactene + okenane) / β -carotane and (renieratane +
626 reneirapurpurane) / β -carotane, it suggests that organisms producing β -paleorenieratene and
627 β -isorenieratene (IX) may also have more complex sources than previously recognized.
628 Although β -isorenieratene has been proven to be synthesized by GSB, β -reneirapurpurin (XI)
629 and, potentially, other bicyclic monoaromatic carotenoids are also intermediates along the
630 biosynthesis pathway leading to aromatic carotenoids in cyanobacteria and streptomyces as
631 well (Graham & Bryant, 2008).

632
633 In this study, the C_{39} carotenoid/ β -carotane ratio across many samples presented a strong
634 correlation with (renieratane+reneirapurpurane) / β -carotane, suggesting a source relationship.
635 Additionally, the C_{39} compound (XXIII) was in higher abundances compared to renieratane
636 (XVI) and reneirapurpurane (XX), suggesting that the depositional environment favored the
637 production of its carotenoid precursor. Furthermore, the C_{38} compound (XXVI) is either
638 comparable or sometimes more abundant compared to its C_{39} counterpart, also speaking to
639 their close linkage. Dimethyl aryl isoprenoids have been reported in high concentrations in
640 hypersaline environments (Damsté et al., 1988; Koopmans et al., 1997; Zhang et al., 2011).
641 All of these observations point to the possibility that high concentrations of C_{38} and C_{39}
642 carotenoids are associated with hypersaline conditions as opposed to PZE. However, their
643 presence in SOSB does not negate the possibility that the SOSB was sulfate rich at the
644 Ediacaran-Cambrian Boundary, but instead provides another potential explanation for the

645 presence of C₄₀ diaromatic carotenoids (e.g., renierapurpurane) and their unusual distribution
646 pattern in this case study. Additionally, based on our observations, C₃₈ and C₃₉ carotenoids
647 found in the deep basinal package of the Athel Fm. are much higher than in the Buah Fm. and
648 the carbonate platform units (Figure 5), making them potentially valuable proxies for oil-
649 source correlation.

650

651 **Conclusions**

652 This study demonstrates that the composition of microbial communities varied significantly
653 during the deposition of the U Shale, Silicilyte and Thuleilat Shale members of the Athel
654 Formation in the Athel deep basin across the Ediacaran-Cambrian Boundary. Specifically,
655 sterane/hopane ratios and other indices indicate shifts in plankton communities and water
656 chemistry (e.g., stratification, nutrients), possibly driven by tectonism and the subsequent
657 episodic connection of the restricted basin to the open ocean. The three units investigated in
658 this study witnessed a progressive increase in water column stratification following fresh
659 seawater intrusion in the U Shale.

660

661 Absent evidence for a significant methane cycle and a weak denitrification-induced nitrogen
662 fixation, our data suggest that sulfate reduction was the preferred terminal respiratory process.
663 The presence, albeit low concentrations, of monoaromatic (e.g., chlorobactene, okenane)
664 carotenoids synthesized by anoxygenic photosynthetic purple and green sulfur bacteria
665 supports episodic or transient euxinic conditions in the shallow photic zone of the Athel sub-
666 basin or, possibly, input from microbial mats. Barely detectable isorenieratane in the Athel
667 deep basin implies minimal deep-water adapting “brown strain” green sulfur bacteria,
668 possibly due to 1) shading of planktonic microbial communities; 2) a ferruginous deep ocean;
669 or minimal synthesis of isorenieratane by GSB under some conditions. Comparatively, the
670 unusual predominance of renieratane and renierapurpurane over isorenieratane, together with
671 abundances of C₃₉ and C₃₈ carotenoids in the Athel deep basin and their strong statistical
672 correlation, identifies cyanobacteria as the most probable contributor to the diaromatic
673 carotenoid inventory. Overall, we propose that the aromatic carotenoids present in Ediacaran
674 strata may have sources other than phototrophic sulfur bacteria. Further detailed
675 investigations of aromatic carotenoids in other taxa such as the cyanobacteria will improve
676 the robustness of paleoenvironmental reconstructions.

677

678 **Acknowledgements**

679 We gratefully acknowledge financial support from the NASA Astrobiology Institute
680 (NNA13AA90A) and the Simons Foundation Collaboration on the Origins of Life
681 (290361FY18). This work benefitted from discussions with Professor Donald Bryant of The
682 Pennsylvania State University, concerning the origins of synechocystin and discussions with
683 Don Rocher and John Zumberge of GeoMark Research Inc. on aromatic carotenoids. We
684 thank the Ministry of Oil and Gas of the Sultanate of Oman for permission to access samples.
685 Petroleum Development Oman contributed subsurface sample access.
686

687

688 Table 1: Major biomarkers applied in this study and their source indications.

689

Parameters	Interpretation	References
Acyclic isoprenoids		
pristane/phytane	water column redox	Didyk et al. (1978)
Hopanes		
Ts/(Ts+Tm)	thermal maturity index/ lithology	Farrimond et al. (1998)
C ₃₁ H S/(S+R)	thermal maturity index	Seifert and Moldowan (1980)
(TNH+DNH)/C ₃₀ H	photic zone sulfur cycle	Schoell et al. (1992)
gammacerane/C ₃₀ H	stratification	Schoell et al. (1994)
2-MHI	cyanobacteria or N ₂ -fixing bacteria	Summons et al. (1999); Kuypers et al. (2004)
3-MHI	intensity of methanotroph	Summons et al. (1994)
C ₃₅ HHI	water column redox condition	Peters and Moldowan (1991)
Sterane		
C ₂₉ dia/(dia+reg)	thermal maturity index	van Kaam-Peters et al. (1998)
C ₂₉ /(C ₂₇ +28+29)	green vs red algae	Schwark and Empt (2006)
C ₃₀ -npc	marine algae; sponges	Moldowan et al. (1990); Love et al. (2009)
C ₃₀ -ipc	sponges	Love et al. (2009)
C ₃₀ -26- methylstigmastane	sponges	Zumberge et al. (2018)
Carotenoids		
chlorobactane	green sulfur bacteria	Grice et al. (1998)
okenane	purple sulfur bacteria	Schaeffer et al. (1997)
isorenieratane	green sulfur bacteria	Summons and Powell (1987); Maresca et al. (2008b)
renieratane	green/purple sulfur bacteria or cyanobacteria	Schaeffle et al. (1977); Hartgers et al. (1993) This Study
renierapurpurane	purple sulfur bacteria or	Schaeffle et al. (1977); (Brocks et al., 2005)

	cyanobacteria	This Study
β -paleorenieratane	green sulfur bacteria	Hartgers et al. (1993); This Study
β -isorenieratane	green sulfur bacteria or cyanobacteria	Grice et al. (1998); Schaeffer et al. (1997); this study
β -renierapurpurane	purple sulfur bacteria or cyanobacteria	Behrens et al. (2000); Schaeffer et al. (1997) This Study
C ₃₉ carotenoid	cyanobacteria	This Study
C ₃₈ carotenoid	cyanobacteria	This Study

690

691

692

693

694

695 Table 2. The average and standard deviation of parameters in each formation, where the
 696 Silicilyte is divided into the upper and lower Silicilyte formations.

