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2.626 Fundamentals of Photovoltaics
Fall 2008

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Light Absorption, Charge Excitation and Transport

Lecture 4 – 2.626

Tonio Buonassisi

Concept Quiz: Summary

Problem 1: *Great!*

Identify which unit denotes “power”, and which denotes “energy”: kWh, kW.

Problem 2: *So-so...*

For a solar system rated at $2 \text{ kW}_{\text{peak}}$ (the peak power), provide an estimate of kW_{ave} (the average power) this system will produce in Boston. (Full credit for answers accurate to within a factor of two).

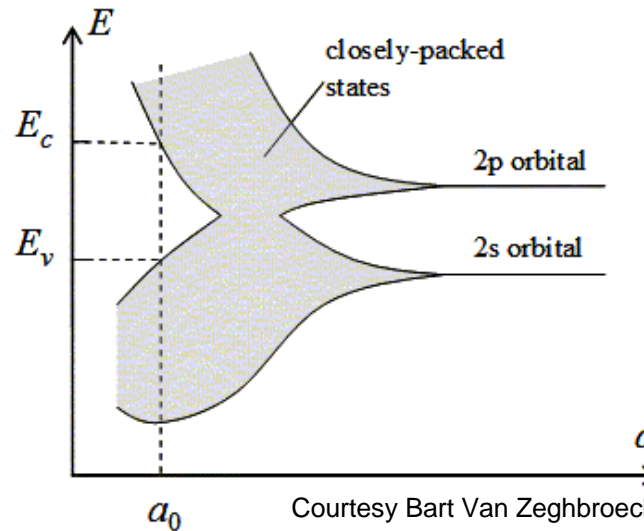
Quick Outline of Today's Lecture

Last class, we discussed *light absorption, charge excitation, and charge carrier mobility* in semiconductors, organic molecules, and nanoparticles.

This class, we discuss *free carrier concentrations, conductivity, and recombination*.

Review: Bandgap

Arises due to interfering atomic electron orbitals.



Courtesy Bart Van Zeghbroeck. Used with permission.

http://ece-www.colorado.edu/~bart/book/book/chapter2/ch2_3.htm

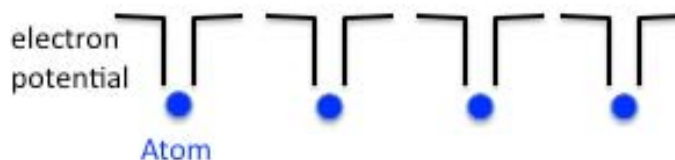
Schematic of a repeating Coulomb potential in a crystal



$$\psi_{n\mathbf{k}}(\mathbf{r}) = e^{i\mathbf{k}\cdot\mathbf{r}} u_{n\mathbf{k}}(\mathbf{r}).$$

Created by repeating crystal potential.

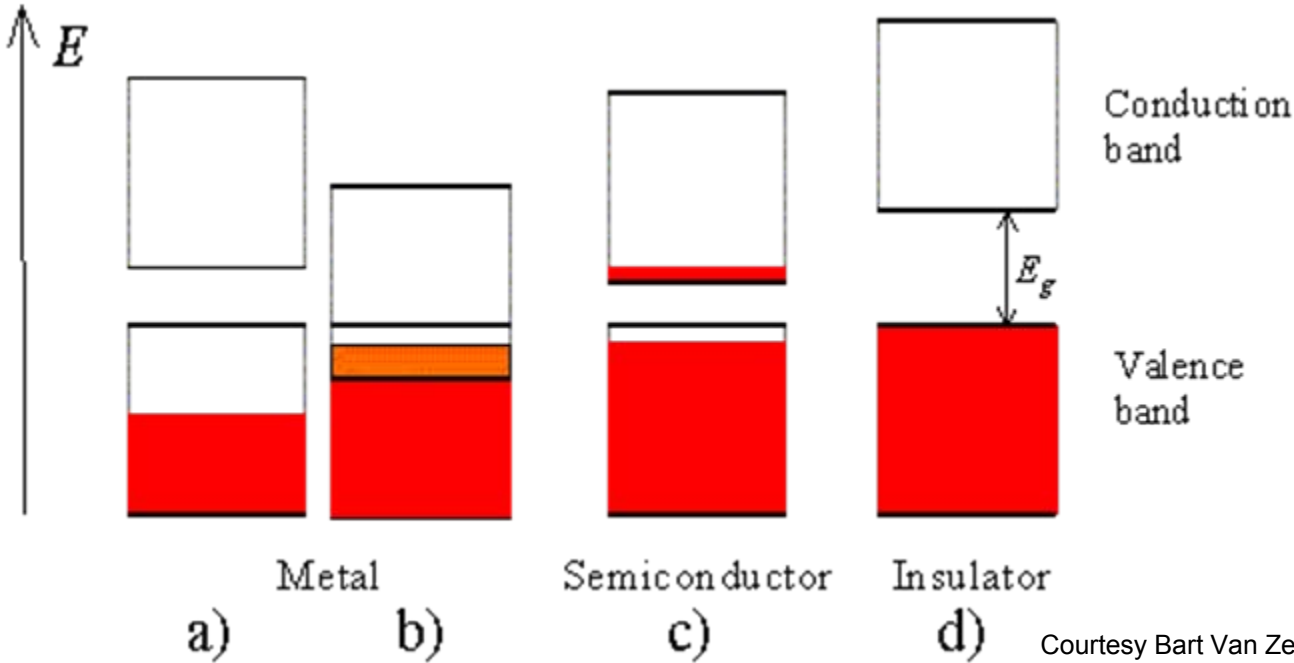
Kronig-Penney idealization of a repeating Coulomb potential in a crystal (easier to solve numerically)



Solve Schrödinger's equation → two possible solutions:

- (1) Electron wavefunction centered on atoms (bound state)
- (2) Electron wavefunction centered between atoms (excited state).

Classes of Materials, based on Bandgap



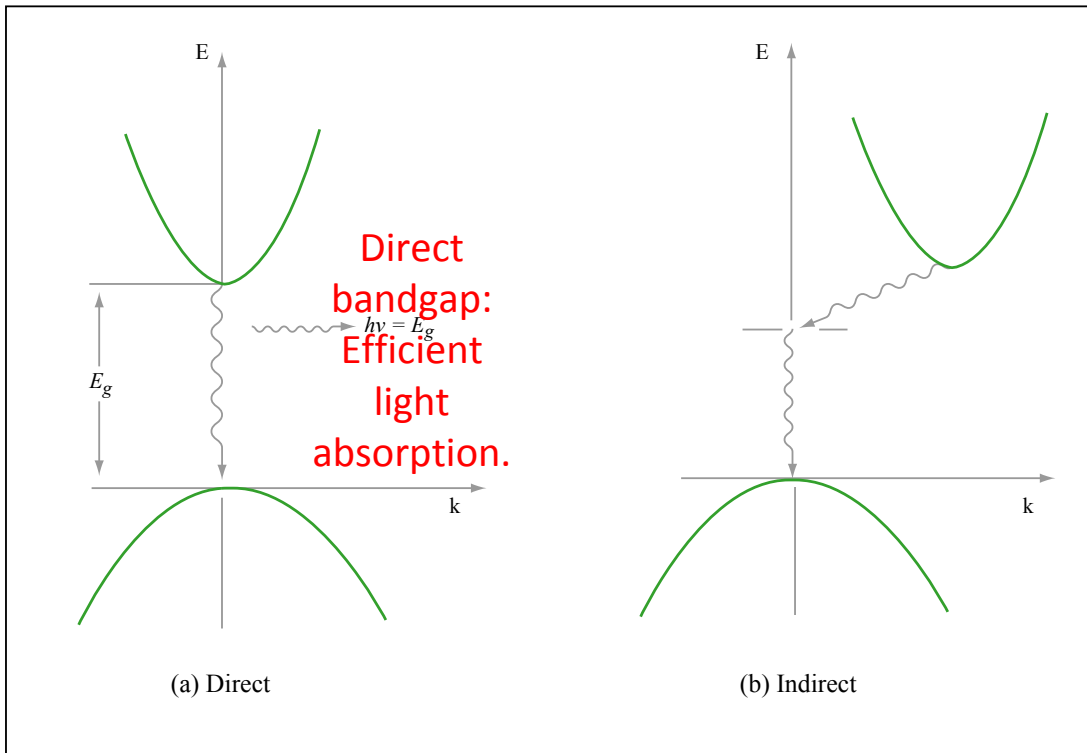
Courtesy Bart Van Zeghbroeck. Used with permission.

http://ece-www.colorado.edu/~bart/book/book/chapter2/ch2_3.htm#fig2_3_7

Review: Optical Absorption in Semiconductor

Material properties described by the **energy band diagram**:

1. Absorption of light.



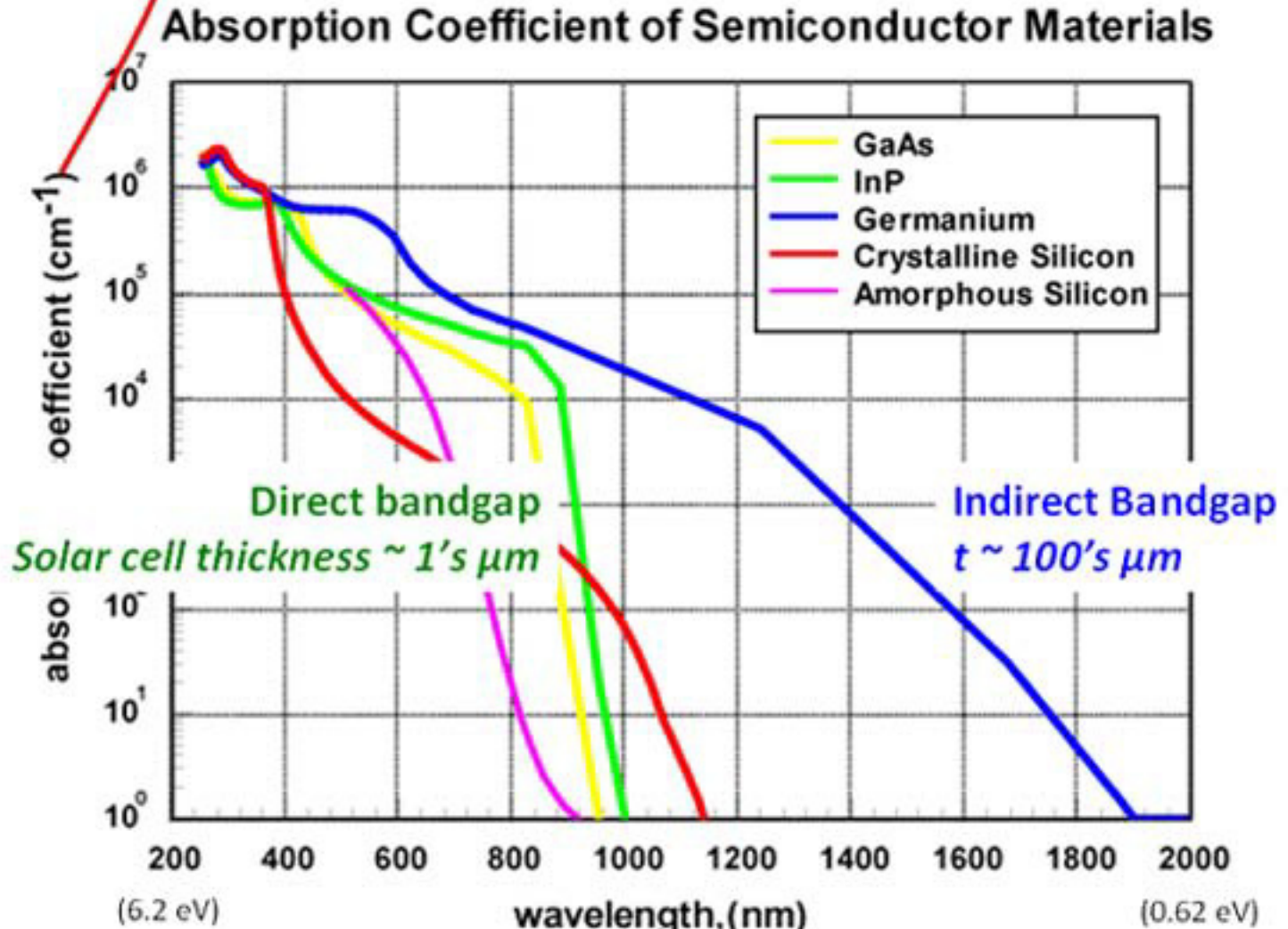
Direct bandgap:
Efficient light absorption.

Indirect bandgap:
phonon-assisted transition.
Less efficient light absorption.

Figure by MIT OpenCourseWare.

Absorption Coefficient (α) for different materials

$$I = I_0 \cdot e^{-\alpha \cdot l}$$

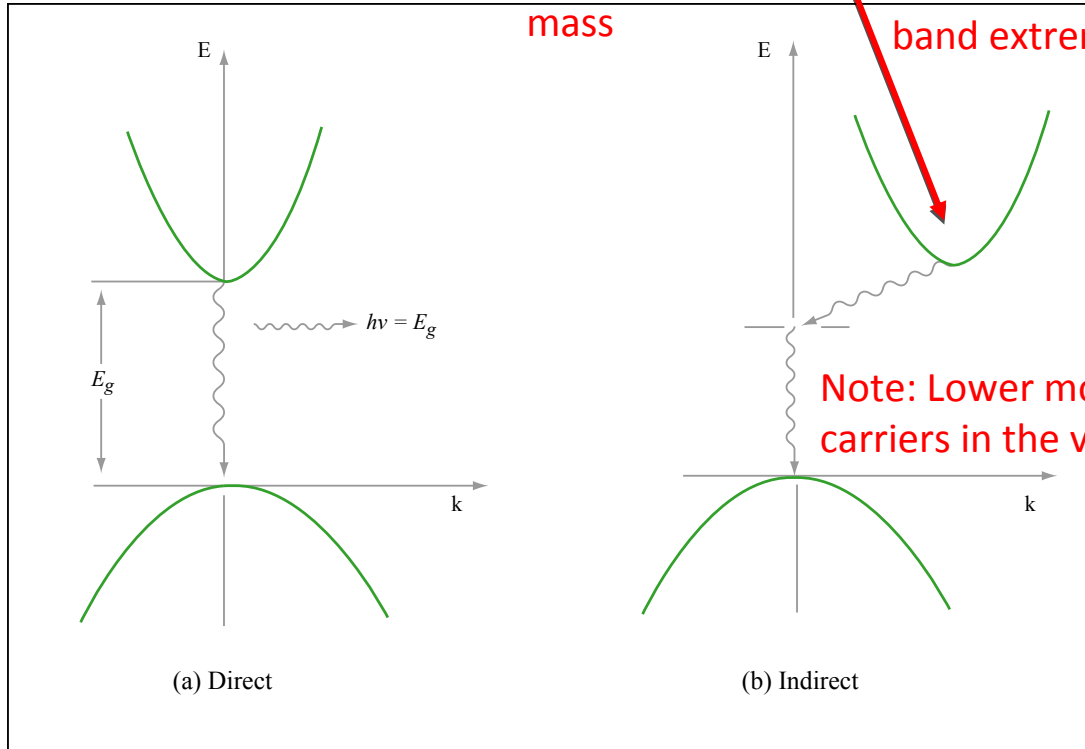


Review: Carrier Mobility

$$\mu = \frac{q}{m \cdot \nu_m} = \frac{1}{m^*} = \frac{1}{\hbar^2} \frac{d^2 E}{dk^2}$$

Effective mass

Curvature at band extremum



Note: Lower mobility for carriers in the valence band!

