## THE GEOCHRONOLOGY OF SOME PLUTONIC ROCKS

## IN **THE**

## CALAIS AREA, MAINE

**by**

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*(1965)*

Submitted in Partial Fulfillment

of the Requirements for the

Degree of Master of Science

## at the

Massachusetts Institute of Technology

January, **1967**

Signature of Author... **.. ..... 0.. ..........** Department of Geology and Geophysics, Manuary 2,1967 Certified **by.....**  $\sim$   $\sim$ .............. O Thesis Supervisor  $21 -$ Accepted **by.....** Chairman, Departmental Committee on Theses

### ABSTRACT

**A** rubidium-strontium and strontium isotopic whole-rock analysis of sixteen rocks from a suite of twenty-seven specimens collected from the plutonic complex in the Calais area in the extreme southeast of Maine has Tielded an age of 401 **± 8** m.y. and an initial ratio **of 0.7093 - 0.0016** from a computer lingqr least squares regression on ten points,(  $\lambda_{\rm ph}87 = 1.39 \times 10^{-7}$ year- **).**

The specimens analyzed include six granites and an associated quartz diorite all of which may be distinguished in the field on petrologic grounds.

The scatter of the data indicates initial inhomogeneities in the granitic bodies, especially in the case of the Baring granite, and there also appears to have been a migration of  $Sr^{87*}$  in the case of the quartz diorite which was sampled in the proximity of the contact with the Meddybemps granite.

The close agreement of the "isochron" with previous K/Ar biotite determinations precludes regional metamorphism of this plutonic body.

Recently published determinations on the Upper Silurian Eastport formation and the Lower Devonian Hedgehog formation (Bottino and Fullagar,1966) give a combined age of 413 **± 5** m.y. The age provided in this study gives an upper limit for the Silurian-Devonian boundary and this age is consistent with field relations and previous age determinations in the region.

## Acknowledgements

The writer wishes to thank his thesis supervisor, Professor H. W. Fairbairn for his interest in this study.

Professor P. M. Hurley has kindly offered to analyze two critical samples **by** isotope dilution methods.

Mr. Stanley **A.** Heath acquainted the author with the laboratory procedure and the idiosyncrasies of mass spectrometers.

Use was made of the IBM 7094 facilities of the Computation Centre, Massachusetts Institute of Technology in the least squares regression analysis of the isochron.

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## TABLE OF **CONTENTS**



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### INTRODUCTION

The plutonic rocks studied form part of an igneous complex located in the extreme southeastern portion of Maine which extends northeastward into southwestern New Brunswick. The specimens collected are, with the exception  $\mathfrak{g}$ -one (R6310), from the Calais and Robbinston quadrangles<sup>1</sup>. (Figure **1).**

This area is of considerable importance in the geochronology of New England in that these granitic rocks are intruded into Late Silurian (Ludlow or Gedinne) fossiliferous slates and are, in turn, overlain **by** the Late Devonian Perry formation, a fossil-bearing sandstone containing detritus from these granites. Previous dates obtained from this region **by** Faul (1960,1963) and Bottino and Fullagar **(1966)** provide close control for the boundary of the Silurian-Devonian periods. Such a study complements paleontological correlations in the area **by** providing "absolute" ages.

## REGIONAL **GEOLOGY**

**1.**

The region studied is underlain **by** metamorphosed Cambro-Ordovician sedimentary and volcanic rocks which are, in turn, unconformably overlain **by** thick sequences of Silurian sediments and volcanic rocks (Alcock, 1946a). In this region, the Cambro-Ordovician rocks (Charlotte group) consist essentially of argillite,quartzite, phyllites, micaceous gneisses and minor

Published **by** the **U. S.** Geological Survey as **15** minute quadrangles on the scale of **1:62,500.** Also available from the **U. S.** Army Corps of Engineers at a scale of 1:24,000 (partial coverage of the area only).