697

	Thuleilat Shale	Upper Silicilyte	Lower Silicilyte	“U” Shale	Buah	Shuram	Carbonate Platform
$\delta^{13}\text{C}$ (‰)	-38.9±1.5	-37.8±0.4	-37.0±0.7	-36.2±1.7	-37.9±0.0	-36.5±1.5	-35.0 ^a
$\delta^{15}\text{N}$ (‰)	3.0±1.2	3.1±0.3	3.3±0.3	2.4±0.5	2.9±1.0	2.5±1.2	5.8 ^a
pristane/phytane	0.55±0.16	0.27±0.12	0.20±0.07	0.31±0.07	0.67±0.04	0.60±0.09	0.43±0.20
Ts/(Ts+Tm)	0.24±0.02	0.23±0.02	0.25±0.02	0.24±0.04	0.16±0.02	0.54±0.25	0.18±0.00
(TNH+DNH)/C ₃₀ H ^a	0.76±0.16	0.77±0.03	0.70±0.35	0.39±0.07	0.21±0.08	0.08±0.08	0.62±0.03
C ₃₀ H $\beta\alpha$ /($\alpha\beta$ + $\beta\alpha$)	0.04±0.00	0.03±0.01	0.03±0.00	0.03±0.00	0.03±0.00	0.03±0.00	0.03±0.00
C ₃₁ H S/(S+R)	0.52±0.01	0.55±0.00	0.56±0.01	0.55±0.01	0.53±0.00	0.56±0.02	0.57±0.00
gammacerane/C ₃₀ H	0.21±0.05	0.19±0.02	0.13±0.05	0.09±0.01	0.30±0.09	0.16±0.08	0.20±0.01
C ₃₅ /(C ₃₁₋₃₅)H (%)	11.6±1.0	12.6±2.4	10.3±2.3	7.5±0.5	10.7±4.0	7.9±2.2	9.9±0.3
2-MHI (%) ^b	11.3±2.2	11.7±1.0	9.5±0.9	7.9±0.5	12.4±0.8	9.1±1.9	5.6±0.1
3-MHI (%) ^c	1.00±0.19	0.66±0.04	0.51±0.08	0.70±0.16	0.83±0.08	0.63±0.14	0.81±0.03
27-nor C ₂₆ /(C ₂₆₋₃₀)S (%) ^d	15.8±3.9	9.4±2.4	6.9±1.5	11.3±3.3	9.7±3.2	8.4±1.9	7.8±0.0
C ₂₇ /(C ₂₆₋₃₀)S (%) ^d	20.0±1.3	18.2±1.3	15.7±1.3	19.6±1.5	22.3±2.4	17.7±3.0	20.6±1.3
C ₂₈ /(C ₂₆₋₃₀)S (%) ^d	12.7±0.1	12.7±0.8	12.3±0.4	13.5±1.0	11.0±0.6	11.3±1.4	11.3±0.4
C ₂₉ /(C ₂₆₋₃₀)S (%) ^d	49.2±4.2	57.4±2.7	63.1±3.3	53.2±5.9	54.7±6.5	60.5±6.0	57.9±0.9
C ₃₀ /(C ₂₆₋₃₀)S (%) ^d	2.3±0.1	2.2±0.2	2.1±0.2	2.4±0.5	2.3±0.3	2.2±0.3	2.4±0.1
C ₃₀ S ipc/npc ^e	0.94±0.02	0.83±0.03	0.96±0.10	0.67±0.15	0.55±0.05	0.84±0.19	0.82±0.09
3 β -me C ₂₉ /C ₂₉ S ^f	0.25±0.01	0.22±0.02	0.31±0.09	0.29±0.04	0.25±0.10	0.35±0.12	0.25±0.00
C ₂₉ S dia/(dia+reg) ^g	0.06±0.02	0.03±0.01	0.02±0.00	0.04±0.01	0.03±0.01	0.11±0.07	0.04±0.00
C ₂₉ S aaa S/(S+R) ^h	0.51±0.01	0.49±0.01	0.50±0.00	0.49±0.02	0.50±0.00	0.51±0.01	0.51±0.00
sterane/hopane ⁱ	0.49±0.06	0.64±0.01	0.61±0.04	0.76±0.09	0.72±0.21	0.59±0.10	0.66±0.11
(chl+oke)/ β -carotane ^j	0.14±0.01	0.18±0.03	0.20±0.07	0.31±0.12	0.18±0.04	0.27±0.09	0.11±0.01
(β -pal+ β -iso) / β -carotane ^k	0.70±0.15	0.47±0.15	0.41±0.08	0.58±0.14	0.29±0.06	0.49±0.15	0.28±0.00
(ren+rnp) / β -carotane ^l	0.23±0.09	0.19±0.03	0.11±0.05	0.14±0.06	0.06±0.02	0.09±0.07	0.05±0.00
C ₃₉ carotenoid / β -carotane	0.68±0.27	0.61±0.08	0.39±0.12	0.33±0.08	0.09±0.02	0.20±0.16	0.10±0.00

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699 ^a only one sample was measured for OC and N isotopes.

700 ^b 2-methylhopane index (2-MHI) is calculated as: C₃₀₋₃₄ 2-methylhopanes/(C₃₀₋₃₄ 2-methylhopanes+ C₃₀₋₃₄
 701 hopanes).

702 ^c 3-methylhopane index (3-MHI) is calculated as: C_{30-32} 3-methylhopanes/(C_{30-32} 3-methylhopanes+ C_{30-32}
703 hopanes).

704 ^d C_{26-30} steranes include diasteranes ($\beta\alpha$ and $\alpha\beta$) and regular steranes (5α , 14α , 17α and 5α , 14β , 17β) of 27-nor
705 C_{26} , C_{27} , C_{28} , C_{29} and C_{30} -npc and C_{30} -ipc.

706 ^e calculated based on 20R 5α , 14α , 17α C_{30} -npc and C_{30} -ipc steranes.

707 ^f calculated based only on regular 3β -me C_{29} and C_{29} steranes.

708 ^g C_{29} S dia/(dia+reg) represents C_{29} diasterane/(C_{29} diasterane+ C_{29} regular sterane)

709 ^h calculated based on 20S and 20R 5α , 14α , 17α C_{29} steranes.

710 ⁱ calculated based on C_{27-29} diasteranes and regular steranes, C_{27-30} hopanes, and C_{31-35} homohopanes.