Material properties described by the **energy band diagram**:

1. Absorption of light.
2. Carrier mobility.

Figure by MIT OpenCourseWare.

μ = mobility

q = charge

ν_m = momentum transfer collision frequency

m = mass

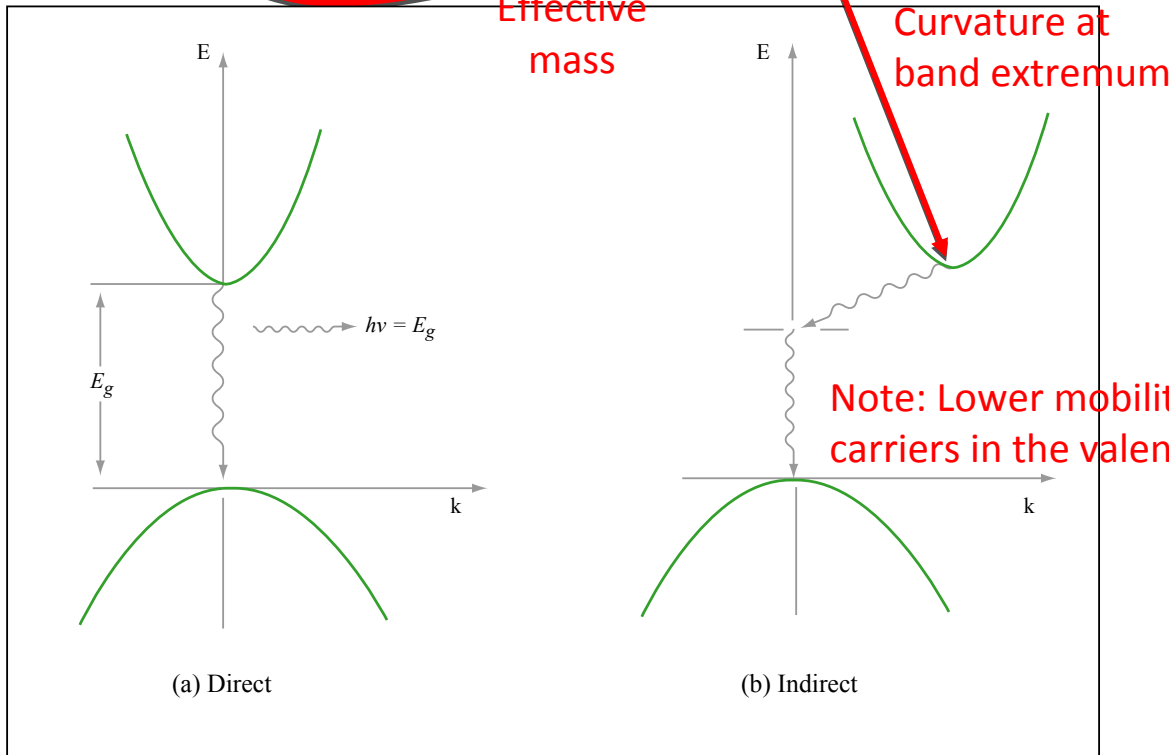
m^* = effective mass

h = Planck's constant

Question: Carriers in *which band* generally have higher mobility?

Review: Carrier Mobility

$$\mu = \frac{q}{m \cdot \nu_m} \quad \frac{1}{m^*} = \frac{1}{\hbar^2} \frac{d^2 E}{dk^2}$$



Material properties described by the **energy band diagram**:

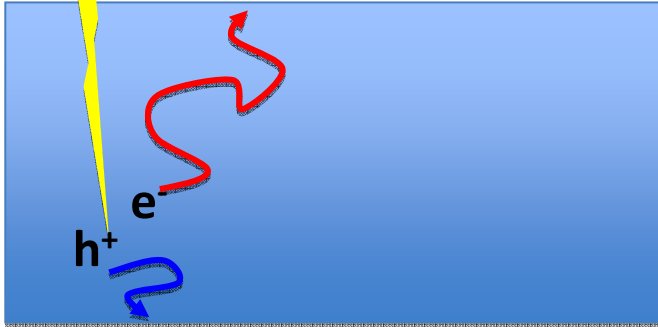
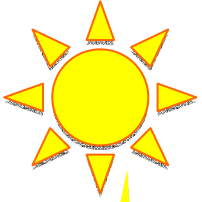
1. Absorption of light.
2. Carrier mobility.

Figure by MIT OpenCourseWare.

μ = mobility
 q = charge
 ν_m = momentum transfer collision frequency
 m = mass
 m^* = effective mass
 h = Planck's constant

Carrier Mobility, Diffusion Length, and Lifetime

Once we know the mobility, we can calculate how far a carrier can move within the crystal... *(and answer the all-important question: Can a photoexcited carrier make it out of the device?)*



Relationship between mobility and diffusivity:

$$\mu = \frac{q}{kT} D$$

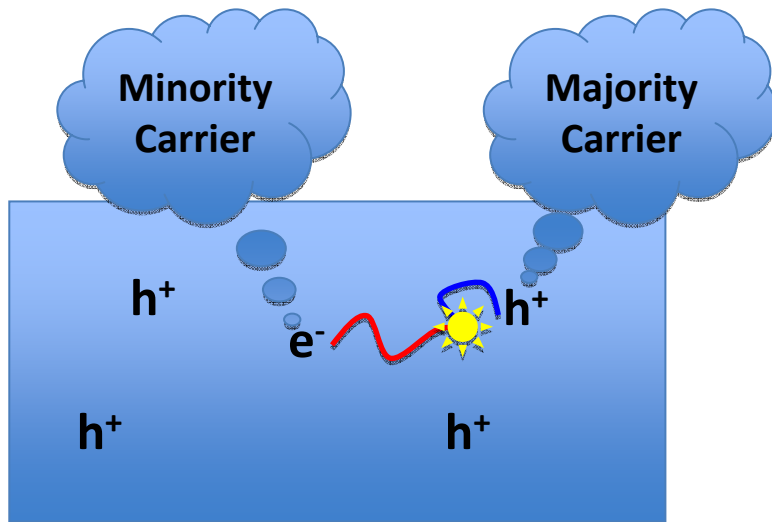
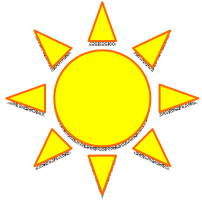
D = diffusivity
 μ = mobility

Relationship between diffusivity and diffusion length:

$$L_{\text{diff}} = \sqrt{D \cdot \tau}$$

L_{diff} = diffusion length
 τ = carrier lifetime

Carrier Lifetime and Recombination



Bulk Minority Carrier Lifetime:

$$\tau = \frac{\Delta n}{R}$$

Δn = Excess minority carrier concentration
 R = Recombination rate

First approximation of minority carrier lifetime, for low injection (e.g., illumination) conditions.

Order of Magnitude:

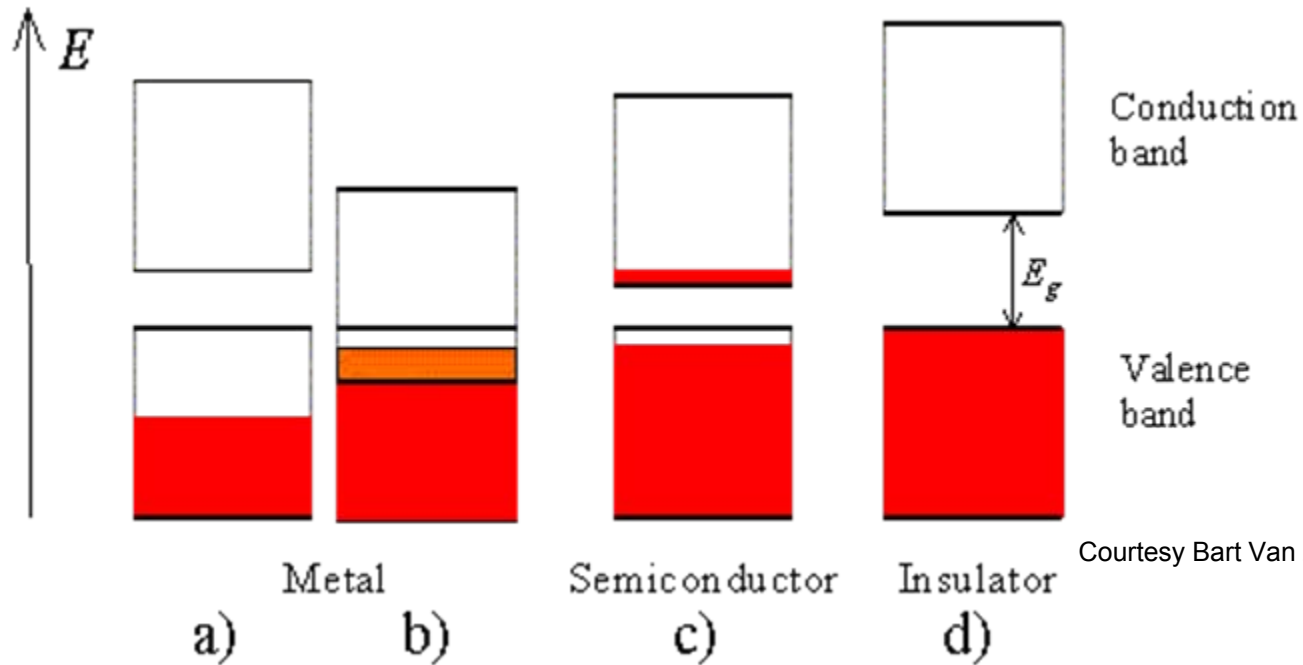
For an indirect bandgap material, τ is typically on the order of a few μs to a few ms.

For a direct bandgap material, τ is typically on the order of 10's of ns to 1 μs .

Conductivity

Question: Which electrons in a material contribute to conductivity?

Classes of Materials, based on Bandgap

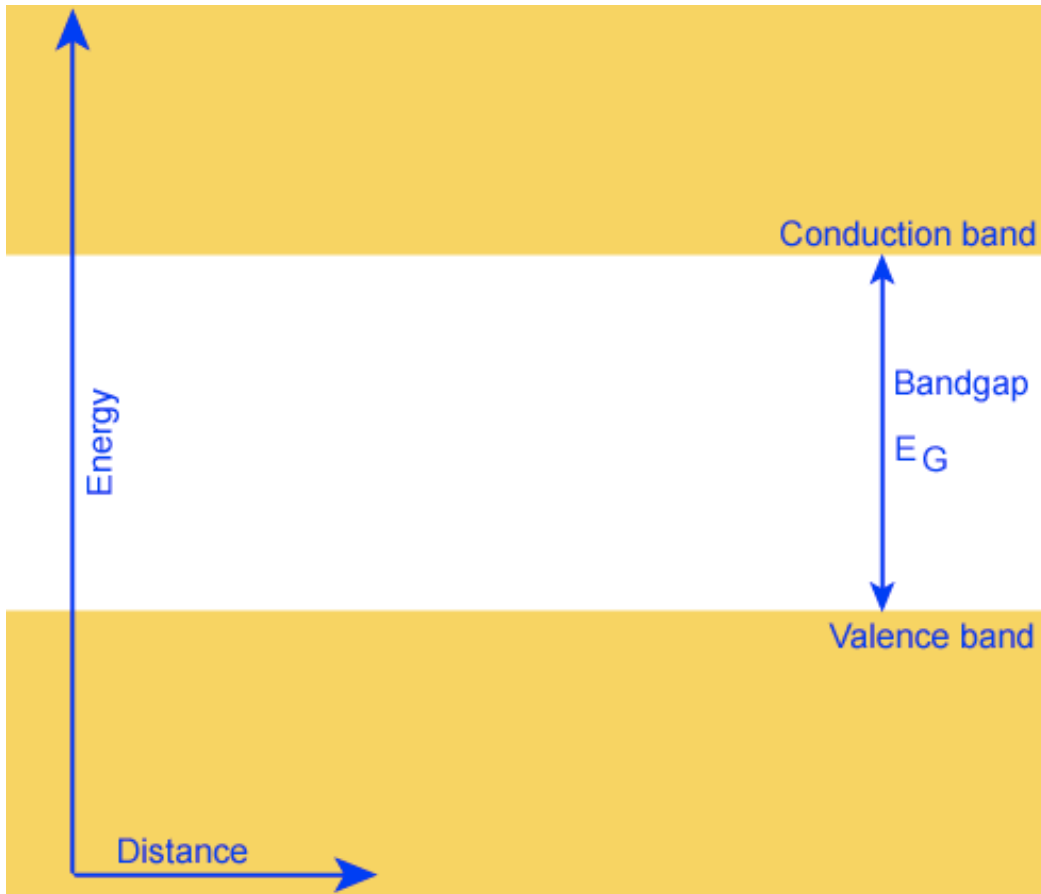


Courtesy Bart Van Zegbroeck. Used with permission.