metamorphosed tuffs and mafic lava flows. Progressive zonal contact metamorphism is invariably observed between the Charlotte group and the younger intrusive rocks. The thickest sequence of the Silurian in the immediate area is the Eastport formation, consisting of thick sequences of basaltic to andesitic tuffs interbedded with shales and argillite. Near contacts with the Red Beach granite, contact metamorphic effects are prominent. Bastin and Williams (1914) assign the Eastport to the Upper Silurian on the basis of paleontology, and recently Boucot et al (1964) and Naylor and Boucot **(1965)** have placed the Eastport in the Ludlow (Upper Silurian) or possibly as Lower Devonian (Gedinnian). Present opinion appears to favor an age younger than Ludlow but older than Gedinnian (Boucot,personal communication, in Bottino and Fullagar, 1966), and the Eastport formation has been placed tentatively in the Skala interval.

Isolated basins of Upper Devonian coarse clastic and volcanic rocks of the Perry formation unconformably overlie the granitic intrusions, and its lower members are coarsely conglomeratic containing poorly sorted angular pebbles of the Red Beach granite, particles of the hornblende phase being more numerous than those of the biotite phase. The Perry formation was dated as Upper Devonian in age from fossil plant remains found in the Eastport quadrangle and also in New Brunswick (Smith and White, 1905). On a regional scale these major rock units form a belted areal pattern with



**FIGURE 3 GEOLOGIC** MAP OF ROBBINSTON **AND CALAIS QUADRANGLES, SOUTHEASTERN MAINE GEOLOGY (AMOS, 1963) AND SAMPLE LOCATIONS**

extreme linear development in a northeast-southwest direction. The intrusive bodies are generally concordant with the regional trends but often show discordant relationships locally within individual bodies.

The detailed petrology of the post-Upper Silurianpre-Upper Devonian rocks was undertaken **by** Amos **(1963).** He recognizes two major groups of rocks:

First, a medium-grained quartz diorite and an associated granodiorite adjacent to the granites form the bulk of the plutonic rocks exposed. Near the contacts with the underlying metasediments of the Charlotte group, the plagioclase is observed to become more calcic, and quartz and biotite are as usually more abundant. Toward contacts with the granites and the granodiorite, the plagioclase in the diorite becomes more sodic and zoning is considerably more prominent. The granodiorite forms a narrow discontinuous belt between the diorite and the granites. Contact relationships reveal that the diorite is younger than rocks of the Charlotte group and the gabbronorite but is older than any of the adjacent granitic rocks.

In the second major group,'no less than five types of granite have been recognized by Amos (1963)on the basis of field and petrologic criteria and these granites invariably are observed to cut the more mafic rocks in the area. Each granite is briefly described below:

## Biotite Granite of Baring:

**A** white, medium-grained subporphyritic biotite granite

forming a large body which extends northeastward into New Brunswick (Alcock,1946a). Near contacts with the older rocks, this granite is invariably more mafic and finer-grained.

## Biotite Granite of Meddybemps:

A light salmon to white medium-grained granite which is locally subporphyritic occurs in the southwest of the map area (Figures2-and **3). A** K/Ar biotite determination **by** Faul **(1960)** on this granite gave an age of 404 my.

## Biotite Granite of Charlotte:

**A** large body of very light tan to light brick-red mediumgrained granite, locally subporphyritic, occurs to the east of the Meddybemps granite, outcropping to the west of the town of Charlotte. On the basis of contact relations, the Charlotte granite is clearly younger than the diorite, gabbro-norite or members of the Eastport formation and a K/Ar biotite determination **by** Faul **(1960)** from the same locality as specimen **#2O** (R6304) of this study gave an age of 406 *my.*

### Red Beach Granite:

This granite consists of two distinct mineralogical types. The biotite granite occurs in the northwestern part and a hornblende granite in the southeastern part. Between the two is a gradational variant of both biotite and hornblende. Both phases of this granite are markedly uniform regionally although there are numerous local variations in texture and composition. The Red Beach granite is felt to be the youngest major rock unit in the plutonic belt since it cuts the Eastport formation and also contributes detritus to the Perry formation.