711 ^j chl: chlorobactene; oke: okenane

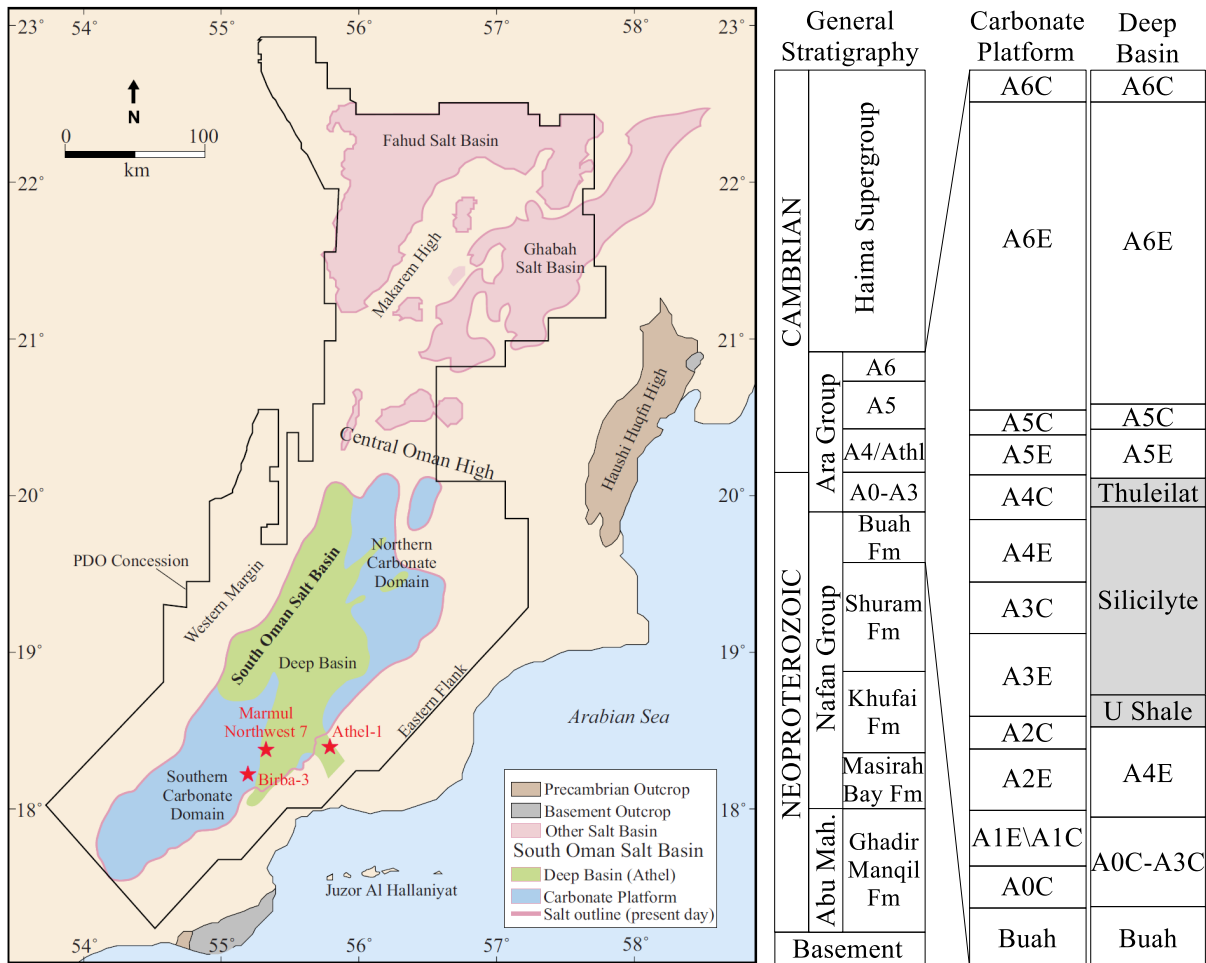
712 ^k β -pal: β -paleorenieratane; β -iso: β -isorenieratane

713 ^l ren: renieratane; rnp: renierapurpurane

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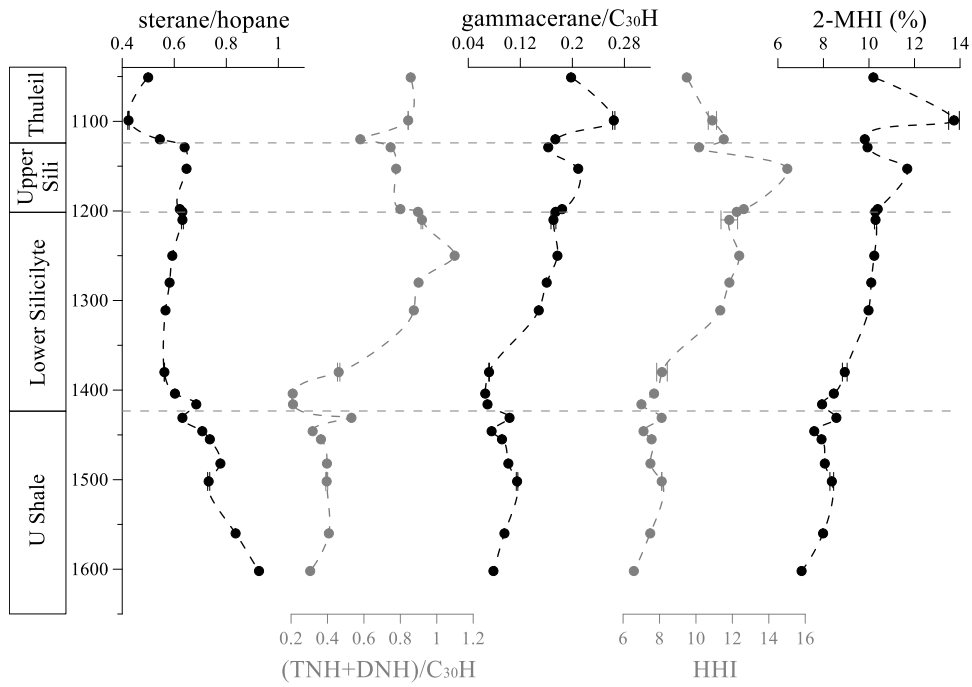
719 Figure 1. Map and stratigraphy of the South Oman Salt Basin, where the A4 carbonates from
 720 the platform are time equivalent to the U Shale + Athel Silicilyte Formation + Thuleilat Shale
 721 in the deep sub-basin. The shaded area represents the time period investigated in this study.
 722 The figures are modified from Al-Siyabi (2005) and Grotzinger and Al-Rawahi (2014).

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728 Figure 2. Downcore profiles of sterane/hopane ratios, selected hopanes, and gammacerane.

729 The numbers represent downcore depths in meters, while the grey dashed lines represent

730 formation boundaries. TNH: 25,28,30-trinorhopane; DNH: 28,30-dinorhopane; HHI:

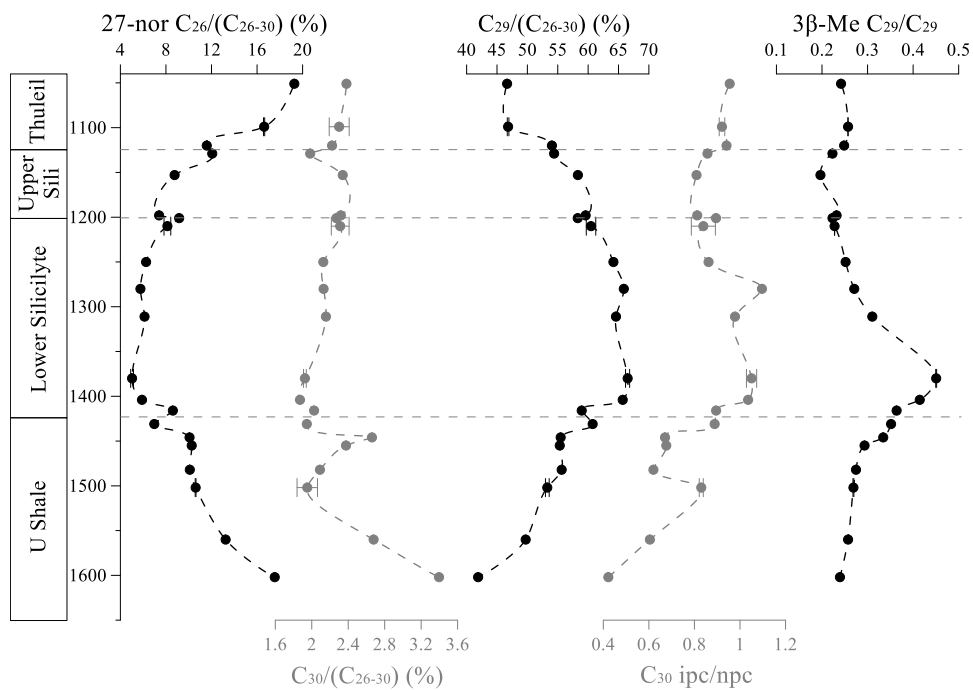
731 homohopane index, defined as the percentage of C₃₅/(C₃₁₋₃₅) homohopanes; 2-MHI: 2-

732 methylhopane index.