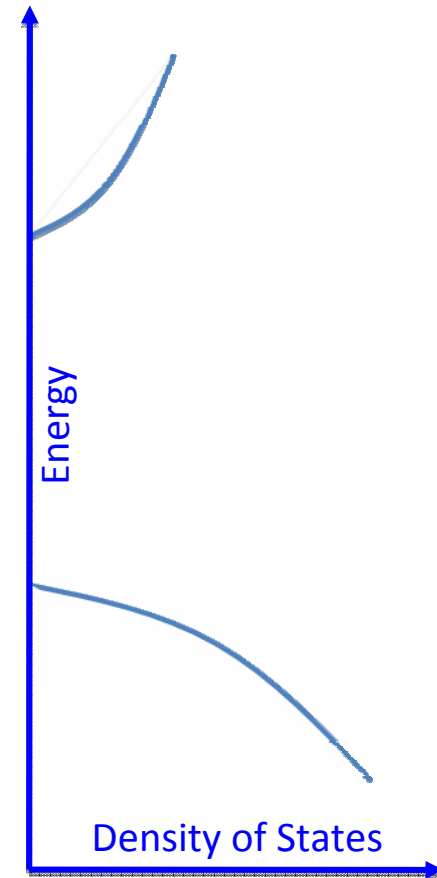
http://ece-www.colorado.edu/~bart/book/book/chapter2/ch2_3.htm#fig2_3_7

Conductivity

Band Diagram (E vs. x)



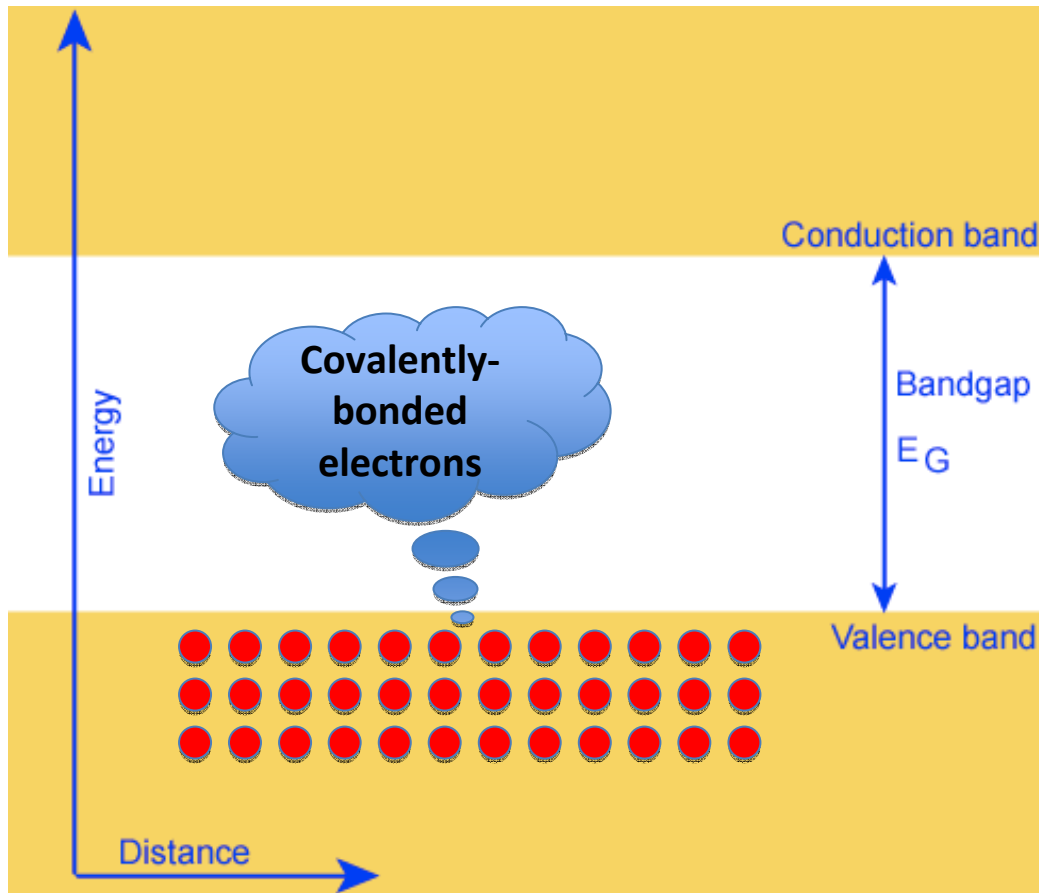
Density of States



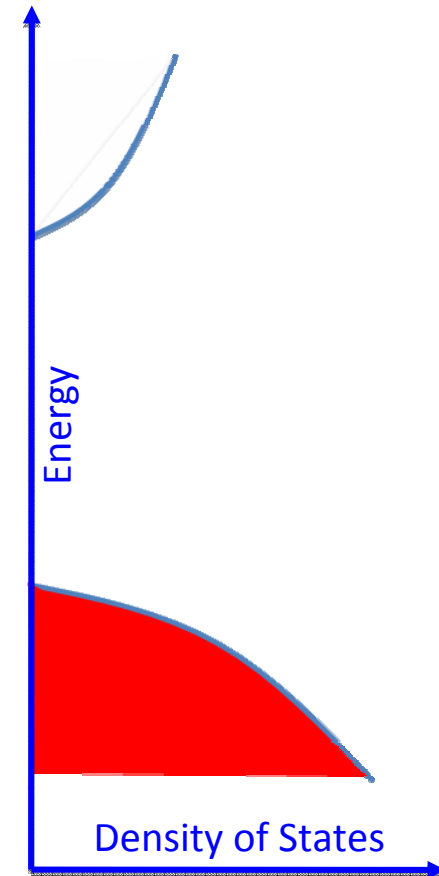
Conductivity: Dependence on Temperature

At absolute zero, no conductivity (perfect insulator).

Band Diagram (E vs. x)



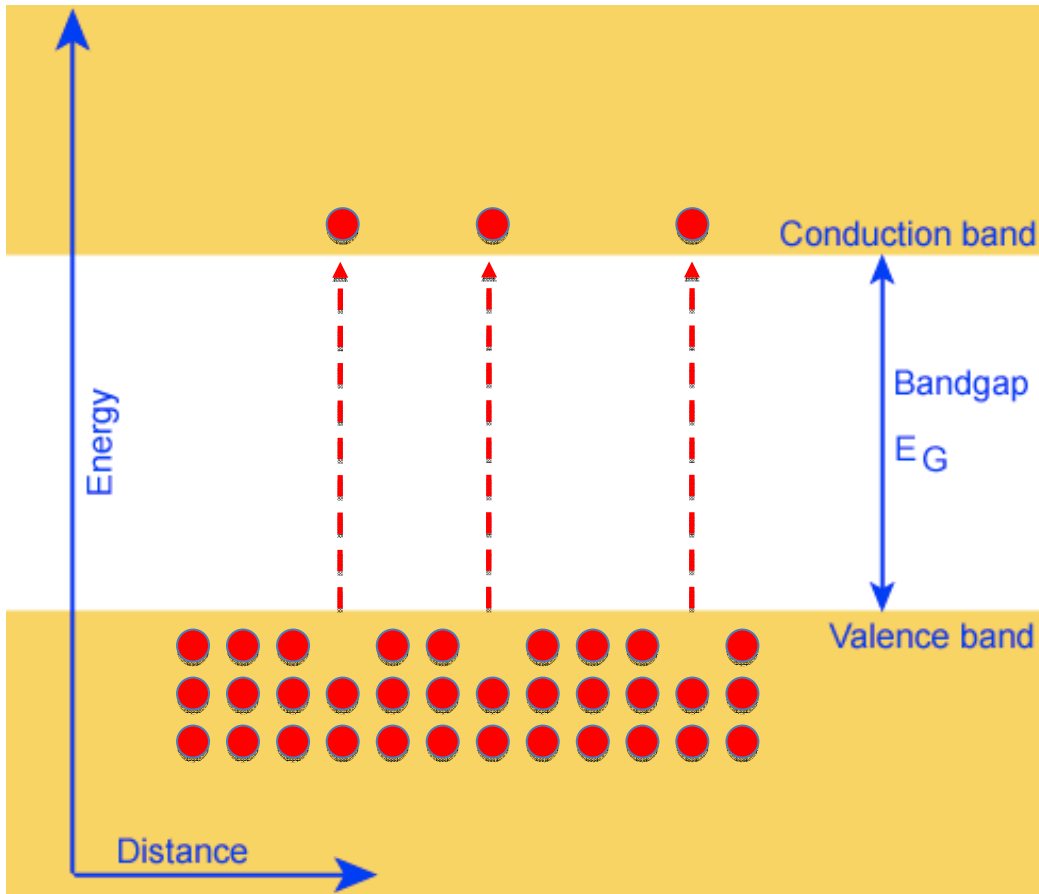
Density of States



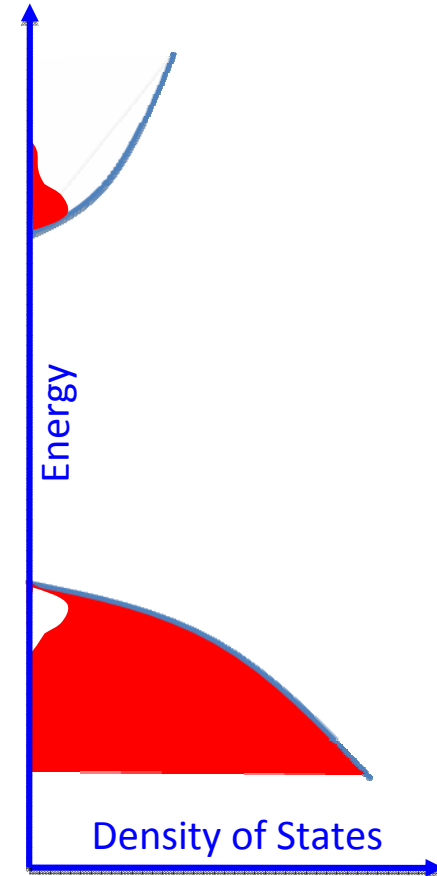
Conductivity: Dependence on Temperature

At $T > 0$ K, some carriers are thermally excited across the bandgap.

Band Diagram (E vs. x)

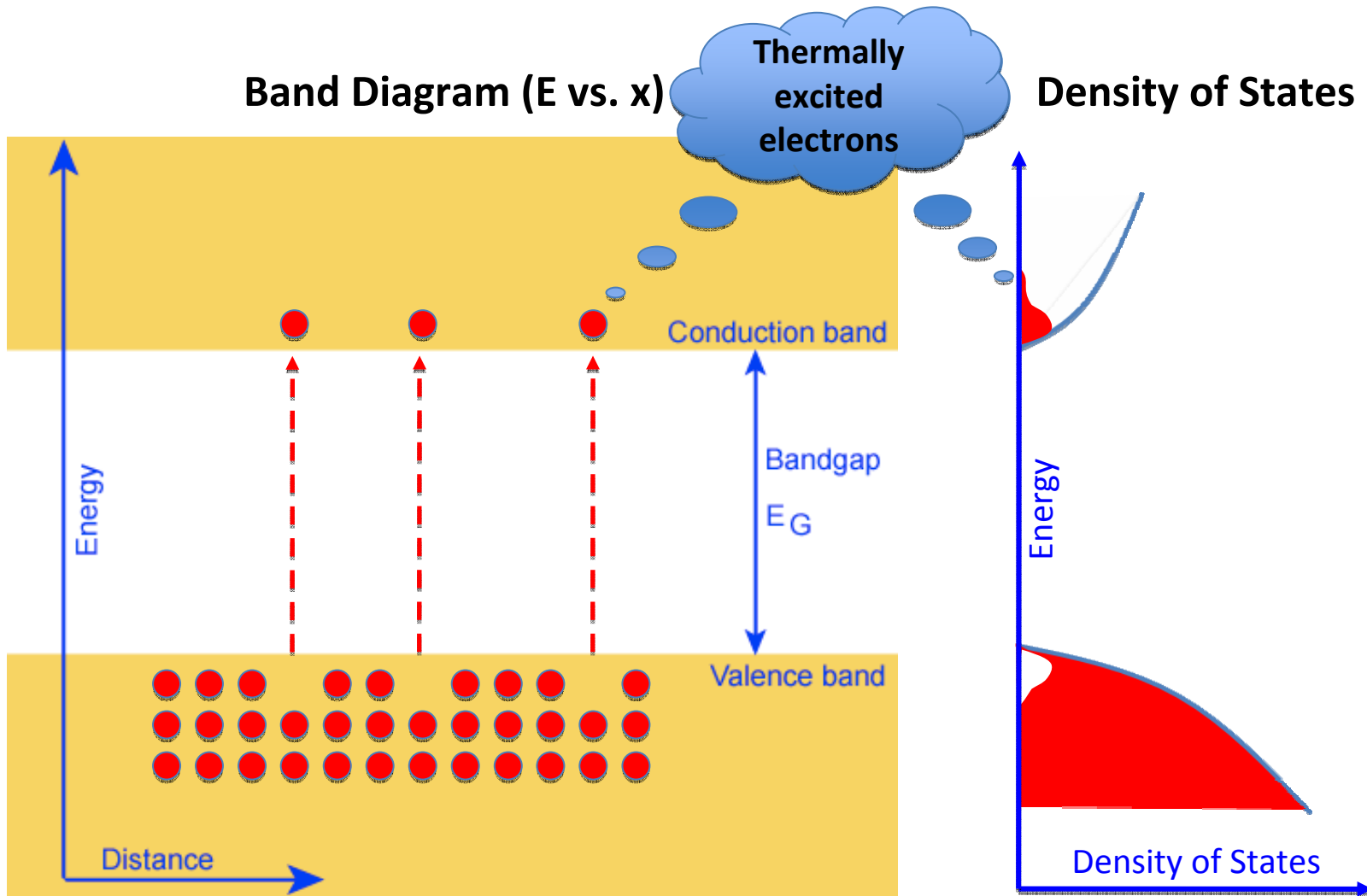


Density of States



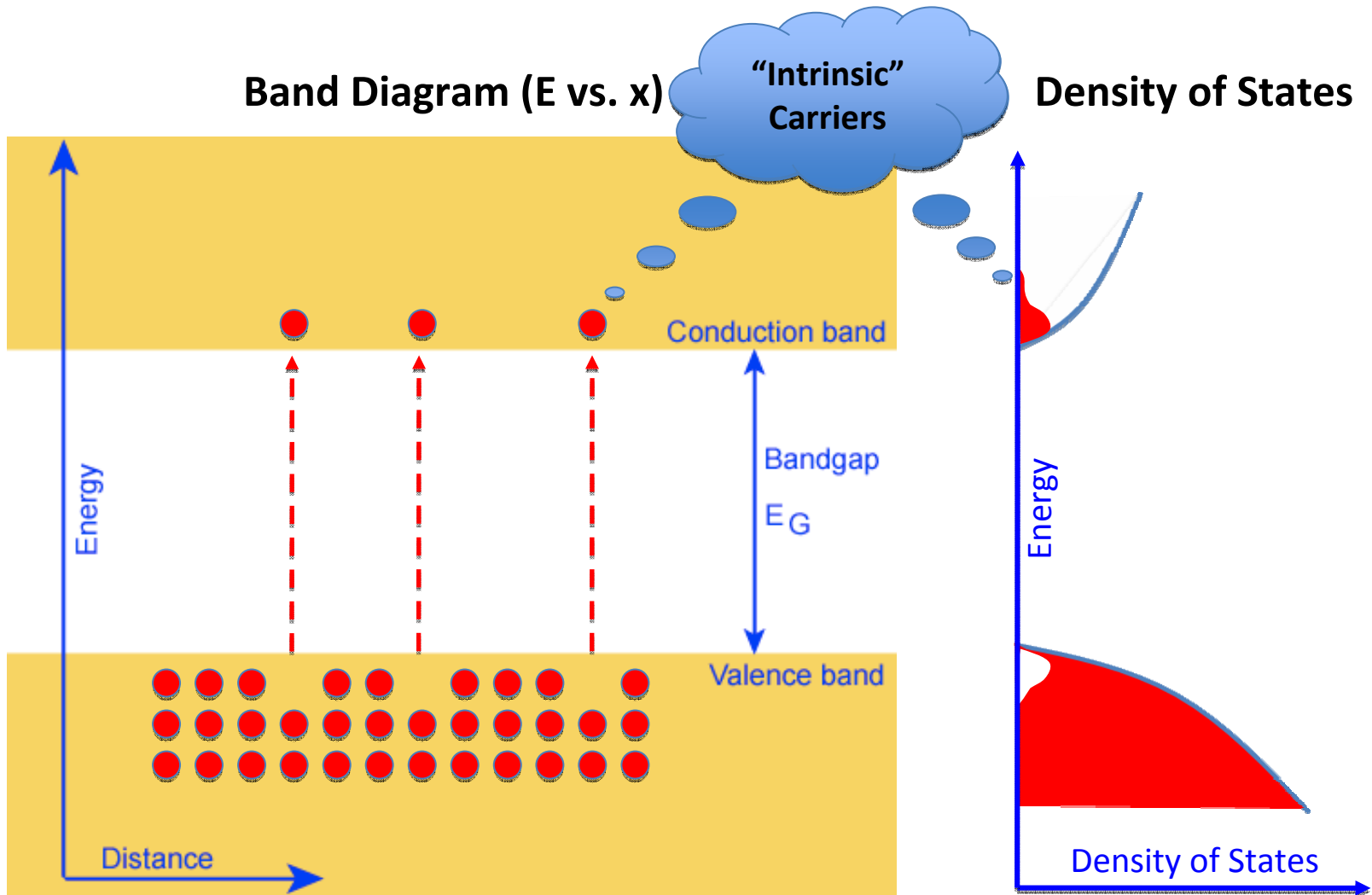
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Conductivity: Dependence on Temperature

