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5.2. Expert Experimentalist Rating: “A Heart as Strong as Iron”

Techniques Checklist:

- Use of a centrifuge

Equipment:

- Disposable UV-Vis cuvettes (1-mL capacity)
- Pipetmen: 20 P, 100 P, 1000 P
- Pipet tips
- Eppendorf tubes (safe-lock)
- Hot plate
- Centrifuge
- Boiling plate or rack to hold Eppendorf tubes
- Large crystallizing dish

Goals:

- From the CC-level experiment, you know the concentration of protein in your sample. Now you will determine the concentration of iron in bovine heart cytochrome c.

Experiment Outline:

The Ferrozine Assay

Ferrozine is an iron-chelating agent. When it forms a complex with ferrous iron (Fe^{II}), it shows a characteristic UV-Vis absorption at 562 nm. By comparing the A₅₆₂ of your sample to a calibration curve of iron standards, you will determine the concentration of iron in your protein sample.

Solutions provided by your TA:

- Fe AA standard (AA = atomic absorption)
- Buffer - 25 mM MOPS, pH 7
- Ultrapure HNO₃ (5 M)
- 75 mM Ascorbic acid solution
- 10 mM Ferrozine solution
- Saturated ammonium acetate solution

1. Preparation of Standards:

- Prepare a fresh set of iron standards in 2 mL Eppendorf tubes, as illustrated below. Carefully label each tube. Also fill 2 tubes with 300 μL of your protein sample.

μL of Fe AA standard (99 $\mu\text{g}/\text{mL}$)	μL of Buffer to add
0	300
6	294
12	288
18	282
24	276

- Add 30 μL of ultrapure HNO_3 (5 M) to each standard and sample tube.
- Place the closed Eppendorf tubes in a rack, and boil them for 30 minutes in a hot water bath (a large Pyrex dish over a heating plate).
- Centrifuge for 1-2 minutes, making sure the centrifuge is properly balanced.
- Remove 300 μL of the supernatant liquid from each tube, and transfer to fresh tubes (labeled!).
- Add 1020 μL of distilled water.
- Add 60 μL of 75 mM ascorbic acid.
- Add 60 μL of 10 mM ferrozine.
- Add 60 μL of saturated ammonium acetate.
- Shake each tube and wait 10-15 minutes, (the solutions should become purplish in color).
- Transfer to a 1.5 mL cuvette, and determine the A_{562} for each standard and your two samples.
- Generate a calibration curve of A_{562} vs. $[\text{Fe}]$ from your standards.
- Determine the $[\text{Fe}]$ in your unknown.

Results

• To obtain your "EE Rating" in Protein Assays and Error Analysis, the line fit for your standard curve must have a 0.995 correlation coefficient or higher. Additionally, the absorbance values for your unknown samples must have a standard deviation of 0.035 or less. Finally, you must determine the number of molecules of iron per molecule of protein.