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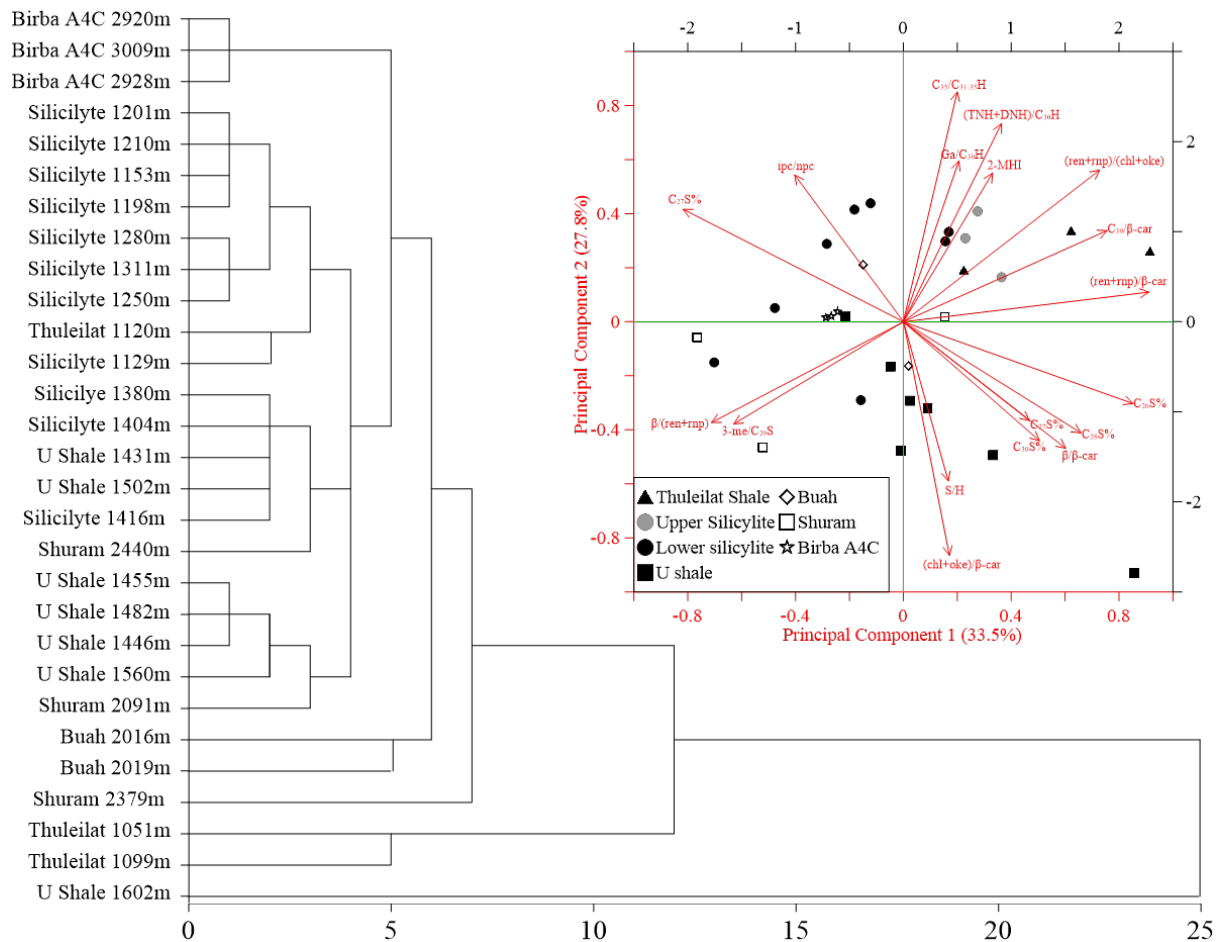
736

737 Figure 3. Downcore profiles of selected sterane indices. C₃₀ ipc: C₃₀ 24-isopropyl cholestane;

738 C₃₀ npc: C₃₀ 24-n-propyl cholestane, 3β-Me C₂₉: 3β-methyl-24-ethyl cholestane.

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743 Figure 4. Comprehensive data analysis, including cluster analysis and principal component

744 analysis (PCA). Cluster analysis is performed based on sterane, hopane, and carotenoid

745 indices. The specific molecular proxies included in the cluster analysis are $C_{28}/(C_{26-30})$

746 steranes, $C_{29}/(C_{26-30})$ steranes, $C_{30}/(C_{26-30})$ steranes, C_{30} sterane ipc/npc, (TNH+DNH)/ C_{30}

747 hopane, 2-MHI, sterane/hopane ratio, (chl+oke)/ β -carotane, (β -pal+ β -iso)/ β -carotane,

748 (ren+rnp)/ β -carotane, C_{39} carotenoid/ β -carotane. Principal component analysis, seen in the

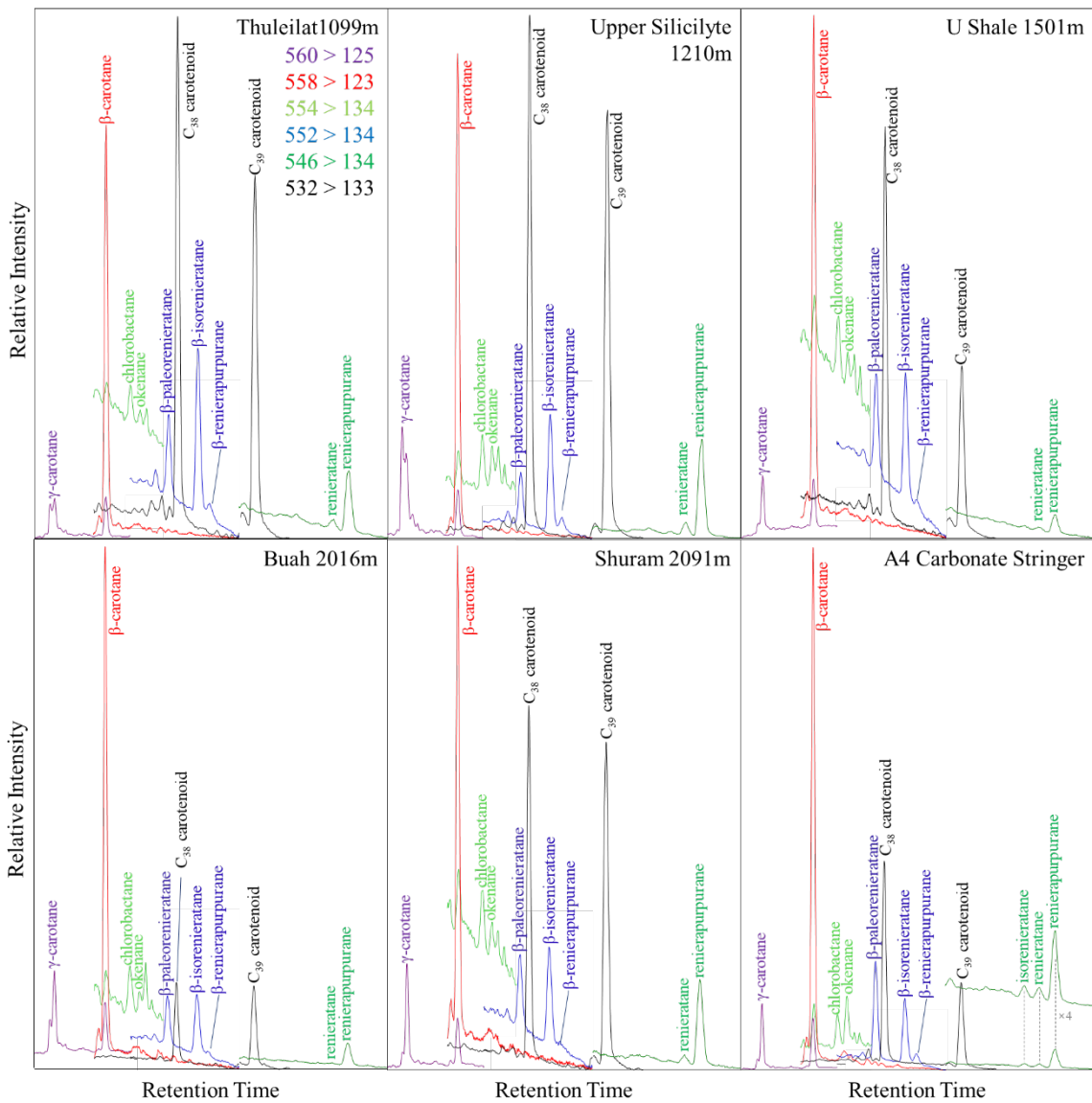
749 upper right corner, is performed with all parameters indicated in the plot. Parameters are