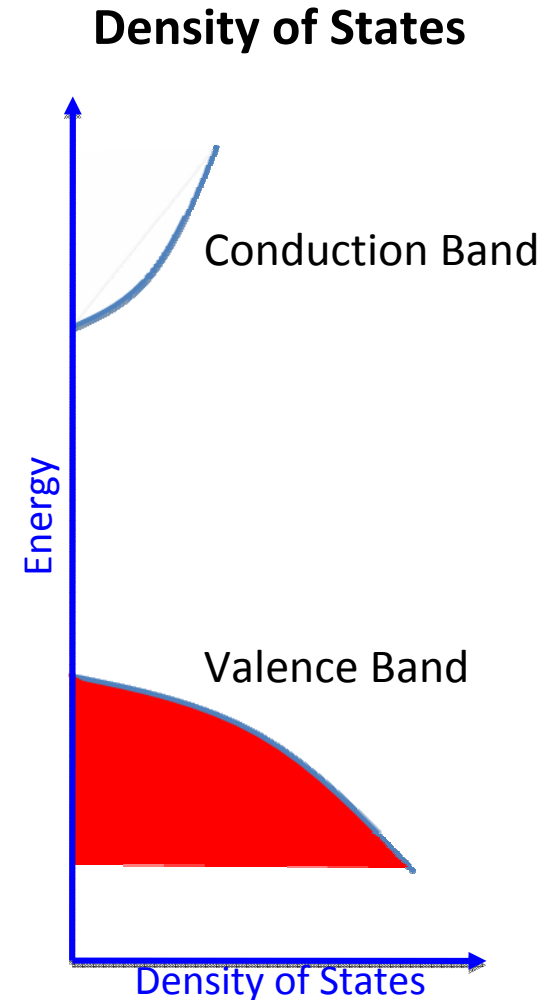
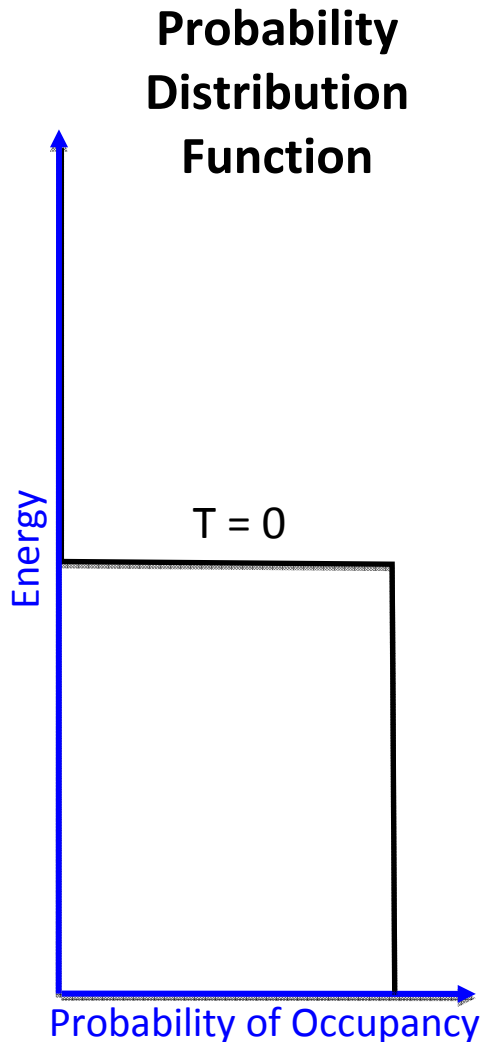
At $T > 0$ K, some carriers are thermally excited across the bandgap.



Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

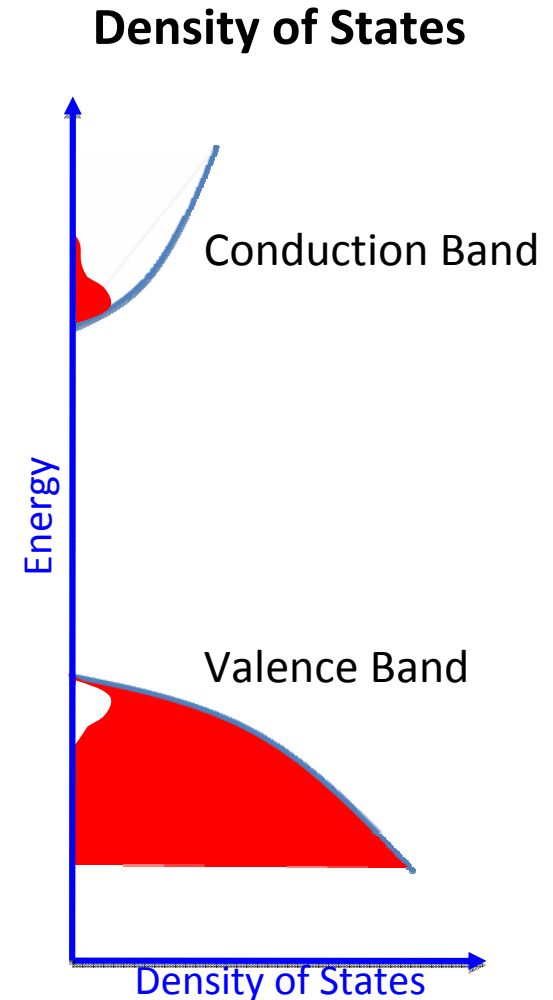
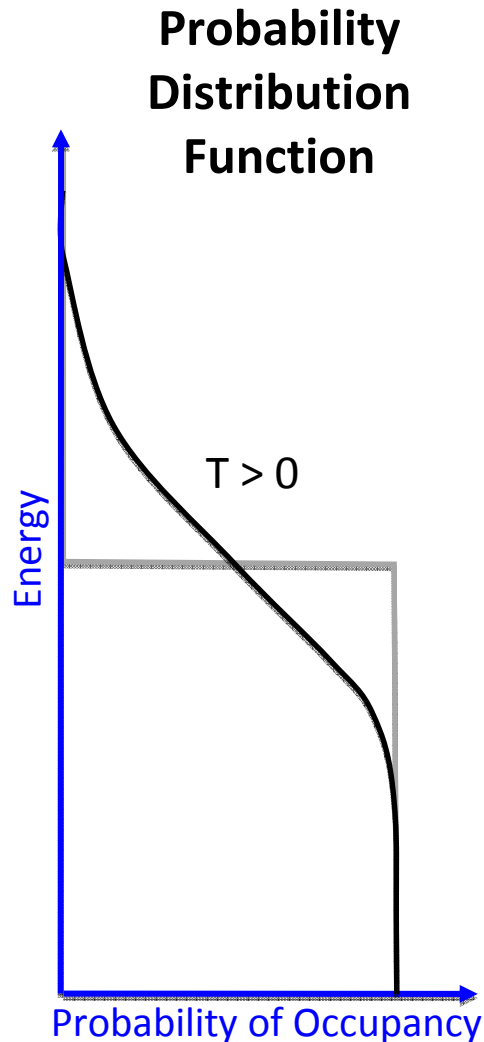
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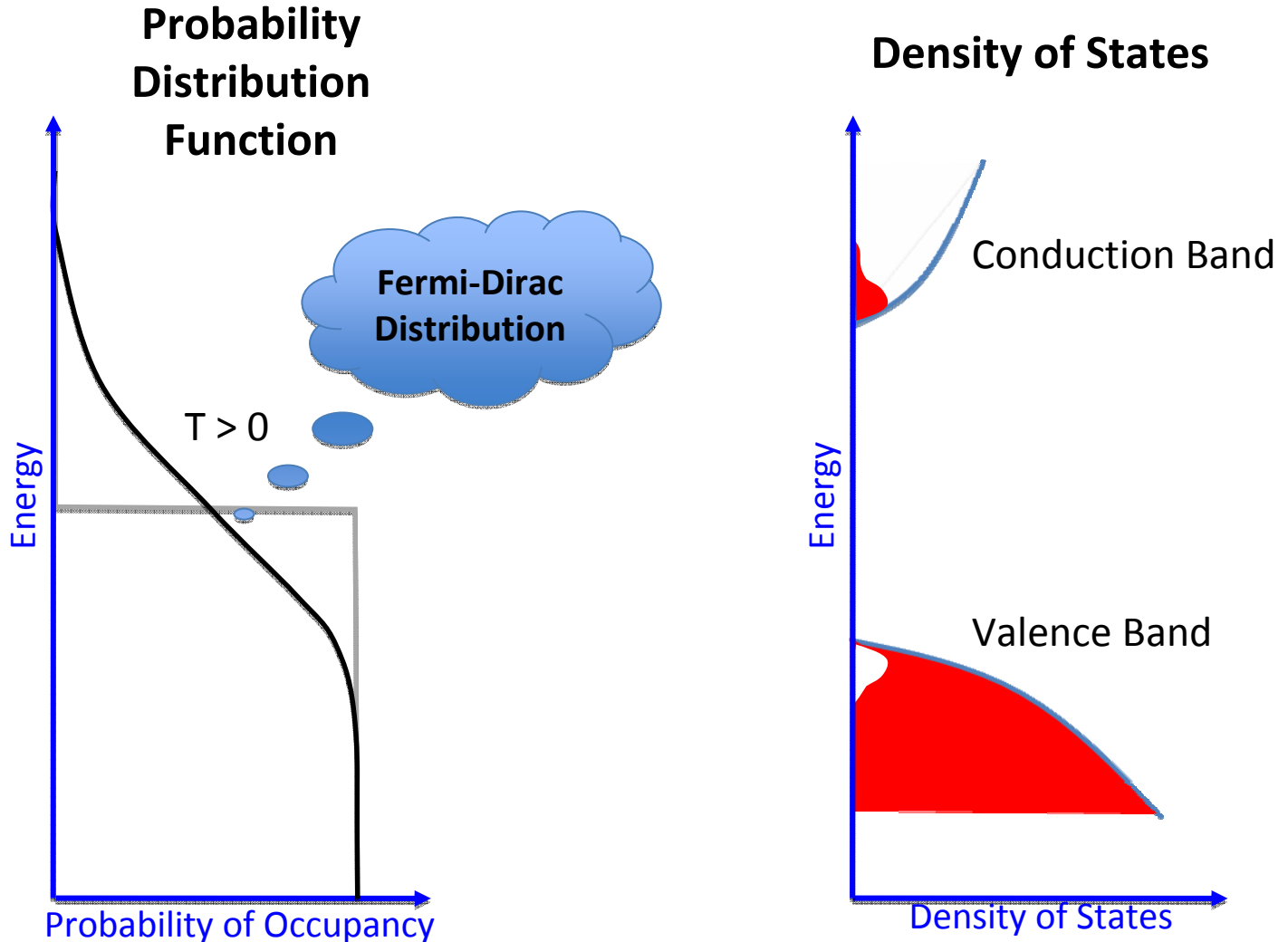
Conductivity: Dependence on Temperature

At a finite temperature, finite conductivity (current can flow).



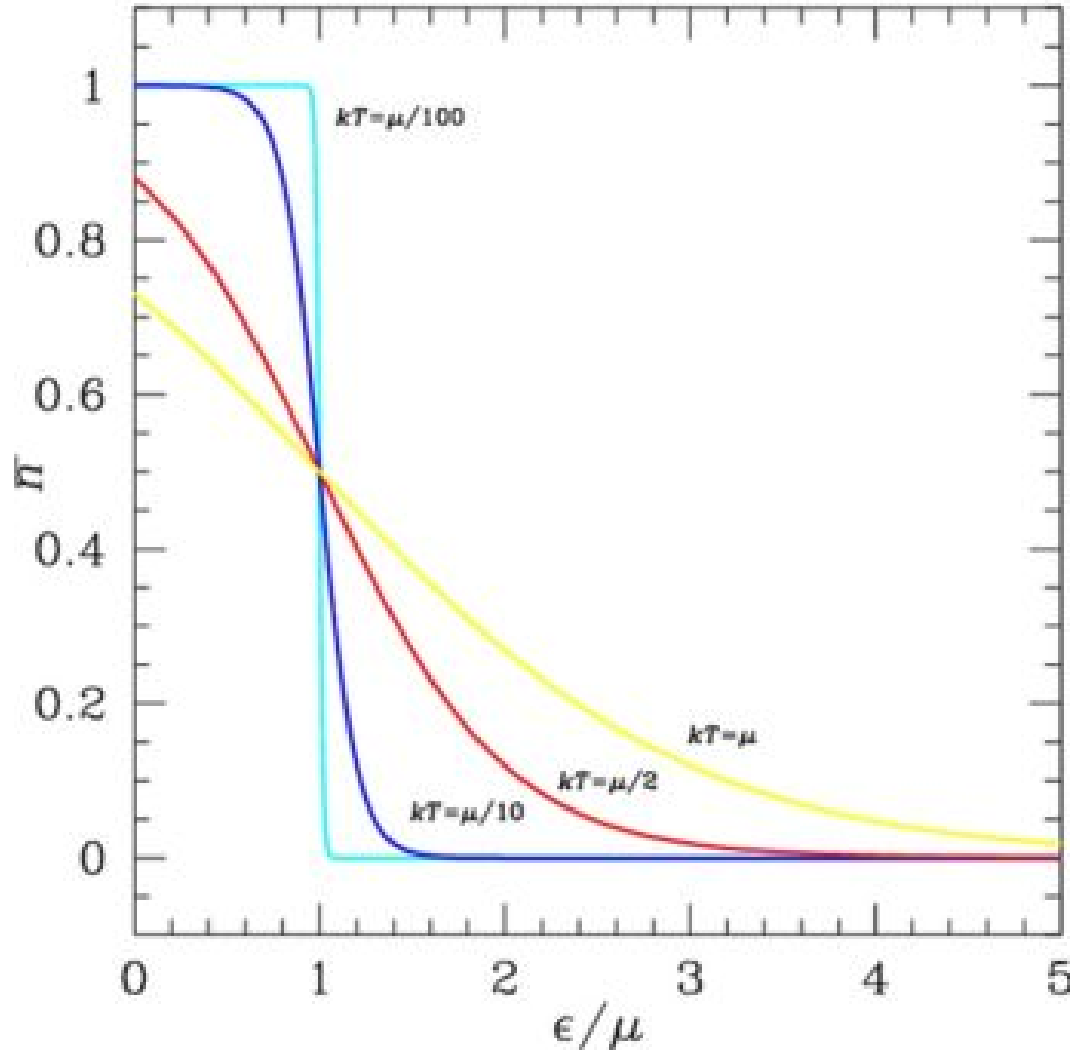
Conductivity: Dependence on Temperature

At a finite temperature, finite conductivity (current can flow).