An average of four K/Ar biotite analyses **by** Faul **(1960)** gave an age of  $404 \pm 8$  m.y.

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## Experimental Procedures:

(i) Sampling:

**All** outcrops sampled were of such a size that there was little chance of sampling glacial boulders in error. Large blocks (about **15"** square) were broken down and smaller fresh hand specimens were removed and placed in small specimen containers at the outcrop. Since these plutonic rocks are for the most part fine-grained and no foliation was evident at the outcrop, the hand specimens chosen are believed to be well over ten times the major inhomogeneity in the rock (such as gneissosity etc. Phinney(1959)). Only fresh material having no signs of weathering or evident veining in the hand specimen was **J** chosen.

## (ii) Laboratory Preparation:

The samples were first passed through a coarse jaw crusher, care being taken to remove particles from previous samples **by** running two blank fragments. The machine was also cleaned **by** means of a jet of compressed air. The coarse powder was rolled and about **5** to **10** grams was reduced to -200 mesh in a Pica Blender Mill (Pitchford Scientific Instruments Corp.) The time required for the reduction was about **10** to **15** minutes. The ground sample was stored in a clean polyethelene vial.

(iii) Mass Spectrometry:

The strontium isotopic ratios of this study were obtained from three instruments of the M.I.T. Geochronology Laboratory, namely; Lulu, Nancy and Iris. **All** are **60** sector, six-inch radius, single filament and single Nier type collector machines utilizing a solid source. The ion beam is amplified **by** a Cary Model **31** vibrating reed electrometer, the output of which is measured on a Brown strip chart recorder. The mass range was scanned **by** a mechanical reversing magnet sweep.

In view of the possibility of drifting characteristics of each mass spectrometer, a standard was run at the beginning of use of a particular machine. These values are reported below normalized to a  $\text{Sr}^{86}/\text{Sr}^{88}$  ratio of 0.1194. (see Table 3)

(iv) Determination of Rb/Sr Ratios **by** X-ray Spectrograph:

In this study all Rb/Sr ratios were determined **by** means of a Norelco X-ray Spectrograph using Mo radiation and a topaz analyzing crystal and a 0.004 inch entrance slit. The x-ray tube was operated at **50** Kv. and **35 ma.** The amplifier gain of the pulse height analyzer was set at **60,** the base line voltage at 1.40 (x6 volts) and the window at **6.0** (x3 volts). The **DC** power supply for the scintillation counter was set at 724 Kv.

Preliminary rapid analysis of all samples collected was made using a **G-1** standard containing about **260** ppm Sr and 220 ppm Rb. This was done in order to obtain a suitable spread in Rb/Sr ratios. These data are given in Table I.

Precise determination of the Rb/Sr weight ratio was

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made by scanning from  $40^{\circ}$  to  $34^{\circ}$  20 at a rate of  $\frac{1}{2}^{\circ}$  20 per minute with a chart speed of **1** per inch. The tracing was usually made twice to lessen ambiguity in the determination of the base line correction. Setting the goniometer at the  $K_{\alpha}$  position for Rb (at  $39.9^{\circ}$  20) and Sr (at  $37.65^{\circ}$  20), the counting rate for lo,000 counts was determined at intervals of 2,000 counts. Also, the counting rate for 2,000 counts was made at  $43^{\circ}$ 20 to determine the anchor point for the base line correction. Before each scan, the zero point was checked to within one half the pen width and the recorder was recalibrated each time. Two runs of three scans each, sixtin all, were made on each sample.

The powdered sample was packed into the holder **by** means of a spatula ground flat on each edge. Each sample was tamped until a flat, smooth surface was obtained.

Each scan was alternated with a dunite standard **(W.-1)** which was enriched in Rb(as RbCl) and Sr (as SrCO<sub>3</sub>) to about **2,500** ppm each. The counting rate was determined for **96,000** counts in three **32,000** count increments. **A** background correction was made and the Rb/Sr ratio was calculated.