750 represented in red arrows, while each sample is displayed using variable symbol. TNH:

751 trisnorhopane; DNH: dinorhopane; chl: chlorobactene; oke: okenane; β -pal: β -

752 paleorenieratane; β -iso: β -isorenieratane; ren: renieratane; rnp: reneirapurpurane.

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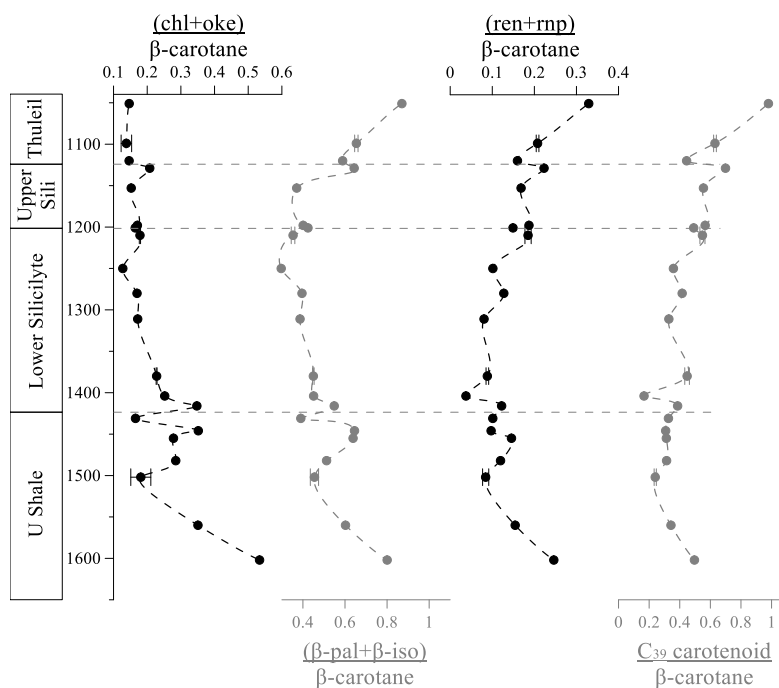


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755 Figure 5. Composite MRM reaction chromatograms of selected saturated and aromatic
 756 carotenoid biomarkers from each representative unit studied in this study. The response of
 757 each compound is shown in intensity normalized to the most intense peak. Chromatograms
 758 are coloured according to their specific reactions in the MRM mode. The upper three samples
 759 from Thuleilat Shale, Silicilyte Formation, and U Shale are shown in comparison with
 760 examples from Buah Formation, Shuram Formation and A4 Carbonate Stringer.

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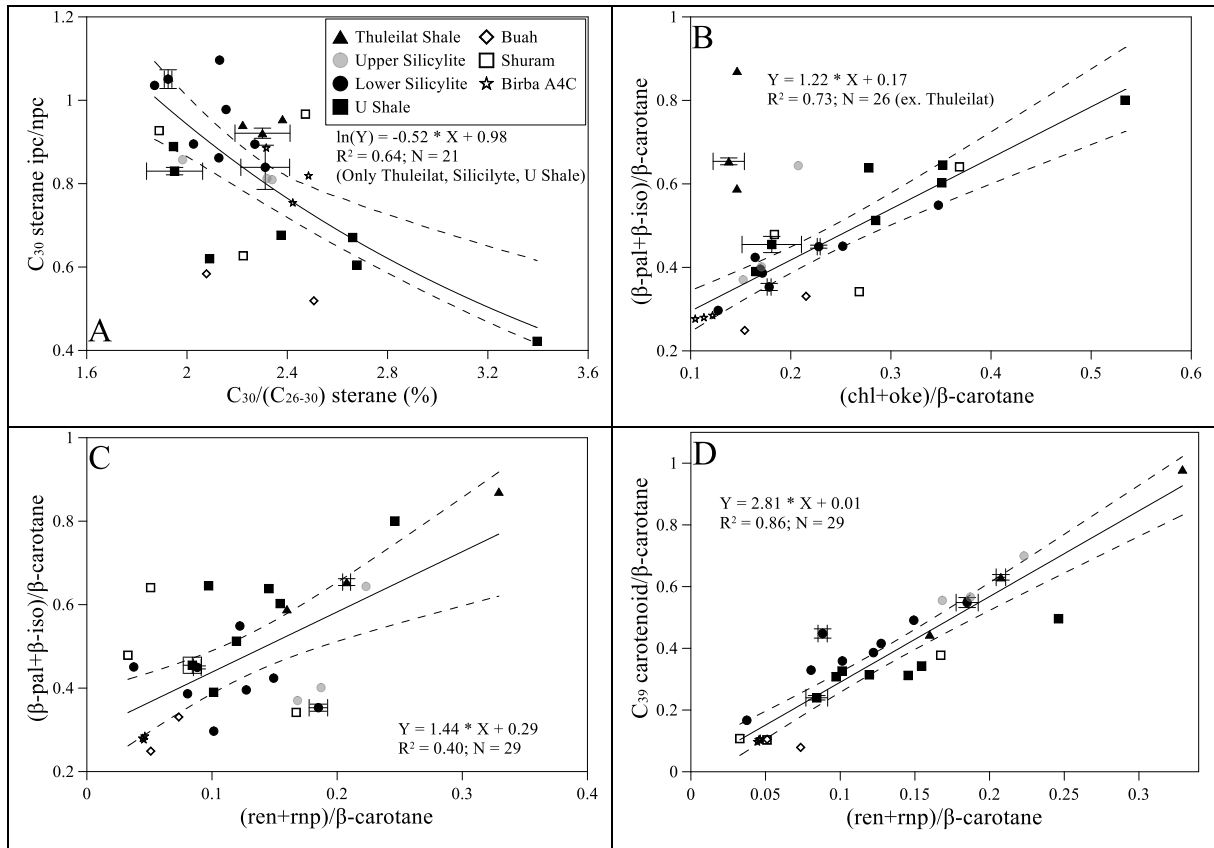


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764 Figure 6. Downcore profiles of selected carotenoid indices. chl: chlorobactene; oke: okenane;
 765 β -pal: β -paleorenieratane; β -iso: β -isorenieratane; ren: renieratane; rnp: renierapurpurane; C₃₉
 766 carotenoid: new compound reported in this study with precursor and product ions of 532Da
 767 and 134Da, respectively.