Fermi-Dirac Distribution

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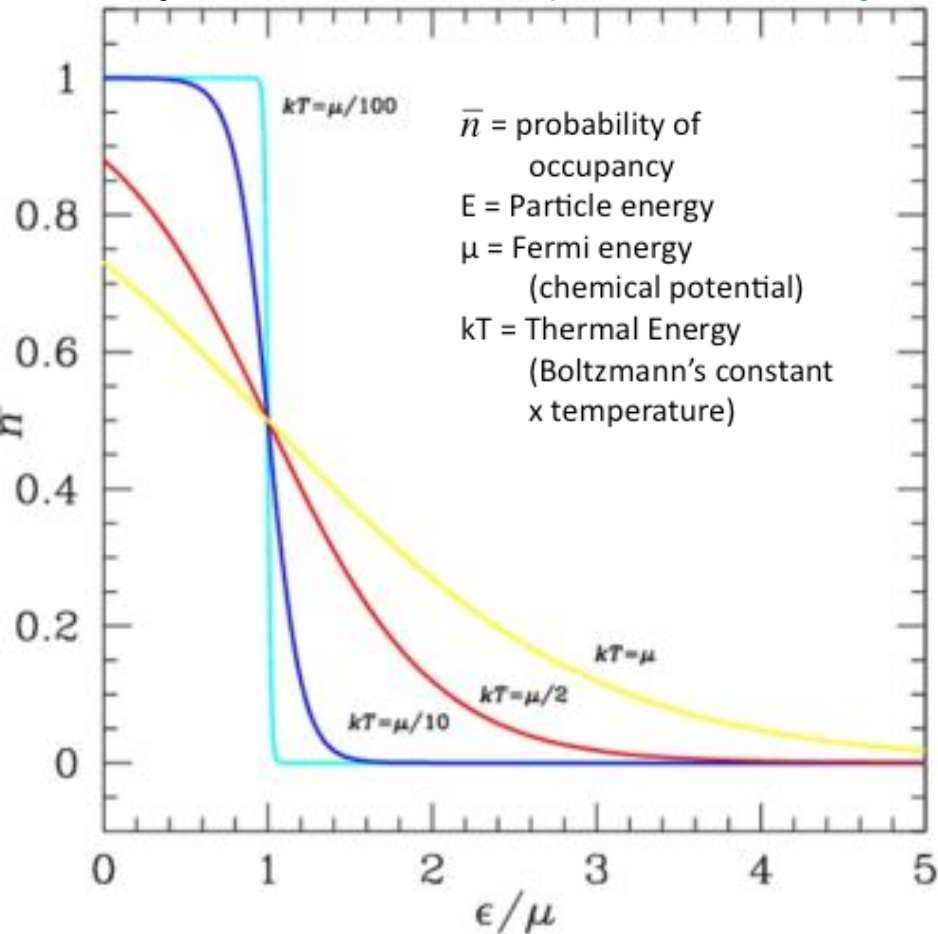


\bar{n} = probability of occupancy
 E = Particle energy
 μ = Fermi energy (chemical potential)
 kT = Thermal Energy (Boltzmann's constant x temperature)

To reduce noise in a Si CCD camera,
should you increase or decrease
temperature?

Lower Temperature = Lower Intrinsic Carrier Concentration

Image from Wikimedia Commons, <http://commons.wikimedia.org>



<http://www.answers.com/topic/semiconductor>

CCD inside a LN dewar

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<http://msowww.anu.edu.au/observing/detectors/wfi.php>

Temperature Dependence of Intrinsic Carrier Concentration

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http://www.tf.uni-kiel.de/matwis/amat/semi_en/kap_2/illustr/n_intrin_temp.gif

Temperature Dependence of Intrinsic Carrier Concentration

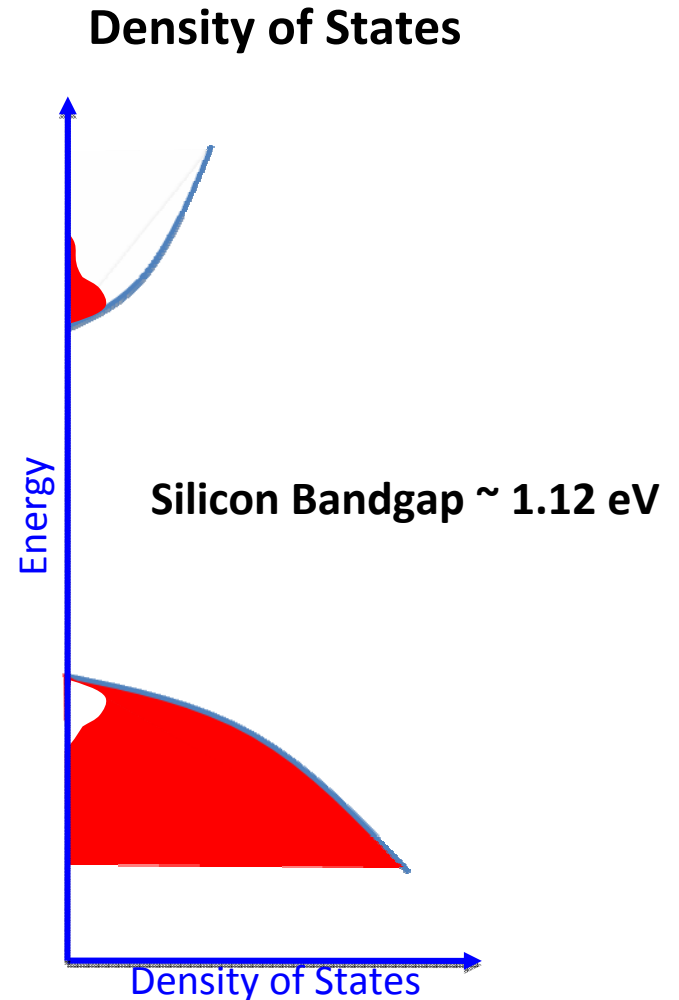
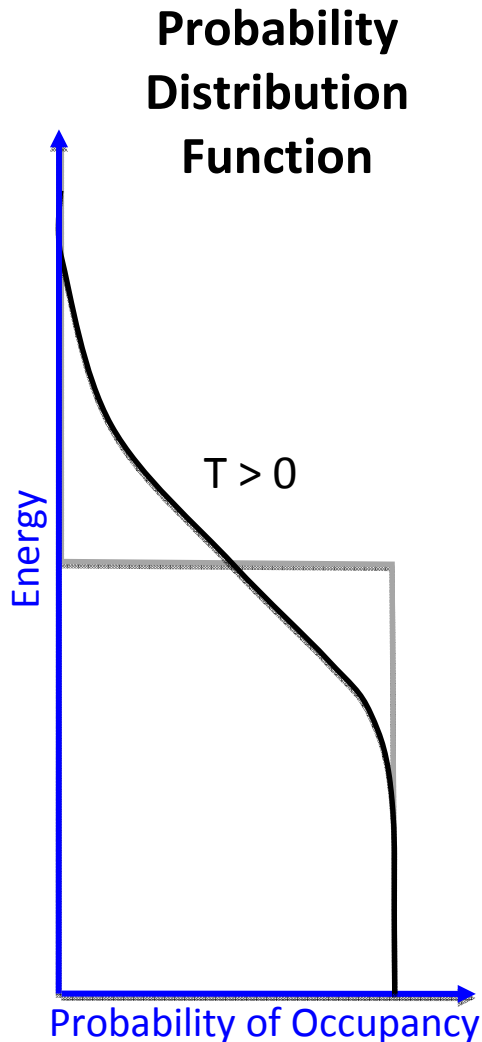
Image removed due to copyright restrictions. Please see

http://www.tf.uni-kiel.de/matwis/amat/semi_en/kap_2/illustr/n_intrin_arrhenius.gif

Question: Transistors made from which semiconductor material experience greater electronic noise at room temperature: Germanium or Silicon?

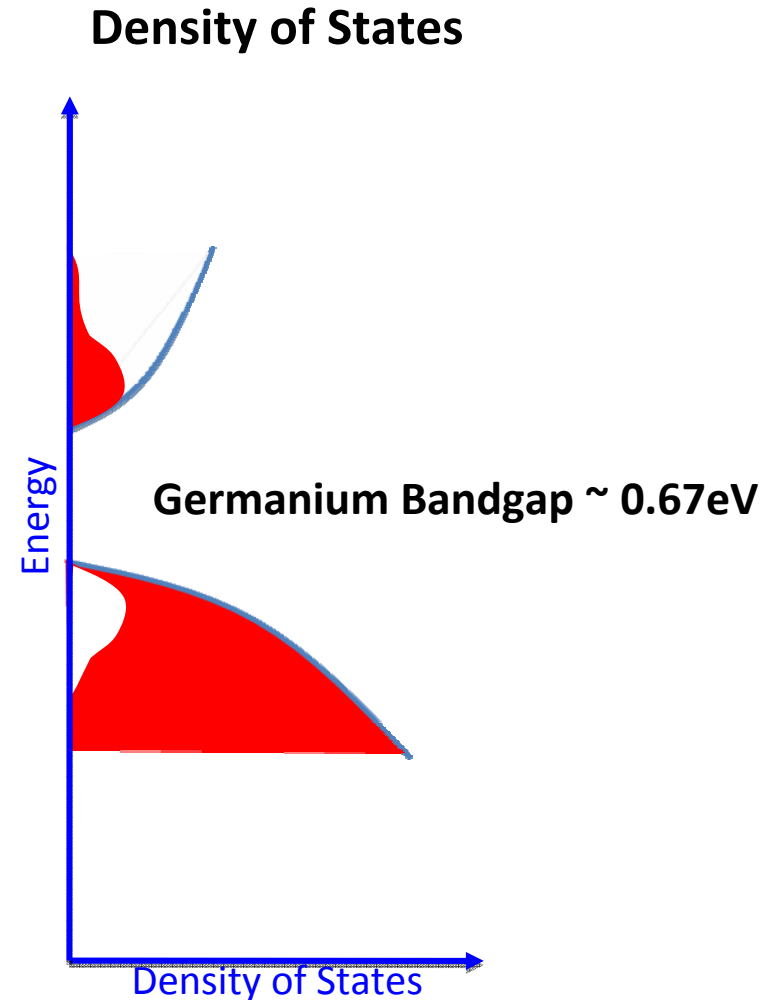
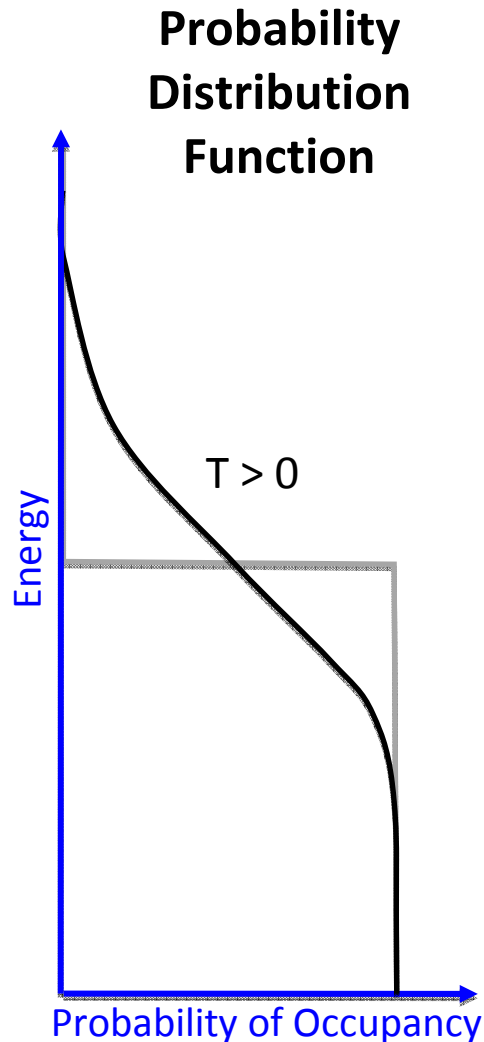
Intrinsic Conductivity: Dependence on Bandgap

At a finite temperature, finite conductivity (current can flow).



Intrinsic Conductivity: Dependence on Bandgap

At a finite temperature, finite conductivity (current can flow).



Intrinsic Conductivity: Dependence on Bandgap

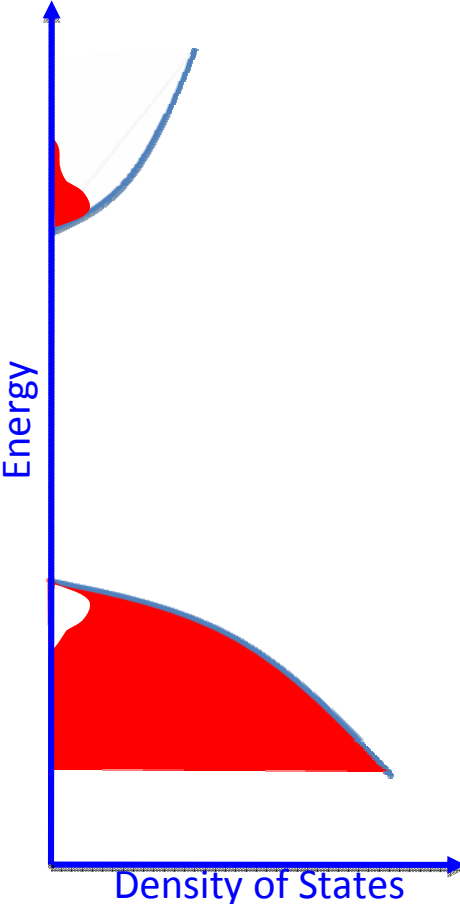
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<http://www.siliconfareast.com/sigegaas.htm>



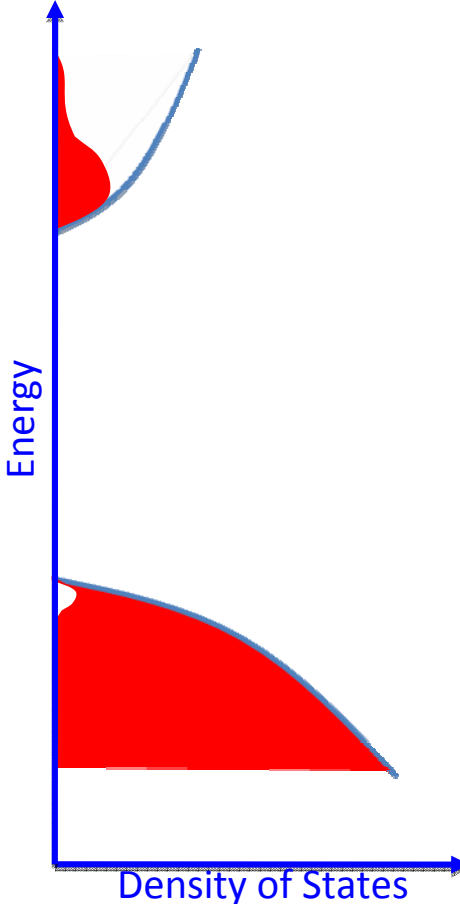
Intentional Modification of Conductivity