### (v) Chemical Preparation:

The following procedure was followed for the dissolution of the silicates and the isolation of the strontium. **1.** Sufficient sample was weighed out on a beam balance such that there was a minimum of **80** u gram Sr in solution. 2. The sample was wet with several milliliters of distilled water.

**3.** Reagent grade HF was added, **10** ml per each **0.5** gm of the powdered sample added followed **by 1.5** ml **HC10** per **0.5** gram sample. The amount of HF added was kept to a minimum since there is approximately 0.02 u gram Sr per gm of sample per analysis contamination from the HF. (see M.I.T. Annual Report **(1962)** p.20,Tables **1** and 2 for degree of contamination). 4. The samples were left for several hours on a steam bath to dissolve (on low heat if left overnight). The samples were then stirred frequently with Teflon stirrers to hasten solution. **5.** When a gel formed after the evaporation of the HF, another small amount was added and allowed to evaporate while stirring. **6.** About **150 ml** of distilled **2 N..HC1** was added to the platinum dishes under high heat, The sample was evaporated again to near dryness.

**7.** The samples were then removed from the steam bath and 20 **ml** of **2N** HCl was added, the solution stirred (with some heating if required) and about 20 ml of distilled water was added. **8.** The solution was allowed to cool for 2 to **3** hours and any Rb salts were allowed to crystallize.

**9.** The samples were filtered using **<sup>a</sup>**high grade ashless filter paper and the clear filtrate was carefully introduced into the column, care being taken not to disturb the level of the column resin. A  $\text{Sr}$ <sup>85</sup> tracer was added (to about 3x to  $4x$ background level).

10. The resin used was Dowex 50W-x8 and the ions were eluted using **2N 1l.**

The order of emergence of the eluted ions is:



**By** means of a flame test, the Ca was discarded after elution and the Sr was collected in nalgene beakers. The beakers containing the most strontium were determined **by** means of a scintillation counter (on a **10** second count). **Of** some **6** to **8** beakers of solution only two were chosen and the remainder discarded.

**11.** The contents of the beakers was evaporated on a steam bath to about **10** ml. then the solution was carefully transferred to a Vycor beaker and was heated on a hot plate to near dryness. Several millilitres of  $HClO_{\mu}$  were added to decompose any carbonaceous residue then several millilitres of distilled **INO3** were added, the solution taken to dryness again, and another final addition of a few millilitres of **HNO <sup>3</sup>**was made after which the solution was taken down to complete dryness.

### DISCUSSION OF RESULTS **AND CONCUISION**

Twenty-seven field samples were collected and from these, sixteen were chosen for isotopic analysis on the basis of their variation in rubidium-strontium ratio (Table **1)** and also on their petrologic variety.

It is seen from Plate I that ten of the samples show a reasonably linear distribution and, though not strictly an isochron, the straight line defined **by** these points gives an average age for this intrusive event.

The scatter of the data is somewhat greater than anticipated from the preliminary work on these rocks (Spooner,1966) especially for specimens of the Baring granite. Specimen R6298 of the Baring granite was collected close to the contact with the quartz diorite so that in all likelihood there has been a migration of radiogenic strontium from within this system. The same case obtains for the quartz diorite, R6305 and R6297 which was, unavoidably sampled from areas containing extensive granite pegmatite veining.

Samples R6306, R6307 and R6308 were taken from the Baring granite proper, well away from border phases; nonetheless, the scatter is indicative of non-homogeneity at the time of emplacement. The smaller-igneous bodies exhibit a greater degree of homogenization as seen from Plate I.

Only ten points showing the least degree of scatter were

used in the least squares regression analysis. Despite the closeness of fit of sample R6306 to the-isochron, the incorporation of this point to the exclusion of R6308, based on geologic criteria. did not seem justifiable.

It is difficult to arrive at a conclusion regarding a comagmatic source for these approximately coeval granitic rocks owing to the large amount of scatter about the "isochron". On the basis of the samples collected however, there appears to be a twelve-fold increase in Rb/Sr ratio with differentiation from the Baring to the **Red** Beach granite, with the Meddybemps and the Charlotte granites between, if a comagmatic assumption is made.