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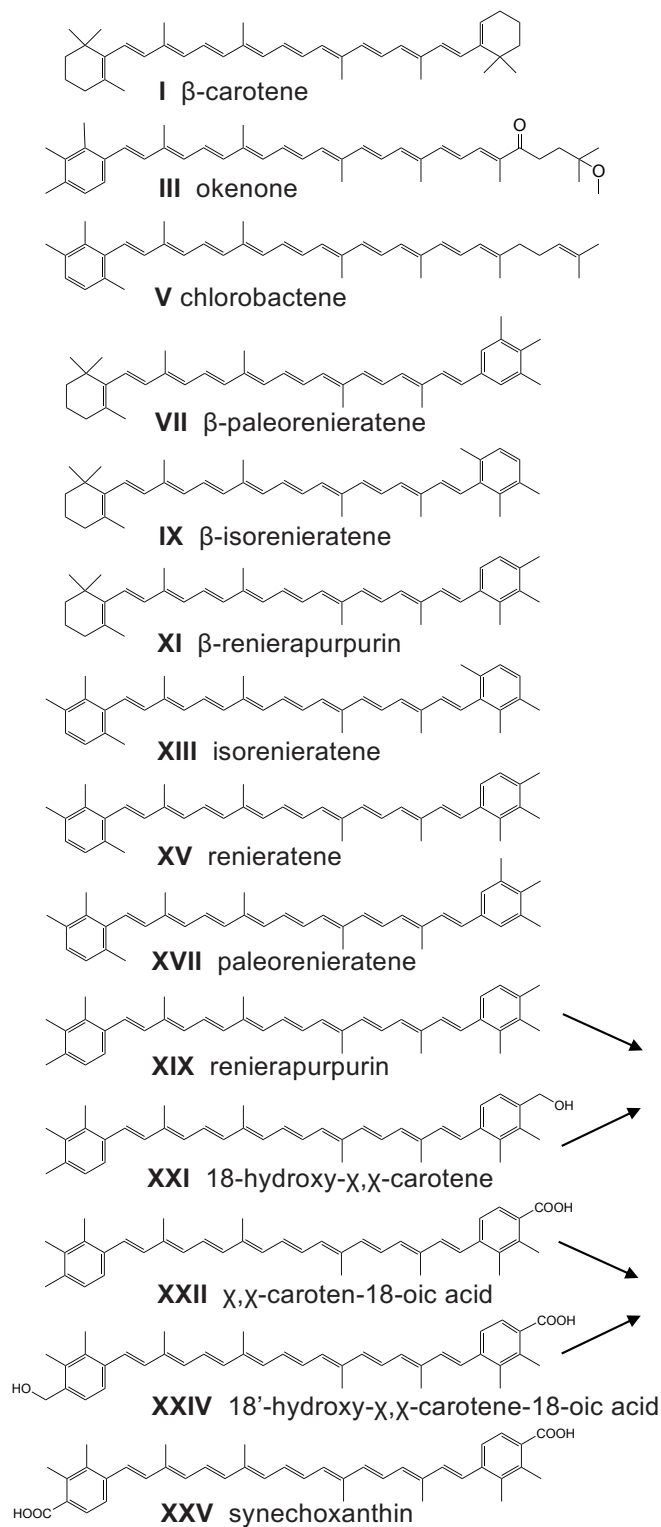
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771 Figure 7. The cross plots of selected sterane and carotenoid-based indices. The regressions are
 772 chosen based on the best fits. The abbreviations are explained in preceding figures and tables.

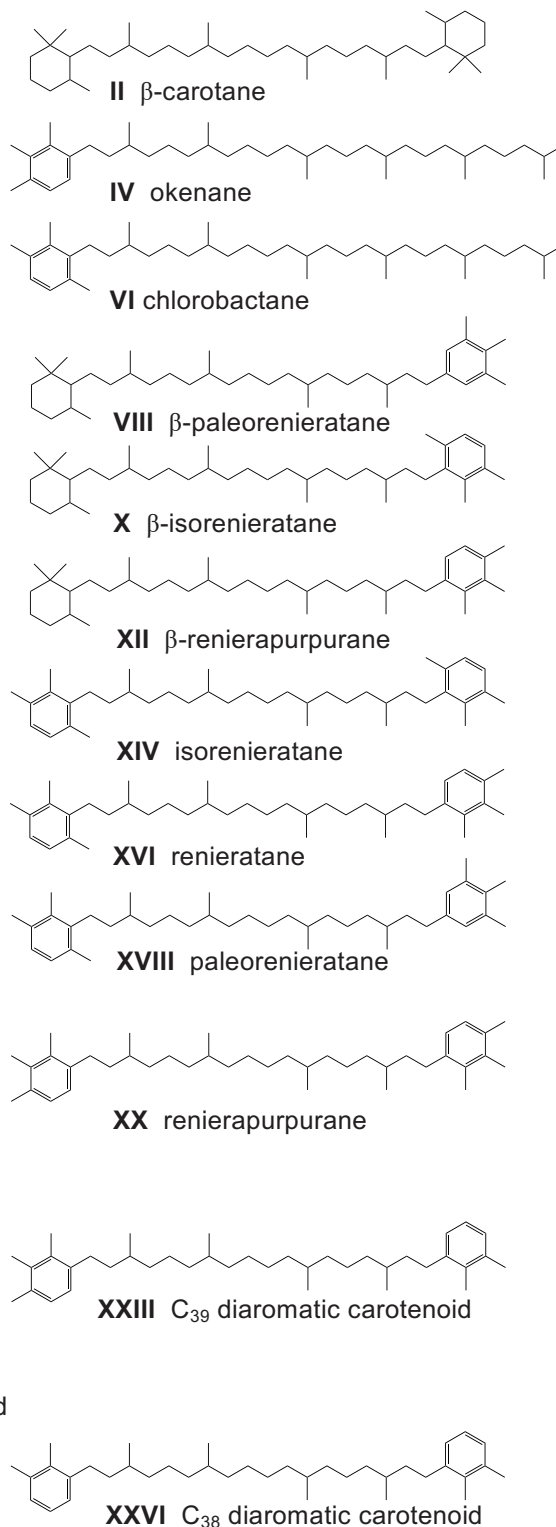
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Biogenic precursor carotenoid



Geostable chemical fossil



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1046 **Supplementary Information**

1047

1048 Table S1. Major compounds analyzed in this study and the precursor and products *m/z* values
1049 used to detect them through MRM transitions.