Increasing Conductivity

**Intrinsic Carrier Concentration
(High Resistivity)**



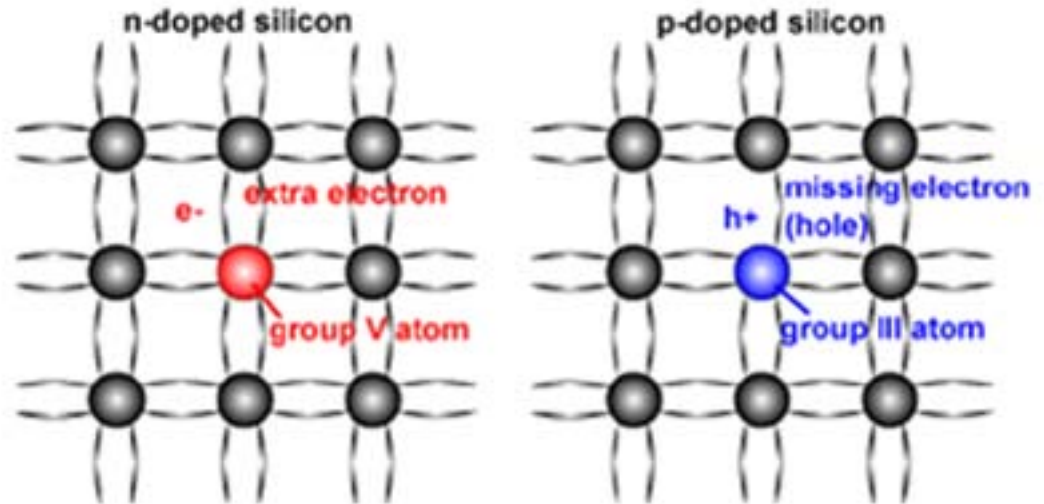
**“Doped” Material
(Low Resistivity)**



Dopant Atoms

Periodic Table

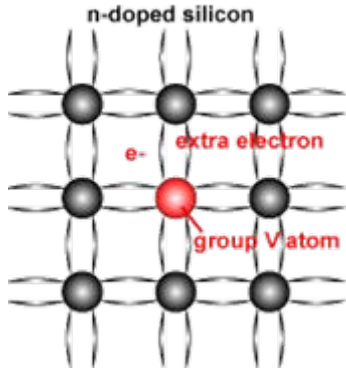
13	14	15	16
5 B	6 C	7 N	8 O
13 Al	14 Si	15 P	16 S
31 Ga	32 Ge	33 As	34 Se
49 In	50 Sn	51 Sb	52 Te
81 Tl	82 Pb	83 Bi	84 Po



<http://pvcdrum.pveducation.org/>

Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

Carrier Binding Energy to Shallow Dopant Atoms

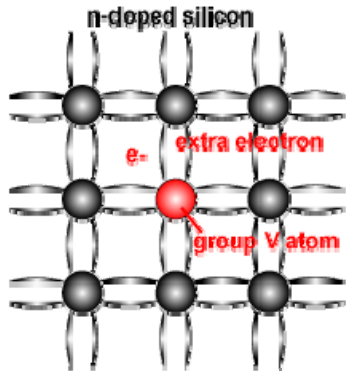


Carrier binding energy to a shallow (hydrogenic) dopant atom:

$$E = E_{\text{H}} \frac{m^*}{m_e} \frac{1}{\epsilon^2} = (13.6 \text{ eV}) \cdot \frac{m^*}{m_e} \frac{1}{\epsilon^2}$$

Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

Carrier Binding Energy to Shallow Dopant Atoms



Carrier binding energy to a shallow (hydrogenic) dopant atom:

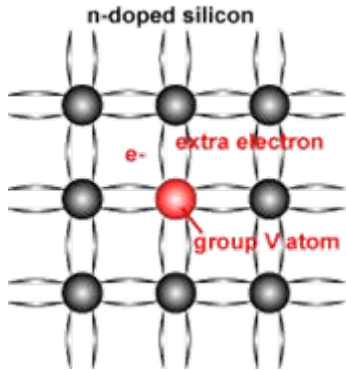
$$E = E_{\text{H}} \frac{m^*}{m_e} \frac{1}{\epsilon^2} = (13.6 \text{ eV}) \cdot \frac{m^*}{m_e} \frac{1}{\epsilon^2}$$

Effective mass correction

Electron screening

Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

Carrier Binding Energy to Shallow Dopant Atoms



Carrier binding energy to a shallow (hydrogenic) dopant atom:

$$E = E_H \frac{m^*}{m_e} \frac{1}{\epsilon^2} = (13.6 \text{ eV}) \cdot \frac{m^*}{m_e} \frac{1}{\epsilon^2}$$

Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

Order of Magnitude Calculation

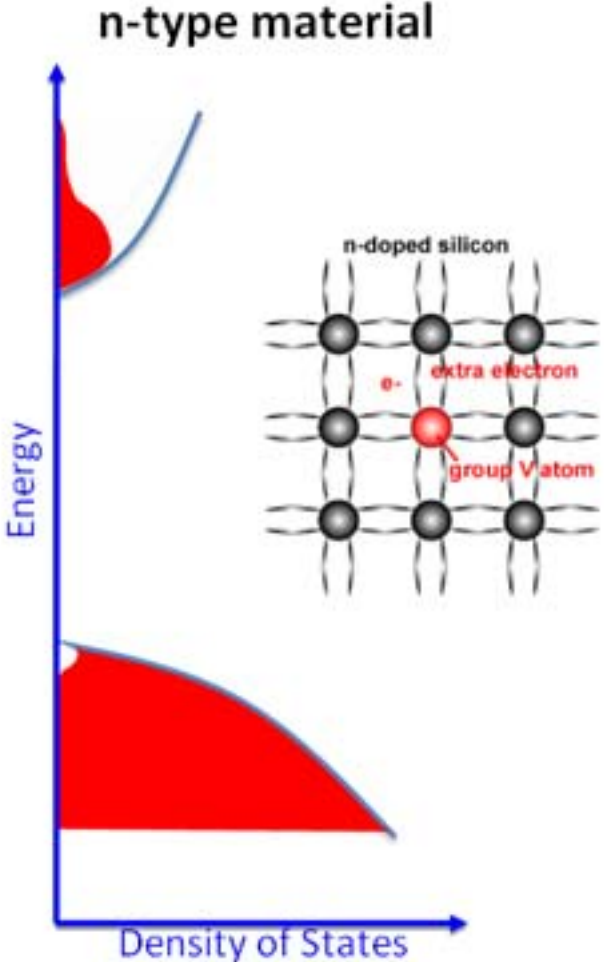
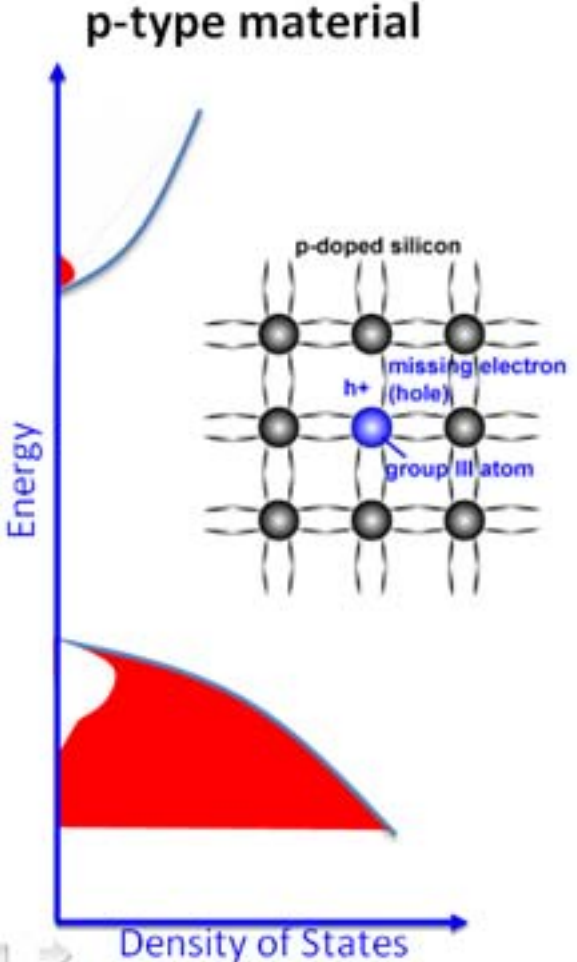
$$\text{Si}(\epsilon=11.7): E_{\text{donor}} \approx 33 \text{ meV}; E_{\text{acceptor}} \approx 75 \text{ meV}$$

$$\text{GaAs}(\epsilon=13.2): E_{\text{donor}} \approx 6 \text{ meV}; E_{\text{acceptor}} \approx 28 \text{ meV}$$

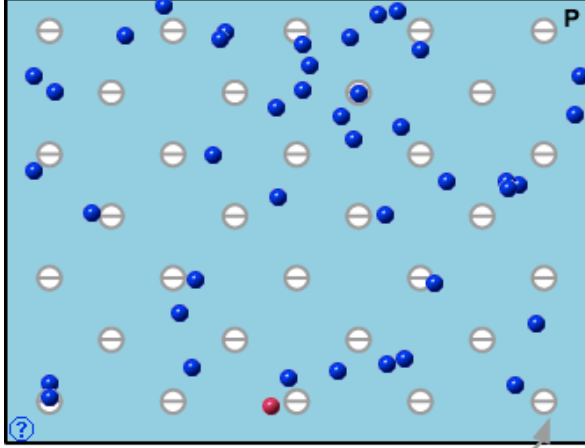
$$kT \text{ at room temperature} \approx 26 \text{ meV}$$

→ most shallow dopants should be ionized at room temperature!

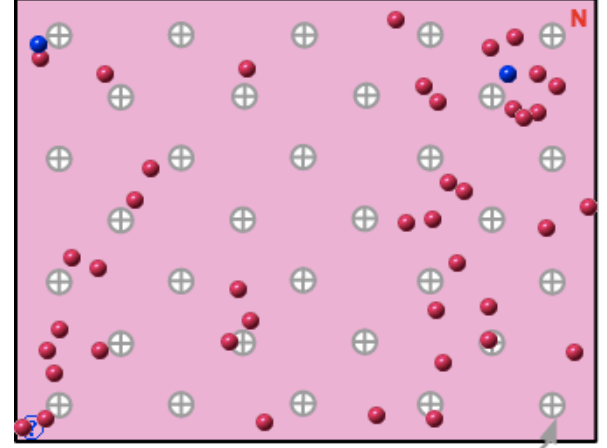
Dopant Atoms



Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

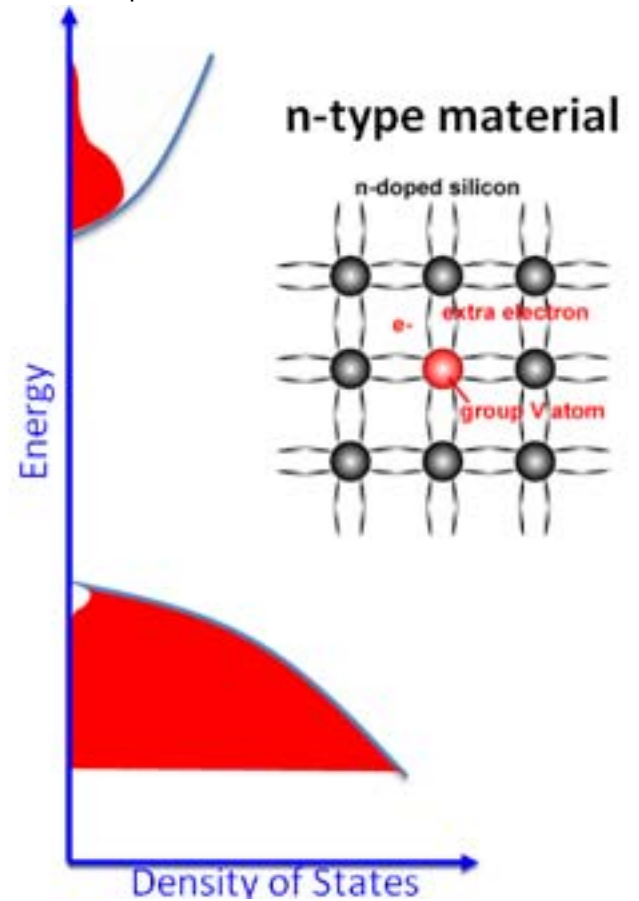
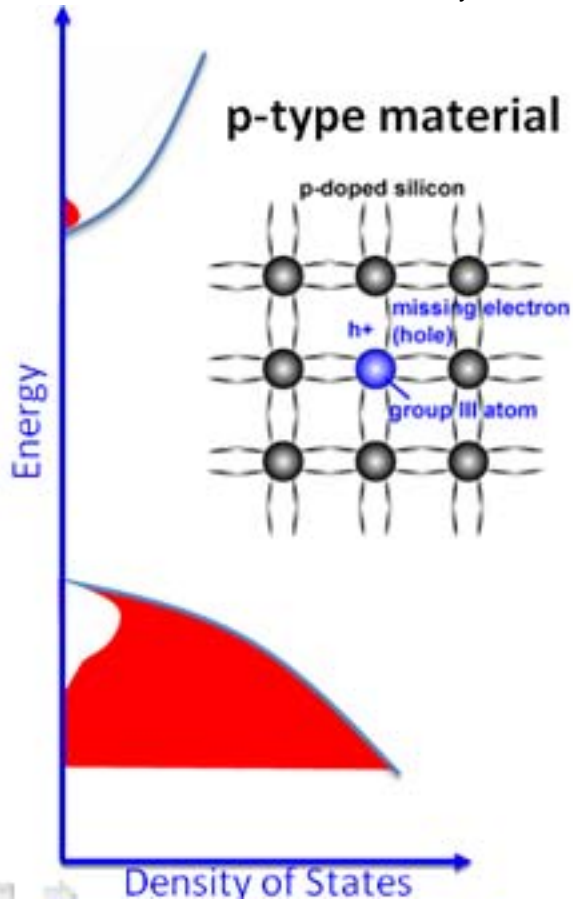


● electrons ● holes ionised acceptor atom



● electrons ● holes ionised donor atom

Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.



Resistivity as a Function of Dopant Concentration

Image removed due to copyright restrictions. Please see Fig. 18a in Sze, S. M., and Ng, Kwok K. *Physics of Semiconductor Devices*. Hoboken, NJ: Wiley-Interscience, 2007.

Resistivity as a Function of Dopant Concentration

Image removed due to copyright restrictions. Please see Fig. 18a in Sze, S. M., and Ng, Kwok K. *Physics of Semiconductor Devices*. Hoboken, NJ: Wiley-Interscience, 2007.

This degree of control (many orders of magnitude!) over the fundamental electrical properties of the material, is what makes semiconductors so versatile.

Resistivity of Polymers can also be Tailored

Comparison of conjugated polymers to other materials

Courtesy of Ilan Gur. Used with permission.

Stretch-aligning polyacetylene and removing chemical defects has increased the conductivity to $10^5 \text{ Ohm}^{-1}\text{cm}^{-1}$. Further improvements are expected.