The IBM 7094 program used in the least squares regression in shom in Appendix II and was provided **by** Derek York **(1966)** of the University of Toronto. **.**

An initial slope **of 0.00560** was chosen on a visually fitted line passing through the coordinates of the arithmetic mean. The least squares slope obtained was **0.00558** with a difference of 0.4<sup> $\frac{4}{6}$ </sup> from the visually estimated slope. Following the procedure for calculating the age of the system given in Appendix I, an age of  $401 \pm 8$  m.y. was obtained with an initial ratio of **0.7093** . 0.0016.

. This result is in good agreement with the  $404$  m.y. K/Ar age obtained on the Meddybemps granite, the 401 m.y. age on the Red Beach granite and the  $406$  m.y. age on the Charlotte granite Faul  $(1960)$ . It is also in accord with the 412  $\pm$  5 m.y. and 413 - **10** m.y. ages obtained **by** Bottino and Fallagar **(1966)** on

the older volcanics of the underlying Eastport formation.

The age obtained in this study confirms a post Silurianpre-Upper Devonian intrusive event at  $401 \stackrel{+}{\smile} 8$ .  $m_ry$ . and in conjunction with the ages obtained **by** Bottino and Fullagar **(1966)** provides a closer limit on the position of the Silurian-' Devonian boundary.

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Approximate Rubidium-Strontium Content of Rocks Studied:				
$M$ . I. T. #	Field $#$	$Rb$ (ppm)	$Sr$ (ppm)	Rb/Sr
R 6285	1	261	30	8.7
R 6286	$\boldsymbol{z}$	235	41	5.7
R 6287	3	250	37	6,8
R 6288	4	271	51	5.3
R 6289	5	230	43	5.3
R 6290	6	261	51	5.1
R 6291	$\boldsymbol{7}$	267	86	3.1
R 6292	8	248	41	6,0
R 6293	9	109	130	0.84
$R$ 6294	$10$	122	126	0.97
R 6295	11	211	.40	5.3
R 6296	$12\,$	75	112	0,67
R 6297	13	30	159	0.19
R 6298	14	47	235	0,20
R 6299	15	162	88	0,18
R 6300	16	77	153	0,50
R 6301	17	130	134	0.97
R 6302	18	220	87	2,5
R 6303	19	180	88	2,0
R 6304	20	220	53	4.2
R 6305	21	17	216	0.08
R 6306	22	126	70	1.8

TABLE I

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 $\mathcal{N}$ 

# TABLE II





\*at *95%* 2a Confidence Level.

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Mass Spectrometric Determinations:



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October 24,1966 Iris



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TABLE IV

Mass Spectrometric Determinations:



\* run number;\*\* (L)= Lulu, (I)= Iris, (N)=Nancy \*\*\*The  $sr^{\circ}$  /Sr values have been normalized to 0.1194.

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Mass Spectrometric Determinations cont'd:



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 $\ddot{\phantom{a}}$  $\ddot{\phantom{1}}$ 

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Mass Spectrometric Deterinations. cont'd:

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 $\overline{a}$ 

 $\mathcal{F}$ 

 $\ddot{\phantom{a}}$ 

Field Description of Samples:



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**A** thin section examination was made on each specimen collected and almost without exception the potash feldspar showed varying degrees of sericitization and incipient alteration to kaolinite, However, in no case was there evidence of secondary veining either **by** quartz or carbonate. It is felt that there has been no redistribution of rubidium or strontium from the alteration (i. W. Fairbairn, personal communication).