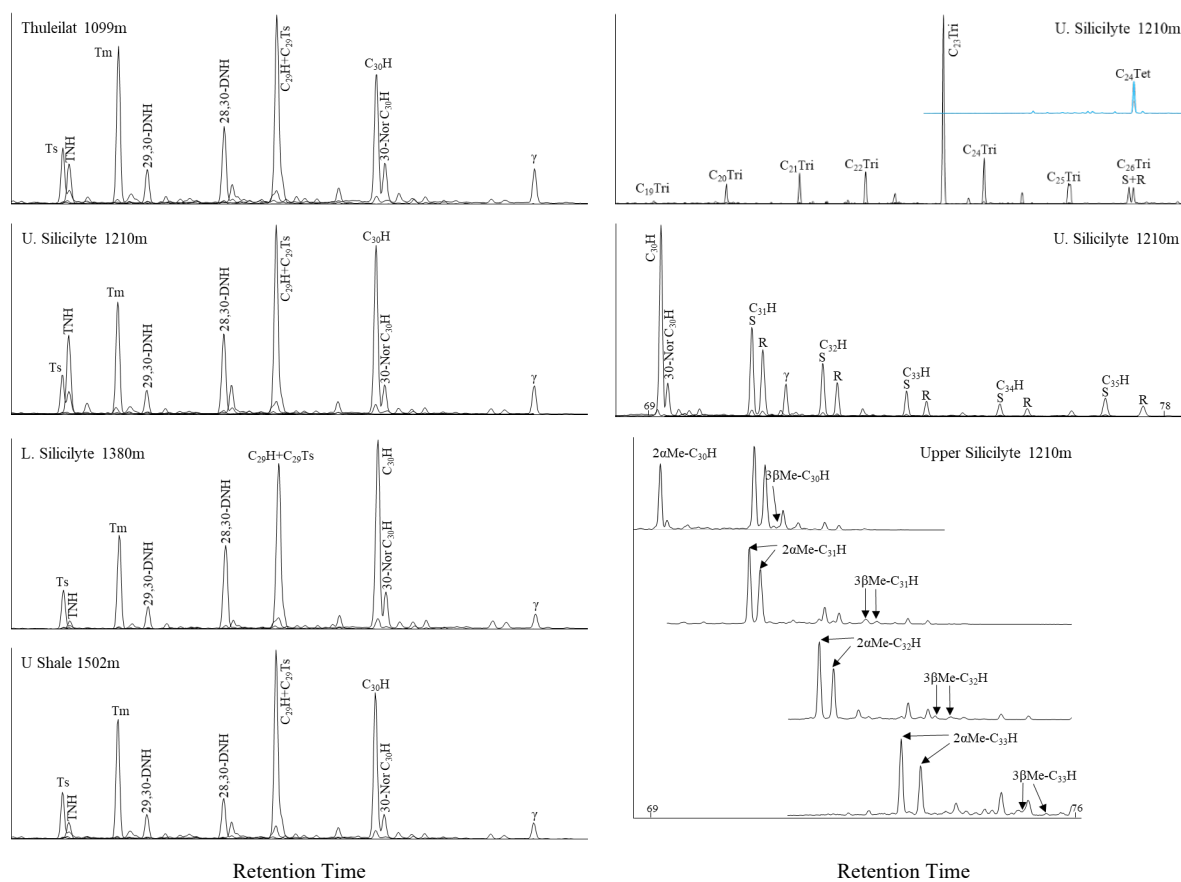
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Target Compound	Precursor	Product	Target Compound	Precursor	Product
C ₂₆ steranes	358	217	C ₂₇ trinor(neo)hopanes	370	191
C ₂₇ steranes	372	217	C ₂₇ trinorhopane	370	177
C ₂₈ steranes	386	217	C ₂₈ dinorhopane	384	191
C ₂₉ steranes	400	217	C ₂₉ norhopane	398	191
C ₃₀ steranes	414	217	C ₃₀ hopanes, gammacerane	412	191
C ₃₀ methyl steranes	414	231	C ₃₁ homohopanes	426	191
β-carotane	558	123	C ₃₂ homohopanes	440	191
γ-carotane	560	125	C ₃₃ homohopanes	454	191
chlorobactane	554	134	C ₃₄ homohopanes	468	191
okenane	554	134	C ₃₅ homohopanes	482	191
isorenieratane	546	134	C ₃₁ methyl hopanes	426	205
renieratane	546	134	C ₃₂ methyl hopanes	440	205
renierapurpurane	546	134	C ₃₃ methyl hopanes	454	205
β-series carotenoids	552	134	C ₃₄ methyl hopanes	468	205
C ₃₉ carotenoids	532	134	C ₃₅ methyl hopanes	482	205
C ₃₈ carotenoids	518	134			

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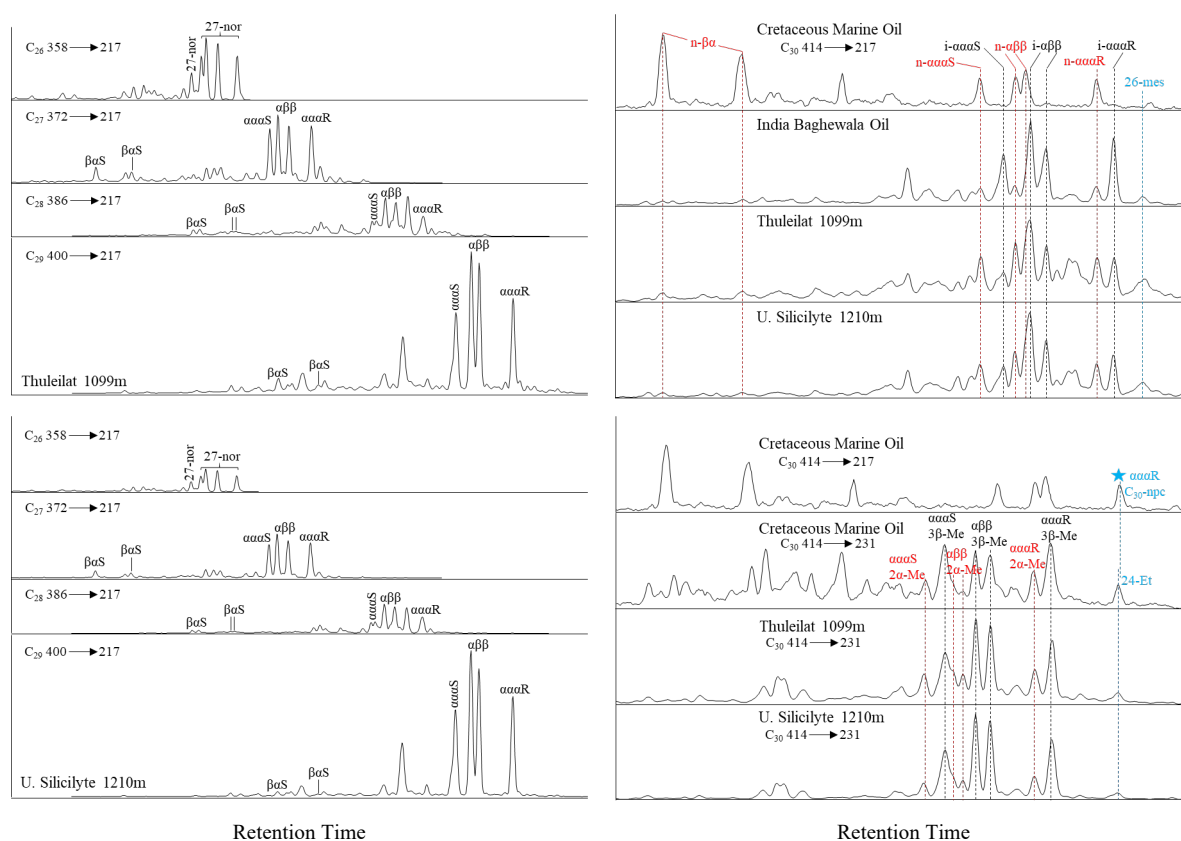


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1056 Figure S1. Representative GCMS data for hopanes, gammacerane, and tricyclic terpanes
 1057 depicted using the summation of the responses of the precursor-product transitions generated
 1058 using MRM data acquisition. C_{27-30} hopanes and gammacerane (γ) are shown in the left panel,
 1059 while tricyclic terpanes, C_{24} tetracyclic terpane, homohopanes, and methylhopanes in the right
 1060 panel. Each group of biomarkers has different range of retention time. Note the numbers in
 1061 homohopane and methylhopane figures are indicative of retention times.