Slide from Ilan Gur, UC Berkeley

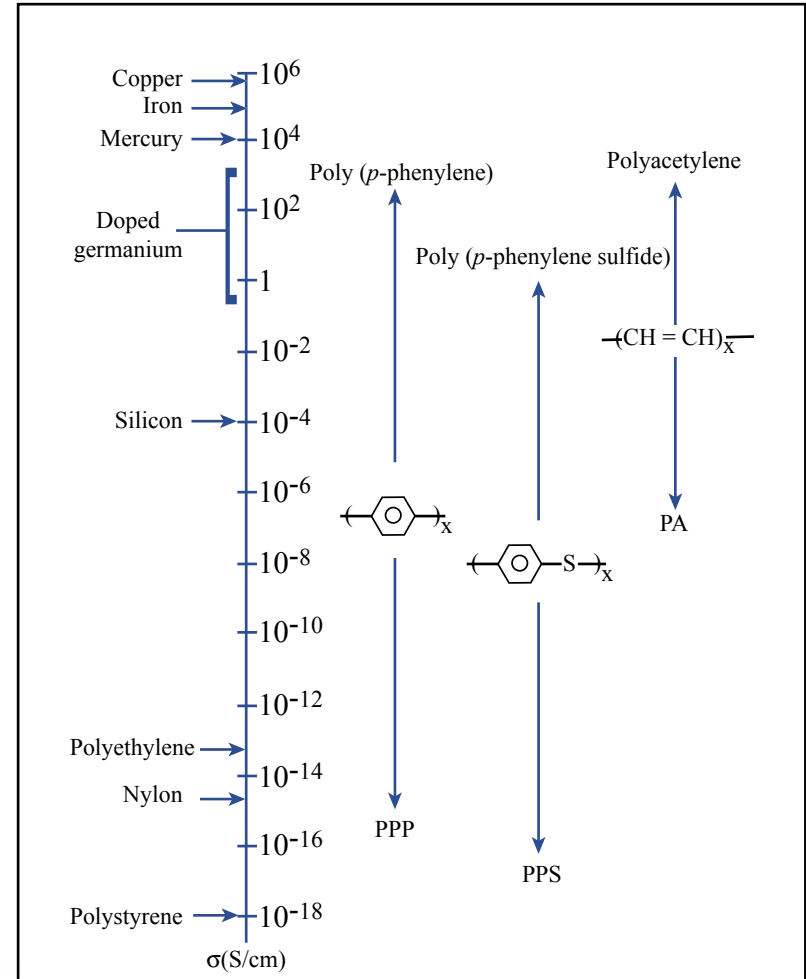
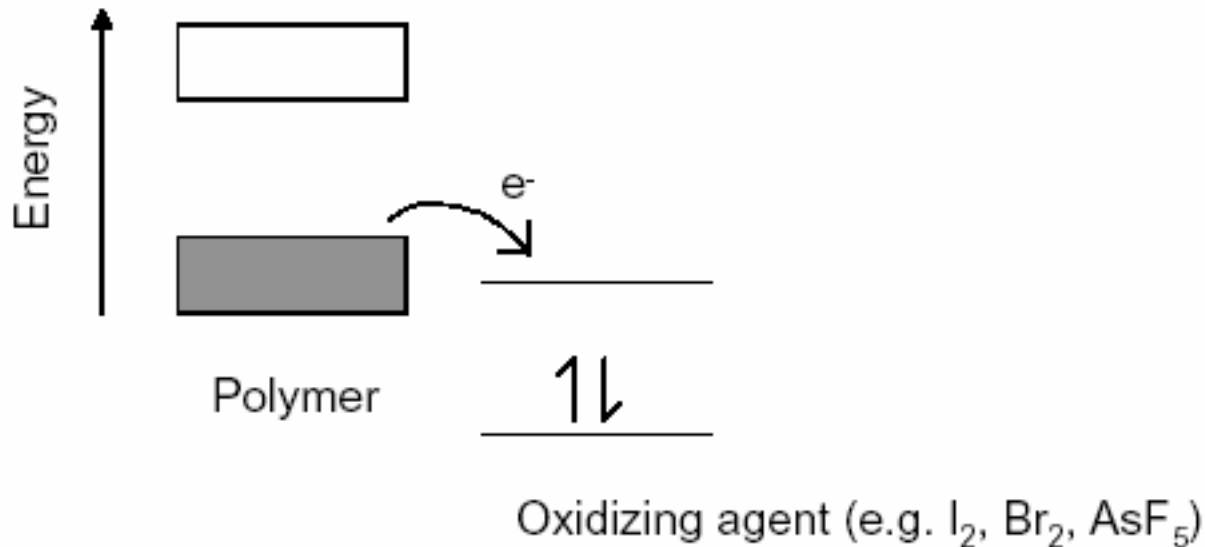


Figure by MIT OpenCourseWare.

P-type Doping of Polymers

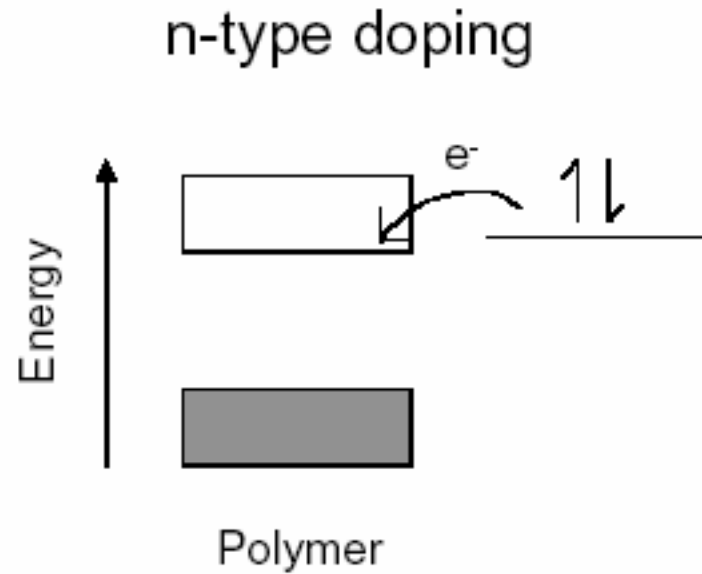
Doping conjugated polymers



Oxidizing agents act as p-type dopants, i.e. they generate holes in the polymer.

Courtesy of Ilan Gur. Used with permission.

N-type Doping of Polymers



Courtesy of Ilan Gur. Used with permission.

Reducing agents such as calcium and lithium can introduce electrons into the conduction band of a conjugated polymer.

Resistivity of Polymers can also be Tailored

Electrical conductivity of *trans*-(CH)_x
as a function of (AsF₅) dopant concentration

Polyacetylene is a semiconductor, but when it is heavily doped, it undergoes an insulator-to-metal transition.

For comparison, the conductivity of copper is 10⁶ (Ohm-cm)⁻¹.

In 2000, Heeger, MacDiarmid and Shirakawa won the Nobel Prize in Chemistry for this experiment and their development of the science and technology of conjugated polymers.

Image removed due to copyright restrictions. Please see Fig. 1 in Chiang, C. K, et al. "Electrical Conductivity of Doped Polyacetylene." *Physical Review Letters* 39 (1977): 1098.

Courtesy of Ilan Gur. Used with permission.

Mobility and Carrier Concentration in Semiconductor

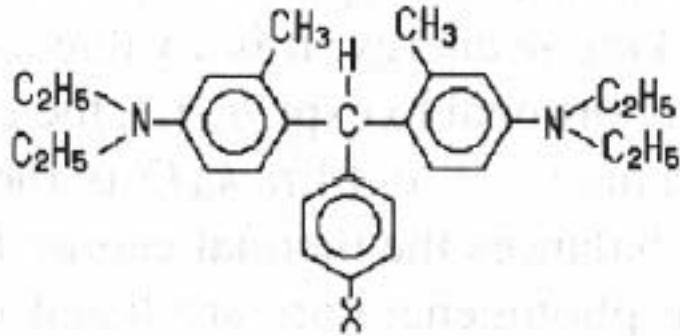
Increased concentration of ionized dopant atoms increases conductivity, but can reduce carrier mobility (due to scattering).

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http://www.tf.uni-kiel.de/matwis/amat/semi_en/kap_2/illustr/mobility_doping.gif.

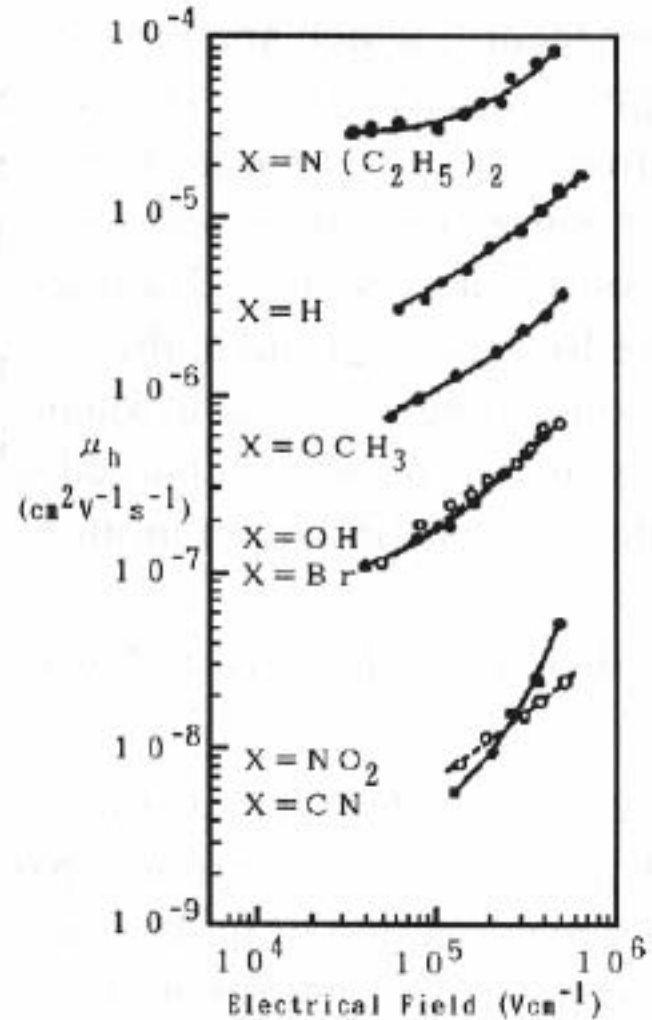
Mobility and Carrier Concentration in Polymer

Courtesy of Ilan Gur. Used with permission.

Effect of substituent on the hole mobility



All samples consisted of 40 % by weight of the molecule shown above blended into polycarbonate.



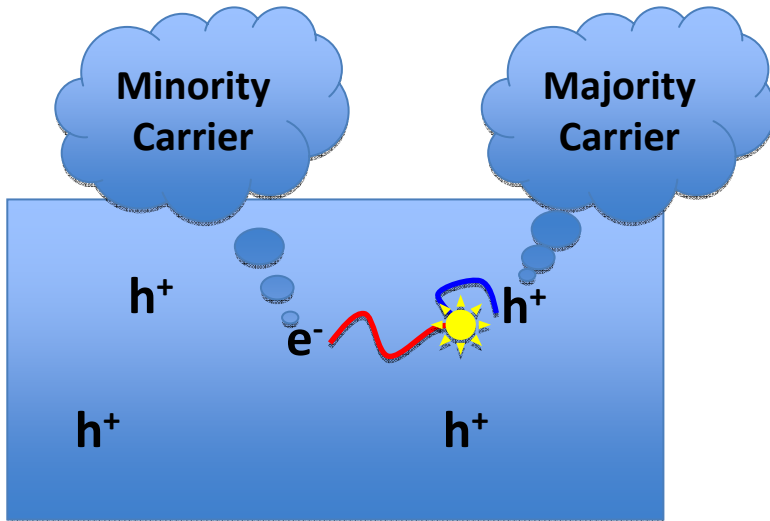
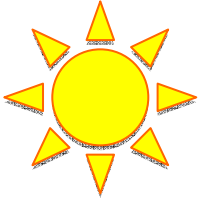
Mobility and Carrier Concentration in Semiconductor

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Please see http://www.tf.uni-kiel.de/matwis/amat/semi_en/kap_2/illustr/mobility_temp.gif.

Carrier Lifetime and Recombination



Bulk Minority Carrier Lifetime:

$$\tau = \frac{\Delta n}{R}$$

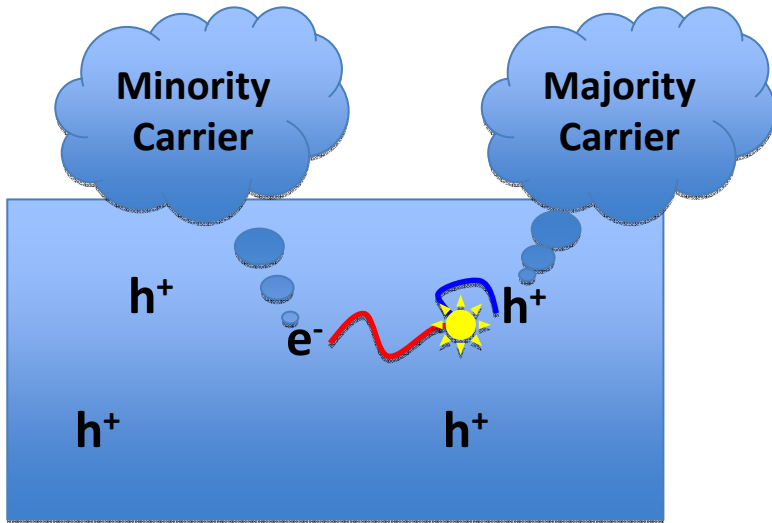
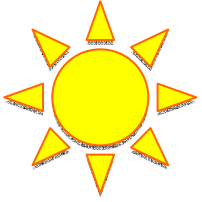
Δn = Excess minority carrier concentration
 R = Recombination rate

First approximation of minority carrier lifetime, for low injection (e.g., illumination) conditions.

More Detailed Calculation:

$$\frac{1}{\tau_{\text{bulk}}} = \frac{1}{\tau_{\text{band}}}$$

Carrier Lifetime and Recombination



Bulk Minority Carrier Lifetime:

$$\tau = \frac{\Delta n}{R}$$

Δn = Excess minority carrier concentration
 R = Recombination rate

First approximation of minority carrier lifetime, for low injection (e.g., illumination) conditions.

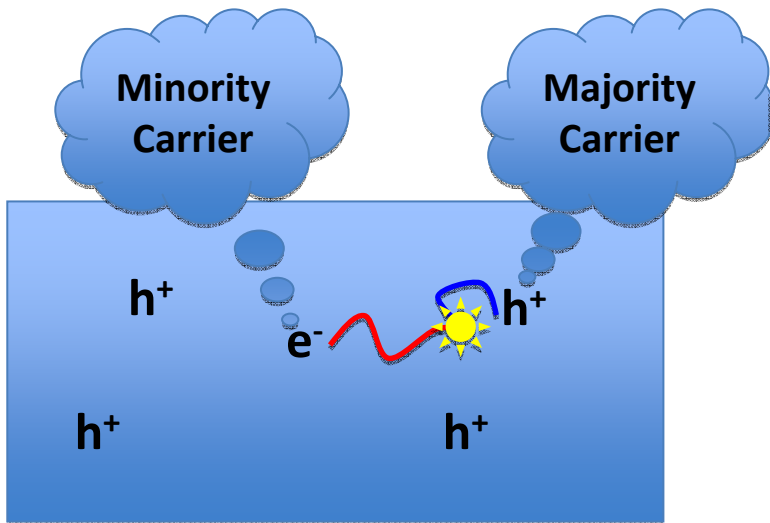
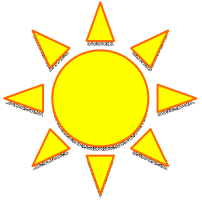
More Detailed Calculation:

$$\frac{1}{\tau_{\text{bulk}}} = \frac{1}{\tau_{\text{band}}} + \frac{1}{\tau_{\text{Auger}}}$$

Dominant under very high injection conditions

$$\tau_{\text{Auger}} = \frac{1}{CN_A}$$

Carrier Lifetime and Recombination



Bulk Minority Carrier Lifetime:

$$\tau = \frac{\Delta n}{R}$$

Δn = Excess minority carrier concentration
 R = Recombination rate

First approximation of minority carrier lifetime, for low injection (e.g., illumination) conditions.