#### APPENDIX I

Rubidium-strontium dating technique is based upon the  $\beta$ -decay of Rb<sup>87</sup>:

$$
Rb \xrightarrow{87} \qquad B \qquad Sr \xrightarrow{87*}
$$

with an approximate half-life **of 50** billion years. In addition to the Sr<sup>87\*</sup> produced radiogenically, natural strontium has four 84 **86 87 88** stable isotopes: Sr , Sr ,Sr , and Sr . In any chemical system that has remained closed with respect to rubidium and strontium, the amount **of** Sr present as a function of time "t" is the initial abundance of  $sr^8$ :  $sr^8$  plus the time integrated 87 87\* result of Rb decay to Sr **\*** Thus we have:

$$
S_r^{87} \bigg]_t = S_r^{87} \bigg]_0 + S_r^{87*} \tag{1}
$$

The basic equation for radioactive decay is:

$$
dN/dt = -\lambda N \tag{2}
$$

**87 87\*** integrating and substituting for Rb and Sr yields:

$$
87^* = Rb \t(e^{\lambda t} - 1)
$$
 (3)

substitution into equation **(1)** and expressing relative to **86** Sr gives:

$$
\frac{87}{86} = \frac{87}{86} + \frac{87}{86} + \frac{87}{86} \text{ (e}^{\lambda t} - 1) \quad (*)
$$
\n
$$
\text{. Sr} \quad \text{. Sr} \quad \text{. Sr} \quad \text{. Br} \quad \text{. Ar}
$$

Where:  $t=$  age of the system since closure with respect to rubidium and strontium.

 $87$  .  $11 -1$  $\lambda$  = decay constant for Rb<sup>oot</sup> in reciprocal years.  $(\lambda = 1.39 \times 10 \times y^6)$ .  $\texttt{Sr} ^{87}/\texttt{Sr} ^{86}$  = the atomic ratio of these to isotopic species when

the system became closed **(** t=0 **)** to Rb and Sr.  $\texttt{Srr}^{87}$  /Sr  $\big|_{+}$  = atomic ratio of Sr  $^{87}$  to Sr  $^{86}$  at t years following the closure of the system. **87** 861 **87 86** Rb /Sr  $\vert_{+}$  = atomic ratio of Rb to Sr at t years following

last closure of the system.

In the B.P.I.\* method, the present day  $Rb^{87}/sr^{86}$  ratio is plotted on the abcissa and the present day  $\text{Sr}$  /Sr atomic ratio is plotted along the ordinate such that the slope defined **by** points falling along a straight line ( an isochron) is given **by:**

$$
\tan \theta = (e^{\lambda t} - 1) \tag{5}
$$

Bernard Price Institute.