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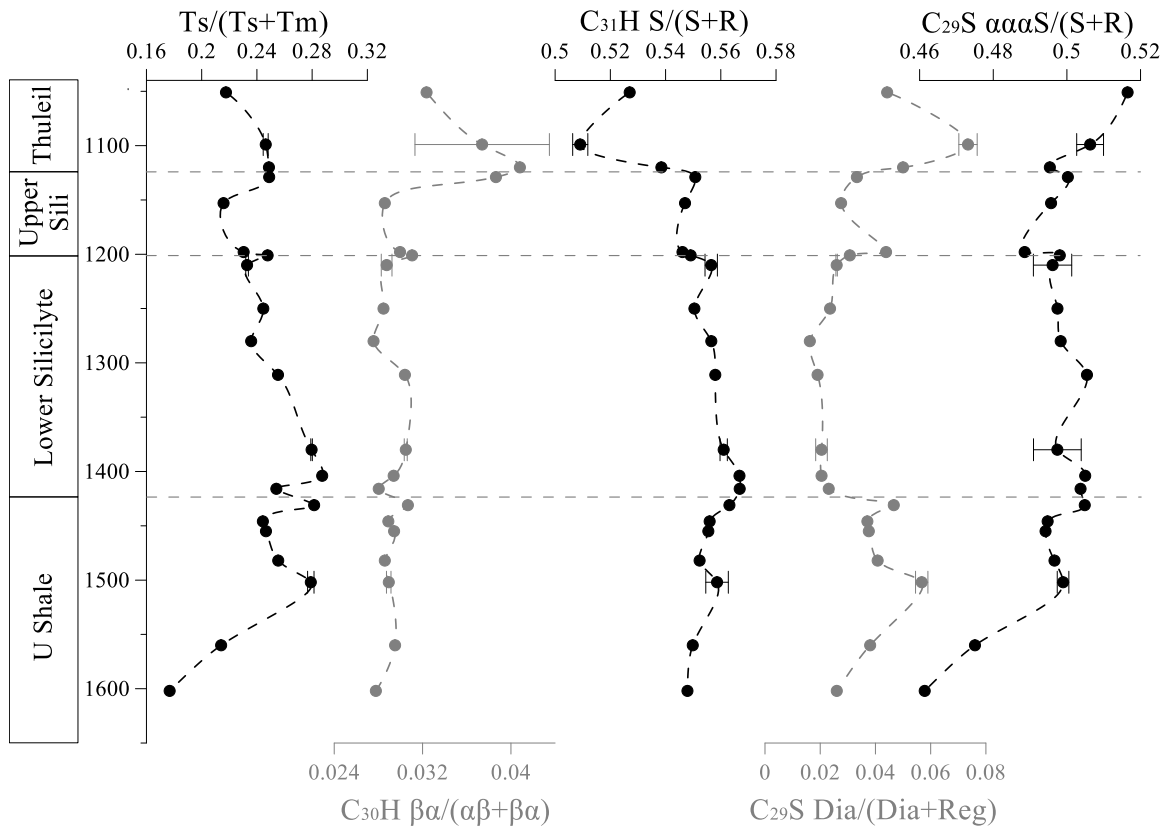
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1066 Figure S2. Representative GCMS data for steranes depicted using the summation of the
 1067 responses of the precursor-product transitions generated using MRM data acquisition. C₂₆-C₂₉
 1068 steranes are shown in the left panel and C₃₀ methyl- and desmethylsteranes in the right panel.
 1069 Each group of biomarkers has different range of retention time.

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1075 Figure S3. The downcore profiles of maturity sensitive parameters, including $Ts/(Ts+Tm)$,
 1076 $C_{30}H \beta\alpha/(\alpha\beta+\beta\alpha)$, $C_{31}H S/(S+R)$, $C_{27-29}S dia/(dia+reg)$, and $C_{29}S \alpha\alpha\alpha S/(S+R)$, where Ts is
 1077 $18\alpha-22,29,30$ -trisorneohopane, Tm is $17\beta-22,29,30$ -trisorhopane, $C_{30}H$ is C_{30} hopane, $C_{31}H$
 1078 is C_{31} homohopane, $C_{29}S$ is C_{29} sterane.

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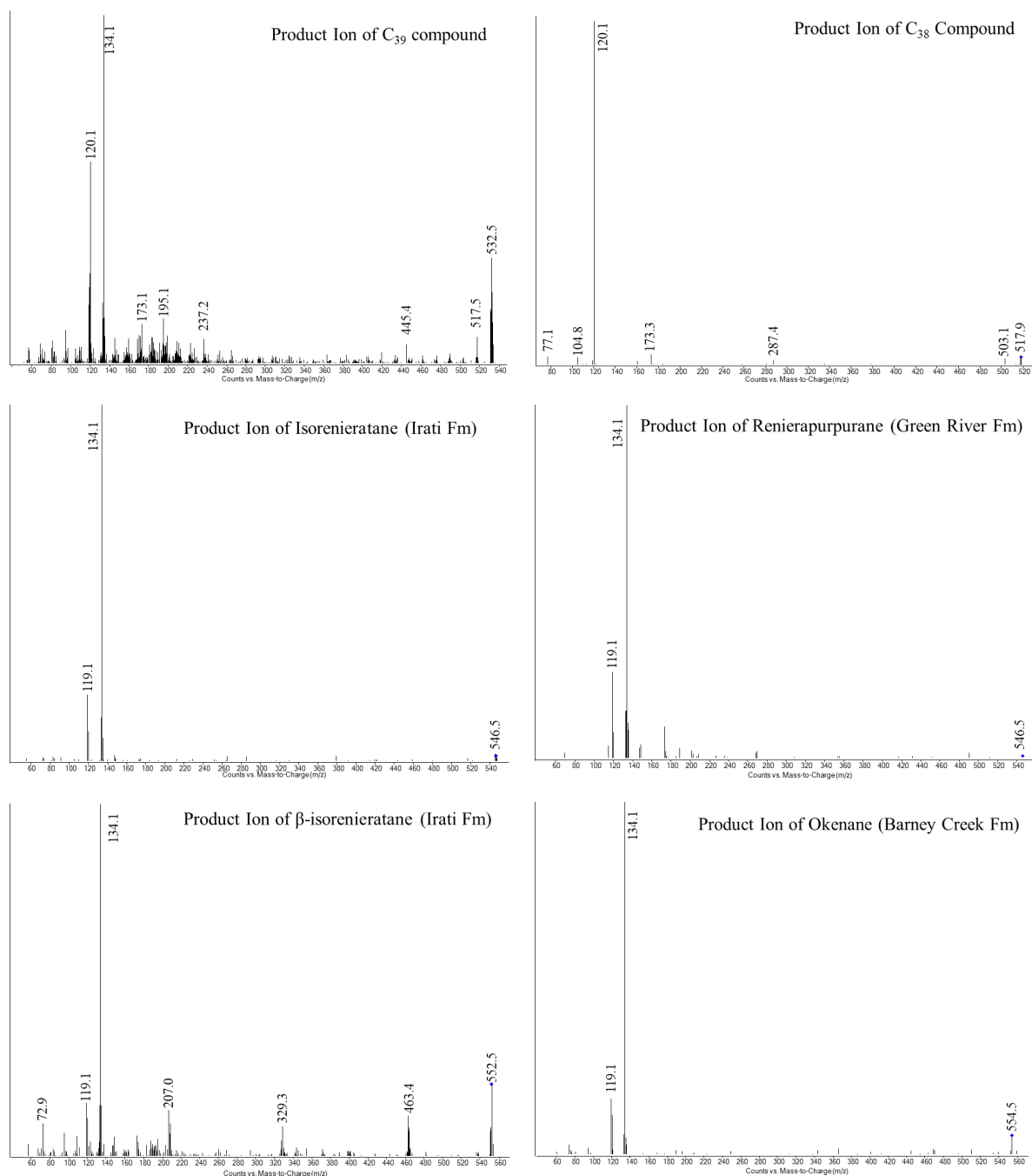
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1087 Figure S4. Product ion spectra of the molecular ions of the C₃₉ and C₃₈ diaromatic carotenoids
1088 proposed to be the diagenetic products of synechocanthin. For comparison, are the product
1089 ion spectra of isorenieratane, renierapurpurane, β -isorenieratane, and okenane from a variety
1090 of reference sediments where they are abundant.

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