More Detailed Calculation:

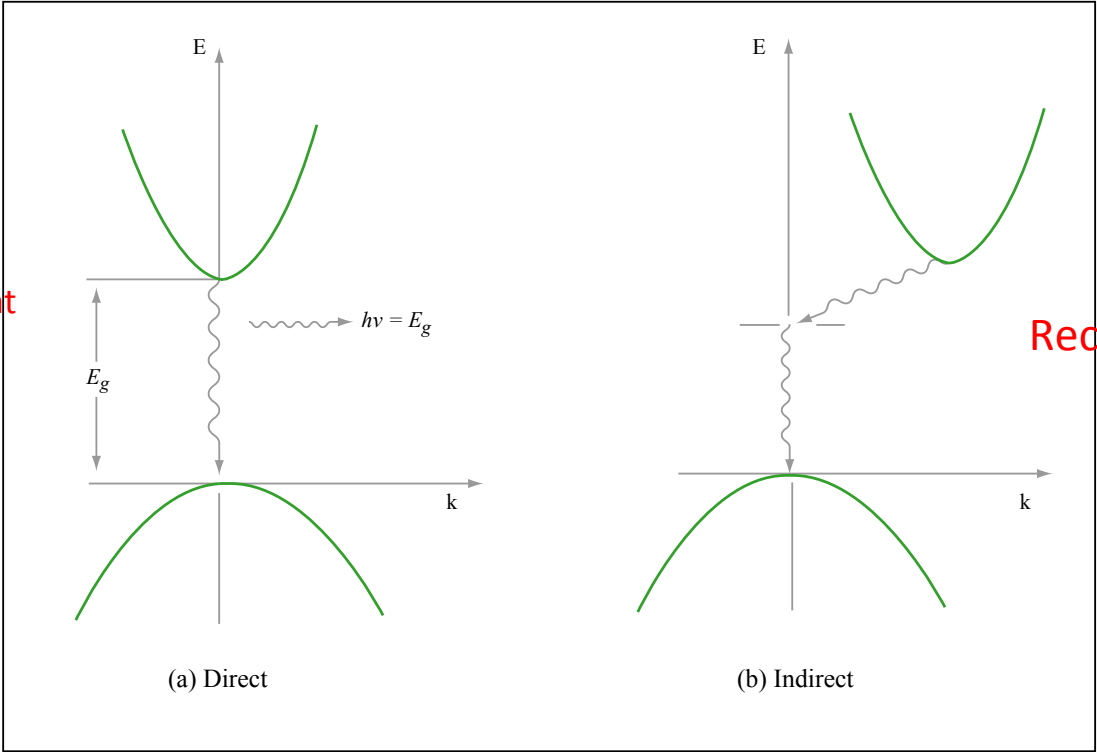
$$\frac{1}{\tau_{\text{bulk}}} = \frac{1}{\tau_{\text{band}}} + \frac{1}{\tau_{\text{Auger}}} + \frac{1}{\tau_{\text{SRH}}}$$



Defects and Carrier Recombination

Direct Bandgap Semiconductor

Indirect Bandgap Semiconductor



Recombination efficient
(no phonon required)
 $\tau \sim \text{ns to } \mu\text{s}$

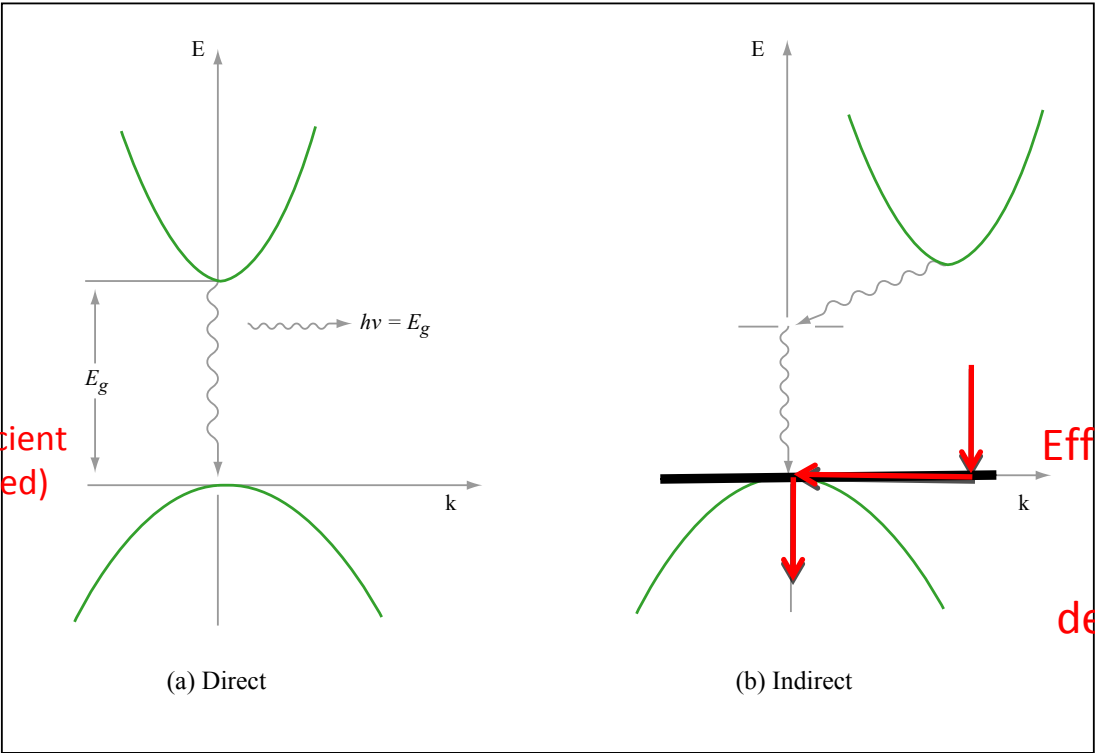
Recombination inefficient
(phonon required)
 $\tau \sim \text{ms}$

Figure by MIT OpenCourseWare.

Defects and Carrier Recombination

Direct Bandgap Semiconductor

Indirect Bandgap Semiconductor



Recombination efficient
(no phonon required)
 $\tau \sim \text{ns to } \mu\text{s}$

Efficient recombination
via defect level!
 $\tau < \mu\text{s}$ with high
defect concentrations

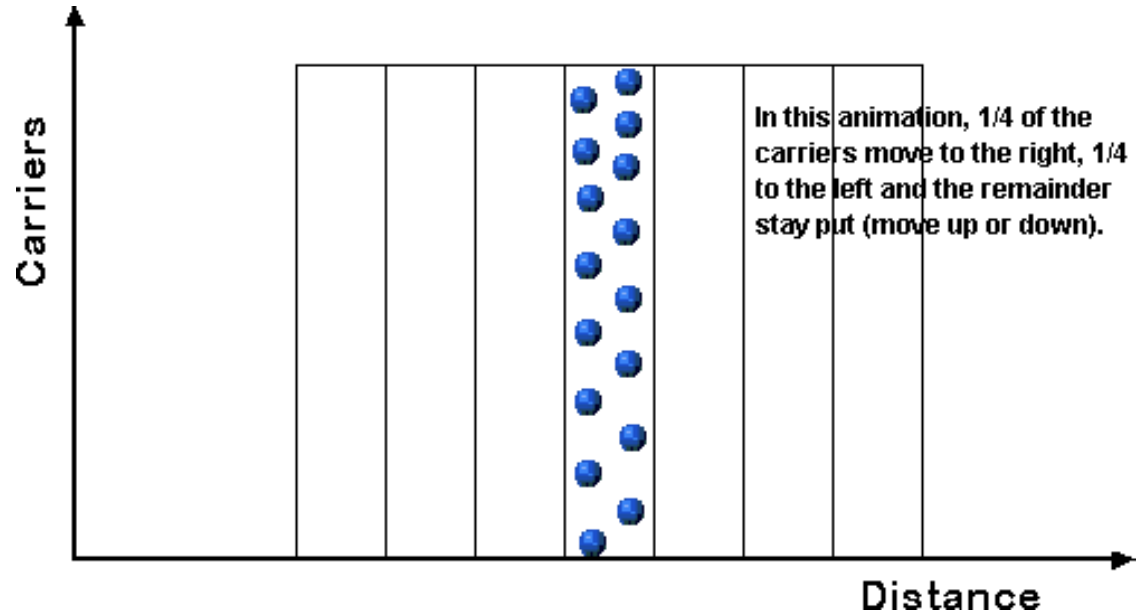
Figure by MIT OpenCourseWare.

Carrier Motion

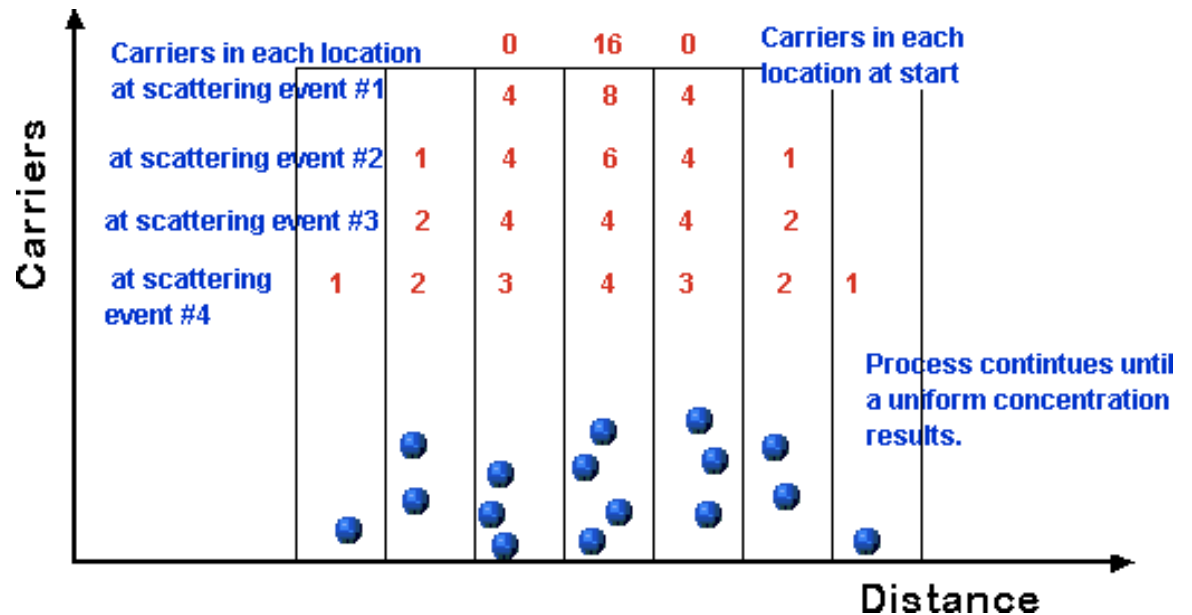
Under equilibrium conditions in a homogeneous material: Individual carriers constantly experience Brownian motion, but the net charge flow is zero.

To achieve net charge flow (current), carriers must move via diffusion or drift.

Diffusion

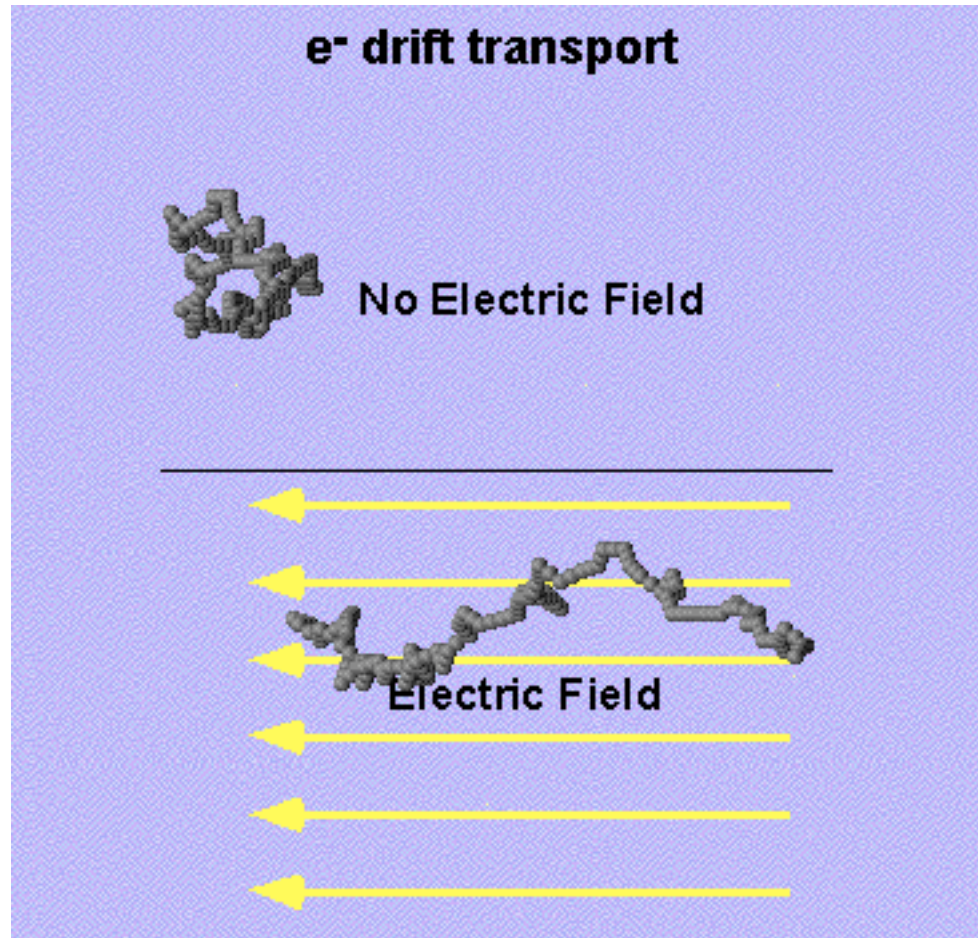


Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.



From PVCDROM

Drift Current



From PVCDROM

Courtesy Christiana Honsberg and Stuart Bowden. Used with permission.

Advanced Concept: Thermalization

Energy Loss Due to Thermalization

Thermalization is an incredibly fast process, difficult to avoid...

Image removed due to copyright restrictions.

Please see http://www.tf.uni-kiel.de/matwis/amat/semi_en/kap_2/illustr/pair_generation.gif.

Energy Losses: First Approx.

Thermalization Losses (bandgap too small)

Images removed due to copyright restrictions.

Please see any diagram of thermalization loss, such as

<https://engineering.purdue.edu/NANOENERGY/research/images/solar.gif>, and any graph of non-absorption losses.

Non-Absorption Losses (bandgap too big!)

Energy Losses: First Approximation

Balance between Thermalization and Non-Absorption Losses

Image removed due to copyright restrictions. Please see any image of photovoltaic efficiency vs. bandgap energy, such as <http://www.grc.nasa.gov/WWW/RT/RT1999/images/5410hepp-f3.jpg>.