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スソ
SIBJOB MAPGO
siBFTC List, DFCK
      0)DIMENSION X(50),Y(50),U(50),V(50),P(50),Q(50),W(50),SQW(50),
    1RESX(50),RESY(50)
 100 READ(5,1)B,N,(X(I),Y(I),P(I),Q(
I) I=1,N)
1 FORMAT (F15.8,Il0/(2F15.7,2E15.
5))
      SUMW=0.
      SUMA = 0.SUMB=0.
      SUMC=0.
      SUMD=0.
      SUME=0.
      SUMS=0.
      SUMT=0.
      XBAR=0.
      YBAR=0.
      DO 2 I = 1 NW(I)=P(I)*Q(I)/(B*B*Q(I)+P(I))
      SQW(T)=W(T)**22 SUMW=SUMW+W(I)
      DO 3 I=1,N
      XBAR=XBAR+W(I)*X(I)/SUMW
 3 YBAR=YBAR+W(I)*Y(I)/SUMW
      DO 4 I = 1, NU(1)=X(1)-XBARV(I)=Y(I)-YBAR
      SUMA=SUMA+SQW(I)*(U(I)**2)/P(I)
      SUJMB=SUMB+SQW(I)*U(I)*V(I)/P(I)
      SUMC=SLJMC+SQW(I)*(V(I)**2)/P(I)
      SUMD=SJMD+W( I )*(U( I ).**2)
 4 SUME=SUME+W(I)*Uj(I)*V( I)
      COA=0.6666667*SUMB/SUMA
      COB=(SUMC-SUMD)/(3.0*SUMA)
      COC=-SUME/SUMA
      CPHI=(COA**3-1.5*COA*COB+0.5*COC)/(COA**2-COB)**1.5
      IF(CPHI**2-1.0)6,6,10
 6 ALPHA=(SQRT(1.0-CPHI**2))/CPHI
      IF(-ALPHA) 7,7,8
 7 PHI=ATAN(ALPHA)
      GO TO 9
 8 PHI=3.1415927+ATAN(ALPHA)
 9 SLOPEA=COA+2.0*SQRT(COA**2-COB)*COS(PHI/3.0)
      SLOPEB=COA+2.0*SQRT(COA**2-COB)*COS((PHI+6.2831854)/3.0)
      SLOPEC=COA+2.0*SQRT(COA**2-COB)*.COS((PHI+12.566371)/3.0)
      GO TO 30
 10 A=3.0*(COB-COA**2)
      C=-2.0*(COA**3)+3*0*COA*COB-COC
      Z=(-C/2.0+SQRT((C**2)/4.0+(A**3)
/27.0))**(1.0/3.0)
      V=(-C/2.0-SQRT((C**2)/4.0+(A**3)
/27.0) )**(1.0/3.0)
      SLOPEA=Z+V+COA
      SLOPEB=0.00000000
      SLOPEC=0.00000000
 30 AINT=YBAR-SLOPEA*XBAR
      BINT=YBAR-SLOPEB*XBAR
      CINT=YBAR-SLOPEC*XBAR
      DO 31 T=1,N
      SUMS=SUMS+W(I)*(SLOPEC*U(I)-V(I)
)**2
      SUMT = SUMT+W(T)*(X(T)**2)RESX(I)=-(SLOPEC)*W(I)*(CINT+SLOPEC*X(I)-Y(I))/(P(I)*X(I))
 31 RESY(I)=W(I)*(CINT+SLOPEC*X(I)-Y
(I))/(Q(I)*Y(I))
      AN=N
      SIGMAB=SQRT(SUMS/(
(AN-2.0)*SUMD))
      SIGMAA=SIGMAB*SQRT
(SUMT/SUMW)
```
**5Z0**

WRITE(6,5)SLOPEA,AINT,SLOPEB,BINT,SLOPEC,CINT,XBAR,YBAR,SIGMAA, 1SIGMAB, B

**5 FORMAT(10X, 8HSLOPEA= ,F15.8,5X,6HAINT= ,F15.8//10X, 8HSLOPEB= ,** lF15\*8i5X,6HBIN **\*T=** ,F1S. 8//1OX,8HSLOPEC= ,F15\*8t5Xq6HCINT= tF15\*8// 210Xs **8H** XBAR= )F15o8,5X,6HYBAR= **9,F15.8//1OX, 8HSIGMAA= 9F15\*8,** 33X9 8HSIGMAB= **9F15o8//1OX9 8H** B= **,F15\*8)**  $WRITE(6, 20)$  ( $RESX(1)$ ,  $RESY(1)$ ,  $I=1$ ,  $N$ )

20 FORMAT(10X,28HRESX RESY//(9X,F11.8,13X,F11.8))  $WRITE(6 \cdot 21)(X(I) \cdot P(I) \cdot Y(I) \cdot Q(I) \cdot I=1 \cdot N)$ <br>FORMAT(56H  $X$  P Q

21 FORMAT(56H X P Y **1//(F15.792X9El5\*5 ,2XFl5o7,2XEl5o5))** WRITE(6999) **99** FORMAT(1H1) **GO** TO **100**

**END**



Geology mapped **by D.** H. Amos in **1950, 1956,** and **1957,** assisted **by** L. **G.** Toler, **C. A.** Chapman, **J. A.** Wehrenberg, and W. **A.** Weeks. Topography **by U. S.** Geological Survey, **U. S.** Army Corps of Enineers, and International Boundary Commission

#### GEOLOGIC MAPI OF THE ROBBINSTON **AND** CALAIS **QUADRANGLES, SOUTHEASTERN MAINE**